

ORIGINAL ARTICLE

Extraction and carrier mediated transport of urea using noncyclic receptors through liquid membrane systems

King Saud University

Arabian Journal of Chemistry

www.ksu.edu.sa www.sciencedirect.com

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Received 15 July 2019; accepted 2 December 2019 Available online 12 December 2019

KEYWORDS

Liquid-liquid extraction; Bulk liquid membrane (BLM) and supported liquid membrane (SLM) transport system; Hydrophobicity; Podands; Recyclization Abstract Extraction and carrier mediated transport through bulk liquid membrane and supported liquid membrane systems have wide applications in separation technology. This paper highlights the use of six noncyclic receptors (podands) having variations in chain length and end group for the removal of urea using liquid membrane system. These receptors R_1 , R_2 , R_3 , R_4 , R_5 , R_6 are diethylene glycol dimethyl ether, diethylene glycol dibutyl ether, diethylene glycol dibenzate, diethylene glycol, triethylene glycol and tetraethylene glycol respectively. The sequence of extraction and transport of urea by BLM system using various receptors is $R_2 > R_3 > R_1 > R_4 > R_5 > R_6$ and $R_6 \approx R_3 > R_5 > R_4 > R_1 > R_2$ respectively. Receptor R_2 containing butyl end group is best extractant while receptor R_6 with flexible backbone is best carrier and this carrier efficiency is used to remove urea using BLM system from the feed phase by recyclization process up to 88.16%. The experimental results influenced by concentration of receptors and urea. Effect of time was also studied.

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1. Introduction

Carrier facilitated transport of biomolecules and metal ions through liquid membrane system using different receptors as an extractant as well as carrier plays a significant role in simulating biological membrane functions and separation tech-

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nologies. Liquid membranes are selective because of high transport efficiency, minimum sample consumption, and economic superiority of liquid membrane over other separation techniques. (Clark et al., 2005; Chakraborty et al., 2009).

Urea is the nitrogenous product of protein metabolism. It is used for preparing formaldehyde-Urea resin -plastics (Chen and Wang, 2017), barbiturates (Dixit et al., 2010), and fertilizers (Rahman et al., 1994; George et al., 1997). Urea is also used at large scale in the paper industry to soften cellulose and is being used to promote healing in infected wounds and other vast applications in the field of medicine (Gnewuch and Sosnovsky, 1997). Hence industrial waste water contains a large amount of urea which could be removed by membrane separation technique. This is also important for the treatment

https://doi.org/10.1016/j.arabjc.2019.12.001

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of brackish and seawater. Azizian and Nabati, 2012; Zhang et al., 2013 Several other methods reported for the removal of urea are adsorption, biological decomposition, chemical oxidation, and enzymatic decomposition etc. (Magne et al., 2002; Sugiyama et al., 2013; Rahimpour, 2004; Nicolau et al., 2014; Wu et al., 2017; Yang et al., 2018). Other then urea, removal of some heavy metals from wastewater using extraction methods has also been reported. (Citak and Tuzen, 2010; Tuzen et al., 2013). Fig. 1 shows the structure of various receptors used and Fig. 2 shows the supported liquid membrane transport system used in our study.

This reactive liquid membrane system has emerged as a novel and effective tool in separation technique and display high selectivity due to specific interaction with guest molecules by the carrier which facilitates the separation through membranes. Supramolecular receptors like crown ethers (Pedersen, 1967) podands (Vögtle and Weber, 1979), lariat ethers and calixarenes (Gutsche, 1989) have found applications for the selective transport of ions and biomolecules through bulk and supported liquid membrane systems. (Anchaliya and Sharma, 2014, 2017; Bhatnagar et al., 2008; Raizada and Sharma, 2013; Robak et al., 2009).

Earlier reports indicate that 86% of urea was removed using urea bioreactor (Nicolau et al., 2014) and 80% urea was removed using urease-immobilized fibers. (Sugiyama et al., 2013). In the present work, the removal of urea through extraction and carrier facilitated transport using chloroform bulk liquid membrane and supported liquid membrane system containing a series of noncyclic receptors is performed and urea was removed from feed phase up to 88.16% by recyclization process. The stripping phase containing urea can be used directly as fertilizer.

2. Experimental

2.1. Reagents and instruments

Diethylene glycol dimethyl ether, Diethylene glycol dibutyl ether, Diethylene glycol dibenzoate, Diethylene glycol, Triethylene glycol, Tetraethylene glycol and P-(N, N- dimethyl amino) benzaldehye were purchased from Fluka (USA). Urea was obtained from Merck (Germany) and used without further purification. Chloroform and ethanol were obtained from





SLM system

Fig. 2 Supported liquid membrane (SLM) system for the transport of urea.

Qualigens (Worli, Mumbai, India). Systronic – spectrophotometer: 106 (Ahmdabad, India) used for the estimation of urea.

2.2. Estimation of urea

A stock solution of p-(N, N- dimethyl amino) benzaldehye was prepared by dissolving 8 g in 400 mL of ethanol and 40 mL of concentrated hydrochloric acid. 1 mL of aqueous urea solution of various concentrations (0.5 M -10 M) was added to 10 mL of this stock solution and the total volume make up to 25 mL in volumetric flask, results in orange colored solution $\lambda_{max} = 440$ nm. The calibration curve was obtained with various concentration of urea and used for the estimation of urea in feed phase and stripping phase.

2.3. Extraction studies

10 mL of aqueous solution of urea (0.5 M-10 M) and 10 mL of receptor (0.1 M) solution in chloroform were taken in a 50 mL beaker and stirred on a magnetic stirrer for 24 h at room temperature. After stirring, the mixture was allowed to stand for 5 min for the separation of two phases, and the aqueous phase was analyzed for extracted urea by determining the difference in the concentration of urea in aqueous phase before and after extraction. The distribution coefficient was calculated shown in Table 2.

2.4. Transport studies

Transport studies for BLM system were performed in a "U" tube glass cell. (Anchaliya and Sharma, 2014)15 mL of chloroform containing (0.1 M) concentration of receptors (R_1 - R_6) was used as membrane phase, feed phase was composed of 10 mL of various concentrations of urea in one limb of the "U" tube and 10 mL of deionised water served as the striping phase in another limb. The membrane phase was constantly stirred for 24 h and the feed phase was replaced by double distilled water keeping the same feed phase and membrane system was stirred for another 24 h. This recyclization process was continued up to 168 h shown in Table 6 and Fig. 7.

In SLM system egg shell membrane was impregnated by dipped overnight in chloroform containing receptors (R_1 - R_6) and used as membrane support which positioned between two cylindrical half-cells. One cell compartment (feed phase) is filled with water containing urea (50 mL) and the other cell compartment (stripping phase) filled with double distilled water (50 mL), separated by membrane. Both phases were stirred on magnetic stirrer (120 rpm) at 25 °C and the striping phase was analyzed after 24 h for the concentration of urea and flux was calculated.

3. Results and discussion

The blank experiments were performed with the concentration of urea from 1 M to 10 M separately in which membrane was devoid of carrier. No detectable amount of urea across chloroform membrane was observed in striping phase which proved that there was no leakage. All measurements were performed in triplicate and average values are shown in the tables.

3.1. Effect of urea concentration

In order to find out the optimum concentration of urea for extraction, we have varied the concentration of urea from 0.5 M to 10 M and the receptor concentration was kept constant at 0.1 M. The amount of urea extracted given in Table 1 and Fig. 3. From the results it is observed that the amount of urea extracted increase with increase in the concentration of urea. A sudden increase in the amount of urea extracted was observed at 7 M.

3.2. Effect of receptor concentration

For optimization of receptor concentration, its concentration was varied from 1×10^{-3} M to 1×10^{-1} M. At the lower concentration, there was no considerable amount of urea

Table 1	Extraction of urea using six non- cyclic receptors.						
Conc of urea			Rec	eptors			
	Amount of urea extracted $(\times 10^{-3}M)$ by R ₁	Amount of urea extracted $(\times 10^{-3}M)$ by R ₂	Amount of urea extracted $(\times 10^{-3}M)$ by R ₃	Amount of urea extracted $(\times 10^{-3}M)$ by R_4	Amount of urea extracted $(\times 10^{-3}M)$ by R_5	Amount of urea extracted $(\times 10^{-3} \text{M})$ by R ₆	
0.5 M	3.50	1.10	2.05	1.95	1.80	1.55	
1 M	25.00	17.50	20.00	25.00	30.00	40.00	
2 M	50.00	37.50	70.00	50.00	30.00	60.00	
3 M	75.00	5.00	120.00	75.00	27.50	20.00	
4 M	170.00	40.00	107.00	95.00	67.50	75.00	
5 M	267.00	42.50	92.00	120.00	107.50	127.00	
6 M	275.00	12.50	70.00	100.00	87.50	112.00	
7 M	2200.00	2400.00	2250.00	1875.00	1850.00	1825.00	
8 M	2250.00	1875.00	1650.00	1555.00	1725.00	1625.00	
9 M	2275.00	1925.00	1700.00	1625.00	1775.00	1650.00	
10 M	2350.00	1975.00	1750.00	1675.00	1800.00	1700.00	

Table 2 Distribution coefficient of Extraction of urea using six non- cyclic receptors.

Conc of urea	Receptors							
	D_u by R_1	D_u by R_2	D_u by R_3	D_u by R_4	D_u by R_5	D_u by R_6		
0.5 M	0.001	0.0008	0.001	0.001	0.001	0.0009		
1 M	0.02	0.02	0.02	0.02	0.30	0.04		
2 M	0.03	0.02	0.04	0.01	0.01	0.38		
3 M	0.03	0.002	0.05	0.03	0.01	0.02		
4 M	0.06	0.03	0.03	0.03	0.02	0.02		
5 M	0.08	0.01	0.02	0.03	0.03	0.38		
6 M	0.07	0.003	0.01	0.25	0.02	0.02		
7 M	0.06	0.81	0.72	0.45	0.55	0.51		
8 M	0.52	0.39	0.335	0.3	0.35	0.32		
9 M	0.51	0.40	0.29	0.3	0.36	0.33		
10 M	0.51	0.40	0.3	0.31	0.36	0.33		



Fig. 3 Amount of urea extracted after 24 h with noncyclic receptors. Conditions: Urea concentration: 0.5 to 10 M, Receptors concentration in chloroform: 0.1 M, Stirring speed = 120 rpm (revolutions per minute) at room temperature.

extracted, therefore the optimal concentration of the receptor is 0.1 M and urea interacts with the receptors and results in the formation of urea- receptor complex in the membrane phase. Receptor R_2 possesses butyl end group is best extractant.

3.3. Effect of time

Fig. 5 shows the time dependence of urea extraction through liquid membrane containing receptor R_2 under the optimized experimental conditions. It is clear from the fig. that the amount of urea extracted was increased with time. Maximum amount of urea extracted was observed within 4–5 h, after this a gradual increase in the amount of extraction up to 20 h and then it becomes constant and observed up to 24 h.

3.4. Back extraction

For back extraction studies, after 24 h of extraction the aqueous phase and organic phase were separated. The organic phase was mixed with 10 mL of double distilled water and the system was stirred for another 24 h and then aqueous



Fig. 4 Amount of urea back extracted after 24 h with noncyclic receptors. Conditions: Urea concentration: 0.5-10 M, Receptors concentration in chloroform: 0.1 M, Stirring speed = 120 rpm at room temperature.



Fig. 5 Extraction of urea with time using receptor R_2 . Conditions: Urea concentration: 7 M, Receptor concentration in chloroform: 0.1 M, Stirring speed = 120 rpm at room temperature.

phase was analyzed for the amount of urea back extracted. From the results shown in Table 3 and Fig. 4, it is clear that for the urea concentration 1 M–6 M the amount of urea back extracted is higher in comparison to 7 M–10 M. This indicates that the back extraction is also concentration dependent.

Amount of urea back extracted $\times 10^{-3}$ M							
Conc of urea			Rece	ptors			
	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆	
0.5 M	1.02	0.85	0.95	0.80	0.80	0.60	
1 M	10.95	16.60	17.50	18.85	18.45	19.35	
2 M	12.17	17.20	18.92	17.47	17.80	18.30	
3 M	23.50	44.50	45.00	22.50	24.65	17.15	
4 M	109.65	37.15	102.15	57.15	59.65	64.65	
5 M	179.65	37.15	87.15	114.65	102.15	122.15	
6 M	197.15	9.65	67.15	89.65	77.15	94.65	
7 M	222.15	412.15	377.15	282.15	302.15	284.65	
8 M	132.15	184.65	242.15	204.65	177.15	189.65	
9 M	145.35	200.00	195.65	195.15	275.15	205.50	
10 M	300.15	255.75	190.15	185.15	215.50	210.50	

 Table 3
 Back extraction of urea using six non- cyclic receptors.

Table 4	I ransport data of urea through bulk liquid membrane by six non- cyclic receptors as carrier.								
Conc of urea	Receptors								
	Amount of urea transported $(\times 10^{-3} \text{ M})$ by R ₁	Amount of urea transported $(\times 10^{-3} \text{ M})$ by R ₂	Amount of urea transported $(\times 10^{-3} \text{ M})$ by R ₃	Amount of urea transported $(\times 10^{-3} \text{ M})$ by R ₄	Amount of urea transported $(\times 10^{-3} \text{ M})$ by R ₅	Amount of urea transported $(\times 10^{-3} \text{ M})$ by R ₆			
1 M	0.27	2.75	0.95	0.03	0.00	0.12			
2 M	2.85	3.70	3.80	0.77	1.05	0.67			
3 M	2.17	2.62	4.60	1.80	2.37	1.32			
4 M	20.75	7.15	17.55	11.67	23.82	22.00			
5 M	33.20	4.90	17.40	14.65	59.65	25.90			
6 M	39.40	12.60	122.90	114.05	110.15	123.15			
7 M	118.65	21.15	124.90	121.40	124.15	124.90			
8 M	121.4	20.9	135.4	131.9	127.4	135.9			
9 M	124.9	22.4	139.15	135.4	131.1	138.9			
10 M	129.9	25.8	144.9	138.15	135.65	145.15			



Fig. 6 Amount of urea transported by BLM system after 24 h with noncyclic receptors. Conditions: Urea concentration (feed phase): 0.5 to 10 M, Receptors concentration in chloroform (membrane phase): 0.1 M, Stirring speed = 120 rpm at room temperature.



Recyclization of urea through bulk liquid membrane Fig. 7 system. Conditions: Urea concentration (feed phase): 7 M, Receptors concentration in chloroform (membrane phase): 0.1 M, Stirring speed = 120 rpm at room temperature.

On the basis of results of back extraction, it is clear that receptors R₁- R₆ have a tendency to complexation- decomplexation under optimum conditions.



Fig. 8 Amount of urea transported by SLM system after 24 h with noncyclic receptors. Conditions: Urea concentration (feed phase): 0.5-10 M, Receptors concentration: 0.1 M, Stirring speed = 120 rpm at room temperature.

3.5. Transport of urea through liquid membrane system

The results of transport of urea by receptors R₁-R₆ through chloroform bulk liquid membrane after 24 h are shown in Table 4 and Fig. 6. The sequence of transport of urea by six receptors was observed $R_6 \approx R_3 > R_5 > R_4 > R_1 > R_2$. The sudden increase in the amount of urea transported was observed at 7 M concentration for receptor R₁, at 9 M for receptor R₂ and for the rest it was observed at 6 M. The results of removal of urea indicate that this technique can be used for the removal of urea from feed phase using these receptors.

The transport efficiency of reactive liquid membranes was improved by structural variations in the receptors i.e. change in end group and by varying chain length of the backbone of receptors. Receptor R_6 possesses a tetraethylene glycol backbone hence flexible with more number of donor sites which enhances the transport efficiency. Receptor R₃ possesses same number of donor sites as R₆ and shows almost same transport efficiency.

The results of transport of urea through supported liquid membrane (SLM) studies with different receptors are shown in Table 7 and Fig. 8. No effective transport was observed in

Table 5	Flux values for transport of urea through bulk liquid membrane by six non- cyclic receptors as carrier.							
Conc of urea			Receptors					
	$ \begin{array}{c} J_m[\times 10^{-7}] \\ (mol/m^2/s) \mbox{ by } R_1 \end{array} $	$\begin{array}{l} J_m[\times 10^{-7}] \\ (mol/m^2/s) \text{ by } R_2 \end{array}$	$\begin{array}{l} J_m[\times 10^{-7}] \\ (mol/m^2/s) \text{ by } R_3 \end{array}$	$\begin{array}{l} J_m[\times 10^{-7}] \\ (mol/m^2/s) \mbox{ by } R_4 \end{array}$	$\begin{array}{l} J_m[\times 10^{-7}] \\ (mol/m^2/s) \text{ by } R_5 \end{array}$	$J_m[\times 10^{-7}] \ (mol/m^2/s)$ by R_6		
1 M	1.00	13.00	4.00	0.00	0.00	0.00		
2 M	14.00	18.00	19.00	3.00	5.00	3.00		
3 M	10.00	13.00	23.00	9.00	11.00	6.00		
4 M	103.00	35.00	87.00	58.00	119.00	110.00		
5 M	166.00	24.00	87.00	73.00	298.00	129.00		
6 M	197.00	63.00	614.00	570.00	550.00	500.00		
7 M	593.00	105.00	624.00	607.00	620.00	624.00		
8 M	652.00	104.00	677.00	659.00	63.00	679.00		
9 M	710.00	523.00	760.00	742.00	720.00	764.00		
10 M	789.00	428.00	849.00	825.00	803.00	850.00		

Table 5	Flux values for transport	of urea through	bulk liquid membrane	by six non- cy	clic receptors as carri

Table 6	Recyclization of urea through bulk liquid membrane	
system.		

Time in hour	Amount of urea in feed phase [%]	Recovery of urea in Stripping phase [%]
24	61.33	38.67
45	55.3	44.7
72	45.92	54.08
96	28.72	71.28
120	22.18	77.82
144	18.25	81.75
168	11.87	88.16

concentration range from 1 M to 5 M of urea. As we increase the concentration of urea from 6 M to 9 M the transport efficiency of six different receptors for urea has also increased. Flux values are shown in Tables 5 and 8 for BLM and SLM system respectively.

4. Conclusion

The efficiency of six different receptors $(R_1 \text{ to } R_6)$ for extraction and transport of urea were studied. Non- cyclic receptor R2 containing butyl end group is best extractant while receptor R_6 with flexible backbone is best carrier. Receptor R_3 and R_6

Table 7 Amount of urea transported after 24 h using six non cyclic receptors in egg shell supported liquid membrane. Source phaseurea solution (50 mL): Conc.6 M to 9 M. Receiving phase -distilled water (50 mL). Membrane -egg shell membrane.

Conc of urea			Rece	ptors		
	Amount of urea transported $(\times 10^{-3} M)$ by R ₁	Amount of urea transported $(\times 10^{-3}M)$ by R ₂	Amount of urea transported $(\times 10^{-3}M)$ by R ₃	Amount of urea transported $(\times 10^{-3} \text{M})$ by R ₄	Amount of urea transported $(\times 10^{-3}M)$ by R ₅	Amount of urea transported $(\times 10^{-3}M)$ by R ₆
6 M	118.9	40.15	369.4	498.15	162.65	160.4
7 M	371.65	73.65	457.65	590.65	383.65	381.65
8 M	432.4	94.15	526.4	684.15	437.65	438.15
9 M	487.4	115.65	594.9	728.15	481.4	491.65

Table 8	8 Flux values for transported results after 24 h using six non cyclic receptors in egg shell supported liquid membrane.						
Conc of			Rece	eptors			
urea	$\frac{J_m[\times 10^{-5}]}{(mol/m^2/s)}$ by R_1	$J_m[\times 10^{-5}]$ (mol/m ² /s) by R ₂	$J_{m}[\times 10^{-5}]$ (mol/m ² /s) by R ₃	$J_{m}[\times 10^{-5}]$ (mol/m ² /s) by R ₄	$J_{m}[\times 10^{-5}]$ (mol/m ² /s) by R ₅	$\begin{array}{c} J_m[\times 10^{-5}]\\ (mol/m^2/s)\\ by R_6 \end{array}$	
6 M	30.43	10.27	94.56	127.52	41.63	41.06	
7 M	95.14	18.85	117.15	151.2	98.21	97.7	
8 M	110.69	24.1	134.75	175.14	112.03	112.16	
9 M	124.77	29.6	152.29	186.4	123.23	125.86	

backbone) play an important role in separation and results may help in designing of more specific carrier for the substrate. The BLM system is more effective for transport of urea than SLM system as 88.16% urea removed from urea contaminated water through BLM by recyclization process.

Acknowledgements

We are thankful to Prof. and Head Dr. Shubha Jain, School of Studies in Chemistry and Biochemistry, Vikram University, Ujjain for providing research facilities.

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