



## REVIEW ARTICLE

# Comprehensive chemical profiling and quantification of Shexiang Xintongning tablets by integrating liquid chromatography-mass spectrometry and gas chromatography-mass spectrometry



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## KEYWORDS

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**Abstract** Shexiang Xintongning tablet (SXXTN) is a traditional Chinese medicine (TCM) preparation for the treatment of coronary heart disease (CHD) angina pectoris. However, due to the complexity of the compounds in SXXTN, the active chemical components responsible for the therapeutic effect are still ambiguous. The purpose of our study was to characterize the chemical profile of SXXTN and quantify the representative chemicals. The high-performance liquid chromatography-

*Abbreviations:* CHD, coronary heart disease; ESI, electrospray ionization; GC-MS, gas chromatograph coupled with mass spectrometry; GC-QQQ MS, gas chromatograph coupled with triple-quadrupole tandem mass spectrometry; HPLC, high-performance liquid chromatography; HRMS, high resolution mass spectrometry; HPLC-QTOF MS, high-performance liquid chromatography coupled with time-of-flight mass spectrometry; HPLC-QQQ MS, high performance liquid chromatography coupled with triple-quadrupole tandem mass spectrometry; LOD, limit of detection; LOQ, limit of quantitation; MRM, multiple reaction monitoring; OA, oleanane; PPD, 20(S)-protopanaxadiol; PPT, 20(S)-protopanaxatriol; RDA, Retro Diels-Alder; RSD, relative standard deviation; SXXTN, Shexiang Xintongning tablet; TCM, traditional Chinese medicine; TLC, thin-layer chromatography; TICs, typical total ion chromatograms

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Multi-component content determination;  
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HPLC-MS

phy coupled with time-of-flight mass spectrometry (HPLC-QTOF MS) method and gas chromatograph coupled with mass spectrometry (GC-MS) method were utilized to identify the chemical constituents of SXXTN. A total of 140 compounds including alkaloids, ginsenosides, organic acids, esters, triterpenes, phthalides and amino acid were identified in accordance with their retention times, accurate masses and characteristic MS/MS fragment patterns. Forty-four volatile components were characterized by GC-MS through NIST database matching. In the further research of quantitative analysis, 40 non-volatile compounds and 17 volatile compounds were determined and successfully applied for detecting in 7 batches of SXXTN samples by high performance liquid chromatography coupled with triple-quadrupole tandem mass spectrometry (HPLC-QQQ MS) and gas chromatograph coupled with triple-quadrupole tandem mass spectrometry (GC-QQQ MS) in multiple reaction monitoring (MRM) mode, respectively. The quantitative methods were verified in linearity, precision, repeatability stability and recovery. The above results indicated that the established method was practical and reliable for synthetical quality evaluation of SXXTN. In addition, our study might supplement the chemical evidence for disclosing the material basis of its therapeutic effects.

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## 1. Introduction

Preparations of TCM formulae have been extensively utilized for clinical medication owing to their therapeutic effects on various diseases and relatively low side effects (Sun et al., 2017). Shexiang Xintongning tablet (SXXTN), a newly hospital preparation which has got a wide application in China to treat coronary heart disease angina pectoris (qi stagnation and blood stasis syndrome) and reportorial clinical studies have shown its efficacy (Shen and Lu, 2005). SXXTN comprised of Artificial Musk, Corydalis Rhizoma (*Corydalis yanhusuo* W. T. Wang.), Ginseng Radix et Rhizoma (*Panax ginseng* C. A. Mey.), Chuanxiong Rhizoma (*Ligusticum chuanxiong* Hort.), Styra (*Liquidambar orientalis* Mill.) and Borneolum Syntheticum. All of the above crude drugs have been reported to be associated with the effect of SXXTN in the treatment of CHD. Here, Musk and Corydalis Rhizoma are reported to reduce infarct size and improve cardiac function (Li et al., 2008; Ling et al., 2010). The mechanisms of Ginseng Radix et Rhizoma in preventing coronary artery disease, myocardial hypertrophy, heart failure and arrhythmia are gradually being revealed (Zheng et al., 2012). Chuanxiong Rhizoma, Styra and Musk have been proved to have the role of anti-myocardial ischemia (Liu et al., 2016; Wang et al., 2019; Wu et al., 2011). Besides, Borneolum Syntheticum as an adjuvant has been reported to provide new possibilities for the treatment of atherosclerosis (Zhang et al., 2017). Muscone, tetrahydropalmatine, ginsenoside, tetramethylpyrazine, cinnamic acid and borneol have been reported as important bioactive components relevant to treatment of CHD. Recently, SXXTN was revealed have the function of reducing oxidative stress-mediated damage and enhancing angiogenesis, and might play an important role in the treatment of myocardial infarction (Li et al., 2020). Obviously, the identification and detection of the main components in SXXTN is the premise and key to reveal its active ingredients. However, the chemical composition of SXXTN is complicated, having both volatile small molecules and non-volatile components such as alkaloids, organic acids and ginsenosides. In previous studies, the chemical constituents of each crude drugs in SXXTN have been reported (He et al., 2018; Zheng et al., 2018; Yang et al., 2021; Gurbuz et al., 2013; Ding et al., 2022; Sun et al., 2014), but little attention was paid to the integral chemical composition of SXXTN. Thin-layer chromatography (TLC) and high-performance liquid chromatography (HPLC) have made powerful contributions for quality control of SXXTN (Zhang et al., 2016). Nevertheless, they were preferred to assay the limited components in SXXTN with difficult access to comprehensive chemical information. Thus, new methods for chemical profiling and quantification of SXXTN are necessary to solve the limitations of the previous techniques.

Comprehensive profiling of chemical constituents in TCM preparations is still facing great challenges on separation, detection and identification due to their structural complexities and diversities. Nowadays, various chromatographic coupled with mass spectrometry techniques, such as GC-MS (Su et al., 2008) and LC-MS (Xu et al., 2015; Luo et al., 2019), are widely used in the study of TCM preparations due to their combined superiorities of high-efficient separation and high-sensitive detection for multi-components in complex samples. On one hand, HPLC-QTOF MS and GC-MS could provide molecular weights and abundant fragment information for structure identification of compounds in TCM preparations. On the other hand, tandem mass spectrometers coupled to LC or GC systems were powerful tools for high-throughput quantitative analysis of TCM preparations owing to their high-selective simultaneous detection of multiple compounds with MRM modes. Therefore, the integration of LC-MS and GC-MS was a potentially effective approach for in-depth chemical profiling and quality control of SXXTN.

In this paper, HPLC-QTOF MS and GC-MS analysis methods were established for the global characterizations of complicated non-volatile and volatile ingredients in SXXTN. Besides, considering the representative components of all relevant single drugs in SXXTN, the abundance and activity of chemicals and the availability of reference substances, 57 compounds were further quantitatively analyzed by HPLC-QQQ MS and GC-QQQ MS method. The aims of our study were comprehensively qualitative and quantitative profiling the chemical basis of SXXTN, which were expected to improve the quality control, promote the pharmacological researches and expand the clinical applications.

## 2. Materials and methods

### 2.1. Reagents and materials

Seven batches of SXXTN were generously provided by Shandong Hongjitang Pharmaceutical Group Co., Ltd. (Shandong, China) and listed in Table S1. A total of 101 reference standards and 3 internal standards were presented in Table S2. All standards were  $\geq 98\%$  by HPLC and  $^1\text{H NMR}$  analyses.

Ultrapure water (18.2 M $\Omega$  cm) for analysis was prepared by a Milli-Q water purification system (Millipore, Bedford, MA, USA). Methanol and acetonitrile (HPLC grade) were provided by Merck (Darmstadt, Germany), and formic acid (HPLC grade) were purchased from ROE (Newark, New Castle, DE, USA). Ethyl alcohol

(HPLC grade) was offered by Sichuan Ruijingte Technology Co., Ltd. (Sichuan, China).

### 2.2. Standard solution and samples preparations

The reference standards were solubilized by 75 % methanol-aqueous solution (*v/v*) to obtain 1.00 mg/mL reserve solution and diluted with appropriate solvent to a range of proper concentrations.

In qualitative analysis, for LC-MS, the SXXTN was ground into powder. SXXTN powder (0.3003 g) was accurately weighed and ultrasonic extracted (40 kHz, 500 W) with 5 mL 75 % methanol-aqueous solution (*v/v*) for 30 min. The extracts were centrifuged (13,000 rpm, 10 min, 4°C) before LC-MS analysis. For GC-MS, the powder (0.3000 g) was accurately weighed, then sonicated for 30 min at 40 kHz with 5 mL of ethanol. The filtrate was filtered by 0.45 µm filter membrane and centrifuged before sampling.

For quantitative analysis, to determine the non-volatile constituents, each batch of SXXTN powder (0.3 g) was weighed in three parallel times, then ultrasonic extracted (40 kHz, 500 W) with 5 mL 75 % methanol for 30 min. The filtrate was filtered by 0.45 µm filter membrane and centrifuged (13000 rpm, 10 min, 4°C). For alkaloids quantification (group A, 24 alkaloids), the supernatant was diluted after adding proper nitidine chloride (IS1, 1.11 µg/mL) as internal standard. For ginsenosides and acids quantification (group B, 14 ginsenosides, cinnamic acid and phenylalanine), the supernatant was diluted after proper saikosaponin C (IS2, 0.985 µg/mL) adding. To determine the volatile constituents, about 0.3 g the powder of SXXTN was extracted with 10 mL ethanol under ultrasonic conditions in ice-water bath for 20 min. The extraction was filtered through syringe filter (0.45 µm) and centrifuged. Isoborneol, borneol, 3-phenylpropyl cinnamate and cinnamyl cinnamate possessed significantly higher abundances in SXXTN comparing with other volatile components, indicating the large differences of contents among various compounds. Therefore, the supernatant was diluted 10 times before injecting to GC-MS for quantitative analysis of high-abundant volatile compounds (Group D), whereas directly injected for others with relatively low-abundances (Group C). And a certain amount of naphthalene (IS3, 23.2 µg/mL) was added to the supernatants as internal standard.

### 2.3. HPLC-QTOF MS analysis conditions

Agilent 1290 HPLC system (Agilent corporation, USA) was used to determine the non-volatile components of SXXTN. A ZORBAX Eclipse Plus C18 column (150 × 2.1 mm, 1.8 µm, Agilent Technologies, Santa Clara, USA) was used for sample separation. The mobile phase consisted of 0.1 % (*v/v*) formic acid in water (A) and acetonitrile (B) with the gradient elution set as follows: 0–3 min, 10 %–12 % B; 3–8 min, 12 %–17 % B; 8–20 min, 17 %–22 % B; 20–30 min, 22 %–35 % B; 30–45 min, 35 %–42 % B; 45–50 min, 42 %–60 % B; 50–57 min, 60 % B; 57–60 min, 60 %–80 % B; 60–68 min, 80 %–100 %. The flow rate was set at 0.4 mL/min, and the column temperature was maintained at 30 °C. Sample volume was 1 µL for injection.

The Q-TOF mass spectrometer equipped with electrospray ionization (ESI) source was used to acquire data in positive

and negative ion modes. The operation conditions were as below: drying gas (N<sub>2</sub>) temperature, 300 °C; drying gas flow, 8.0 L/min; nebulizer gas (N<sub>2</sub>) pressure, 35 psig; sheath gas (N<sub>2</sub>) temperature, 350 °C; sheath gas flow, 11.0 L/min; capillary voltage positive ion mode, 4000 V; negative ion mode, 3500 V; fragmentor voltage, 120 V; skimmer voltage, 65 V. Full-scan MS and MS/MS data was collected over the *m/z* range of 50–1500 using extended dynamic range. Collision energy of secondary mass spectrometry was set as 15 eV, 30 eV and 45 eV.

### 2.4. GC-MS analysis conditions

Compound identification was performed by Agilent 7890B GC system combined with Agilent 5977 Mass Selection Detector. Samples were separated by Agilent HP-5MS (30 m × 0.25 mm, 0.25 µm) column. The carrier gas was high purity helium, and the flow rate was 1 mL/min. Initial column temperature was 60 °C, and programmed to rise at 20 °C/min to 85 °C (1 min held), 5 °C/min to 100 °C (5 min held), 15 °C/min to 150 °C (6 min held), 5 °C/min to 200 °C (4 min held), rising at 5 °C/min to 280 °C (5 min held). The injection volume was 1 µL and the splitting ratio was 30:1. The temperature of injector and aux heaters was controlled at 250°C and 280°C, respectively. MS quadrupole and ion source temperature were maintained at 150°C and 230°C, severally. MS data were recorded at 70 eV and acquired in full scan mode over the range of *m/z* 40–600.

### 2.5. HPLC-QQQ MS analysis conditions

The quantitative analysis was performed on the Shimadzu LCMS-8050 triple quadrupole tandem mass spectrometry detector (Shimadzu, Kyoto, Japan) with an Agilent Zorbax Eclipse Plus C18 column (2.1 × 150 mm, 1.8 µm, Agilent Technologies, Santa Clara, USA). For group A, 0.1 % (*v/v*) formic acid water (A) and acetonitrile (B) were used as mobile phases, and the gradient elution procedure was as follows: 0–12 min, 19 %–20 % B; 12–14 min, 20 %–35 % B; 14–16 min, 35 %–90 % B; 16–19 min, 90 %–100 % B. For group B, the mobile phase was water (A) and acetonitrile (B), with the following gradient elution: 0–3 min, 10 %–12 % B; 3–6 min, 12 %–35 % B; 6–14 min, 35 %–36.5 % B; 14–15 min, 36.5 %–90 % B; 15–19 min, 90 %–100 % B. The flow rate was maintained at 0.4 mL/min, with the injection volume 2 µL for all samples. The MS conditions were as below: capillary voltage, 4000 V; drying gas temperature, 300 °C. The flow rate of drying gas (N<sub>2</sub>) and nebulizer gas (N<sub>2</sub>) was 10.0 L/min and 3.0 L/min, severally. Analytes were determined in MRM modes, and the optimized parameters were shown in [Table S3 and S4](#).

### 2.6. GC-QQQ MS analysis conditions

The quantitative analysis of volatile components was operated on an Agilent 7890B gas chromatography coupling to Agilent 5977A mass spectrometry (Agilent, Santa Clara, CA, USA). For group C, the initial column temperature was 60 °C, and programmed to rise at 20 °C/min to 85 °C, 5 °C/min to 100 °C (5 min held), 15 °C/min to 150 °C, 5 °C/min to 180 °C (1 min held), finally rising at 15 °C/min to 280 °C (2 min held). For group D, the initial column temperature was set at 100 °C,

and programmed to rise at 10 °C/min to 110 °C, 3 °C/min to 120 °C, 55 °C/min to 265 °C, finally rising at 18 °C/min to 280 °C (2 min held). The injection volume was 1 µL and the splitting ratio was 10:1. The MRM parameters for all analytes are presented in Table S5 and S6. Other analytical conditions refer to Section 2.4.

### 3. Results and discussion

#### 3.1. Qualitative analysis of SXXTN based on diagnostic ion strategy by HPLC-QTOF MS

The HPLC-QTOF MS conditions of the mobile phase systems (methanol-aqueous, acetonitrile-aqueous, and acetonitrile-aqueous with 0.1 % formic acid), gradient program, column temperature (25 °C, 30 °C, and 35 °C) and the flow rate (0.2, 0.3 and 0.4 mL/min) were optimized in order to obtain overall constituents of SXXTN with good resolution within a short analysis. The total peak area was adopted as a criterion for optimization. Ultimately, the optimum conditions mentioned in Section 2.3 were preferred.

Diagnostic ion strategy is regarded as a powerful approach for rapid characterization of chemicals in TCMs based on the principle that similar chemical constituents have similar cleavage rules and the fragmentation information, which is applicable for the identification of structural analogues in complex TCMs and formulae (Wang et al., 2017). In our study, by comparing with the reference standards, the known compounds were marked. On the basis of MS/MS analysis of authentic compounds, the characteristic fragmentation pathways of compounds with the same carbon skeleton were presented,

the obtained rules were further applied to the structural characterization of its derivatives. For other unknown compounds, identification based on MS/MS spectra and relevant literature or online databases, including PubChem search (<https://pubchem.ncbi.nlm.nih.gov/>) and the Human metabolome database (<https://www.hmdb.ca/>). The typical total ion chromatograms (TICs) of SXXTN by HPLC-QTOF MS in both of positive and negative ion modes were displayed in Fig. 1. Totally, 140 compounds were identified based on diagnostic ion strategy, including 60 alkaloids, 34 ginsenosides, 21 organic acids, 12 phthalides, 10 triterpenes, 2 esters and 1 amino acid. The chemical structures and detailed information of compounds could be viewed in Fig. 2 and Table 1, respectively. The MS/MS spectra and fragmentation pathways of the representative chemicals were shown in Fig S1 and Fig S2.

##### 3.1.1. Identification of alkaloids in SXXTN

Sixty alkaloids in SXXTN demonstrated quasi-molecular ions  $[M + H]^+$  or  $[M]^+$  in positive ion mode and listed in Table 1, mostly originated from *Corydalis Rhizoma* and identified as four main types, including tetrahydroproberberines, berberines, protopines and aporphines.

A total of 19 tetrahydroprotoberberine-type (21, 22, 24, 28, 29, 33, 34, 35, 38, 39, 42, 47, 48, 50, 51, 58, 59, 60, 72) and 3 protopine-type alkaloids (36, 44, 69) were tentatively identified or unambiguously characterized with the characteristic cleavage pathway of Retro Diels-Alder (RDA) reaction, which can be used to distinguish them from other types of alkaloids (Yuan et al., 2016). In MS/MS of tetrahydropalmatine (48) shown as Fig. S1A, the fragment ion with the strongest intensity was located at  $m/z$  192.1019  $[M + H - C_{10}H_{12}O_2]^+$ , and it

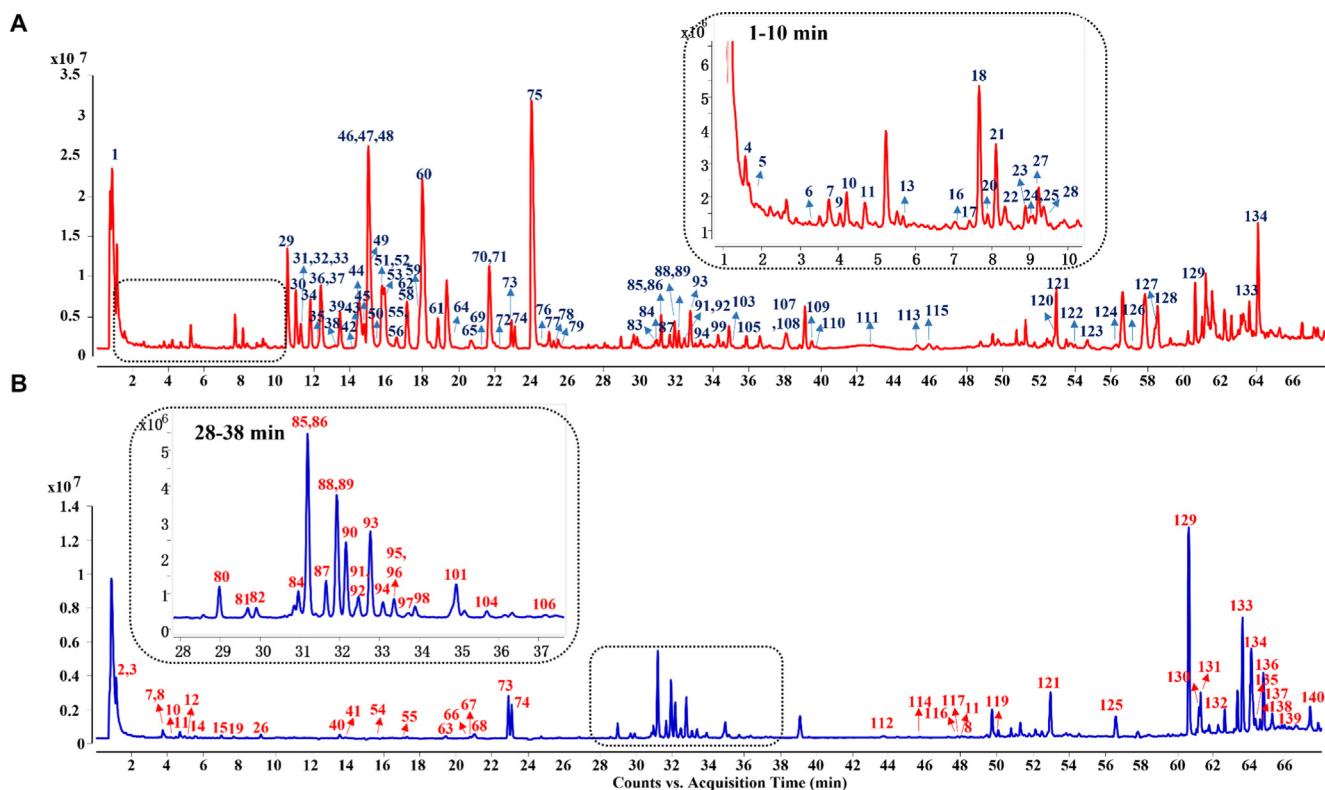


Fig. 1 Total ion current chromatograms of SXXTN in positive ion mode (A) and negative ion mode (B) by HPLC-QTOF MS.

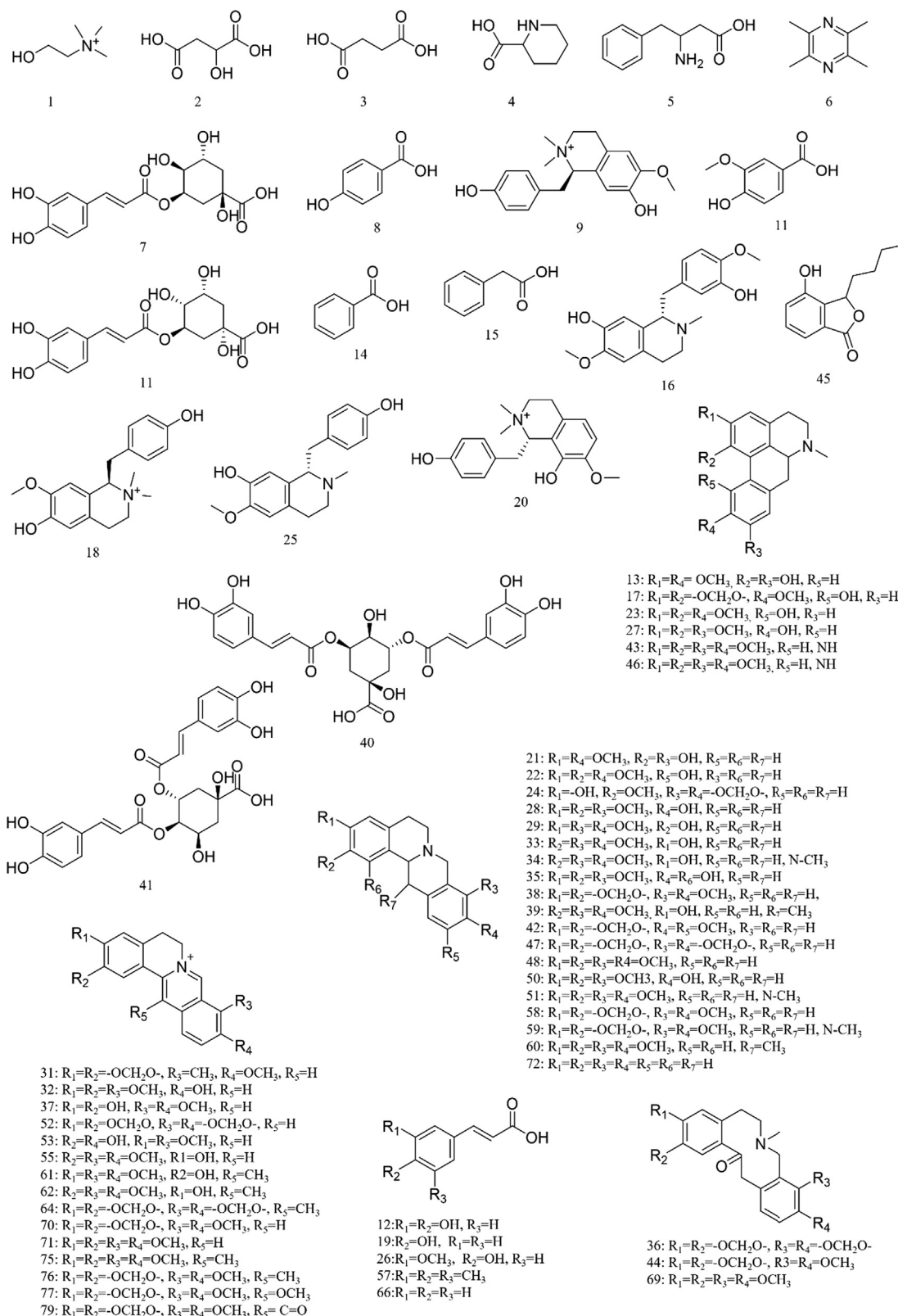


Fig. 2 Structures of chemical constituents from SXXTN.

was found that the complementary fragment ion  $m/z$  165.0909  $[M + H-C_{11}H_{13}NO_2]^+$  was corresponding to the RDA reaction of C ring. The detailed fragmentation pathways of tetrahydropalmitine (**48**) were displayed in Fig. S2A. For

protopine-type alkaloids, C-14 position is linked to oxygen to form carbonyl, which is easy to dehydrate and forms stable fragment ions, thus distinguishing it from tetrahydroberberine-type alkaloids (Yuan et al., 2016).

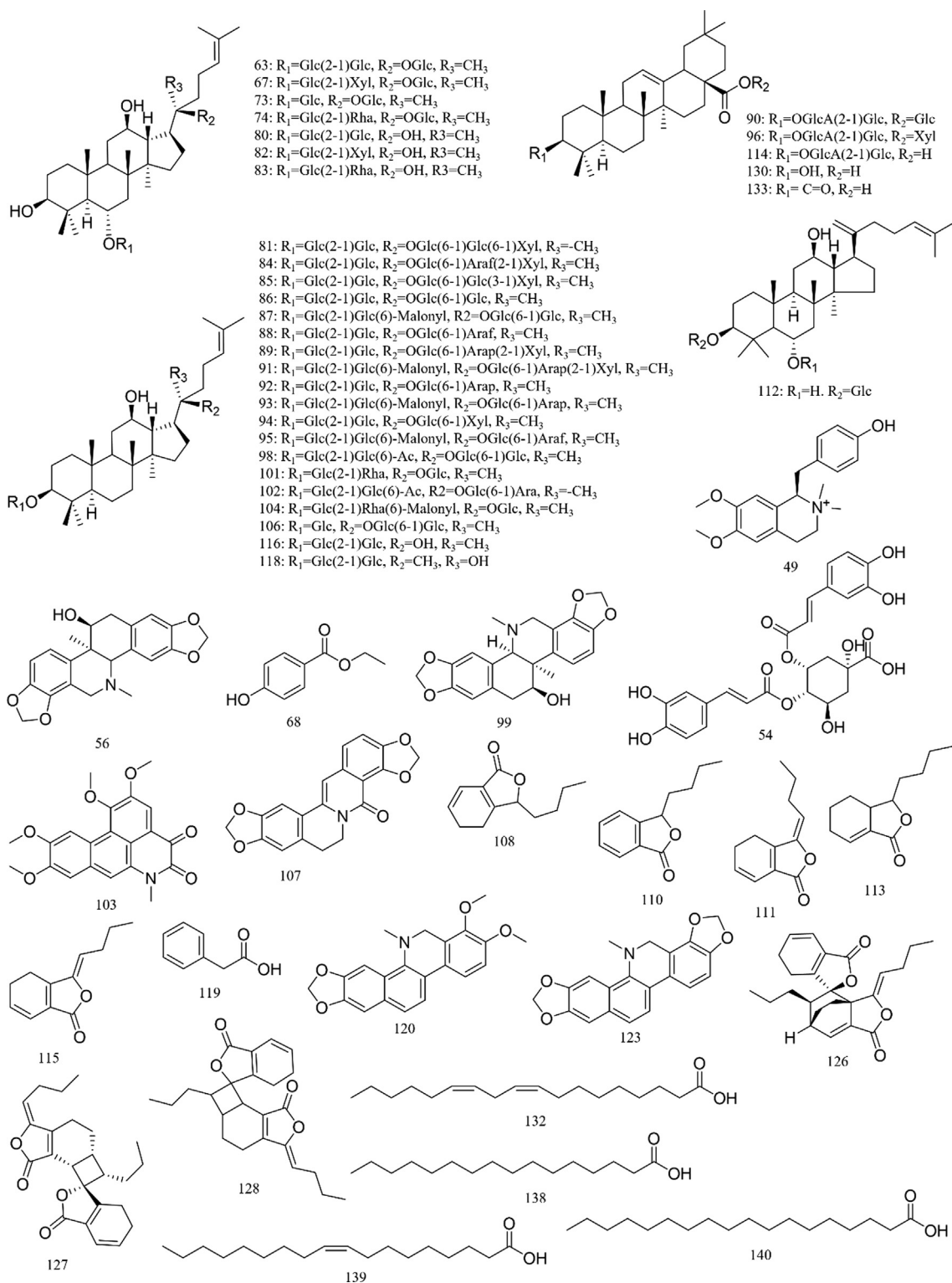


Fig. 2 (continued)

Taking protopine (36) as an example (Fig. S1B), the product ions at  $m/z$  336.1232 [ $M + H - H_2O$ ]<sup>+</sup> and  $m/z$  188.0709 [ $M + H - C_9H_8O_2 - H_2O$ ]<sup>+</sup> may be formed by neutral losses of H<sub>2</sub>O from molecular ions and  $m/z$  206.0813 (Fig. S2B).

Fifteen protoberberine-type (31, 32, 37, 52, 53, 55, 61, 62, 64, 70, 71, 75, 76, 77, 79) and six aporphine-type alkaloids (13, 17, 23, 27, 43, 46) were identified in SXXTN with the cleavage pathway based on the fragmentation of substituents

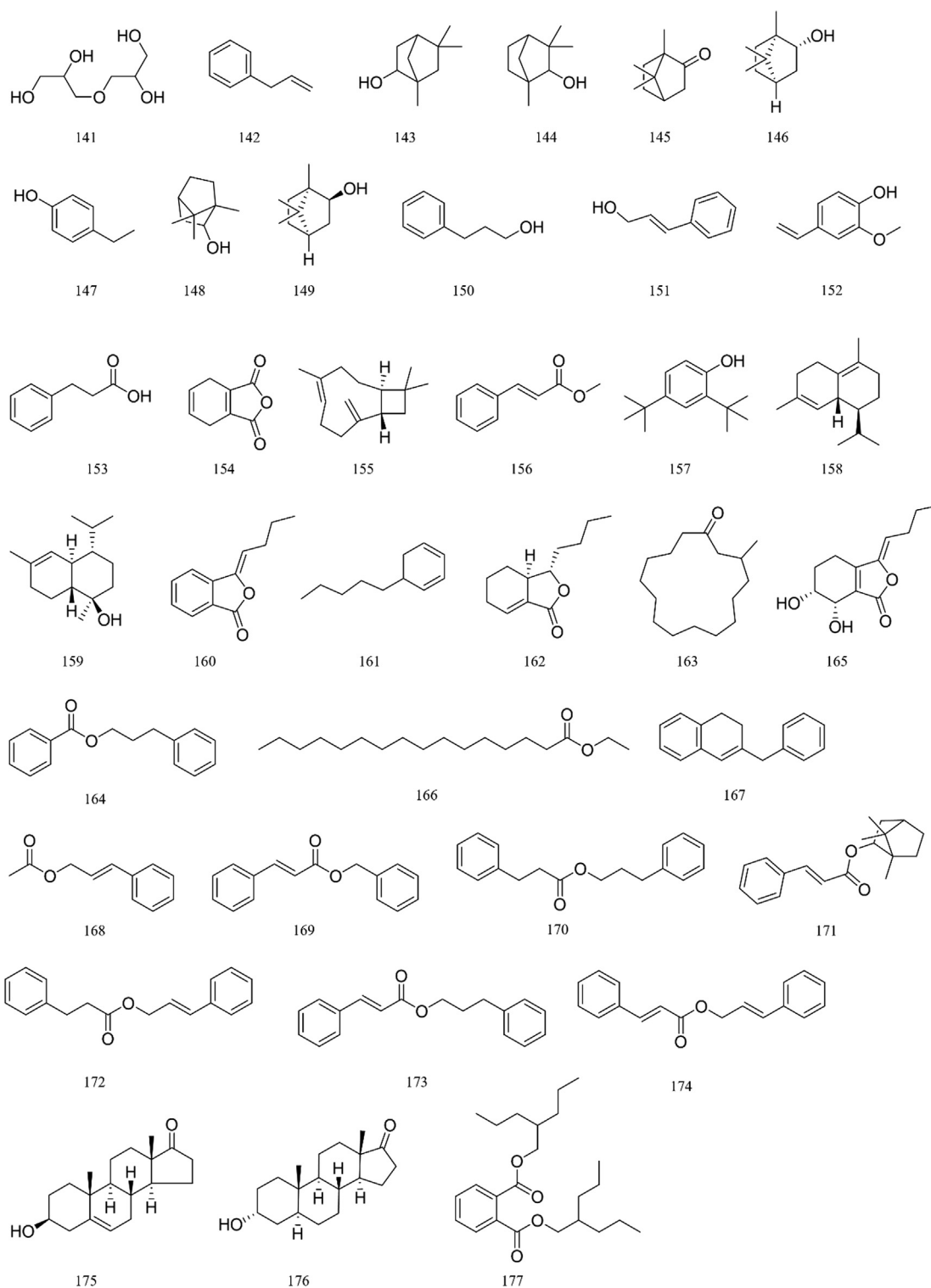


Fig. 2 (continued)

(Yuan et al., 2016) as displayed in Fig. S2C. For protoberberine-type alkaloids, usually losing 15 Da ( $-\text{CH}_3$ ) substituent as seen in MS/MS spectrum (Fig. S1C) of berberine (70), the main product ions appeared at  $m/z$  320.0919  $[\text{M}-\text{CH}_4]^+$  and  $m/z$  321.0978  $[\text{M}-\text{CH}_3]^+$ . In addition, the

successive losses of  $\text{CH}_3$  and  $\text{CO}$  were the characteristic cleavage pathway of this alkaloid. For aporphine-type alkaloids, the fragment ions with the highest relative abundance usually appear when the methoxy group at 31 Da is lost. For example, fragment ion  $m/z$  297.1108  $[\text{M} + \text{H}-\text{OCH}_3]^+$  was found in

**Table 1** Characterization of chemical constituents of SXXTN by HPLC-QTOF MS.

No.	t <sub>R</sub> (min)	Formula	Precursor ions ( <i>m/z</i> )	Diff (ppm)	Fragment ions ( <i>m/z</i> )	Identification	Structural Types
1	0.88	C <sub>5</sub> H <sub>14</sub> NO <sup>+</sup>	104.1069 [M + H] <sup>+</sup>	-0.87	58.0658,60.0813	Choline	Alkaloid
2*	1.06	C <sub>4</sub> H <sub>6</sub> O <sub>5</sub>	133.0143 [M-H] <sup>-</sup>	0.40	115.0036,89.0252,71.0148	Malic acid	Organic acid
3	1.33	C <sub>4</sub> H <sub>6</sub> O <sub>4</sub>	117.0192 [M-H] <sup>-</sup>	-1.13	99.9252,73.0303	Succinic acid	Organic acid
4*	1.46	C <sub>6</sub> H <sub>11</sub> NO <sub>2</sub>	130.0867 [M + H] <sup>+</sup>	3.42	70.0653,84.0814,56.0510	<i>DL</i> -pipecolic acid	Organic acid
5*	1.56	C <sub>9</sub> H <sub>11</sub> NO <sub>2</sub>	166.0864 [M + H] <sup>+</sup>	0.87	120.0806,103.0547	Phenylalanine	Amino acid
6*	3.28	C <sub>8</sub> H <sub>12</sub> N <sub>2</sub>	137.1067 [M + H] <sup>+</sup>	-4.50	55.0550,80.0475	Tetramethylpyrazine	Alkaloid
7	3.74	C <sub>16</sub> H <sub>18</sub> O <sub>9</sub>	353.087 [M-H] <sup>-</sup>	-2.28	191.0562,179.0339	Neochlorogenic acid	Organic acid
8*	3.89	C <sub>7</sub> H <sub>6</sub> O <sub>3</sub>	137.024 [M-H] <sup>-</sup>	-3.05	93.0348,65.0413	4-Hydroxybenzoic acid	Organic acid
9	4.07	C <sub>19</sub> H <sub>24</sub> NO <sub>3</sub>	314.1755 [M] <sup>+</sup>	1.37	269.1173,175.0748,107.0491	Magnocurarine	Alkaloid
10*	4.22	C <sub>16</sub> H <sub>18</sub> O <sub>9</sub>	353.0901 [M-H] <sup>-</sup>	6.50	191.0563,179.0364,173.0456	Chlorogenic acid	Organic acid
11*	4.75	C <sub>8</sub> H <sub>8</sub> O <sub>4</sub>	167.0341 [M-H] <sup>-</sup>	-5.28	152.3304,123.0429,108.0130	Vanillic acid	Organic acid
12*	5.04	C <sub>9</sub> H <sub>8</sub> O <sub>4</sub>	179.0352 [M-H] <sup>-</sup>	1.22	135.0458	Caffeic acid	Organic acid
13	5.69	C <sub>19</sub> H <sub>21</sub> NO <sub>4</sub>	328.1547 [M + H] <sup>+</sup>	1.11	265.0853,297.1108,282.0882,165.0713	Isoboldine	Alkaloid
14	5.71	C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>	121.0291 [M-H] <sup>-</sup>	-3.33	92.0281,76.9491	Benzoic acid	Organic acid
15	7.04	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	135.0456 [M-H] <sup>-</sup>	3.31	120.0225,92.0278	Phenylacetic acid	Organic acid
16	7.07	C <sub>19</sub> H <sub>24</sub> NO <sub>4</sub>	330.1702 [M] <sup>+</sup>	0.65	299.1269,192.1032,175.0186, 143.0016	Reticuline	Alkaloid
17	7.44	C <sub>19</sub> H <sub>19</sub> NO <sub>4</sub>	326.1386 [M + H] <sup>+</sup>	-0.26	295.0969,263.0696,235.0752	Bulbocapnine	Alkaloid
18	7.66	C <sub>19</sub> H <sub>24</sub> NO <sub>3</sub>	314.1759 [M] <sup>+</sup>	2.64	269.1173,237.0907,175.0745, 107.0490	Lotusine	Alkaloid
19	7.71	C <sub>9</sub> H <sub>8</sub> O <sub>3</sub>	163.0402 [M-H] <sup>-</sup>	0.81	119.0497	4-Hydroxycinnamic acid	Organic acid
20	7.92	C <sub>19</sub> H <sub>24</sub> NO <sub>3</sub>	314.1743 [M] <sup>+</sup>	-2.45	237.0885,209.0961,107.0488	Oblongine	Alkaloid
21*	8.13	C <sub>19</sub> H <sub>21</sub> NO <sub>4</sub>	328.1549 [M + H] <sup>+</sup>	1.72	178.0862,163.0627,151.0755	Scoulerine	Alkaloid
22	8.39	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1697 [M + H] <sup>+</sup>	-0.83	178.0856,326.1402	Corytenchine	Alkaloid
23*	8.83	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1699 [M + H] <sup>+</sup>	-0.25	279.1015,311.1278,342.1699	Isocorydine	Alkaloid
24	9.06	C <sub>19</sub> H <sub>19</sub> NO <sub>4</sub>	326.1393 [M + H] <sup>+</sup>	1.89	178.0854,151.0730	Cheilanthifoline	Alkaloid
25	9.10	C <sub>18</sub> H <sub>21</sub> NO <sub>3</sub>	300.1596 [M + H] <sup>+</sup>	0.60	269.1175,237.0921,192.1025	<i>N</i> -Methylcoclaurine	Alkaloid
26*	9.19	C <sub>10</sub> H <sub>10</sub> O <sub>4</sub>	193.0505 [M-H] <sup>-</sup>	-0.69	178.0277,149.0594,134.0374	Ferulic acid	Organic acid
27	9.21	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1698 [M + H] <sup>+</sup>	-0.54	192.1020,148.0753	Lirioferine	Alkaloid
28*	9.36	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1700 [M + H] <sup>+</sup>	-0.25	327.1472,165.0909,192.1016	Corydalmine	Alkaloid
29*	10.57	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1707 [M + H] <sup>+</sup>	2.09	327.1472,326.1414,178.0866	Tetrahydrocolumbamine	Alkaloid
30	11.02	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1856 [M + H] <sup>+</sup>	-0.10	341.1609,326.1389,308.1276, 192.1020,177.0783	<i>N</i> -Methyltetrahydropalmatrubine	Alkaloid
31	11.24	C <sub>19</sub> H <sub>16</sub> NO <sub>4</sub>	322.1074 [M + H] <sup>+</sup>	-2.57	307.0839,294.2059,279.0888	Berberrubine	Alkaloid
32	11.29	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1857 [M + H] <sup>+</sup>	0.18	341.1617,326.1390,192.1018, 165.0909,150.0672	<i>N</i> -Methylcorydalmine	Alkaloid



**Table 1** (continued)

No.	t <sub>R</sub> (min)	Formula	Precursor ions ( <i>m/z</i> )	Diff (ppm)	Fragment ions ( <i>m/z</i> )	Identification	Structural Types
33*	11.35	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1695 [M + H] <sup>+</sup>	-1.42	326.1387,178.0860,163.0629, 151.0725,119.0489	Corypalmine	Alkaloid
34	11.83	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1860 [M + H] <sup>+</sup>	1.03	341.1590,326.1380,308.1283,192.1020	<i>N</i> -Methylcorypalmine	Alkaloid
35	12.26	C <sub>20</sub> H <sub>23</sub> NO <sub>5</sub>	358.1652 [M] <sup>+</sup>	0.84	356.1856,340.1516	Capaurimine	Alkaloid
36*	12.41	C <sub>20</sub> H <sub>19</sub> NO <sub>5</sub>	354.1343 [M + H] <sup>+</sup>	1.98	336.1229,206.0812,189.0777,275.0705	Corydinine	Alkaloid
37*	12.45	C <sub>19</sub> H <sub>18</sub> NO <sub>4</sub> <sup>+</sup>	324.1228 [M] <sup>+</sup>	-0.72	307.9500,280.0005,309.0006	Demethyleneberberine	Alkaloid
38	13.17	C <sub>20</sub> H <sub>21</sub> NO <sub>4</sub>	340.1538 [M + H] <sup>+</sup>	-1.57	324.1229,309.1100,296.1274	Sinactine	Alkaloid
39	13.41	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1855 [M + H] <sup>+</sup>	-0.38	341.1632,340.1554,326.1415	Corybulbine	Alkaloid
40	13.58	C <sub>25</sub> H <sub>24</sub> O <sub>12</sub>	515.118 [M-H] <sup>-</sup>	-2.91	353.0853,191.0556	Isochlorogenic acid A	Organic acid
41	13.84	C <sub>25</sub> H <sub>24</sub> O <sub>12</sub>	515.1182 [M-H] <sup>-</sup>	-6.14	353.0867,191.0541	Isochlorogenic acid B	Organic acid
42	14.20	C <sub>20</sub> H <sub>20</sub> NO <sub>4</sub> <sup>+</sup>	338.1390 [M] <sup>+</sup>	0.93	322.1055,380.0926,294.1122,280.0937	Tetrahydrocorysamine	Alkaloid
43*	14.40	C <sub>20</sub> H <sub>23</sub> NO <sub>4</sub>	342.1679 [M + H] <sup>+</sup>	-0.25	325.1432,294.1250,279.1035,251.1113	Norglaucine	Alkaloid
44*	14.54	C <sub>21</sub> H <sub>23</sub> NO <sub>5</sub>	370.1659 [M + H] <sup>+</sup>	2.70	188.0706,290.0939,321.1141,352.1548	Allocriptopine	Alkaloid
45	14.79	C <sub>12</sub> H <sub>14</sub> O <sub>3</sub>	207.1016 [M + H] <sup>+</sup>	0.14	189.0893,175.0196,123.0433,67.0544	4-Hydroxy-3-butylphthalide	Phthalide
46*	14.93	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1846 [M + H] <sup>+</sup>	-2.91	294.1254,310.1206,325.1436	Glaucine	Alkaloid
47*	14.96	C <sub>19</sub> H <sub>17</sub> NO <sub>4</sub>	324.1243 [M + H] <sup>+</sup>	2.05	176.0713,294.1251,149.0579,	Tetrahydrocoptisine	Alkaloid
48*	15.04	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1879 [M + H] <sup>+</sup>	2.71	192.1032,165.0918,194.1271,326.1479	Tetrahydropalmatine	Alkaloid
49	15.09	C <sub>20</sub> H <sub>25</sub> NO <sub>3</sub> <sup>+</sup>	328.1916 [M + H] <sup>+</sup>	2.68	283.1348,251.1070,236.0850	6- <i>O</i> -methylotusine	Alkaloid
50*	15.36	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	356.1856 [M + H] <sup>+</sup>	-0.38	354.1478,325.1338,194.2615	Yuanhunine	Alkaloid
51	15.75	C <sub>22</sub> H <sub>27</sub> NO <sub>4</sub>	370.2008 [M + H] <sup>+</sup>	-1.31	354.1693,206.1174,190.0871,165.0900	<i>N</i> - Methyltetrahydropalmatine	Alkaloid
52*	15.79	C <sub>19</sub> H <sub>14</sub> NO <sub>4</sub> <sup>+</sup>	320.0926 [M] <sup>+</sup>	2.70	292.0972,262.0880,234.0919	Coptisin	Alkaloid
53*	15.90	C <sub>20</sub> H <sub>20</sub> NO <sub>4</sub> <sup>+</sup>	338.1395 [M] <sup>+</sup>	2.41	323.1142,322.1015	Columbamine	Alkaloid
54	16.36	C <sub>25</sub> H <sub>24</sub> O <sub>12</sub>	515.1167 [M-H] <sup>-</sup>	-5.44	353.0848,191.0513	Isochlorogenic acid C	Organic acid
55*	16.58	C <sub>20</sub> H <sub>20</sub> NO <sub>4</sub> <sup>+</sup>	338.1390 [M] <sup>+</sup>	0.93	323.1167,294.1143,322.1093	Jatrorrhizine	Alkaloid
56	16.70	C <sub>12</sub> H <sub>14</sub> O <sub>3</sub>	207.1015 [M + H] <sup>+</sup>	-0.34	189.0818,161.0967	Senkyunolide F	Phthalide
57	17.10	C <sub>12</sub> H <sub>14</sub> O <sub>5</sub>	237.0753 [M-H] <sup>-</sup>	-6.53	193.0849,108.0193	Trimethoxycinnamic acid	Organic acid
58*	17.16	C <sub>20</sub> H <sub>21</sub> NO <sub>4</sub>	340.1556 [M + H] <sup>+</sup>	3.72	176.0716,149.0608	Canadine	Alkaloid
59	17.93	C <sub>21</sub> H <sub>24</sub> NO <sub>4</sub> <sup>+</sup>	354.1708 [M] <sup>+</sup>	2.30	165.0906,190.0876	<i>N</i> -Methylcanadine	Alkaloid
60*	18.03	C <sub>22</sub> H <sub>27</sub> NO <sub>4</sub>	370.2023 [M + H] <sup>+</sup>	2.74	355.1790,192.1032,176.0731,165.0912	Corydaline	Alkaloid
61	18.86	C <sub>21</sub> H <sub>22</sub> NO <sub>4</sub> <sup>+</sup>	352.1543 [M] <sup>+</sup>	-0.10	337.1308,322.1064,309.1345,293.1041	13-Methylcolumbamine	Alkaloid
62	19.38	C <sub>21</sub> H <sub>22</sub> NO <sub>4</sub> <sup>+</sup>	352.1550 [M] <sup>+</sup>	1.89	337.1322,322.1101,336.1254	Dehydrocorybulbine	Alkaloid
63*	19.45	C <sub>48</sub> H <sub>82</sub> O <sub>19</sub>	1007.538 [M + COOH] <sup>-</sup>	-5.16	961.5223,799.4737,637.4235,475.3723	20- <i>O</i> -glucoginsenoside Rf	Ginsenoside
64*	19.58	C <sub>20</sub> H <sub>16</sub> NO <sub>4</sub> <sup>+</sup>	334.1069 [M] <sup>+</sup>	0.35	291.0887,261.0785,147.0680	Worenine	Alkaloid
65	20.73	C <sub>20</sub> H <sub>22</sub> NO <sub>5</sub> <sup>+</sup>	356.1503 [M] <sup>+</sup>	2.95	338.1383,322.1085,308.1241,192.0662, 164.0828,149.0594	Pseudotetrahydropalmatine	Alkaloid
66*	20.73	C <sub>9</sub> H <sub>8</sub> O <sub>2</sub>	147.0447 [M-H] <sup>-</sup>	-3.08	119.0485,117.0334,103.0543	Cinnamic acid	Organic acid
67*	20.78	C <sub>47</sub> H <sub>80</sub> O <sub>18</sub>	977.5267 [M + COOH] <sup>-</sup>	-6.11	931.5112,799.4755,637.4223,475.3682	Notoginsenoside R <sub>1</sub>	Ginsenoside
68*	21.28	C <sub>9</sub> H <sub>10</sub> O <sub>3</sub>	165.0562 [M-H] <sup>-</sup>	2.92	137.0213,92.0268	Ethylparaben	Ester

(continued on next page)

**Table 1** (continued)

No.	t <sub>R</sub> (min)	Formula	Precursor ions ( <i>m/z</i> )	Diff (ppm)	Fragment ions ( <i>m/z</i> )	Identification	Structural Types
69	21.37	C <sub>22</sub> H <sub>27</sub> NO <sub>5</sub>	386.1955 [M + H] <sup>+</sup>	-1.81	368.1833,190.0847,178.0980	Muramine	Alkaloid
70*	21.58	C <sub>20</sub> H <sub>18</sub> NO <sub>4</sub>	336.1237 [M] <sup>+</sup>	1.98	321.1008,306.0783,320.0931	Berberine	Alkaloid
71*	21.96	C <sub>21</sub> H <sub>22</sub> NO <sub>4</sub> <sup>+</sup>	352.1551 [M] <sup>+</sup>	2.17	337.1324,322.1101,308.1302,294.1139, 279.0938	Palmatine	Alkaloid
72	22.23	C <sub>22</sub> H <sub>26</sub> NO <sub>4</sub> <sup>+</sup>	368.1836 [M] <sup>+</sup>	-5.35	352.1524,338.1259,192.0987	Tetrahydroprotoberberine	Alkaloid
73*	22.90	C <sub>42</sub> H <sub>72</sub> O <sub>14</sub>	845.4884 [M + COOH] <sup>-</sup>	-2.38	799.4690,637.4205,475.3705,161.0433	Ginsenoside Rg <sub>1</sub>	Ginsenoside
74*	23.09	C <sub>48</sub> H <sub>82</sub> O <sub>18</sub>	991.545 [M + COOH] <sup>-</sup>	-3.35	945.5269,783.4777,637.4213,475.3705	Ginsenoside Re	Ginsenoside
75*	24.01	C <sub>22</sub> H <sub>24</sub> NO <sub>4</sub>	366.1705 [M] <sup>+</sup>	1.41	351.1488,350.1417,336.1260	Dehydrocorydaline	Alkaloid
76*	24.47	C <sub>21</sub> H <sub>20</sub> NO <sub>4</sub> <sup>+</sup>	350.1379 [M] <sup>+</sup>	-2.24	334.1062,306.1124,320.0961	13-Methylberberine	Alkaloid
77	25.01	C <sub>22</sub> H <sub>24</sub> NO <sub>4</sub> <sup>+</sup>	366.17 [M] <sup>+</sup>	-0.91	336.1226,351.1454	13-Methoxyberberine	Alkaloid
78	25.24	C <sub>19</sub> H <sub>14</sub> NO <sub>4</sub> <sup>+</sup>	320.0922 [M] <sup>+</sup>	1.45	292.0947,262.0843,234.0893	Coptisin isomer	Alkaloid
79	25.51	C <sub>20</sub> H <sub>18</sub> NO <sub>5</sub> <sup>+</sup>	352.1197 [M] <sup>+</sup>	4.97	336.0877,322.0695,306.0756,292.0591	13-Oxoberberine	Alkaloid
80*	28.91	C <sub>42</sub> H <sub>72</sub> O <sub>14</sub>	845.4891 [M + COOH] <sup>-</sup>	-1.55	799.4723,637.4220,475.3721,161.0441	Ginsenoside Rf	Ginsenoside
81*	29.69	C <sub>59</sub> H <sub>100</sub> O <sub>27</sub>	1239.6357 [M-H] <sup>-</sup>	-1.79	1107.5968,1077.5859	Notoginsenoside R <sub>4</sub>	Ginsenoside
82*	29.90	C <sub>41</sub> H <sub>70</sub> O <sub>13</sub>	815.4785 [M + COOH] <sup>-</sup>	-1.65	769.4610,637.4229,475.3727, 161.0449,391.2853	Notoginsenoside R <sub>2</sub>	Ginsenoside
83*	30.84	C <sub>42</sub> H <sub>72</sub> O <sub>13</sub>	829.4945 [M + COOH] <sup>-</sup>	-1.20	783.4794,637.4245,475.3739, 391.2800,161.0439	20( <i>S</i> )-Ginsenoside Rg <sub>2</sub>	Ginsenoside
84*	30.93	C <sub>58</sub> H <sub>98</sub> O <sub>26</sub>	1245.6046 [M + Cl] <sup>-</sup>	0.45	1209.6246,1077.5795	Ginsenoside Ra <sub>2</sub>	Ginsenoside
85*	31.18	C <sub>59</sub> H <sub>100</sub> O <sub>27</sub>	1239.6358 [M-H] <sup>-</sup>	-1.17	1107.8589,864.3445,783.5100	Ginsenoside Ra <sub>3</sub>	Ginsenoside
86*	31.19	C <sub>54</sub> H <sub>92</sub> O <sub>23</sub>	1107.5962 [M-H] <sup>-</sup>	0.49	945.5333,783.4762,179.0540	Ginsenoside Rb <sub>1</sub>	Ginsenoside
87	31.66	C <sub>57</sub> H <sub>94</sub> O <sub>26</sub>	1193.594 [M-H] <sup>-</sup>	-1.72	1159.5908,1107.5793,1089.5701, 945.5294	Malonylginsenoside Rb <sub>1</sub>	Ginsenoside
88*	31.92	C <sub>53</sub> H <sub>90</sub> O <sub>22</sub>	1123.5884 [M + COOH] <sup>-</sup>	-1.94	1077.5685,945.5308,784.4864	Ginsenoside Rc	Ginsenoside
89*	31.92	C <sub>58</sub> H <sub>98</sub> O <sub>26</sub>	1245.6046 [M + Cl] <sup>-</sup>	0.45	1209.6278,945.5503	Ginsenoside Ra <sub>1</sub>	Ginsenoside
90*	32.14	C <sub>48</sub> H <sub>76</sub> O <sub>19</sub>	955.4849 [M-H] <sup>-</sup>	-6.18	793.4271,631.3664,523.3719,455.3456	Ginsenoside Ro	Ginsenoside
91	32.39	C <sub>61</sub> H <sub>100</sub> O <sub>29</sub>	1295.6251 [M-H] <sup>-</sup>	-2.05	1251.6282,1209.6131,1191.6035, 1059.5610	Malonylginsenoside Ra <sub>1</sub> /Ra <sub>2</sub>	Ginsenoside
92	32.47	C <sub>56</sub> H <sub>92</sub> O <sub>25</sub>	1163.5837 [M-H] <sup>-</sup>	-1.54	1119.5849,1077.5763,1059.5647, 927.5217	Malonylginsenoside Rb <sub>2</sub>	Ginsenoside
93*	32.77	C <sub>53</sub> H <sub>90</sub> O <sub>22</sub>	1123.5874 [M + COOH] <sup>-</sup>	-2.83	783.4804,945.5378,149.0458	Ginsenoside Rb <sub>2</sub>	Ginsenoside
94*	33.09	C <sub>53</sub> H <sub>90</sub> O <sub>22</sub>	1113.5633 [M + Cl] <sup>-</sup>	1.37	1077.4828,945.5401,783.4882, 621.4349	Ginsenoside Rb <sub>3</sub>	Ginsenoside
95	33.37	C <sub>56</sub> H <sub>92</sub> O <sub>25</sub>	1163.5826 [M-H] <sup>-</sup>	-2.56	1119.5877,1077.5761,1059.5664	Malonylginsenoside Rc	Ginsenoside
96*	33.50	C <sub>47</sub> H <sub>74</sub> O <sub>18</sub>	925.4793 [M-H] <sup>-</sup>	-1.01	763.4258,569.3849	Pseudoginsenoside RT <sub>1</sub>	Ginsenoside
97	33.72	C <sub>56</sub> H <sub>92</sub> O <sub>25</sub>	1163.5808 [M-H] <sup>-</sup>	-4.03	1119.5819,1077.5723,1059.5612, 927.5204	Malonylginsenoside Rb <sub>2</sub> /Rc isomer	Ginsenoside
98	33.90	C <sub>56</sub> H <sub>94</sub> O <sub>24</sub>	1185.5829 [M + Cl] <sup>-</sup>	0.00	1149.6073,1107.5942,1089.5846	Quinquenoside R <sub>1</sub>	Ginsenoside
99	34.23	C <sub>21</sub> H <sub>22</sub> NO <sub>5</sub> <sup>+</sup>	368.1492 [M] <sup>+</sup>	-4.21	353.1250,338.1017,336.1226	Corynoline	Alkaloid
100	34.84	C <sub>56</sub> H <sub>92</sub> O <sub>25</sub>	1163.5859 [M-H] <sup>-</sup>	0.35	1119.5911,1077.5854,783.4934	Malonylginsenoside Rb <sub>2</sub> /Rc isomer	Ginsenoside
101*	34.94	C <sub>48</sub> H <sub>82</sub> O <sub>18</sub>	991.5454 [M + COOH] <sup>-</sup>	-3.35	945.5265,783.4762,621.4260,459.3774	Ginsenoside Rd	Ginsenoside
102	35.12	C <sub>55</sub> H <sub>92</sub> O <sub>23</sub>	1119.5956 [M-H] <sup>-</sup>	-0.06	1077.5746,1059.5686,937.1230	Ginsenoside RS <sub>2</sub>	Ginsenoside
103	35.17	C <sub>21</sub> H <sub>19</sub> NO <sub>6</sub>	382.1282 [M + H] <sup>+</sup>	-0.82	336.0869,308.0974,265.0691	Pontevedrine	Alkaloid

**Table 1** (continued)

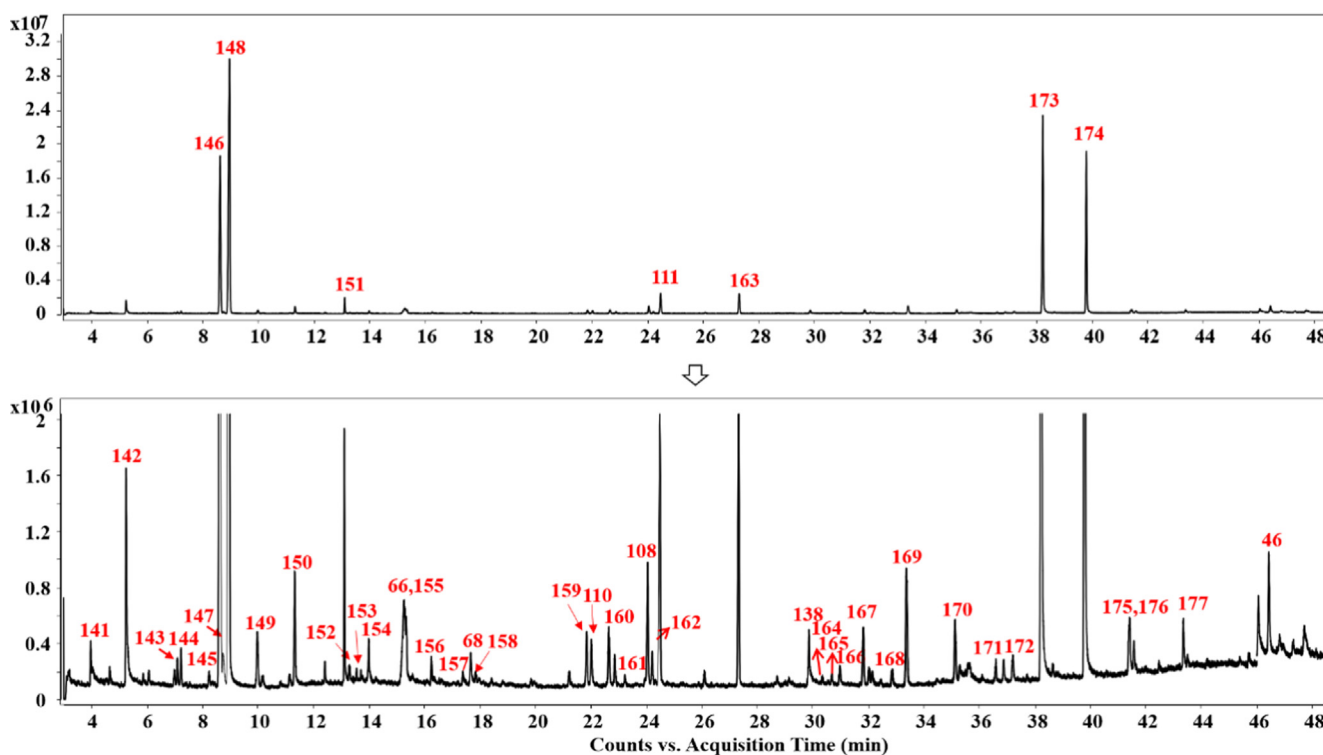
No.	t <sub>R</sub> (min)	Formula	Precursor ions ( <i>m/z</i> )	Diff (ppm)	Fragment ions ( <i>m/z</i> )	Identification	Structural Types
104	35.65	C <sub>51</sub> H <sub>84</sub> O <sub>21</sub>	1031.5382 [M–H] <sup>–</sup>	–4.88	987.5375,945.5286,927.5203, 783.4779,765.4668	Malonyl Ginsenoside Rd	Ginsenoside
105	35.86	C <sub>22</sub> H <sub>24</sub> NO <sub>4</sub> <sup>+</sup>	366.1687 [M] <sup>+</sup>	–3.51	350.1374,336.1240,322.1413,308.1290	Dehydrocorydaline isomer	Alkaloid
106*	37.22	C <sub>48</sub> H <sub>82</sub> O <sub>18</sub>	991.547 [M + COOH] <sup>–</sup>	–1.33	945.5397	Gypenoside XVII	Ginsenoside
107*	37.95	C <sub>19</sub> H <sub>13</sub> NO <sub>5</sub>	336.0869 [M + H] <sup>+</sup>	0.75	308.0913,293.0668,250.0864	8-Oxycoptisine	Alkaloid
108*	38.05	C <sub>12</sub> H <sub>16</sub> O <sub>2</sub>	193.1228 [M + H] <sup>+</sup>	2.56	105.0706,137.0609,147.1169	Senkyunolide A	Phthalide
109	39.48	C <sub>22</sub> H <sub>24</sub> NO <sub>4</sub> <sup>+</sup>	366.1695 [M] <sup>+</sup>	–1.32	350.1388,336.1237,322.1447,308.1251	Dehydrocorydaline isomer	Alkaloid
110*	39.70	C <sub>12</sub> H <sub>14</sub> O <sub>2</sub>	191.1066 [M + H] <sup>+</sup>	–0.29	135.0455,145.1003	Butylphthalide	Phthalide
111*	42.85	C <sub>12</sub> H <sub>14</sub> O <sub>2</sub>	191.1059 [M + H] <sup>+</sup>	–3.96	145.1010,173.0957,117.0695	( <i>E</i> )-Ligustilide	Phthalide
112	43.70	C <sub>36</sub> H <sub>60</sub> O <sub>8</sub>	665.4265 [M + COOH] <sup>–</sup>	–0.78	655.3976,569.2387,327.1338	Ginsenoside Rk <sub>3</sub> /Rh <sub>4</sub>	Ginsenoside
113*	45.28	C <sub>12</sub> H <sub>18</sub> O <sub>2</sub>	195.1382 [M + H] <sup>+</sup>	1.25	177.1269,149.1309,125.0595	Sedanolide	Phthalide
114*	45.71	C <sub>42</sub> H <sub>66</sub> O <sub>14</sub>	793.4335 [M–H] <sup>–</sup>	–5.65	613.3633,523.3703,455.3451	Zingibroside R <sub>1</sub>	Ginsenoside
115*	45.95	C <sub>12</sub> H <sub>14</sub> O <sub>2</sub>	191.1073 [M + H] <sup>+</sup>	–1.34	145.1008,173.0960,112.9674,117.0694	( <i>Z</i> )-Ligustilide	Phthalide
116*	47.90	C <sub>42</sub> H <sub>72</sub> O <sub>13</sub>	829.4918 [M + COOH] <sup>–</sup>	–4.45	783.4861,621.4345,113.0268, 459.3739,161.0423	20( <i>S</i> )-Ginsenoside Rg <sub>3</sub>	Ginsenoside
117	48.02	C <sub>42</sub> H <sub>66</sub> O <sub>14</sub>	793.4377 [M–H] <sup>–</sup>	–0.35	613.3632,569.3760,455.3473	Zingibroside R <sub>1</sub> isomer	Ginsenoside
118*	48.22	C <sub>42</sub> H <sub>72</sub> O <sub>13</sub>	819.4670 [M + Cl] <sup>–</sup>	0.37	783.4903,621.4341,459.3756	20( <i>R</i> )-Ginsenoside Rg <sub>3</sub>	Ginsenoside
119	50.05	C <sub>18</sub> H <sub>18</sub> O <sub>3</sub>	281.118 [M–H] <sup>–</sup>	1.71	163.0350,145.0262,117.0316	Isoeugenyl phenylacetate	Ester
120*	52.84	C <sub>21</sub> H <sub>19</sub> NO <sub>4</sub>	350.1385 [M + H] <sup>+</sup>	–0.53	335.1146,319.1191,334.1083	Dihydrochelerythrine	Alkaloid
121	52.94	C <sub>30</sub> H <sub>47</sub> O <sub>4</sub> <sup>–</sup>	471.3487 [M] <sup>–</sup>	1.52	393.3162,71.0506	2-Hydroxyoleanolate or isomer	Triterpene
122	53.96	C <sub>24</sub> H <sub>30</sub> O <sub>4</sub>	383.2227 [M + H] <sup>+</sup>	2.65	191.1063,149.0599	Senkyunolide P or isomer	Phthalide
123*	54.68	C <sub>20</sub> H <sub>15</sub> NO <sub>4</sub>	334.1097 [M + H] <sup>+</sup>	0.65	319.0841,304.0967,279.1013	Dihydrosanguinarine	Alkaloid
124	56.19	C <sub>24</sub> H <sub>30</sub> O <sub>4</sub>	383.2211 [M + H] <sup>+</sup>	–1.53	191.1063,149.0597	Senkyunolide P or isomer	Phthalide
125	56.57	C <sub>30</sub> H <sub>47</sub> O <sub>4</sub> <sup>–</sup>	471.3480 [M] <sup>–</sup>	0.03	359.2921,162.8327	2-Hydroxyoleanolate or isomer	Triterpene
126*	57.08	C <sub>24</sub> H <sub>28</sub> O <sub>4</sub>	381.2059 [M + H] <sup>+</sup>	–0.36	191.1075,173.0972,279.1508	Tokinolide B	Phthalide
127*	58.38	C <sub>24</sub> H <sub>28</sub> O <sub>4</sub>	381.2070 [M + H] <sup>+</sup>	2.53	191.107,267.1386,141.1136	Riligustilide	Phthalide
128*	58.55	C <sub>24</sub> H <sub>28</sub> O <sub>4</sub>	381.2069 [M + H] <sup>+</sup>	2.27	191.107,141.1136	Angelicide	Phthalide
129	60.63	C <sub>30</sub> H <sub>46</sub> O <sub>4</sub>	469.3316 [M–H] <sup>–</sup>	–1.56	305.1903,164.8363	Glycyrrhetic acid or isomer	Triterpene
130*	61.18	C <sub>30</sub> H <sub>48</sub> O <sub>3</sub>	455.3528 [M–H] <sup>–</sup>	–0.59	410.3532	Oleanolic acid	Triterpene
131	61.28	C <sub>30</sub> H <sub>46</sub> O <sub>4</sub>	469.3314 [M–H] <sup>–</sup>	–1.99	423.3197,211.1525	Glycyrrhetic acid or isomer	Triterpene
132*	63.34	C <sub>18</sub> H <sub>32</sub> O <sub>2</sub>	279.2327 [M–H] <sup>–</sup>	–0.91	261.2193,59.0146	Linoleic acid	Organic acid
133	63.59	C <sub>30</sub> H <sub>46</sub> O <sub>3</sub>	453.3376 [M–H] <sup>–</sup>	0.40	407.3308,325.2544,100.9336	Oleanonic acid or isomer	Triterpene
134	64.04	C <sub>30</sub> H <sub>46</sub> O <sub>3</sub>	453.3376 [M–H] <sup>–</sup>	0.40	407.3316,97.0653	Oleanonic acid or isomer	Triterpene
135	64.08	C <sub>30</sub> H <sub>46</sub> O <sub>3</sub>	453.3374 [M–H] <sup>–</sup>	–0.04	407.3316,97.0661	Oleanonic acid or isomer	Triterpene
136	64.31	C <sub>30</sub> H <sub>46</sub> O <sub>3</sub>	453.3380 [M–H] <sup>–</sup>	1.28	407.3244,97.0693	Oleanonic acid or isomer	Triterpene

(continued on next page)

**Table 1** (continued)

No.	t <sub>R</sub> (min)	Formula	Precursor ions ( <i>m/z</i> )	Diff (ppm)	Fragment ions ( <i>m/z</i> )	Identification	Structural Types
137	64.58	C <sub>30</sub> H <sub>46</sub> O <sub>4</sub>	469.3319 [M-H] <sup>-</sup>	-0.92	336.1460,141.8654	Glycyrrhetic acid or isomer	Triterpene
138*	64.76	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	255.2329 [M-H] <sup>-</sup>	-0.21	237.2247,116.9283	Palmitic acid	Organic acid
139*	65.26	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	281.2490 [M-H] <sup>-</sup>	1.41	116.9279	Oleic acid	Organic acid
140*	67.38	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	283.2622 [M-H] <sup>-</sup>	-7.25	265.2464,211.6753,141.7728	Octadecanoic acid	Organic acid

\* Compared with a reference standard.

**Fig. 3** Total ion current chromatograms of SXXTN by GC-MS.

isoboldine (**13**), showing a loss of 31 Da ( $-\text{OCH}_3$ ). Due to the loss of  $\text{NH}_2\text{CH}_3$  (Fig. S2D), a crucial characteristic ion at  $m/z$  325.1445 was obtained in glaucine (**46**) (Fig. S1D).

### 3.1.2. Identification of ginsenosides in SXXTN

A total of 34 ginsenosides in SXXTN were displayed in Table 1, mostly from Ginseng Radix et Rhizoma and demonstrated quasi-molecular ions  $[\text{M}-\text{H}]^-$  or  $[\text{M} + \text{COOH}]^-$  in negative ion mode due to formic acid in the mobile phase. Based on their aglycone, ginsenosides can be classified into three main categories: 20(*S*)-protopanaxadiol (PPD), 20(*S*)-protopanaxatriol (PPT) and oleanane type (OA) saponin.

19 PPD-type ginsenosides (**81**, **84**, **85**, **86**, **87**, **88**, **89**, **91**, **92**, **93**, **94**, **95**, **98**, **101**, **102**, **104**, **106**, **116**, **118**) were characterized and prone to produce  $[\text{20}(\text{S})\text{-protopanaxadiol-H}]^-$  ( $\text{C}_{30}\text{H}_{51}\text{O}_3$ ) characteristic aglycone fragment ions at  $m/z$

459.38 (Yang et al., 2021). For example, in MS/MS spectrometry (Fig. S1E), Compound **101** gave abundant ion at  $m/z$  783.4762 ( $[\text{M}-\text{H}-\text{Glc}]^-$ ),  $m/z$  621.4260 ( $[\text{M}-\text{H}-2\text{Glc}]^-$ ) and  $m/z$  459.3774 ( $[\text{20}(\text{S})\text{-protopanaxadiol-H}]^-$ ,  $\text{C}_{30}\text{H}_{51}\text{O}_3$ ), resulting from sequential eliminations of sugar residues (Fig. S2E). A total of 7 PPT-type ginsenosides (**63**, **67**, **73**, **74**, **80**, **82**, **83**) were tentatively identified and clearly marked with characteristic ions at  $m/z$  475.37 ( $[\text{20}(\text{S})\text{-protopanaxatriol-H}]^-$ ,  $\text{C}_{30}\text{H}_{51}\text{O}_4$ ) (Yang et al., 2021). As presented in Fig. S1F and Fig. S2F, ginsenoside Rg<sub>1</sub> (**73**) produced abundant characteristic ions including  $m/z$  673.4209 ( $[\text{M}-\text{H}-\text{Glc}]^-$ ) and  $m/z$  475.3713 ( $[\text{20}(\text{S})\text{-protopanaxatriol-H}]^-$ ). Three OA-type ginsenosides (**90**, **96**, **114**) were unambiguously elucidated as ginsenoside Ro, pseudoginsenoside RT<sub>1</sub> and zingibroside R<sub>1</sub> by comparison with reference standards with characteristic ions at  $m/z$  455.35 ( $[\text{Oleanolicacid-}$

**Table 2** Characterization of chemical constituents of SXXTN by GC-MS.

No.	Rt (min)	Identification	Match	Formula	Structural Types
141	4.03	Glycerin	91.1	C <sub>3</sub> H <sub>8</sub> O <sub>3</sub>	Alcohol
142	5.26	Allylbenzene	82.7	C <sub>9</sub> H <sub>10</sub>	Hydrocarbon
143	7.06	Bicyclo [2,2,1] heptan-2-ol,1,5,5-trimethyl	85.2	C <sub>10</sub> H <sub>18</sub> O	Monoterpenoid
144*	7.19	Fenchol	93.6	C <sub>10</sub> H <sub>18</sub> O	Monoterpenoid
145*	8.19	Camphor	90.3	C <sub>10</sub> H <sub>16</sub> O	Monoterpenoid
146*	8.58	Isoborneol	98.2	C <sub>10</sub> H <sub>18</sub> O	Monoterpenoid
147*	8.71	Phenol, 4-ethyl-	90.7	C <sub>8</sub> H <sub>10</sub> O	Phenol
148*	8.92	Borneol	97.4	C <sub>10</sub> H <sub>18</sub> O	Monoterpenoid
149	9.95	Bicyclo [2.2.1] heptan-2-ol, 1,7,7-trimethyl-, (1 <i>S</i> -endo)-	88.3	C <sub>10</sub> H <sub>18</sub> O	Monoterpenoid
150*	11.31	3-Phenylpropanol	96.5	C <sub>9</sub> H <sub>12</sub> O	Alcohol
151*	13.11	Cinnamyl alcohol	98.5	C <sub>9</sub> H <sub>10</sub> O	Alcohol
152*	13.28	2-Methoxy-4-vinylphenol	91.4	C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	Phenol
153*	13.57	Hydrocinnamic acid	96.1	C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	Organic acid
154	13.96	1,4-Cyclohexadiene-1,2-dicarboxylic anhydride	86.5	C <sub>8</sub> H <sub>6</sub> O <sub>3</sub>	Anhydride
66*	15.27	Cinnamic acid	72.1	C <sub>9</sub> H <sub>8</sub> O <sub>2</sub>	Organic acid
155*	15.34	Caryophyllene	88.7	C <sub>15</sub> H <sub>24</sub>	Sesquiterpene
156*	16.25	Ethyl cinnamate	77.7	C <sub>11</sub> H <sub>12</sub> O <sub>2</sub>	Organic acid ester
157*	17.37	2,4-Di- <i>tert</i> -butylphenol	84.6	C <sub>14</sub> H <sub>22</sub> O	Phenol
68*	17.66	Ethylparaben	95.1	C <sub>9</sub> H <sub>10</sub> O <sub>3</sub>	Organic acid ester
158	17.84	$\Delta$ -Cadinene	80.7	C <sub>15</sub> H <sub>24</sub>	Sesquiterpene
159*	21.84	Cadinol	91.7	C <sub>15</sub> H <sub>26</sub> O	Sesquiterpene
110*	22.02	Butylphthalide	93.1	C <sub>12</sub> H <sub>14</sub> O <sub>2</sub>	Phthalide
160*	22.66	<i>Z</i> -Butylidenephthalide	92.4	C <sub>12</sub> H <sub>12</sub> O <sub>2</sub>	Phthalide
161	23.23	5-Pentylcyclohexa-1,3-diene	87.4	C <sub>11</sub> H <sub>18</sub>	Hydrocarbon
108*	24.06	Senkyunolide A	89.2	C <sub>12</sub> H <sub>16</sub> O <sub>2</sub>	Phthalide
162*	24.22	Neocnidilide	92.0	C <sub>12</sub> H <sub>18</sub> O <sub>2</sub>	Phthalide
111*	24.46	( <i>E</i> )-Ligustilide	94.9	C <sub>12</sub> H <sub>14</sub> O <sub>2</sub>	Phthalide
163*	27.32	Muscone	95.7	C <sub>16</sub> H <sub>30</sub> O	Cyclic ketone
138	29.92	Palmitic acid	82.1	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	Organic acid
164	30.36	3-Phenylpropyl benzoate	92.6	C <sub>16</sub> H <sub>22</sub> O <sub>4</sub>	Organic acid ester
165	30.69	Hexadecanoic acid, ethyl ester	88.4	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Organic acid ester
166	31.00	Senkyunolide H	84.1	C <sub>12</sub> H <sub>16</sub> O <sub>4</sub>	Phthalide
167	31.83	3-benzyl-1,2-dihydronaphthalene	70.3	C <sub>17</sub> H <sub>17</sub>	Hydrocarbon
168	32.86	( <i>Z</i> )-Cinnamyl benzoate	96.0	C <sub>16</sub> H <sub>14</sub> O <sub>2</sub>	Organic acid ester
169*	33.39	Benzyl cinnamate	95.7	C <sub>16</sub> H <sub>14</sub> O <sub>2</sub>	Organic acid ester
170	35.15	Benzenepropanoic acid, 3-phenylpropyl ester	94.3	C <sub>18</sub> H <sub>20</sub> O <sub>2</sub>	Organic acid ester
171	36.62	Borny cinnamate	91.5	C <sub>19</sub> H <sub>24</sub> O <sub>2</sub>	Organic acid ester
172	36.89	Benzenepropanoic acid, 3-phenyl-2-propenyl ester	94.8	C <sub>18</sub> H <sub>18</sub> O <sub>2</sub>	Organic acid ester
173*	38.26	3-Phenylpropyl cinnamate, ( <i>E</i> )-	97.4	C <sub>18</sub> H <sub>18</sub> O <sub>2</sub>	Organic acid ester
174*	39.86	Cinnamyl cinnamate	97.4	C <sub>18</sub> H <sub>16</sub> O <sub>2</sub>	Organic acid ester
175*	41.40	Prasterone	76.3	C <sub>19</sub> H <sub>28</sub> O <sub>2</sub>	Steroid
176*	41.43	Androsterone	73.3	C <sub>19</sub> H <sub>30</sub> O <sub>2</sub>	Steroid
177	41.58	Phthalic acid, di(2-propylpentyl) ester	87.0	C <sub>24</sub> H <sub>38</sub> O <sub>4</sub>	Organic acid ester
46	46.06	Glucine	89.7	C <sub>21</sub> H <sub>25</sub> NO <sub>4</sub>	Alkaloid

\* Compared with a reference standard.

H], C<sub>30</sub>H<sub>47</sub>O<sub>3</sub>) (Yang et al., 2021) as shown in MS/MS spectrum (Fig. S1G) of ginsenoside Ro (90). The fragmentation pathways were shown in Fig. S2G.

Aglycones can be identified by discovering diagnostic ions and neutral loss can be observed to determine the number and type of glycosidic bond cleavage of ginsenosides. As shown in Fig. S1H, ginsenoside Rb<sub>2</sub> (93) appeared *m/z* 945.5378 ([M-H-Ara]<sup>-</sup>) and *m/z* 783.4803 ([M-H-Ara-Glc]<sup>-</sup>) in MS/MS spectrum.

### 3.1.3. Identification of organic acids in SXXTN

In total, 21 organic acids (2, 3, 4, 7, 8, 10, 11, 12, 14, 15, 19, 26, 40, 41, 54, 57, 66, 132, 138, 139, 140) were identified from SXXTN and shown in Table 1. Organic acids were easy to gen-

erate fragment ions in MS/MS with losing CO, CO<sub>2</sub>, -COOH, H<sub>2</sub>O, etc. (Yan, Wang, 2014). For example, fragment ions *m/z* 178.0273 [M-H-CH<sub>3</sub>]<sup>-</sup>, *m/z* 149.0561 [M-H-CO<sub>2</sub>]<sup>-</sup>, *m/z* 134.0371 [M-H-CH<sub>3</sub>-CO<sub>2</sub>]<sup>-</sup> and *m/z* 160.8423 [M-H-CH<sub>3</sub>-H<sub>2</sub>O]<sup>-</sup> were observed in secondary mass spectrometry of ferulic acid (26) (Fig. S1I) following specific cleavage pathways (Fig. S2H). In the MS/MS spectrometry of cinnamic acid (66) (Fig. S1J), the fragment ions with the highest abundance were observed to be *m/z* 103.0548 [M-H-CO<sub>2</sub>]<sup>-</sup>, which conformed to the cleavage characteristics of organic acids.

### 3.1.4. Identification of phthalides in SXXTN

Totally, 12 phthalides were tentatively identified or unambiguously authenticated, including 9 monomeric phthalides (45, 56,

**Table 3** Contents of 57 analytes determined in SXXTN samples by HPLC-QQQ MS and GC-QQQ MS ( $\mu\text{g/g}$ , Mean  $\pm$  SD, n = 3).

No.	Components	B1	B2	B3	B4	B5	B6	B7
<i>non-volatile components</i>								
5	Phenylalanine	86.83 $\pm$ 2.32	86.5 $\pm$ 3.25	64.25 $\pm$ 6.71	61.17 $\pm$ 5.4	29.33 $\pm$ 2.88	20.92 $\pm$ 1.01	85.92 $\pm$ 7.67
6	Tetramethylpyrazine	1.36 $\pm$ 0.05	1.47 $\pm$ 0.13	2.28 $\pm$ 0.05	1.5 $\pm$ 0.08	2.64 $\pm$ 0.05	3.14 $\pm$ 0.05	3.19 $\pm$ 0.05
21	Scoulerine	65.31 $\pm$ 6.33	71.92 $\pm$ 2.5	82.53 $\pm$ 2.59	74.67 $\pm$ 1.67	103.25 $\pm$ 5.29	107.08 $\pm$ 6.98	107.58 $\pm$ 2.62
23	Isocorydine	17.97 $\pm$ 0.65	19.44 $\pm$ 0.46	20.53 $\pm$ 0.67	19.72 $\pm$ 0.61	28.89 $\pm$ 0.68	28.97 $\pm$ 0.47	29.72 $\pm$ 0.57
28	Corydalmine	47.47 $\pm$ 1.46	50.17 $\pm$ 1.52	56.75 $\pm$ 1.18	51.81 $\pm$ 0.77	51.25 $\pm$ 4.1	51.67 $\pm$ 2.82	51.53 $\pm$ 1.5
29	Tetrahydrocolumbamine	631.94 $\pm$ 12.95	621.94 $\pm$ 38.75	710.56 $\pm$ 17	602.22 $\pm$ 9.18	988.89 $\pm$ 26.79	1043.61 $\pm$ 2.55	1040.56 $\pm$ 44.92
33	Corypalmine	68.64 $\pm$ 0.6	73.22 $\pm$ 2.53	78.36 $\pm$ 1.64	75.42 $\pm$ 2.3	127.64 $\pm$ 1.77	137.22 $\pm$ 5.23	133.61 $\pm$ 3.72
36	Protopine	849.11 $\pm$ 12.48	869.92 $\pm$ 34.43	935.56 $\pm$ 35.06	860.64 $\pm$ 15.65	1062.14 $\pm$ 19.22	1125.22 $\pm$ 42.62	1125.03 $\pm$ 27.57
37	Demethyleneberberine	8.14 $\pm$ 0.05	8.5 $\pm$ 0.36	9.22 $\pm$ 0.21	8.58 $\pm$ 0.17	7.81 $\pm$ 0.13	8.25 $\pm$ 0.17	8.11 $\pm$ 0.34
43	Norglaucine	10063.33 $\pm$ 154.52	9121.39 $\pm$ 263.53	9767.78 $\pm$ 337.78	8266.94 $\pm$ 90.49	7842.22 $\pm$ 509.16	7953.89 $\pm$ 104.36	8121.94 $\pm$ 194.84
44	Allocryptopine	544.53 $\pm$ 27.26	534.5 $\pm$ 14.24	575.03 $\pm$ 20.11	530.03 $\pm$ 13.11	631.28 $\pm$ 27.8	672.31 $\pm$ 28.3	684.06 $\pm$ 22.77
46	Glaucine	910 $\pm$ 5.46	955.56 $\pm$ 31.9	1068.33 $\pm$ 48.18	929.44 $\pm$ 9.66	1009.17 $\pm$ 88.93	1036.11 $\pm$ 14.82	1061.11 $\pm$ 29.27
48	Tetrahydropalmatine	432.22 $\pm$ 14.2	422.78 $\pm$ 18.95	486.94 $\pm$ 22.12	418.89 $\pm$ 8.67	646.39 $\pm$ 57.4	674.17 $\pm$ 11.67	683.61 $\pm$ 12.48
52	Coptisin	513.14 $\pm$ 18.5	533.58 $\pm$ 19.6	631.47 $\pm$ 17.92	528.25 $\pm$ 20.18	683.81 $\pm$ 46.08	770.03 $\pm$ 47.76	768.31 $\pm$ 12.25
53	Columbamine	414.25 $\pm$ 8.33	426.61 $\pm$ 12.44	466.33 $\pm$ 14.99	422.61 $\pm$ 18.19	385.47 $\pm$ 25.7	417.39 $\pm$ 23.39	418.83 $\pm$ 8.24
55	Jatrorrhizine	24.75 $\pm$ 2.32	24.64 $\pm$ 3.56	28.42 $\pm$ 1.98	26.58 $\pm$ 1.69	28.64 $\pm$ 2.79	31.25 $\pm$ 2.35	32.92 $\pm$ 1.08
58	Canadine	103.61 $\pm$ 3.37	103.33 $\pm$ 3.63	120.28 $\pm$ 2.41	104.72 $\pm$ 0.48	171.39 $\pm$ 12.14	183.89 $\pm$ 2.55	183.89 $\pm$ 2.55
60	Corydaline	445.28 $\pm$ 11.71	455.83 $\pm$ 15.83	537.22 $\pm$ 17.02	443.89 $\pm$ 5.55	839.17 $\pm$ 50.26	885 $\pm$ 23.11	897.78 $\pm$ 7.74
66	Cinnamic acid	482.5 $\pm$ 15.52	529.5 $\pm$ 20.11	505 $\pm$ 31.83	515.33 $\pm$ 16.33	477.33 $\pm$ 14.1	515.25 $\pm$ 17.37	308.08 $\pm$ 18.32
67	Notoginsenoside R <sub>1</sub>	50.25 $\pm$ 0.25	53.58 $\pm$ 1.28	50.83 $\pm$ 4.94	49.25 $\pm$ 2.7	24.75 $\pm$ 2.14	26.25 $\pm$ 1.39	23.08 $\pm$ 1.01
70	Berberine	171.39 $\pm$ 4.86	176.39 $\pm$ 2.51	190.11 $\pm$ 4.03	170.92 $\pm$ 0.75	197.5 $\pm$ 4.01	203.06 $\pm$ 4.76	201.83 $\pm$ 2.35
71	Palmatine	588.44 $\pm$ 12.35	603.11 $\pm$ 6.88	638.22 $\pm$ 13.26	581.14 $\pm$ 4.43	540.64 $\pm$ 9.63	560.03 $\pm$ 13.97	555.5 $\pm$ 2.28
73	Ginsenoside Rg <sub>1</sub>	803.67 $\pm$ 14.68	831.42 $\pm$ 17.47	777.58 $\pm$ 34.06	731.5 $\pm$ 37.15	929.33 $\pm$ 27.19	915.58 $\pm$ 18.06	867.67 $\pm$ 54.56
74	Ginsenoside Re	565.17 $\pm$ 19.12	579.67 $\pm$ 36.71	547.42 $\pm$ 43.84	521.58 $\pm$ 32.77	559.42 $\pm$ 23.51	562.67 $\pm$ 11.2	826.83 $\pm$ 81.78
75	Dehydrocorydaline	1666.11 $\pm$ 18.15	1677.22 $\pm$ 49.14	1840.28 $\pm$ 40.79	1677.22 $\pm$ 25.69	1548.06 $\pm$ 73.42	1607.5 $\pm$ 28.83	1554.44 $\pm$ 17.8
76	13-Methylberberine	11.92 $\pm$ 0.17	11.92 $\pm$ 0.17	12.89 $\pm$ 0.24	11.81 $\pm$ 0.05	8.58 $\pm$ 0.14	8.86 $\pm$ 0.1	9 $\pm$ 0.08
80	Ginsenoside Rf	102 $\pm$ 3.36	111 $\pm$ 7.15	102.83 $\pm$ 10.26	90.5 $\pm$ 6.43	135.25 $\pm$ 3.5	135.83 $\pm$ 7.52	108.08 $\pm$ 6.79
83	Ginsenoside Rg <sub>2</sub>	92.83 $\pm$ 2.74	98.42 $\pm$ 5.84	96.58 $\pm$ 7.38	90 $\pm$ 2.61	77.25 $\pm$ 1.32	79 $\pm$ 2.84	60.92 $\pm$ 3.17
84	Ginsenoside Ra <sub>2</sub>	71.83 $\pm$ 1.91	74.25 $\pm$ 2.82	76.33 $\pm$ 3.4	72 $\pm$ 2.14	83.67 $\pm$ 0.88	84.67 $\pm$ 1.91	80.17 $\pm$ 1.89
85	Ginsenoside Ra <sub>3</sub>	211 $\pm$ 0.43	199.25 $\pm$ 16.69	190.67 $\pm$ 3.39	188.33 $\pm$ 9.7	203.92 $\pm$ 3.69	214.67 $\pm$ 6.57	290.83 $\pm$ 27.19
86	Ginsenoside Rb <sub>1</sub>	1025.25 $\pm$ 6.51	1045.08 $\pm$ 54.55	1030.92 $\pm$ 80.79	960.92 $\pm$ 29.16	708.08 $\pm$ 22.64	716.33 $\pm$ 15.95	571.33 $\pm$ 22.92
88	Ginsenoside Rc	281.42 $\pm$ 4.23	294.17 $\pm$ 8.08	288.5 $\pm$ 13.56	267.58 $\pm$ 8.38	227.75 $\pm$ 0.9	226.5 $\pm$ 0.66	221.58 $\pm$ 10.04
89	Ginsenoside Ra <sub>1</sub>	155.58 $\pm$ 3.83	167.67 $\pm$ 9.87	169.25 $\pm$ 10.4	148.33 $\pm$ 3.22	184.5 $\pm$ 9.79	187.33 $\pm$ 7.04	180.58 $\pm$ 5.11
93	Ginsenoside Rb <sub>2</sub>	379.33 $\pm$ 7.69	398.17 $\pm$ 20.89	394.67 $\pm$ 31.29	357.75 $\pm$ 12.89	282.5 $\pm$ 12.67	281.42 $\pm$ 5.58	225.75 $\pm$ 5.91
94	Ginsenoside Rb <sub>3</sub>	37.75 $\pm$ 1	39.17 $\pm$ 2.01	41.75 $\pm$ 3.12	34.33 $\pm$ 2.7	24.5 $\pm$ 2.41	24.42 $\pm$ 0.95	227.75 $\pm$ 6.71
101	Ginsenoside Rd	248.33 $\pm$ 10.47	261.92 $\pm$ 8.61	261.25 $\pm$ 20.12	236.75 $\pm$ 9.85	178.5 $\pm$ 7.7	180.17 $\pm$ 4.69	142.25 $\pm$ 6.29
107	8-Oxycoptisine	27.08 $\pm$ 3.18	25.36 $\pm$ 0.67	27.75 $\pm$ 0.38	25.06 $\pm$ 0.38	27.81 $\pm$ 0.42	28.86 $\pm$ 0.27	28 $\pm$ 0.38
116	20(S)-Ginsenoside Rg <sub>3</sub>	57.67 $\pm$ 2.36	65.42 $\pm$ 3.15	64.58 $\pm$ 1.42	60.08 $\pm$ 2.47	15.83 $\pm$ 0.14	15.33 $\pm$ 0.38	14.5 $\pm$ 0.5
120	Dihydrochelerythrine	36.69 $\pm$ 4.46	34.36 $\pm$ 0.87	36.33 $\pm$ 0.22	35.44 $\pm$ 0.59	49.53 $\pm$ 0.63	49.08 $\pm$ 0.33	49.64 $\pm$ 0.79
123	Dihydrosanguinarine	31.5 $\pm$ 2.58	32 $\pm$ 0.58	34.17 $\pm$ 0.46	32.89 $\pm$ 0.42	45.86 $\pm$ 0.21	46.67 $\pm$ 0.38	45.5 $\pm$ 0.87

Table 3 (continued)

No.	Components	B1	B2	B3	B4	B5	B6	B7
<i>volatile components</i>								
144	Fenchol	31.79 ± 0.72	24.14 ± 0.74	31.36 ± 0.23	28.89 ± 1.00	132.64 ± 1.08	233.88 ± 1.23	212.26 ± 0.61
145	Camphor	122.36 ± 0.58	99.94 ± 1.17	130.70 ± 0.35	115.32 ± 0.86	40.82 ± 0.14	78.08 ± 1.08	67.93 ± 0.08
146	Isoborneol	10746.58 ± 56.15	8297.41 ± 30.66	10784.14 ± 10.58	11041.49 ± 41.12	6992.38 ± 22.46	10192.18 ± 55.59	10537.26 ± 85.96
147	4-Ethylphenol	111.73 ± 2.12	101.78 ± 2.15	110.57 ± 0.64	107.77 ± 1.56	92.31 ± 5.60	122.53 ± 3.56	120.49 ± 0.66
148	Borneol	16154.89 ± 17.28	12716.32 ± 71.79	16158.12 ± 26.98	16557.61 ± 73.28	11800.29 ± 15.21	15303.33 ± 71.80	16110.40 ± 39.41
150	3-Phenylpropanol	287.24 ± 5.17	269.94 ± 5.95	295.45 ± 2.43	282.39 ± 4.51	323.49 ± 4.19	342.09 ± 10.25	332.53 ± 1.81
151	Cinnamyl alcohol	677.68 ± 11.97	659.66 ± 7.87	687.77 ± 9.42	676.31 ± 1.38	775.47 ± 15.66	782.07 ± 17.59	775.85 ± 9.96
155	Caryophyllene	99.41 ± 0.63	88.88 ± 0.23	101.31 ± 0.57	96.25 ± 1.05	68.95 ± 0.09	108.27 ± 0.63	102.55 ± 0.62
110	Butylphthalide	120.17 ± 0.72	119.28 ± 0.8	120.54 ± 0.52	118.96 ± 0.03	175.46 ± 0.95	166.65 ± 0.62	166.47 ± 0.78
160	Z-Butylidene-phthalide	470.46 ± 8.24	461.27 ± 6.60	469.83 ± 4.71	462.35 ± 5.09	918.91 ± 4.81	838.41 ± 5.49	834.35 ± 3.20
108	Senkyunolide A	445.75 ± 1.01	445.75 ± 5.70	453.88 ± 4.66	436.02 ± 2.77	1001.42 ± 2.58	963.05 ± 5.76	953.08 ± 6.39
162	Neocnidilide	114.94 ± 0.39	111.39 ± 1.46	112.35 ± 0.51	111.00 ± 0.5	209.89 ± 0.45	195.90 ± 0.17	196.40 ± 1.01
111	(E)-Ligustilide	708.93 ± 4.13	683.43 ± 5.57	698.54 ± 2.74	689.34 ± 2.44	1470.36 ± 15.50	1400.27 ± 10.26	1388.91 ± 5.15
163	Muscone	1185.77 ± 14.67	1092.55 ± 4.44	1174.19 ± 3.1	1145.85 ± 16.89	1258.91 ± 21.03	1206.96 ± 16.67	1173.77 ± 4.76
169	Benzyl cinnamate	663.55 ± 3.06	634.83 ± 5.35	664.60 ± 2.63	652.38 ± 1.63	694.71 ± 2.69	689.84 ± 4.29	665.64 ± 1.52
173	3-Phenylpropyl Cinnamate	3532.42 ± 24.74	3194.93 ± 115.13	3503.38 ± 37.00	3493.68 ± 142.94	3855.59 ± 47.58	3730.66 ± 119.27	3828.99 ± 29.46
174	Cinnamyl cinnamate	4781.76 ± 48.54	4442.54 ± 117.11	4790.28 ± 38.97	4753.69 ± 164.10	5193.83 ± 65.96	4995.83 ± 171.5	5131.81 ± 42.73

108, 110, 111, 113, 122, 124) and 3 phthalide dimers (126, 127, 128). Monomeric phthalide compounds with a phthalide structure unit as the core, are prone to neutral loss of H<sub>2</sub>O, CO, CO<sub>2</sub> and alkyl radicals or alkyl chains (CH<sub>3</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>3</sub>H<sub>6</sub>, C<sub>4</sub>H<sub>8</sub>, etc.) (Yan et al, 2022) as shown in Fig. S2I. In the MS/MS spectrometry of senkyunolide A (1 0 8), the product ion *m/z* 175.1123 was produced by the precursor ion loss of H<sub>2</sub>O. On this basis, the characteristic fragment *m/z* 147.1167 was produced by the successive loss of CO and *m/z* 137.0595 was the production of alkyl radical C<sub>4</sub>H<sub>8</sub> lost by precursor ions (Fig. S1K). The phthalide dimer compounds are formed by the polymerization of two phthalide monomers. They are induced to dissociate into monomeric phthalide in MS/MS, and the highest intensity ions at *m/z* 191.11 are often produced (Zhang et al., 2018). Take Angelicide (128) as example, the fragment ion with highest abundance was observed at *m/z* 191.1066 in MS/MS spectrometry (Fig. S1L). On this basis, the cleavages of phthalide skeletons could also be observed in the MS/MS spectra of phthalide dimers, which were similar to monomeric phthalides.

### 3.2. GC-MS qualitative analysis of SXXTN

In preceding reports, the volatile components in SXXTN such as Artificial Musk, Chuanxiong Rhizoma, Styraax and Borneolum Syntheticum have been revealed (Ding et al., 2022; He et al., 2018; Gurbuz et al., 2013; Sun et al., 2014), whereas little attention was paid to the volatile components in the intact SXXTN prescription. In this work, we supplemented the information of volatile chemicals and improved the global characterizations of complicated ingredients in SXXTN.

The GC-MS conditions of the temperature program, splitting ratio (10:1, 30:1 and 50:1) and the injector temperature (250°C, 280°C and 300°C) were optimized in the direction of analyzing comprehensive volatile constituents of SXXTN with well separation performance in a short analysis. The total peak area was calculated as a criterion for optimization. The final conditions were described in Section 2.4.

The TICs of SXXTN by GC-MS can be viewed in Fig. 3. There were 44 volatile compounds tentatively identified from SXXTN, based on the mass spectrometric data of reference standards, the mass spectral library (NIST17) and the literature, including 12 organic acid esters, 6 monoterpenes, 6 phthalides, 3 organic acids, 3 sesquiterpenes, 3 alcohols, 3 alkanes, 3 hydrocarbons, 2 steroids, 1 macrocyclic ketone, 1 alkaloid and 1 anhydride (Table 2). The chemical structures were illustrated in Fig. 2. Among them, seven compounds (66, 68, 110, 108, 111, 138, 46) have been identified in the previous LC-MS analysis. According to the comparison with authentic standards, 25 components were clearly marked. Isoborneol (146), borneol (148), cinnamyl alcohol (151), (E)-ligustilide (111), muscone (163), 3-phenylpropyl cinnamate (173) and cinnamyl cinnamate (174) exhibited relatively high abundances during GC-MS analysis of SXXTN.

### 3.3. Quantification of 40 non-volatile compounds in SXXTN by HPLC-QQQ MS

Forty confirmed non-volatile chemicals were further quantified by the optimized HPLC-QQQ MS method (Table S3-S4) to evaluate the quality of SXXTN. According to the difference

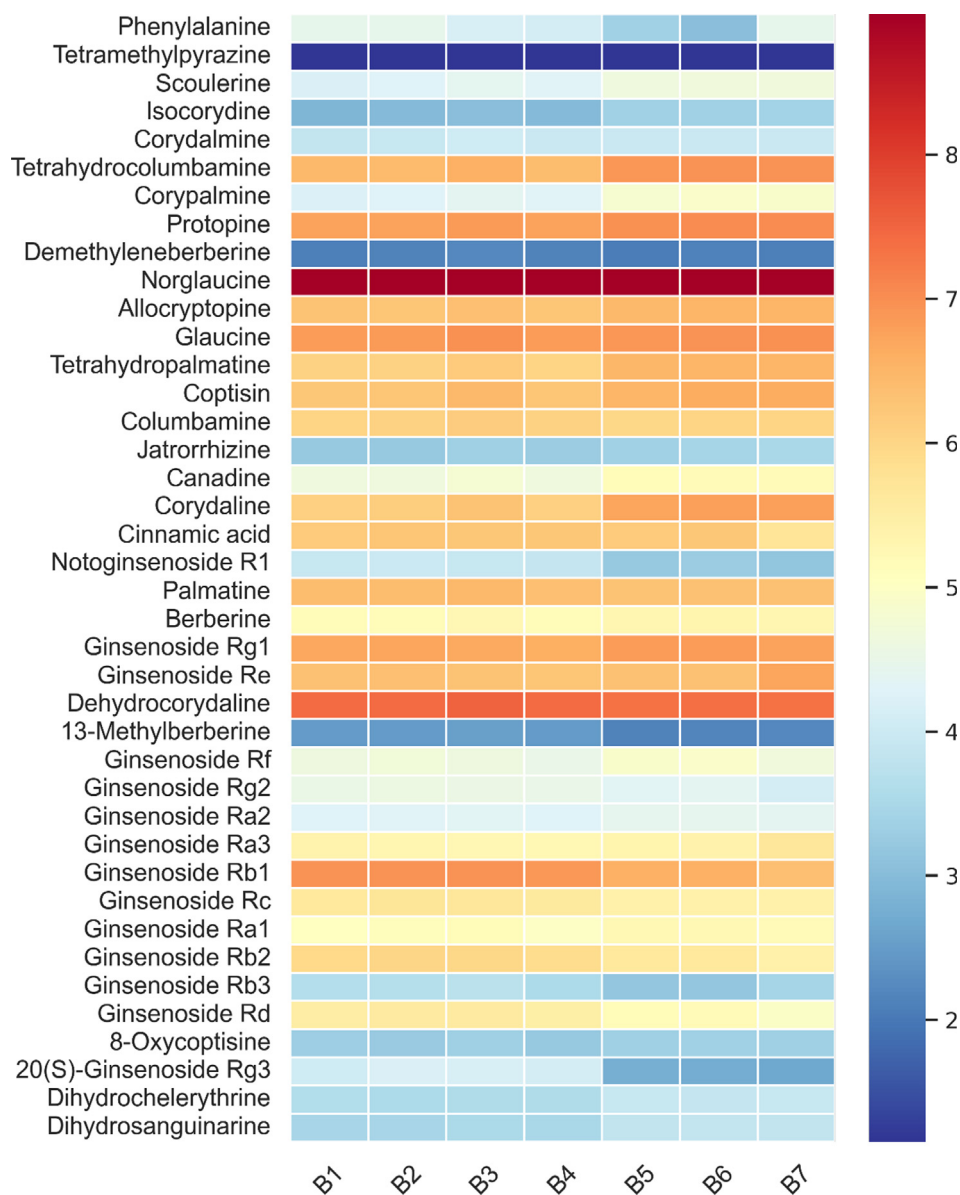


Fig. 4 The content distribution heatmap of 40 non-volatile compounds in 7 batches of SXXTN.

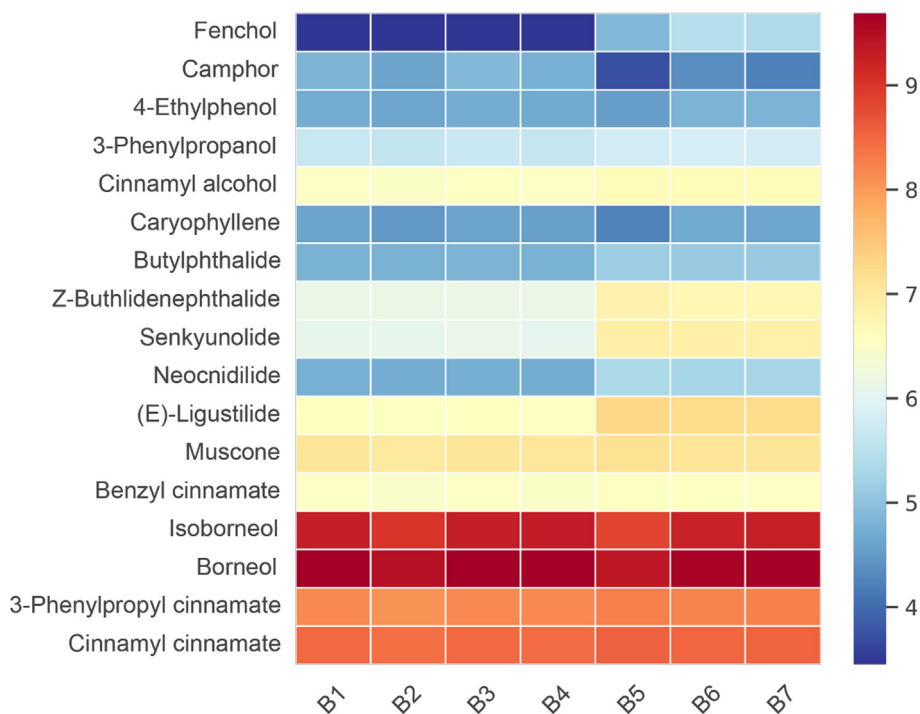
in the response of the components in positive and negative ion modes, two MRM methods with different polarity were established for Quantification. The typical MRM chromatograms of analytes were illustrated in Fig S3.

Nice linearity with coefficients of determination ( $R^2 > 0.9900$ ) were obtained for the 40 analytes. Limit of detection (LOD) and limit of quantitation (LOQ) tests were performed and listed in Table S7. As exhibited in Table S8, relative standard deviations (RSD) of repeatability, intra- and inter-day precision were 1.27 % – 4.79 %, 1.07 % – 5.41 % and 1.18 % – 9.43 %, respectively. Besides, all analytes could remain stable within 24 h under 4°C, with the RSD ranging from 0.60 % – 6.14 % and recoveries of 40 compounds were ranged from 80.36 % – 117.13 % with the RSD ranging 2.38 % – 12.61 %. Consequently, the established HPLC-QQQ MS approach was proved as a sensitive, repeatable and accurate

tool for the quantification of non-volatile compounds in SXXTN.

According to the established HPLC-QQQ MS quantitative analysis method, the content of 40 compounds in 7 batches of SXXTN provided by the enterprise was determined as shown in Table 3 and Fig. 4. The total content of 40 analytes in each batch was 2.03 % – 2.30 %. Among them, the components with higher content ( $> 1.00$  mg/g) were norglaucine (43), glaucine (46), dehydrocorydaline (75) and ginsenoside Rb<sub>1</sub> (86). In the previous reported, aforementioned compounds presented promising effects for myocardial protection (Wen et al., 2022; Han et al., 2012; Zheng et al., 2017; Kong et al., 2018). For example, studies have found that intraperitoneal injection of dehydrocorydine in ApoE<sup>-/-</sup> mice can not only inhibit the development of atherosclerosis, but also improve aortic compliance and plaque stability (Wen et al., 2022).





**Fig. 5** The content distribution heatmap of 17 volatile compounds in 7 batches of SXXTN.

Tetrahydropalmatine can activate PI3K/Akt/eNOS/NO pathway, increase the expression of HIF-1 $\alpha$  and VEGF, and inhibit iNOS-derived NO production in myocardium. This effect may reduce the accumulation of inflammatory factors (including TNF- $\alpha$  and MPO) and reduce the degree of apoptosis (Han et al., 2012). It has also been reported that ginsenoside Rb<sub>1</sub> can improve heart failure, which may be achieved by regulating the mitochondrial membrane in cardiomyocytes (Kong et al., 2018).

### 3.4. Quantification of 17 volatile compounds in SXXTN by GC-QQQ MS

We established rapid and accurate quantitative methods for detecting the contents of the major volatile compounds in SXXTN. Seventeen confirmed compounds including 5 phthalides, 3 organic acid esters, 3 alcohols, 3 monoterpenes, 1 sesquiterpene, 1 phenol and 1 macrocyclic ketone were determined by GC-QQQ MS with naphthalene (IS3) as internal standards. The optimized conditions were shown in Table S5-S6 and typical MRM chromatograms of 17 analytes were illustrated in Fig S4.

The optimized GC-QQQ MS method was validated in the aspect of linearity, LODs, LOQs, precision, repeatability, stability and recovery and the results were presented in Table S9 and Table S10. Reasonable correlation coefficient values ( $R^2 > 0.9904$ ) indicated good correlations between investigated standards concentrations and their peak areas within the ranges tested. The ranges of LODs and LOQs for all the analytes were 0.002 - 2.642  $\mu\text{g/mL}$ , and 0.013 - 6.653  $\mu\text{g/mL}$ , respectively. The RSDs of repeatability, intra- and inter-day precision were 1.63% - 9.66%, 0.44% - 9.06%, 0.80% - 8.06%, respectively. All analytes could remain stable within 24 h under 4 $^{\circ}\text{C}$ , with the RSD ranging 1.14% - 7.97%. The

developed method had good accuracy with the recoveries were between 90.20% and 123.51%. These results provided that the established method was accurate, reproducible, and reliable for assessing the quality of volatile compounds in SXXTN.

According to the established GC-QQQ MS quantitative analysis method, the contents of main volatile components in 7 batches of SXXTN provided by the enterprise were determined as displayed in Table 3 and Fig. 5. The total content of analytes in each batch was 3.34 % - 4.26 % in SXXTN. Borneol (146), isoborneol (148), cinnamyl cinnamate (174), 3-phenylpropyl cinnamate (173) and muscone (163) were the predominant components and were closely related to the anti-coronary heart disease and angina pectoris effect of SXXTN (Liu et al., 2017; Wu et al., 2011; Wang et al., 2020).

Combined with the quantitative analysis of non-volatile components, the total contents of 57 main components in 7 batches of SXXTN were 5.50 % - 6.49 %. The percentages of different structural types of chemicals in the total 57 analytes were as follows: monoterpenes and sesquiterpenes accounted for the largest proportion (42 %), followed by alkaloids (26 %), organic acids and esters (18 %), ginsenosides (6 %) and phthalide (5 %). Both volatile and non-volatile components should be taken into consideration for quality evaluation of SXXTN. More batches of samples are more conducive to assessing the consistency and stability of SXXTN.

## 4. Conclusion

In view of the current deficiencies in the constituent research and quality control of SXXTN, efficient, stable and reliable LC-MS and GC-MS methods were established in our study. A total of 177 chemical components were identified from SXXTN and content of 57 components in 7 batches of SXXTN was further determined. To the best of our knowledge, this is the initial report on the comprehensive profiling

of chemical constituents in SXXTN by LC-MS and GC-MS. The evaluation approach provided much more qualitative and quantitative information of multi-components in SXXTN than other single-marker quality assessments. In all, this study provided comprehensive material basis of SXXTN, which could be beneficial to improve the quality control. Furthermore, it could facilitate the pharmacological research and clinical application of SXXTN in some degree.

### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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### Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.arabjc.2022.104527>.

### References

- Ding, M., Fan, J.L., Huang, D.F., Jiang, Y., Li, M.N., Zheng, Y.Q., Yang, X.P., Li, P., Yang, H., 2022. From non-targeted to targeted GC-MS metabolomics strategy for identification of TCM preparations containing natural and artificial musk. *Chin. Med.* 17, 41.
- Gurbuz, I., Yesilada, E., Demirci, B., Sezik, E., Demirci, F., Baser, K. H., 2013. Characterization of volatiles and anti-ulcerogenic effect of Turkish sweetgum balsam (*Styrax liquidus*). *J. Ethnopharmacol.* 148, 332–336.
- Han, Y., Zhang, W., Tang, Y., Bai, W.L., Yang, F., Xie, L.P., Li, X. Z., Zhou, S.M., Pan, S.Y., Chen, Q., Ferro, A., Ji, Y., 2012. 1-Tetrahydropalmatine, an Active Component of *Corydalis yanhusuo* W.T. Wang, Protects against Myocardial Ischaemia-Reperfusion Injury in Rats. *PLoS. One.* 7, e38627.
- He, M., Yan, P., Yang, Z.Y., Zhang, Z.M., Yang, T.B., Hong, L., 2018. A modified multiscale peak alignment method combined with trilinear decomposition to study the volatile/heat-labile components in *Ligusticum chuanxiong* Hort - *Cyperus rotundus* rhizomes by HS-SPME-GC/MS. *J. Chromatogr. B. Analyt. Technol. Biomed. Life. Sci.* 1079, 41–50.
- Kong, H.L., Hou, A.J., Liu, N.N., Chen, B.H., Dai, S.N., Huang, H. T., 2018. The effects of ginsenoside Rb<sub>1</sub> on fatty acid  $\beta$ -oxidation, mediated by AMPK, in the failing heart. *Iran. J. Basic. Med. Sci.* 21, 731–737.
- Li, J., Cao, G.Y., Zhang, X.F., Meng, Z.Q., Gan, L., Li, J.X., Lan, X. Y., Yang, C.L., Zhang, C.F., 2020. Chinese Medicine She-Xiang-Xin-Tong-Ning, Containing *Moschus*, *Corydalis* and *Ginseng*, Protects from Myocardial Ischemia Injury via Angiogenesis. *Am. J. Chin. Med.* 48, 107–126.
- Li, Y., Zhang, J., Li, L., 2008. Comparison of the therapeutic effects of different compositions of muskone in the treatment of experimental myocardial infarct in rats and analgesia in mice. *Phytother. Res.* 22, 1219–1223.
- Ling, H., Wu, L., Li, L., 2010. *Corydalis yanhusuo* rhizoma extract reduces infarct size and improves heart function during myocardial ischemia/reperfusion by inhibiting apoptosis in rats. *Phytother. Res.* 20, 448–453.
- Liu, F., Huang, Z.Z., Sun, Y.H., Li, T., Yang, D.H., Xu, G., Su, Y.Y., Zhang, T., 2017. Four Main Active Ingredients Derived from a Traditional Chinese Medicine *Guanxin Shutong* Capsule Cause Cardioprotection during Myocardial Ischemia Injury Calcium Overload Suppression. *Phytother. Res.* 31 (3), 507–515.
- Liu, X., Li, X., Ji, S., Cui, X., Li, M., 2016. Screening of Bioactive Ingredients in *Ligusticum Chuanxiong* Hort for Protection against Myocardial Ischemia. *Cell. Physiol. Biochem.* 40, 770–780.
- Luo, X., Chen, X., Shen, X., Yang, Z., Du, G., 2019. Rapid identification and analysis of the active components of traditional Chinese medicine Xiaoxuming decoction for ischemic stroke treatment by integrating HPLC-Q-TOF/MS and RRLC-QTRAP MSn method. *J. Chromatogr. B. Analyt. Technol. Biomed. Life. Sci.* 1124, 313–322.
- Shen, X.X., Lu, W.X., 2005. Observation of the Blood Rheology and Clinical Effects of Shexiangxintongning tablet on 60 Cases with Coronary Heart Disease Combined with Angina Pectoris. *Chin. J. Nat. Med.* 7, 114–117.
- Su, S., Hua, Y., Duan, J.A., Shang, E., Tang, Y., Bao, X., Lu, Y., Ding, A.W., 2008. Hypothesis of active components in volatile oil from a Chinese herb formulation, Shao-Fu-Zhu-Yu decoction, using GC-MS and chemometrics. *J. Sep. Sci.* 31 (6–7), 1085–1091.
- Sun, X.M., Liao, Q.F., Zhou, Y.T., Deng, X.J., Xie, Z.Y., 2014. Simultaneous determination of borneol and its metabolite in rat plasma by GC-MS and its application to pharmacokinetic study. *J. Pharm. Anal.* 4 (5), 345–350.
- Sun, J.N., Sun, W.Y., Dong, S.F., 2017. Discussion on efficacy evaluation thought and method for innovation medicine of Chinese herbal compound formula based on clinical application characteristics. *Zhongguo Zhongyao Zazhi.* 42 (5), 852–855.
- Wang, Y., Gou, D., 2019. Relationship between dose and effects of Suhexiang against myocardial ischemia of model rats. *Acta Chin. Med.* 34, 2157–2163.
- Wang, Y.F., Li, Z.R., Liu, B.X., Wu, R.M., Gong, H.F., Su, Z.H., Zhang, S.D., 2020. Isoborneol Attenuates Low-Density Lipoprotein Accumulation and Foam Cell Formation in Macrophages. *Drug Des. Dev. Ther.* 14, 167–173.
- Wang, Q., Lu, Z.W., Liu, Y.H., Wang, M.L., Fu, S., Zhang, Q.Q., Zhao, H.Z., Zhang, Z.X., Xie, Z.Y., Huang, Z.H., Yu, H.H., Zhou, W.J., Gao, X.Y., 2017. Rapid analysis on phenolic compounds in *Rheum palmatum* based on HPLC-Q-TOF/MSE combined with diagnostic ions filter. *China J. Chin. Mater. Med.* 42 (10), 1922–1931.
- Wen, B., Dang, Y.Y., Wu, S.H., Huang, Y.M., Ma, K.Y., Xu, Y.M., Zheng, X.L., Dai, X.Y., 2022. Antiatherosclerotic effect of dehydrocorydaline on ApoE(-/-) mice: inhibition of macrophage inflammation. *Acta. Pharmacol. Sin.* 6, 1408–1418.
- Wu, Q.B., Li, H.T., Wu, Y., Shen, W.X., Zeng, L., Cheng, H.B., He, L., 2011. Protective effects of muscone on ischemia-reperfusion injury in cardiac myocytes. *J. Ethnopharmacol.* 138 (1), 34–39.
- Xu, W., Huang, M., Li, H., Chen, X., Zhang, Y., Liu, J., Xu, W., Chu, K., Chen, L.D., 2015. Chemical profiling and quantification of Gua-Lou-Gui-Zhi decoction by high performance liquid chromatography/quadrupole-time-of-flight mass spectrometry and ultra-performance liquid chromatography/triple quadrupole mass spectrometry. *J. Ethnopharmacol.* 138 (1), 34–39.
- Yan, N., Wang, S.F., 2014. Analysis on chemical constituents in Danggui-Shaoyao-San by LC-Q-TOF-MS and LC-IT-MSn. *Chin. Tradit. Herb. Drugs* 45 (8), 1056–1062.
- Yan, H.L., Zhou, Y.L., Tang, F., Wang, C.J., Wu, J., Hu, C.J., Xie, X. F., Peng, C., Tan, Y.Z., 2022. A comprehensive investigation on the chemical diversity and efficacy of different parts of *Ligusticum chuanxiong*. *Food. Funct.* 13, 1092–1107.

- Yang, Y., Yang, Y., Qiu, H., Ju, Z., Shi, Y., Wang, Z., Yang, L., 2021. Localization of constituents for determining the age and parts of ginseng through ultraperformance liquid chromatography quadrupole/time of flight-mass spectrometry combined with desorption electrospray ionization mass spectrometry imaging. *J. Pharm. Biomed. Anal.* 193, 113722.
- Yuan, L., Yin, J., Tian, M., Xie, J.B., Wang, Y., Hou, Z.G., Li, Y.B., Zhang, Y.J., 2016. The classification and identification of complex chemical compositions based on UPLC-Q-TOF/MS using yanhusuo herb as an example. *Anal. Methods*. 8 (10), 2274–2281.
- Zhang, Q.Q., Huo, M.Q., Zhang, Y.L., Qiao, Y.J., Gao, X.Y., 2018. A strategy to improve the identification reliability of the chemical constituents by high-resolution mass spectrometry-based isomer structure prediction combined with a quantitative structure retention relationship analysis: Phthalide compounds in Chuanxiong as a test case. *J. Chromatogr. A*. 1552, 17–28.
- Zhang, X., Xu, F., Liu, L., Feng, L., Wu, X., Shen, Y., Sun, Y., Wu, X., Xu, Q., 2017. (+)-Borneol improves the efficacy of edaravone against DSS-induced colitis by promoting M2 macrophages polarization via JAK2-STAT3 signaling pathway. *Int. Immunopharmacol.* 53, 1–10.
- Zhang, Q., Zhang, J.H., Song, X., Bi, L.J., Li, H.Y., Li, Y., Wu, Y., Li, X.F., 2016. A detection method for Shexiangxintongning preparation. CN104897811B.
- Zheng, X., Wang, S., Zou, X., Jing, Y., Yang, R., Li, S., Wang, F., 2017. Ginsenoside Rb1 improves cardiac function and remodeling in heart failure. *Exp. Anim.* 66 (3), 217–228.
- Zheng, S.D., Wu, J.H., Wu, D.L., 2012. Roles and mechanisms of ginseng in protecting heart. *Chin. J. Integr. Med.* 18, 548–555.
- Zheng, X., Zheng, W., Zhou, J., Gao, X., Liu, Z., Han, N., Yin, J., 2018. Study on the discrimination between *Corydalis Rhizoma* and its adulterants based on HPLC-DAD-Q-TOF-MS associated with chemometric analysis. *J. Chromatogr. B*. 1090, 110–121.