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### **ORIGINAL ARTICLE**

# Nanoemulsion of *Myrtus communis* essential oil and evaluation of its larvicidal activity against *Anopheles stephensi*



## Samira Firooziyan<sup>a,b</sup>, Mahmoud Osanloo<sup>c</sup>, Hamid Reza Basseri<sup>a</sup>, Seyed Hasan Moosa-Kazemi<sup>a</sup>, Habib Mohammadzadeh Hajipirloo<sup>d</sup>, Amir Amani<sup>e,f,\*</sup>, Mohammad Mehdi Sedaghat<sup>a,\*</sup>

- <sup>a</sup> Department of Medical Entomology and Vector Control, School of Public Health, Tehran University of Medical Sciences, Tehran, Iran
- <sup>b</sup> Urmia Health Center, Disease Control Unit, Urmia University of Medical Sciences, Urmia, Iran
- <sup>c</sup> Department of Medical Nanotechnology, School of Advanced Technologies in Medicine, Fasa University of Medical Sciences, Fasa, Iran
- <sup>d</sup> Medical Parasitology Department, School of Medicine, Urmia University of Medical Sciences, Urmia, Iran
- <sup>e</sup> Natural Products and Medicinal Plants Research Center, North Khorasan University of Medical Sciences, Bojnurd, Iran
- <sup>f</sup> Medical Biomaterials Research Center (MBRC), Tehran University of Medical Sciences, Tehran, Iran

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#### **KEYWORDS**

Myrtus communis; Nanoemulsion; Larvicide; Anopheles stephensi **Abstract** *Purpose:* Excessive use of chemical insecticides has caused environmental pollution and vector resistance. Herbal essential oils with larvicidal properties are good alternatives to synthetic insecticides.mIn this study, larvicide properties of *Myrtus communis* essential oil and its nanoemulsion against *Anopheles stephensi* were investigated.

*Methods:* Components of *Myrtus communis* essential oil were identified by GC–MS. Nanoemulsion of essential oil was made with Tween 80, Span 20, and water. Dynamic light scattering (DLS) and transmission electron microscopy (TEM) determined particle size and morphology of nanoemulsions. The larvicide activity compared with bulk essential oil.

*Results:* A total of 107 *M. communis* essential oil compounds were discovered. The morphology of a selected nanoemulsion was spherical.  $LC_{50}$  and  $LC_{90}$  of *M. communis* essential oil were

\* Corresponding authors.

E-mail addresses: amani76@gmail.com (A. Amani), sedaghat@hotmail.co.uk (M.M. Sedaghat). Peer review under responsibility of King Saud University.



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1878-5352 © 2022 The Author(s). Published by Elsevier B.V. on behalf of King Saud University. This is an open access article under the CC BY-NC-ND license (http://creativecommons.org/licenses/by-nc-nd/4.0/). calculated as 26.1 and 46.2  $\mu$ g/ml, respectively. The larvicide activity of nanoemulsion increased by 40% compared to the bulk essential oil. The nanoemulsion's larvicide activity (100%) lasted up to 3 days, while the essential oil had larvicide properties only for up to 24 h.

*Conclusions: Myrtus communis* essential oil was found to be an effective larvicide and classified as an active larvicide. The residual efficacy of the nanoformulation of *M. communis* significantly increased compared with the bulk essential oil.

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#### 1. Introduction

Malaria remains a preventable cause of serious illness and death worldwide. The disease is caused by Plasmodium parasites, which are transmitted to other people through the bites of infected Anopheles mosquitoes (Ward 2008). Infants, children under 5, pregnant women, HIV / AIDS patients, migrant workers, and travelers are at higher risk for malaria (Osanloo et al., 2019a, 2019b, 2019c). Anopheles stephensi is a major vector of the disease. This species is the most abundant Anopheles species in Iran's malaria-prone areas (Osanloo et al., 2017a, 2017b, Firooziyan et al., 2021). According to the World Health Organization, until 2011, the distribution of A. stephensi was limited to certain countries in Southeast Asia and the Arabian Peninsula. Since then, this vector has expanded to Djibouti (2012), Ethiopia (2016), Sri Lanka (2017), and Sudan (2019). In the Horn of Africa, the vector appears to be expanding from neighboring countries (Djibouti) to neighboring countries (Osanloo et al., 2019a, 2019b, 2019c). Anopheles stephensi quickly adapts to the local environment. That is why it spreads to new lands (Osanloo et al., 2018). In addition, it resists several classes of insecticides and poses potential challenges to be controlled (Organization 2005, Fathian et al., 2015, Osanloo et al., 2019a, 2019b, 2019c). The danger of pesticides to ecosystems, human health, and other living organisms have become apparent, which has led to increased attention to herbal pesticides, especially plant essential oils, in recent years (Khanavi et al., 2013, Osanloo et al., 2019a, 2019b, 2019c).

The main problem of using essential oils in nature is their high volatility and consequent instability. To deal with this problem, nanoformulations should increase the stability and efficiency of the desired compound (Guan et al., 2010, Osanloo et al., 2017a, 2017b, Osanloo et al., 2018, Osanloo et al., 2019a, 2019b, 2019c, Firooziyan et al., 2021). Due to the small size of the droplets and the resulting transparency and long-term physical stability (without any apparent coagulation, precipitation, or biphasic), Nanoemulsions cause more adsorption and reduced material consumption (Naseema et al., 2021). The term nanocapsule with dimensions of 10 to 1000 nm is used (Hemingway et al., 2006). Nanoencapsulation of the active component prevents chemical reactions between the active substance and light, moisture, and oxygen. It also reduces side effects, prolongs the shelf-life, and controls the release of the active substance (Hemingway et al., 2006, Sell 2006, Bergeson 2010). For example, the controlled release of insecticides may provide several months of efficacy in vector control applications (Yadegarinia et al., 2006). Preparation and fabrication of plant-based nanoemulsions could be an interesting alternative to chemical pesticides for controlling mosquitoes. Nanoemulsions are also non-irritation to the skin, have minimal toxicity, and are not flammable (Messaoud et al., 2005, Gurpreet and Singh 2018).

*Myrtus communis* is a small shrub that normally grows to a height of 1 to 3 m. This plant has an anti-hair loss, antifungal, antibacterial, anti-inflammatory, and anti-dandruff properties (Tuberoso et al., 2006, Akin et al., 2010).

The essential oils of *M. communis* showed different insecticidal activity against *Culex pipiens, Aedes albopictus, Pediculus humanis capitis, Mediterranean flour moth Ephestia kuehniella Zeller, the Indian meal moth Plodia interpunctella Hubner* and the bean weevil Acanthoscelides obtectus Say (Traboulsi et al., 2002, Ayvaz et al., 2010,

Sumbul et al., 2011). In addition, the repellent properties of this essential oil have been reported against *A. stephensi* (Tavassoli et al., 2011, Kayedi et al., 2014).

In this study, the larvicidal activity of *M. communis* essential oil (MEO) against *A. stephensi* was investigated and for the first time, a nanoformulation of *M. communis* essential oil (MNE) was prepared. The larvicidal properties of MNE were compared with MEO according to the guidelines of the World Health Organization (Organization 2005).

#### 2. Materials and methods

#### 2.1. Materials

MEO (100% purity) was prepared from Green Plants of Life Company (Iran), Tween 80%, Span 20, and Ethanol were from Merck Chemicals (Germany).

#### 2.2. Analysis of MEO compounds

NIST & Wiley libraries were used for the identification. The calculation was performed with a Flame ionization detector (FID). GC-MS analysis (Agilent Technologies, 7890A) was used to identify MEO compounds. The mass selection detector was 5975C VL MSD with a Triple-Axis detector and the Ion source was Electron Impact (EI) 70 eV. The column type was Rtx 5 MS with a length of 30 m and an inner diameter of 0.25 mm. The conditions and temperature program are given in Table 1.

#### 2.3. Mosquito raring

Conditions

Anopheles stephensi mosquitoes were raised in the insectary at 29  $\pm$  1 °C with a relative humidity of 70  $\pm$  5% under 12 h

 Table 1
 The conditions and temperature program of GC-MS.

1. Injection port and ion source temperature	230 °C
2. Carrier gas	He 99.999%
3. Sample volume	0.2 µL
Temperature Program	
Initial temperature (°C)	40
Initial time (min)	1
Program rate (°C/min)	3
Final temperature (°C)	270
Final time (min)	10
Split ratio (ml/min)	100
Septum purge (ml/min)	-
Flow rate (ml/min)	1

lightness/12 h darkness. Mosquitoes were fed with sugar (10%) and defibrillated with sheep blood. The larvae were fed with fish flakes. Third and early fourth instar larvae were selected for larvicide tests that had been eaten the day before.

#### 2.4. Bioassay test

Serial dilutions of the MEO were prepared using ethanol. According to the WHO guideline, water chlorine-free (room temperature, pH 7) was used for the bioassay of larvae. One ml of diluted MEO or MNE was added to 249 ml of water. Then 25 healthy larvae were added to the beaker. The number of live and dead larvae per beaker was counted after 24 h.

#### 2.5. Residual effect

To test residual properties, 1 ml of MEO or MNE was added to 249 ml of water and 25 live larvae were added. After 24 h and reading the test results, without changing the solution, all larvae (dead or alive) were removed and 25 new live larvae were added to the beaker. The larvae were exchanged for up to 8 days, and the results were read each day.

All tests were repeated 4 times. In each replicate, two control groups were considered to have 1 ml of ethanol instead of MEO or MNE.

#### 2.6. Statistical analysis

Lethal concentrations of 50% and 90% (LC50 and LC90) were calculated using Minitab software (Pennsylvania State University by researchers Barbara F. Ryan, Thomas A. Ryan, Jr., Brian L. Joiner in 1972) and compared using an independent sample test by SPSS. The regression line was plotted using Excel 2007 software (the Microsoft Corporation in 1985).

#### 2.7. Preparation of MNE

First, Tween 80, Span 20 and the MEO (10 min, 600 rpm) were mixed in colorless glass vials at room temperature. The water was then added dropwise and stirred for 38 min. The solutions were placed in a dark cupboard at room temperature for 24 h. Vials with signs of phase separation, precipitation, or creaminess were discarded. The particle size (PS) of homogeneous solutions was measured by dynamic light scattering (DLS) (K-ONE.LTD, Korea). Transmission electron microscopy (TEM) (Zeiss, Germany) was used to confirm PS and examine particle morphology.

#### 3. Results

#### 3.1. Chemical composition of MEO

A total of 107 compounds were identified in the MEO (Table 2). The major compounds identified were  $\alpha$ -Pinene (34.199%), dl-Limonene (16.587%), 1,8-Cineole (8.301%), Linalool (8.223%), and Linalyl acetate (4.945%).

#### 3.2. Determination of droplet size and morphology of MNE

Ten MNEs were prepared. From DLS results, a solution with  $d50 \sim 200$  nm,  $d90 \sim 400$  nm, and span < 1 was selected as the best nanoformulation (F4) for the next steps. Fig. 1 shows the DLS of the selected sample.

#### 4. Morphology of MNE

The TEM image of F4 is given in Fig. 2. The MNE droplets were well-formed, and the particles were almost spherical.

#### 4.1. MEO larvicide properties

The larvicide effect started from 12.5  $\mu$ g/ml and increased with increasing concentration and at 100  $\mu$ g/ml all larvae were killed. Based on probit analysis, calculated LC<sub>50</sub> and LC<sub>90</sub> values for the essential oil were 26.1 (17.56–39.44) and 46.2 (38.62–61.18)  $\mu$ g/ml, respectively (Fig. 3).

#### 4.2. Comparison of larvicidal properties MNE with MEO

To compare the larvicide properties of MNE and MEO against *A. stephensi* larvae, equal concentrations of MEO and MNE (F4) were used in the test. As shown in Fig. 4, after 24 h, MNE had 40% more efficacy compared with MEO. Based on the *t*-test, p < 0.001 was obtained.

The MNE showed better larvicidal properties in the 8-day test compared to its MEO and was able to kill all larvae in up to three days. After the third day, larval mortality decreased and reached 44%. MEO was able to control 100% of the larvae only for the first day, and the larval mortality rate decreased from 92% to 16% from the second day to the sixth day. MEO did not affect the larvae on the seventh and eighth days of the experiment (Fig. 5).

#### 5. Discussion

Chemical pesticides cause serious damage to the environment and non-target animals (Guan et al., 2010). In addition, frequent use of pesticides has led to pests resistance. This has reduced the number of pesticides suitable for use in vector control programs, especially in the control of malaria and Aedes-borne diseases (Hemingway et al., 2006, Sell 2006). Nanoformulations, by providing faster and more absorption in the target pest, can increase the efficacy of pesticides (Bergeson 2010, Gurpreet and Singh 2018). Nano-formulations of plant pesticides could be an interesting alternative in the fight against agricultural pests and vector-borne diseases.

In this study, the number of identified components (i.e. 107) in MEO was more than in previous studies (i.e. 16 to 70) (Vanhaelen and Vanhaelen-Fastré 1980, Chalchat et al., 1998, Asllani 2000, Messaoud et al., 2005, Tuberoso et al., 2006, Yadegarinia et al., 2006, Akin et al., 2010, Zomorodian et al., 2013, Rasooli et al., 2018). Comparison and study of ingredients of EO are very important since different numbers of biologically active substances affect the biological activity of EO (Koutsaviti et al., 2015). Various factors can affect the chemical composition of the studied EOs. They

No.	Retention Index	Compound	%	No.	Retention Index	Compound	%
1	932	alpha Pinene	34.2	55	1423	1-Cyclohexyl-1-butyne	0.1
2	1137	dl-Limonene	16.6	56	1264	2,6-Octadienal, 3,7-dimethyl	0.1
3	1026	1,8-Cineole	8.3	57	1639	Aromadendrene	0.1
4	1214	Linalool	8.2	58	1188	Bicyclooct-1-ene, 7-exo-ethenyl	0.1
5	1373	Linalyl acetate	4.9	59	954	Camphene	0.1
6	1020	Benzene, 1-methyl-4-(1-methylethyl)-	3.4	60	1639	Phenacetic acid	0.1
7	1186	3-Cyclohexene-1-methanol	2.7	61	1365	2-Cyclohexen-1	0.1
8	1361	2,6-Octadien-1-ol, 3,7-dimethyl-, acetate, (Z)	2.7	62	1739	1H-Inden	0.1
9	1257	(+)-4-Carene	2.2	63	1844	2,5-Dihydro-5-methoxy-2-furanone	0.1
10	1001	Delta.3-Carene	0.9	64	1316	Camphene	0.1
11	1058	Bicyclo	0.9	65	1218	D-Fenchyl alcohol	0.1
12	1452	alpha-Humulene	0.8	66	1654	alpha-Farnesene	0.1
13	1570	trans-Caryophyllene	0.8	67	1407	Acetic acid	0.1
14	1590	Globulol	0.7	68	1578	1-Cycloheptene	0.1
15	1603	Propanoic acid	0.7	69	1065	Sabinene	0.0
16	1583	Caryophyllene oxide	0.6	70	1215	trans-Carveol	0.0
17	1324	Myrtenyl acetate	0.6	71	1740	alphaBisabolene epoxide	0.0
18	1492	Benzene, 1,2-dimethoxy-4-(2-propenyl)	0.5	72	1653	trans-beta-Farnesene	0.0
19	1088	Bicyclo heptane, 6,6-dimethyl-2- methylene	0.5	73	1417	3-Methylenebicyclo octane-2-one	0.0
20	1141	Dimethyl ether	0.5	74	1632	2(1H)-Naphthalenone	0.0
21	1639	Aromadendrene	0.5	75	1474	exo-2-Hydroxycineole acetate	0.0
22	1709	1H-Cycloprop azulen	0.5	76	1792	Naphthalene	0.0
23	1608	12-Oxabicyclo	0.4	77	1186	3-Cyclohexene-1-methanol	0.0
24	1252	Benzene, 1-methoxy-4-(2-propenyl)	0.4	78	1559	Germacrene B (CAS)	0.0
25	1086	alpha Terpinolene	0.4	79	1590	1,2-Benzenedicarboxylic acid, 3-nitro	0.0
26	1299	3-Cyclohexen-1	0.4	80	961	Verbenene	0.0
27	1342	4,6-Diethyl-2-methoxypyrimidine	0.4	81	1880	Phosphonous dichloride	0.0
28	1577	1H-Cycloprop[e]azulene	0.3	82	1052	Bicyclo heptane, 7,7-dimethyl-2- methylene	0.0
29	1633	Benzamide, 3,4-fluoro	0.3	83	1674	10-Methyl-2,5;3,10-diepoxybicyclo decane	0.0
30	1434	Neryl acetete neryl	0.3	84	1444	Naphthalene	0.0
31	1370	Phenol, 2-methyl-5-(1-methylethyl)	0.3	85	1582	6,6-Dimethyl-3-oxatricyclo	0.0
32	1090	beta-Myrcene	0.3	86	1578	Spathulenol	0.0
33	1088	3-ethylpropadiene	0.3	87	1407	Longifolene-(V4)	0.0
34	1135	trans-Pinocarveol	0.2	88	1326	Limonene dioxide 2	0.0
35	1044	1,3,6-Octatriene, 3,7-dimethyl-, €	0.2	89	1489	beta-Selinene	0.0
36	1389	Neoisolongifolene	0.2	90	1283	Borneol	0.0
37	1280	CIS-Verbenol	0.2	91	1086	alpha-Terpinene	0.0
38	1310	2-Cyclohexen-1-one	0.2	92	1099	alpha-PINENE	0.0
39	1592	Veridiflorol	0.1	93	1122	3-Cyclopentene-1-acetaldehyde, 2	0.0
40	1590	Epiglobulol	0.1	94	1803	6,7-Dimethoxy-5- hydroxymethylbenzofuran	0.0
41	1294	l-Phellandrene	0.1	95	1871	Phosphorous acid, tributyl ester	0.0
42	1531	1,6,10-Dodecatrien-3-ol, 3,7,11- trimethyl-	0.1	96	1350	Citronellyl acetate	0.0
43	1099	3-Oxatricyclo octane, 2,7,7-trimethyl-	0.1	97	1639	caryophylla-4(12),8(13)-dien-5beta-ol	0.0
44	1379	trans-Geraniol	0.1	98	650	2-Propanone (CAS)	0.0
45	1489	Eudesma-4(14),11-diene	0.1	99	1783	2-Naphthalenemethanol	0.0
46	1032	cis-Ocimene	0.1	100	1007	Acetic acid	0.0
47	1054	gamma Terpinene	0.1	101	1566	1H-Cyclopropa[a]naphthalene	0.0
48	1316	Camphene	0.1	102	1112	Bicyclo hept-2-ene, 3,7,7-trimethyl	0.0
49	1632	1-Ethyl-3,4,5,6,12,12b-hexahydro	0.1	103	1394	Benzene, 1-methyl-3-(1-methylethyl)	0.0
50	1170	cis-Linaloloxide	0.1	104	788	3-Pentanone, 2,4-dimethyl	0.0
51	1654	2-Octene, 2,3,7-trimethyl-	0.1	105	931	1,2,4,4-Tetramethylcyclopentene	0.0
52	1374	Isoledene	0.1	106	1008	Bicyclo hept-2-ene, 2,7,7-trimethyl	0.0
53	1319	Bicyclo hept-3-en-2-one	0.1	107	1008	Delta3-Carene	0.0
54	1496	Ledene	0.1				100

Table 2Identified components of MEO using GC-MS.







Fig. 2 TEM image of the MNE particles.



Fig. 3 Probit regression line of *A. stephensi* larvae exposed to different interval concentrations of MEO.

include geographical location, the plant parts studied, type and method of extraction and/ or analysis of the EOs, method of drying, storage conditions, fruit ripening stage, type of the plant studied (wild / lab), cultivar and genotype of the plants



Fig. 4 Comparison of larvicidal effects of MNE and MEO against *A. stephensi* larvae after 24 h.

(Parra and Amo-Marco 1998). It is difficult to establish a relationship between the larvicidal efficiency and the components of the essential oil, as the interactions and synergistic effects between the components can affect the activity of the MEO (Felipe et al., 2008, Scotti et al., 2014). However, it is well known that multi-component formulations are less likely to slow resistance in vectors. This is the main advantage of using EOs as pesticides.

Although the effect of many substances is still unknown, what is certain is that the vector resistance to pesticides is more common to a formulation with an effective component rather than a multi-component formulation (Intirach et al., 2012, Worthington and Melander 2013, Araujo et al., 2016).

A few studies have been performed on the larvicidal properties of MEO against important health vectors. A comparable result was obtained by Koutsavitia et al. (2015). They collected various taxa of the *M. communis* from different parts of Greece and extracted essential oils. All samples of Greek essential oils were tested against Culex pipiens. They reported moderate to weak larvicide properties compared with our results (LC<sub>95</sub> and LC<sub>50</sub> were 160 and 76.6 mg / l, respectively). Such differences may be attributed to the different chemical compositions of EOs were prepared from plants, although there was not a significant relationship between the chemical composition of different taxa of this plant in this study and their larvicidal activity (Koutsaviti et al., 2015). In 2010, Conti and colleagues investigated the larvicide properties of essential oils against Aedes albopictus. The results were shown at a concentration of 300 µg/ml only 36.7% of larvae were killed (Conti et al., 2010) while our results indicated 100% mortality at 50  $\mu$ g/ml.



Fig. 5 Comparison of larvicidal properties MNE and MEO against A. stephensi larvae in 8 days.

Plants EOs have been categorized based on the LC50 into six categories (Vatandoost et al. 2012). According to their suggestion, plants EOs with the LC50 values of less than 1 consider as extremely-active, 1-5 very active, 5-50 active, 50-100 moderately active, 100-200 slightly active, and more than 200 non-active respectively. The larvicide properties of MEO are active in this classification and can compete with chemical larvicides, we used nanoemulsion to increase the stability of the EO in nature. From our findings, the MNE increased the larvicide properties against A. stephensi by ~ 40% compared to the bulk EO (Osanloo et al., 2017a, 2017b). This is due to smaller particles which contribute to improved penetration into the larvae. Also, the residual effect test showed complete larvicide activity for 3 days for the MNE and 1 day for the MEO. A similar pattern of results has been obtained previously. Firooziyan et al. (2021) investigated the larvicide properties of cinnamon essential oil and nanoemulsion against A. stephensi larvae. The results were shown a 32% increase in the larvicide effects of nanoemulsion of cinnamon essential oil compared to cinnamon essential oil. Nanoemulsion of cinnamon essential oil had a high larvicide effect for up to 72 h, while after 24 h the residual effect of cinnamon essential oil decreased (Firooziyan et al., 2021). Osanloo et al. (2021) achieved similar results. Nanoemulsion of Artemisia dracunculus essential oil increased the larvicide properties of its essential oil against A. stephensi from two days to nine days (Osanloo et al., 2019a, 2019b, 2019c). A comparable result was obtained by Volpato et al. (2016). They investigated the effect of cinnamon essential oil and nanoemulsion against Alphitobius diaperinus. Nanoemulsion at a concentration of 5% caused 70% mortality of larvae after two days and had a threefold effect compared to treatment with essential oil (Volpato et al., 2016). Balasubramani et al. (2017) also showed increased larvicide activity of nanoformulation of Vitex negundo L.essential oil compared to its essential oil against Aedes aegypti (Balasubramani et al., 2017).

#### 6. Conclusion

The study showed that nanoemulsion of essential oil may play an important role in the control of larvae. The nanoemulsion also showed increased stability for the essential oils the main challenge when using essential oils as larvicides.

#### **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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