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REVIEW ARTICLE

A critical review on quercetin bioflavonoid and its derivatives: Scope, synthesis, and biological applications with future prospects



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Poor aqueous solubility

Abstract Lifestyle is an important factor in health, and researchers are trying to see the adverse effect on human health of a change in lifestyle. Many people follow unhealthy lifestyles, which affect human physical and mental health. Millions of people suffer from various problems like cardiovascular disease, overweight, malnutrition, hypertension, stress, violence, etc. There are many medicines that can cure these diseases, but they have side effects. When a man consumes green leaves and vegetables in his diet, his chances of reducing the risk of any disease increase. Every plant has primary and secondary metabolites. Flavonoid is a secondary metabolite found in various plants that plays an important role in the survival of the plant by performing physiological activities and being able to withstand adverse effects from the environment. Quercetin is one such bioflavonoid found in many plants, like onions, citrus fruits, parsley, etc. It comes under the category of flavone, with five hydroxy groups and a chromen-4-one skeletal structure. It has various applications, which include anti-cancer, anti-bacterial, anti-viral, anti-inflammatory, anti-diabetic, etc. However, the use of quercetin is limited due to its poor bioavailability, poor aqueous solubility, lower stability in the gastrointestinal tract, and first-pass effect. To enhance the application of quercetin, various derivatives of quercetin and its metal complex are required that are water soluble, which helps them reach the target site without any first-pass effect and provides them with stability. This review is divided into three main sections: The first section explains about flavonoids and their types; the second section explains about quercetin, its various methods of synthesis, its derivatives, and its metal complex; and the last section represents the complete applications of quercetin, with the drawbacks and how to overcome or deal with the drawbacks.

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1. Introduction

As individuals, we might have wondered how plants survive. But the answer is simple: plants survive by producing primary and secondary metabolites with other external sources like light, water, quality of soil, etc., which involve complex mechanisms that aid the survival of plants through photosynthesis and metabolic cycles. Many plants have a unique pigment called flavonoid, which cannot be produced in the human body since it is a phytochemical. Flavonoids are polyphenolic compounds, which are bioactive secondary metabolites (Mosunova et al., 2021). The main difference between primary and secondary metabolites is that secondary metabolites are not required for plant growth or reproduction; these functions are carried out by primary metabolites, mainly proteins, lipids, carbohydrates, etc. But secondary metabolites play a significant role in the survival of the plant (Mathesius, 2018). Some of the secondary metabolites are terpenoids, alkaloids, phenolic compounds, flavonoids, steroids, etc (Kessler and Kalske, 2018). Almost 50,000 secondary metabolites have been discovered in the plant kingdom (Pang et al., 2021) and medicinal plants, which mainly depend on secondary metabolites for their action (Teoh, 2016). Flavonoids are made of an aromatic ring with at least one hydroxyl group (Panche et al., 2016). They have variable phenolic structures and are found in parts of the plant like the bark, stem, flower, fruits, etc (Havsteen, 2002). Flavonoids are classified based on their molecular structure; the four main classes of flavonoids are flavanone, flavone, catechin, and anthocyanin, whose general structures are depicted in (Fig. 1) (Rice-Evans et al., 1996).

It was found that the human diet includes a higher number of flavonoids, the most abundant flavonoid is flavonol. Flavonoids give flowers their flavour, colour, and aroma; they also protect enzymes and prevent fat oxidation (Tomas-Barberan and Clifford, 2000). Since flavonoids play an important role in the aroma of flowers, this attracts pollinators, and ultimately pollination occurs in plants; they also help in fruit dispersion, which helps in seed germination and finally leads to the growth and development of the seedling (Dudek et al., 2016). Stress is one of the major issues in plants during drought conditions. Flavonoids act as a unique UV filter that help plants cope with biotic and abiotic stresses (Takahashi and Ohnishi, 2004). In addition to being drought resistant, flavonoids are also helpful in increasing freezing tolerance; sometimes, due to mutation, there might be a positive or negative effect depending upon the situation; the flavonoid character can be enhanced or diminished, but if it is diminished, it becomes hard for the plant to survive and to perform the function (Jorgensen, 1995; Pandey and Rizvi, 2009). Cotton is one of the major crops of India; however, researchers work on cotton to improve its resistance and defence against pathogens, herbivores, etc. *Bacillus thuringiensis* (Bt) cotton is widespread for its resistant ability, but further growing research uses strategies utilising flavonoids in pest management and

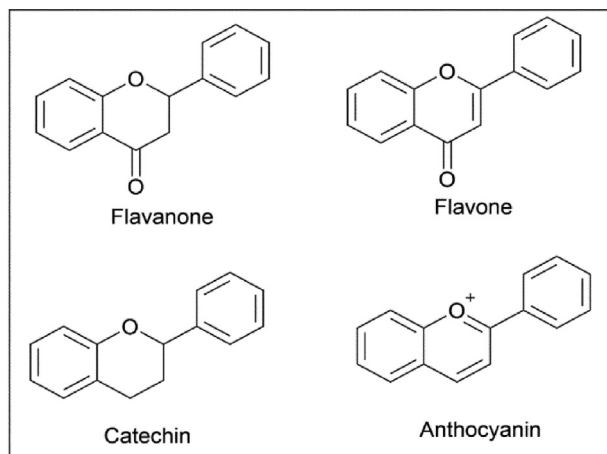


Fig. 1 General structure of four main class of flavonoids.

developing the plant with desired characters. In addition to *Bacillus thuringiensis* (Bt) cotton flavonoids are important to improve the pest resistance of bananas (Hölscher et al., 2016; Nix et al., 2017; Zafar et al., 2020). Flavonoids play an important role in plant-bacteria interactions, since they are structure-specific and control the functions of nodulation, gene expression, and infection of nitrogen-fixing rhizobia (Liu and Murray, 2016; Peters et al., 1986; Spaink et al., 1989). Apart from inducing flavonoids to increase resistance, there are natural sources of flavonoids that are available. Different classes of flavonoids in various food sources are listed below in (Table 1) (Kumar and Pandey, 2013a; Nijveldt et al., 2001).

Flavonoids having functional hydroxyl groups mediate antioxidant effects by scavenging free radicals or by chelating metal ions (Kumar et al., 2013b; Kumar and Pandey, 2013b). Prevention of radical generation might be crucial during the chelation of metals, which can damage target biomolecules (Kumar et al., 2013a; Leopoldini et al., 2006). Flavonoids have good health promoting properties since they have high antioxidant properties both *in vitro* and *in vivo* and can induce a protective system in humans (Cook, 1996; Rice-evans et al., 1995). They can regulate growth factors in plants, such as auxin. However, in the case of bacteria and fungi, there is the presence of biosynthetic genes that can produce flavonoids (Agati et al., 2012; Fangchuan et al., 2011). Apart from flavanones, flavones, and chalcones, flavonoids are also in the form of isoflavones and flavanols, which can be produced by bacteria with antioxidant properties (Song et al., 2014; Sülsen et al., 2017).

Flavonoids are biosynthesized from derivatives of acetic acid or phenylalanine using a specific pathway known as the shikimic acid pathway. Flavonols and flavones have many compounds called 2-benzo- γ -pyrone (narrow-sense flavonoids) (Wang et al., 2018c).

1.1. Flavanones

Flavanones are a type of flavonoid that are generally found in citrus fruits such as lemon, orange, etc (Table 1). They are responsible for the bitter taste of the fruit and peel in citrus fruits. Hesperitin and naringenin come under this category (Agrawal, 2011; Terahara, 2015). The main structural difference between flavanones and flavones is that flavones have a double bond present between the 2 and 3 positions, which is absent in flavanones (Fig. 1) (Iwashina, 2013). Isomerization of 2' hydroxychalcones gives 2-arylchroman-4-ones by a ring closure mechanism with a stereogenic centre. When comparing natural

Table 1 Flavonoids in various food source.

Class	Compounds	Dietary source
Flavanones	Hesperidin	Oranges, Grapefruits,
	Naringenin	Lemon and other
	Naringin	citrus fruits
	Taxifolin	
Flavone	Quercetin	Onions, parsley, lettuce, olives
	Rutin	Cranberries, Buckwheat
	Chrysin	Berries, Fruit skins
	Apigenin	Apple skin, red wine
	Luteolin	Celery
	Kaempferol	Broccoli
	Myricetin	Fruit peels
	Sibelin	Grapes
Catechins	Epicatechin	Tea
	Catechin	Red wine
	Epigallocatechin	Tea
Anthocyanins	Apigenidin	Cherry
	Cyanidin	Berries
	Peonidin	Red graphs
	Pelargonidin	Raspberries
	Petunidin	Red wine

flavanones, they are optically active and have an S configuration (Brahmachari, 2008). According to studies, sweet oranges contain 19.6 mg of aglycones per 100 g edible fruit, and the striking fact is that flavanones are 95% when measured. In the case of tangerines, flavanones were found to be 96% (Peterson et al., 2006). The concentration of flavanones will differ when measured in different fruits and vegetables. Natural flavanones are linked with sugars, in the form of 7-O-glycosides (Dias et al., 2021). Flavanones have free radical-scavenging properties, which show good health benefits like (YAO et al., 2004) anti-inflammatory action, anti-tumour activity, antioxidant activity, control of cholesterol levels, etc (Lambert et al., 2005; McPhail et al., 2003; Woo, 2013; Zhang et al., 2005).

1.2. Flavone

Flavones are mainly found in onions, parsley, lettuce, olives, cranberries, buckwheat, etc (Table 1). Chamomile and parsley have the highest flavone content, with quercetin, rutin, chrysin, etc. as some examples of flavones. The main source of flavone is from plants; mainly, they are conjugated as 7-O-glycosides with acetyl moieties (Gebhardt et al., 2005). They can be hydrolyzed with enzymes or acids even before analysis (Engelhardt et al., 1993; Hostetler et al., 2017). They have a double bond between the 2 and 3 positions and are oxidised at the C4 position (Martens and Mithöfer, 2005). They act as primary pigments in white or cream-colored flowering plants and co-pigments in blue-coloured flowering plants (Harborne and Williams, 2000). They have anti-inflammatory and antioxidant activity (Soto-Blanco, 2022). A considerable amount of flavone helps in cardiovascular and neuropathological diseases (Arredondo et al., 2015). Flavones are a signal molecule for initiating the colonisation of roots by nitrogen-fixing bacteria and fungi (mycorrhiza) (Rolfe, 1988; Siqueira et al., 1991). They protect plants by acting as natural pesticides against insects, bacteria, and fungi (McNally et al., 2003).

1.3. Catechins

Catechins are polyphenols found in tea, red wine, etc (Table 1). They account for 75% of the polyphenol compounds in tea leaves, with many chemical features, for example the hydroxy group (Singh et al., 2011b). Epicatechin, catechin, and epigallocatechin are a few types of catechin (Jin et al., 2006). They help with anti-allergic activities. Since allergy is an abnormal immune cell response, by controlling the amount of histamine binding to histamine receptors, we can prevent allergy. Studies are conducted with catechin, which was extracted from tea and analysed; it was confirmed that it prevents the binding of histamine to receptors (OHMORI et al., 1995). Also help in anti-inflammatory, anti-microbial, antioxidant, and anti-arthritis activities (Aoshima et al., 2009; Bae et al., 2020; Foyet et al., 2015; Thring et al., 2009). Catechins play an important role in protecting the human body against UV radiation, promoting cell activity (Kim et al., 2015b; Zhang et al., 2017). Catechins are extensively used in medical, pharmaceutical, cosmetic, etc (Thornfeldt, 2005).

1.4. Anthocyanins

Anthocyanins is also a type of polyphenolic compound found in cherries, berries, red grapes, raspberries, etc (Table 1). They are glycosylated polyphenolic compounds with different colours ranging from red, orange, blue, and even purple (Tanaka and Ohmiya, 2008). Since they are water-soluble, they are stored in cell vacuoles. There are about 600 anthocyanins identified, among which apigenidin, cyanidin, peonidin, pelargonidin, and petunidin are prominent ones (Kong et al., 2003; Liu et al., 2018; Smeriglio et al., 2016). Anthocyanins help plants withstand adverse conditions for both biotic and abiotic stress (Ahmed et al., 2014). Due to their attractive colour, humans have various ben-

efits like anti-obesity, anti-inflammatory properties, antioxidants, cardio-vascular activities, etc (Bors et al., 1990; He et al., 2011; Khoo et al., 2017; Rechner and Kroner, 2005; Tsuda et al., 2003).

1.5. Others

Isoflavonoids, neoflavonoids, chalcones, etc. are subgroups of flavonoids. Isoflavonoids are mainly found in soyabeans, beans, and lentils (legume seeds) (Gacek, 2014). They have anti-carcinogenic potential and antioxidant activities (Bolca, 2014; Popa and Rusu, 2017; Spagnuolo et al., 2015). Neoflavonoids have a 4-phenylchromen backbone and are mainly found in the bark of *Mesua thwaitesii*; they are mainly anti-inflammatory (Inuma et al., 1987; Nishimura et al., 2000; Wu et al., 2011). Chalcones are also referred to as open-chain flavonoids because of the absence of one ring from the basic flavonoid structure. They are found in berries like strawberries, bearberries, etc (Rozmer and Perjési, 2016; Rudrapal et al., 2021). Arbutin, chalconaringenin, phloridzin, etc., are some of the examples of chalcone (Gomes et al., 2017). They help with neurological disorders and cardiovascular diseases (Cheng et al., 2014; García-Calderón et al., 2020).

Flavonoids have a unique structure and function that allows them to interact with specific enzymes, resulting in an effective polyparamacological behaviors. This way, flavonoids show good activity against many diseases and protect us from foods we consume that have rich flavonoid content (Balasuriya and Rupasinghe, 2012; Falcone Ferreyra et al., 2012; Gao et al., 2015; Giuliani et al., 2014; Krych and Gebicka, 2013; Rakers et al., 2014).

In general, if flavonoids are present in the daily diet, which helps in the prevention of cardiovascular disease (Egert and Rimbach, 2011), one of the most important functions of flavonoids is that they have anti-cancer activity (Chandra Tiwari and Husain, 2017; Lin et al., 2005; Ozcan et al., 2014; Saxena et al., 2012). Several epidemiological studies provide evidence for the consumption of fresh fruits and vegetables that have high concentrations of flavonoids that show good activity against stroke (Pietta, 2000). When we see the structure-activity relationship of bioactive flavonoids, which have good anti-bacterial and anti-viral activity, they show therapeutic activities against *Escherichia coli*, influenza virus, hepatitis C virus, and canine distemper virus (Carvalho et al., 2013; Mierziak et al., 2014; Nguyen et al., 2013; Shibata et al., 2014) by attributing chemical structure by hydroxylation, methoxylation, etc (Kim et al., 2011). Flavonoids, in addition to their anti-cancer, anti-bacterial, and anti-virus properties, play an important role in the treatment of diabetics, as the disease is now seen in both children and adults. Diabetes is a serious issue because it is a chronic disease and has adverse effects on the human body, mainly kidney failure, stress, slow wound healing, eye problems, high blood glucose levels, being overweight or obese, etc. Flavonoids, help in maintaining diabetics and act as a barrier from their adverse effects on the human body (Ha et al., 2018; Moller, 2001; Nazir et al., 2018; Vu et al., 2020), also showed that people consuming approximately 5–80 g per day of fruits and vegetables containing flavonoids had a reduced risk of myocardial infection. Studies have also shown that people consuming 600 g of fruits and vegetables containing flavonoids have a decreased risk of coronary heart disease (Borges et al., 2010; Lock et al., 2005). One of the greatest challenges faced by neuropsychologists is that today a lot of people suffer from Alzheimer's disease and atherosclerosis. Alzheimer's disease is a progressive neurologic disorder that causes the brain to shrink, and it is usually associated with dementia; whereas, atherosclerosis is a condition that can narrow or harden the arteries, mainly due to cholesterol lining called atheromatous plaque deposition over time. Once the arteries are blocked, blood flow is reduced, which can lead to heart attacks, strokes, etc. In such cases, flavonoids are used to treat this complex puzzle called Alzheimer's disease and atherosclerosis to a certain extent, but a complete cure is not possible (Bondi et al., 2017; Castañeda-Ovando et al., 2009; Lee et al., 2016, 2009; Rafeian-

Kopaei et al., 2014). In this review, quercetin's structural aspects, method of synthesis, and main applications, which are required in the medical field, are discussed.

2. Quercetin

Quercetin (3,4,5,7-tetrahydroxyflavonol) belongs to the class of flavonoids, which are naturally occurring polyphenols. The name quercetin is derived from Quercetum (oak forest), named after Quercus in 1857. Quercetin is a yellow solid with a bitter taste; it is sparingly soluble in alcohol and soluble in glacial acetic acid but not soluble in water since it is hydrophobic in nature (Baghel et al., 2012; Scholz and Williamson, 2007). Quercetin is composed of two benzene rings linked by a heterocyclic pyrone ring, as shown in (Fig. 2). The IUPAC nomenclature of quercetin is 2-(3',4'-dihydroxyphenyl)-3,5,7-trihydrochromen-4-one. They are mainly present in onions, apples, berries, red wine, green tea, etc (Almeida et al., 2018; Anand David et al., 2016b; Ding et al., 2021; Sakai et al., 2022; Shi et al., 2021; Tu et al., 2022). They are also found in medicinal plants like *Hypericum perforatum*, *Ginkgo biloba*, etc. The quercetin content of certain selected foods is listed below in (Table 2 and Table 3) (Häkkinen et al., 1999; Wiczkowski et al., 2008; Williamson and Manach, 2005). Quercetin has a keto-carbonyl group, and salt can be generated with strong acid since the oxygen atom at the first carbon is basic (Yang et al., 2020). Quercetin has a lot of applications, like being anti-inflammatory, treating cardiovascular disease, being anti-cancer, having anti-angiogenic activity, being anti-depressive, etc (Babaei et al., 2018; Chirumbolo, 2010; Dajas, 2012; Horowitz and Zunszain, 2015; Li et al., 2020; Salehi et al., 2020; Suganthi et al., 2016; Vauzour, 2014; Wang et al., 2019; Yang et al., 2018). Quercetin also has antioxidant properties with free radical scavenging, xanthine oxidase inhibitors, modulation of gene expression, reduction of leukocyte immobilization, NOS (Nitric Oxide Synthase) inhibitors, etc (Arai et al., 2000; Chang et al., 1993; Chondrogianni et al., 2010; Costantino et al., 1999; IIO et al., 1986; Mandel et al., 2004; Mohd Sairazi and Sirajudeen, 2020; Rahman et al., 1990; Shutenko et al., 1999). Recently, quercetin was also granted GRAS (Generally Recognized as Safe) status by the United States Food and Drug Organization (de Barros et al., 2022). The dose of quercetin is usually 400–600 mg of a coated tablet, taken one–three times between meals. In one experiment, 500–1000 mg of quercetin were administered for an

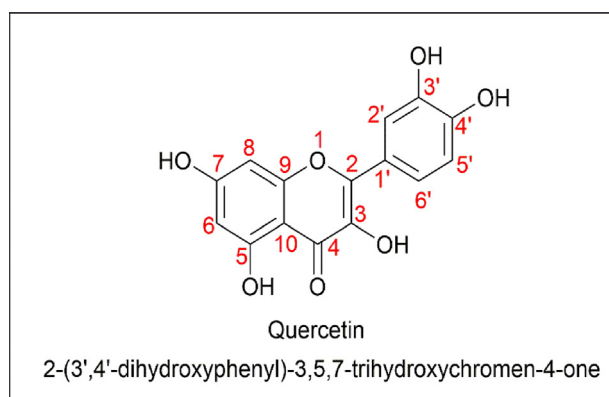


Fig. 2 Structure of Quercetin.

Table 2 Quercetin content in certain selected foods.

Food Source	Quercetin Content (mg/100 g)
Green tea leaves, dry	255.55
Black tea leaves, dry	204.66
onions	13.27
Spinach	4.28
Apple with skin	4.42
Broccoli	3.21
Red Wine	0.84

Table 3 Pharmacological activity of quercetin from various family of plants.

Family	Pharmacological Activity	Examples
Apiaceae	Lowers blood pressure	<i>Apium graveolens</i>
Amaryllidaceae	Cardioprotective, antioxidant	<i>Allium cepa</i>
Moringaceae	Antibacterial, antihypertensive	<i>Moringa oleifera</i>
Apiaceae	Wound healing	<i>Centella asiatica</i>
Hypericaceae	Neurological effects	<i>Hypericum perforatum</i>
Brassicaceae	Reduce risk of stroke, neuropathy, prevents cancer	<i>Brassica oleracea</i>
Brassicaceae	Reduce the risk of cancer	<i>Nasturtium officinale</i>
Rosaceae	Decrease the risk of cardiovascular disease	<i>Malus domestica</i>
Ericaceae	Urinary tract infections	<i>Vaccinium oxycoccus</i>
Theaceae	Antiviral, antidiabetic	<i>Camellia sinensis</i>

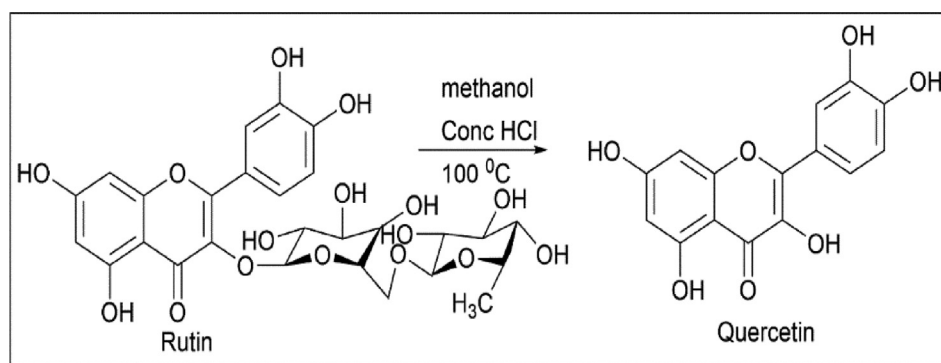
upper respiratory tract infection (Heinz et al., 2010a, 2010b; Henson et al., 2008; Li et al., 2016; NIEMAN et al., 2009). As said earlier, quercetin is not soluble in water, so in certain cases, a protein-digesting enzyme is extracted from pineapple, which increases the absorption. Most of the time, quercetin is blended with one or more additives (Horwitz, 2018; Rodríguez De Luna et al., 2020).

3. Synthesis

3.1. Synthesis of quercetin

Elena Vladimirovna Vetrova et al. (Vetrova et al., 2017) used the conventional hydrolysis method of rutin. This procedure involves dissolving 0.10 g of rutin in 1.5 ml of methanol and adding 0.25 ml of concentrated hydrochloric acid. The reaction was kept at 100 °C for 3 h. Once the hydrolysis is complete, the precipitate is filtered and dried at 80 °C for 3 h (Scheme 1).

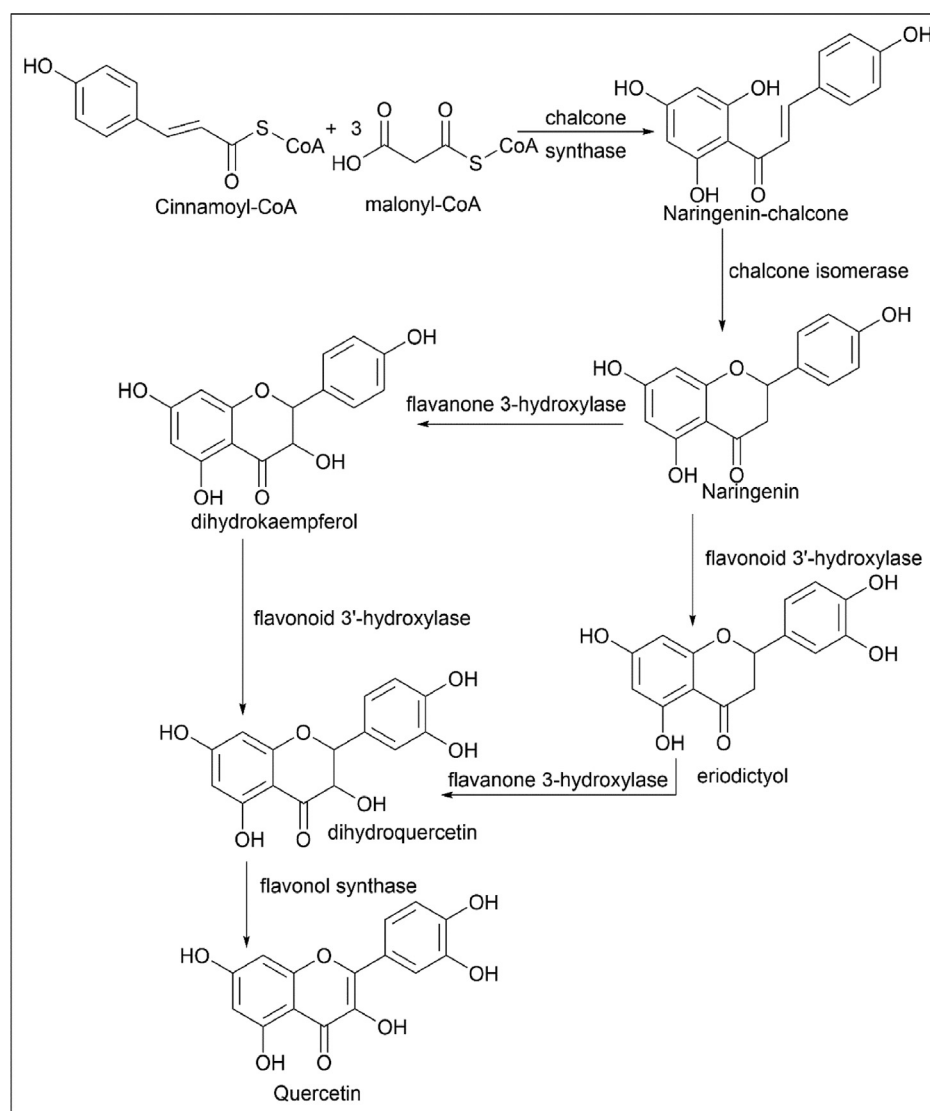
The biosynthesis of quercetin is done by using cinnamoyl-CoA, as shown by Ajay Sharma et al. (Sharma et al., 2018; Singh et al., 2021) cinnamoyl-CoA and malonyl-CoA used as starting material, and in the presence of chalcone synthase, which acts as a catalytic enzyme, results in the formation of naringenin-chalcone. Naringenin-chalcone is treated with chalcone isomerase for the isomerisation reaction, which leads to the formation of naringenin. Naringenin is treated with fla-



Scheme 1 Synthesis of Quercetin by hydrolysis of Rutin.

vanone 3-hydroxylase and flavonoid 3'-hydroxylase, which give dihydrokaempferol and eriodictyol, respectively, by hydroxylation reactions at the C3 and C3' positions. The resultant compound, dihydrokaempferol, was treated with flavonoid 3'-hydroxylase, and eriodictyol was treated with

flavanone 3-hydroxylase so that hydroxylation takes place at the C3' and C3 positions of dihydrokaempferol and eriodictyol, which results in the formation of dihydroquercetin. It is further reacted with flavonol synthase, resulting in the formation of quercetin by forming a double bond (Scheme 2).



Scheme 2 Biosynthesis of Quercetin using Cinnamoyl-CoA and Malonyl-CoA.

3.2. Derivatives of quercetin

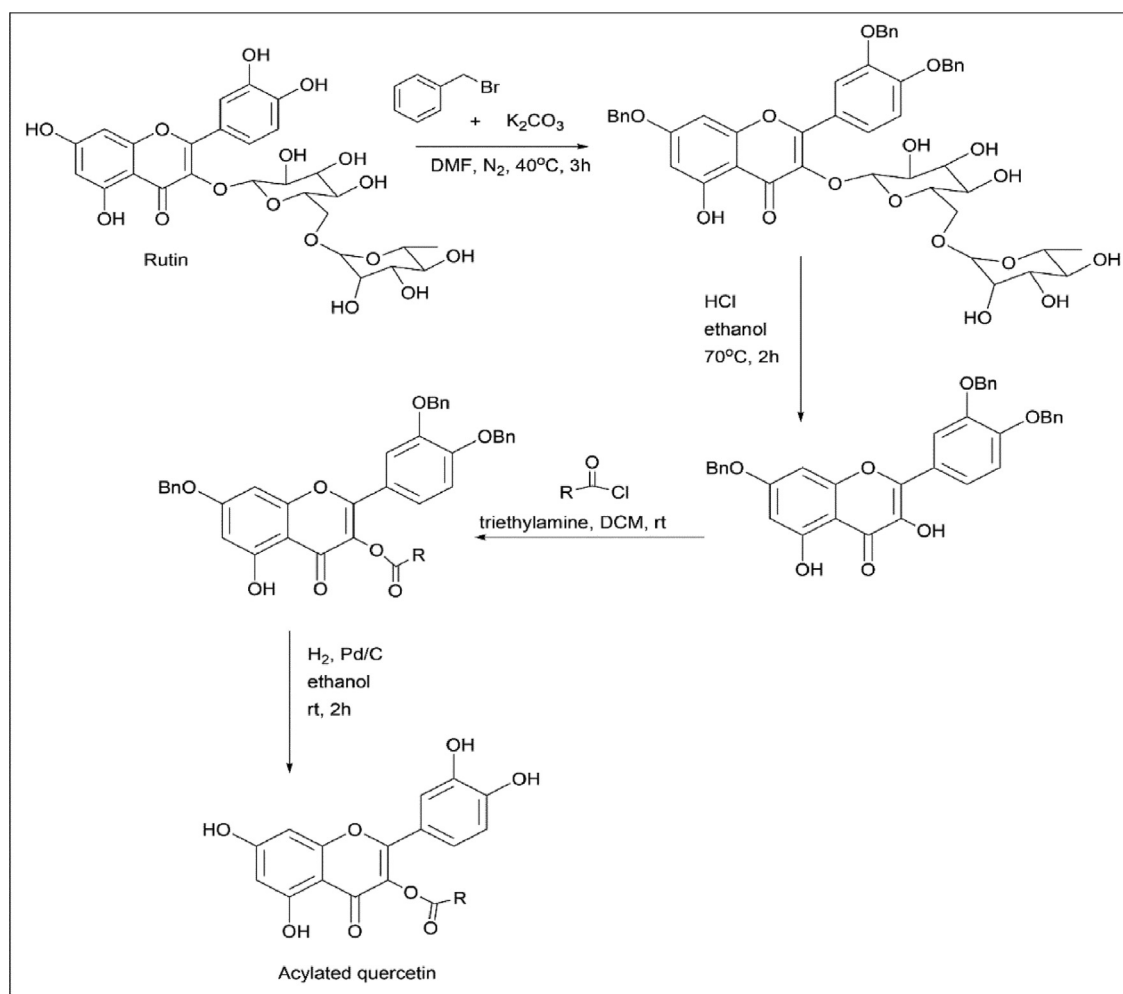
Duan *et al.* (Duan *et al.*, 2017) synthesised an acylated quercetin analogue using rutin as the starting material. 40 mmol of rutin and 133 mmol of K_2CO_3 were added to 160 ml of DMF under nitrogen conditions at room temperature for about one hour. For the above solution, 133 mmol of benzyl bromide was added dropwise and stirred for 3 h at 40 °C under nitrogen condition, then the pH was adjusted to 6.0 using acetic acid at 10% (v/v) in an ice bath. About 300 ml of deionized water was added and filtered. For 600 ml of 95% (v/v) of ethanol, the filtered residue was added at 70 °C HCl was added, and hydrolysis continued at 70 °C for 2 h. The suspension was filtered and washed with ice water until its pH became neutral to yield hydrolysate. In 250 ml of DCM, the hydrolysate was dissolved, followed by the addition of equimolar chlorides (butyryl chloride, propionyl chloride, etc) and TEA. After the reaction was completed, it was extracted with 1 mol/L HCl. A solution of saturated aqueous solution of $NaHCO_3$ and deionized water was used to wash the organic layer and dry it. In 5000 ml of ethanol/dioxane (1:1), the acylated compound was dissolved, Pd/C was added, and the mixture was stirred at room temperature under hydrogen at atmospheric pressure for

about 2 h. The Pd/C was removed by filtering, and an acylated quercetin crude product was obtained (Scheme 3).

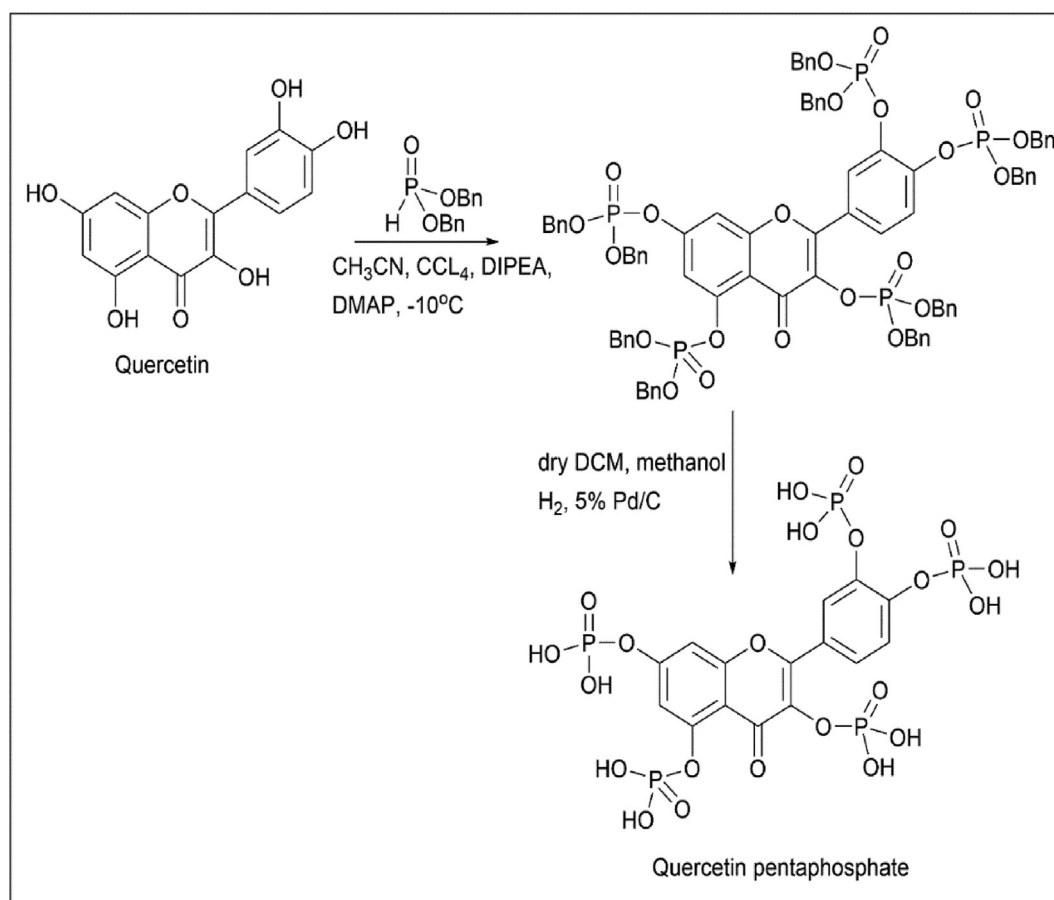
Osonga *et al.* (Osonga *et al.*, 2017) used quercetin as the starting material, to which dibenzyl phosphate was added, which gets substituted on the OH group of quercetin in the presence of methyl cyanide and carbon tetrachloride. Once the reaction is over, dry DCM is added, H_2 is added, 5% Pd/C is added so that the benzyl group is converted to an OH group to form quercetin pentaphosphate (Scheme 4).

Mihyang Kim *et al.* (Kim *et al.*, 2015a) alkylated quercetin to form pentamethoxyquercetin. 3 mmol of quercetin was dissolved in 120 ml of DMF, 90.1 mmol of K_2CO_3 and methyl iodide were added to the reaction mixture and stirred for 3 h at room temperature. Ethyl acetate was used for extraction and washed with brine, dried over anhydrous $MgSO_4$, and then filtered and concentrated under reduced pressure. Purified using vacuum liquid chromatography using hexane and ethyl acetate to yield pentamethoxyquercetin (Scheme 5).

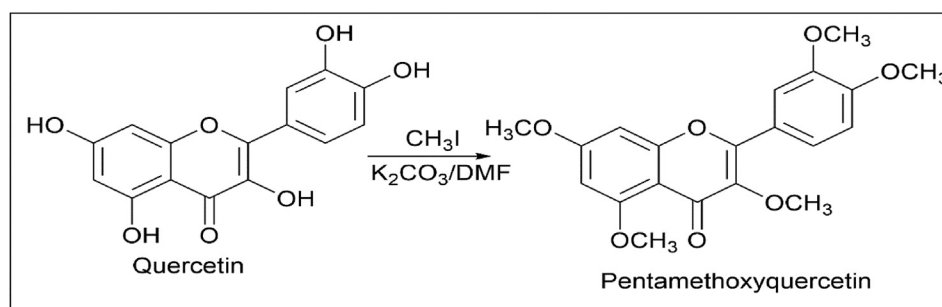
Martina Danihelova *et al.* (Danihelová *et al.*, 2012) used a similar method for alkylation of quercetin using 1-chloropropan-2-ol. Quercetin was dissolved in DMF; K_2CO_3 and 1-chloropropane was added. Once the reaction is over, the compound is filtered and dried (Scheme 6).



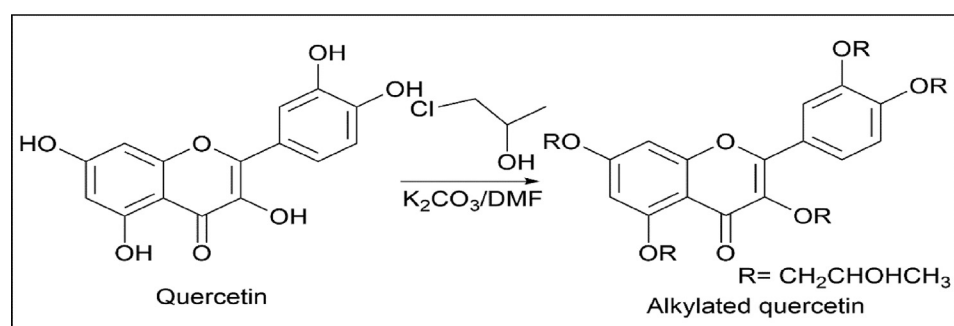
Scheme 3 Synthesis of Acylated Quercetin.



Scheme 4 Synthesis of Quercetin Pentaphosphate.



Scheme 5 Synthesis of Pentamethoxyquercetin.



Scheme 6 Synthesis of Alkylated Quercetin.

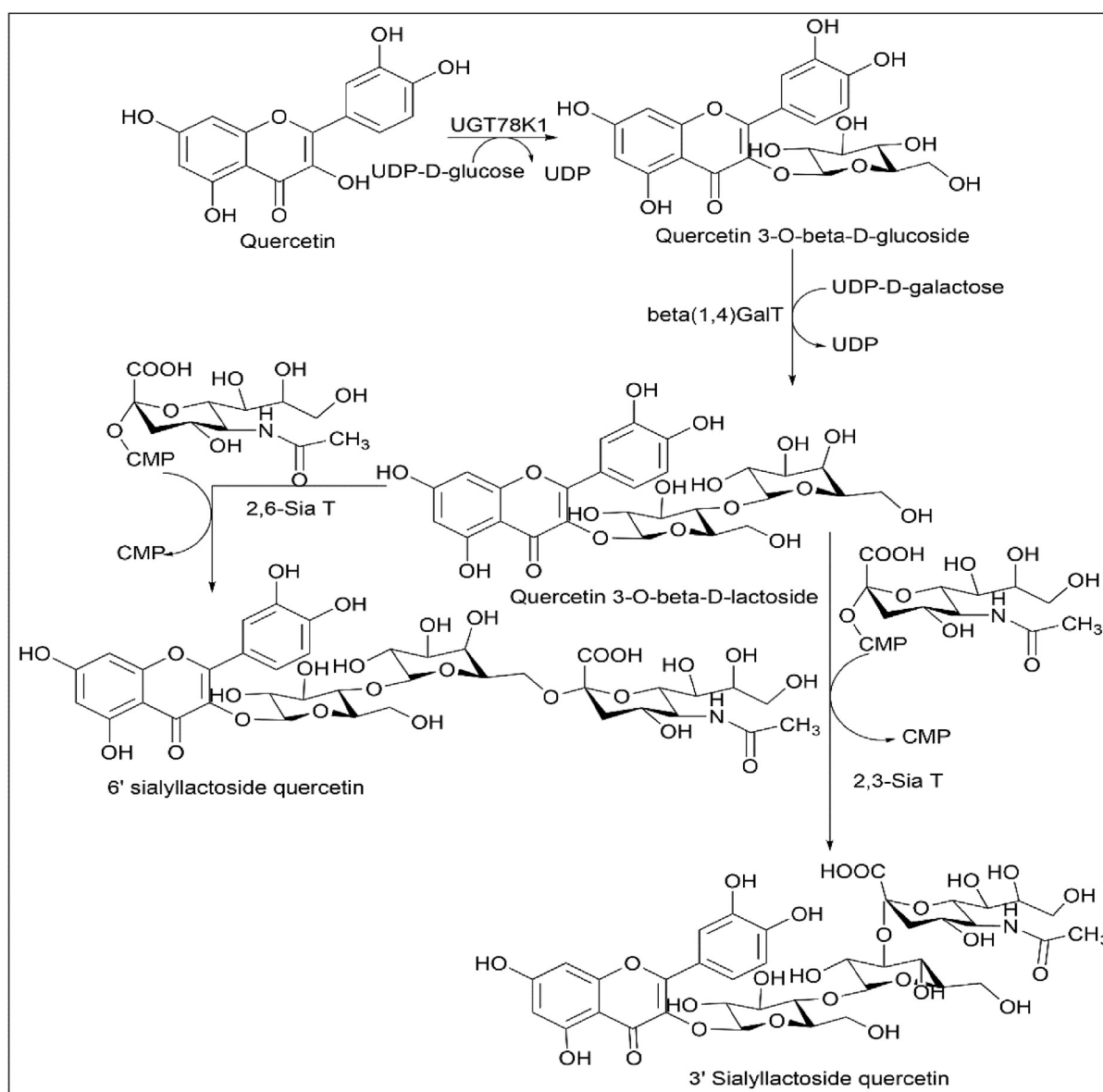
Darsandhari *et al.* (Darsandhari *et al.*, 2019) used enzymatic synthesis of quercetin derivatives. Here, three derivatives of quercetin were synthesised, namely quercetin 3-O- β -D-glucoside, quercetin 3-O- β -D-lactoside, 3' sialyllactosyl quercetin and 6' sialyllactosyl quercetin.

The synthesis of quercetin 3-O- β -D-glucoside is done by one pot synthesis using 100 mM tris buffer, 20 mM $MgCl_2$, 150 mM acetyl phosphate, 50 mM glucose 1-phosphate, 2 mM UMP, 8 mM quercetin, and 1 mM ATP. The enzymes used is ACK, GalU, and UGT78K1, incubated at 37 °C (Scheme 7).

The synthesis of quercetin 3-O- β -D-lactoside was also prepared using enzyme, 50 mM of tris-Cl buffer (pH 7), 10 mM of UDP- α -D-lactoside, $MnCl_2$, and 20% crude β (1,4) GalT. To this, 2 mM quercetin 3-O- β -D-glucoside was added and incubated for 2 h at 37 °C (Scheme 7).

The synthesis of 3' sialyllactosyl quercetin and 6' sialyllactosyl quercetin involves 25 mM tris-HCl buffer (pH 7.5), 20 mM $MgCl_2 \cdot 6H_2O$, and 5 mM quercetin 3-O- β -D-lactoside dissolved in DMSO. CMP-NeuNAc was used as a donor along with the sialyltransferase enzymes, 10 mM of CMP-NeuNAc and 50 μ g/ml of sialyltransferase were added. The reaction mixture was incubated at 37 °C, unreacted protein was removed by centrifugation (Scheme 7).

Rubin Thapa Magar *et al.* (Magar and Sohng, 2020) used a different strategy of synthesising a derivative of quercetin using gram-negative bacteria called *Escherichia coli* by metabolic engineering. *E. coli* is designed or engineered in such a way that it expresses only the desired biocatalysts, which are isolated from various sources while suppressing other genes in *E. coli*. This engineered *E. coli* is used for the biotransformation of quercetin. Engineered *E. coli* produce nucleoside



Scheme 7 Synthesis of Quercetin 3-O- β -D-glucoside, Quercetin 3-O- β -D-lactoside, 3' sialyllactosyl quercetin and 6' sialyllactosyl quercetin.

diphosphate sugar (NDP-sugar) and glycosyltransferases (GTs). For the methylation of quercetin O-methyl transferase (OMTs) use S-adenosylmethionine, releasing S-adenosylhomocysteine, and glycosyltransferases use NDP-sugar for the formation of quercetin glycosides (Scheme 8).

Moon-Hee Choi *et al.* (Choi *et al.*, 2021) used quercetin to synthesise a novel derivative called 3,7-dioleylequercetin by a modified method of Kato *et al.* Quercetin and oleyl bromide were used as starting materials, and generally the reaction takes place by S_N2 reaction. Simultaneously, reactions occur on the 3,7, and 4' positions of quercetin OH groups. The synthesis was performed by adjusting the equivalent ratio so that the oleyl moiety attaches to the 3, 7 OH groups. Once the reaction is complete, oleyl moieties are obtained (Scheme 9).

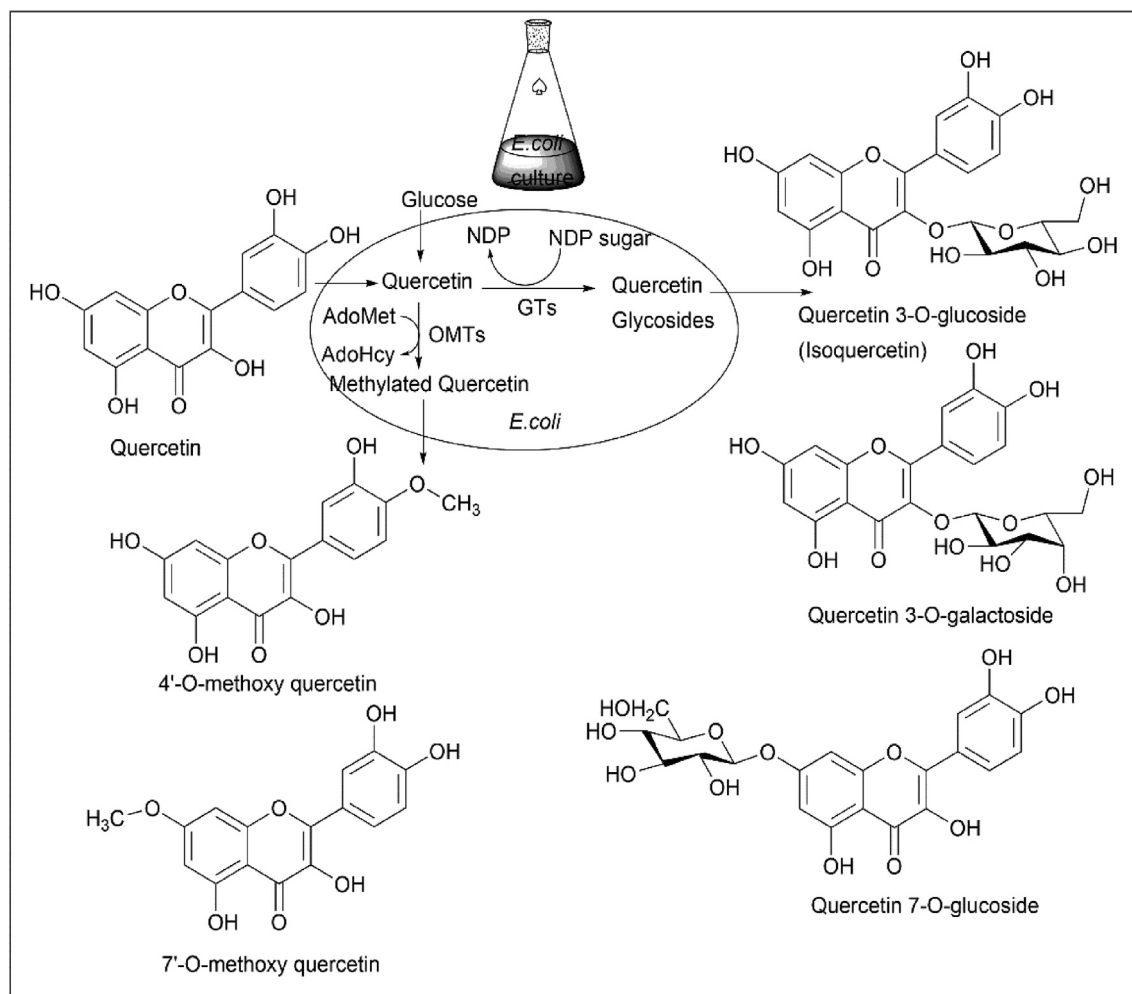
Mohammed Kajjout *et al.* (Kajjout and Rolando, 2011) protected and synthesised a derivative of quercetin. Quercetin was treated with dichlorodiphenylmethane at 180 °C for its protection (Scheme 10). The synthesis of 3-O- β -D-glucoside and 3-O- β -D-glucuronide is done by using protected quercetin as starting material and treating it with 2-(acetoxymethyl)-6-bromotetrahydro-2H-pyran-3,4,5-triyl triacetate in the presence of K_2CO_3 /DMF, which forms a complex structure by replacing the Br group of 2-(acetoxymethyl)-6-bromotetrahydro-2H-pyran-3,4,5-triyl triacetate, which is treated with ben-

zyl bromide, which acts as a protection for the OH group of the quercetin complex. The quercetin complex is treated with sodium methoxide and methanol to convert all O-acetyl groups to OH groups, which further react with H_2 /Pd/C and NaOCl for 3-O- β -D-glucoside and 3-O- β -D-glucuronide, respectively (Scheme 11).

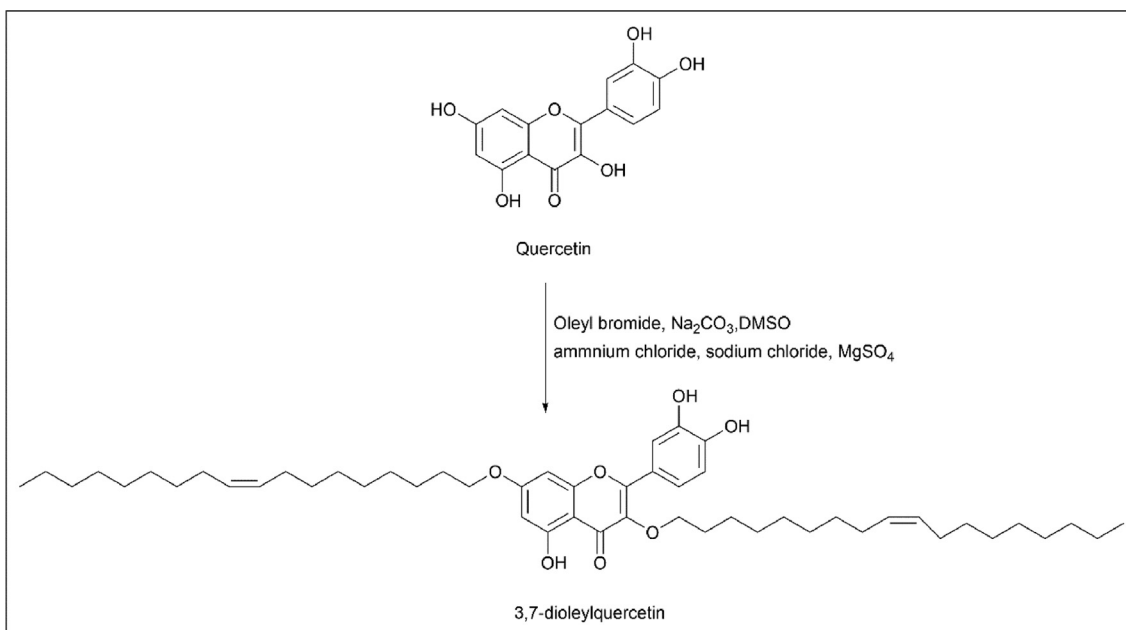
Ketan V. Hirpara *et al.* (Hirpara *et al.*, 2009) synthesised a fullerene derivative of quercetin. Quercetin was reacted with methyl iodide and K_2CO_3 to give tetra-methoxy quercetin and penta-methoxy quercetin, respectively. Tetra-methoxy quercetin was reacted with *p*-TsCl to give a tosylated product called tetramethylated monotosylated quercetin. Penta-methoxy quercetin was reacted with aluminium bromide so that one methoxy group was converted to an OH group by the demethylation process (Scheme 12).

Tetra-methoxy quercetin was treated with 3-iodo-1-propanol to give a propanol derivative of tetra-methoxy quercetin, further reacting with methyl malonyl chloride to give corresponding malonate esters, which on reacting with C_{60} by the Bingel reaction (cyclopropanation reaction) yield a fullerene derivative of quercetin (Scheme 13).

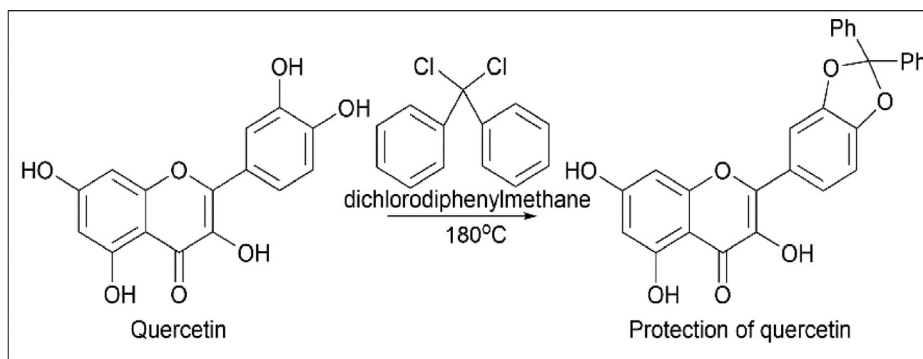
Komei Kato *et al.* (Kato *et al.*, 2016) synthesised isorhamnetin (3'-O-methylquercetin) and azaleatin (5-O-methylquercetin) using quercetin. Quercetin was treated with



Scheme 8 Synthesis of derivative of Quercetin using *E. coli*.



Scheme 9 Synthesis of 3,7-dioleylquercetin.



Scheme 10 Protection of Quercetin.

3 equivalents of benzyl bromide and 3 equivalents of K₂CO₃ to give the protected product 3,4',7-tri-O-benzylquercetin, which further reacted with methyl iodide, hydrogen, and palladium to give isorhamnetin (3'-O-methylquercetin) (Scheme 14).

Quercetin was treated with 4 equivalents of benzyl bromide and 4 equivalents of K₂CO₃ to give 3,3',4',7-tetra-O-benzyl quercetin, further reacted with methyl iodide, hydrogen, and palladium to give Azaleatin (5-O-methylquercetin) (Scheme 15).

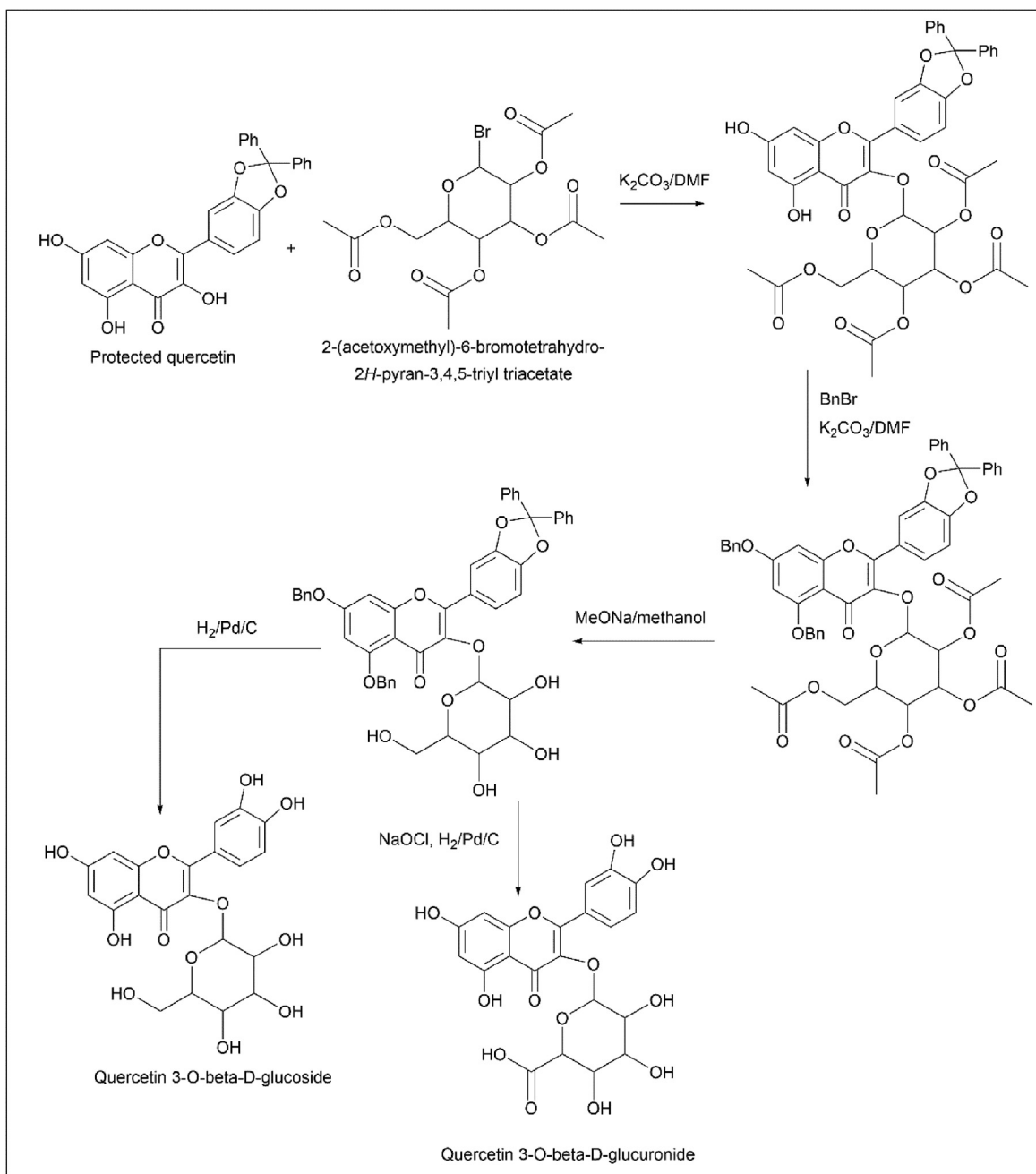
Evgeny V. Buravlev *et al.* (Buravlev *et al.*, 2018) synthesised the C-8 monoaminomethyl derivative of quercetin. Quercetin was reacted with amine in 1,4-dioxane or ethanol, aqueous formaldehyde was treated with hydrochloric acid, and the removal of solvated organic solvents was done. Heterocyclic secondary amine was used as an amine source in the reaction (Scheme 16).

Tomas C. Tempesti *et al.* (Tempesti *et al.*, 2012) synthesised quercetin-CF₃ using rutin as the starting material. 0.08 mmol of rutin and 0.04 mmol of 2-chloro-5-(trifluoromethyl)-1,3-dini-

trobenzene was dissolved in 5 ml of DMF and stirred with 100 mg of K₂CO₃ at 80 °C for about 15 h. The solution was extracted using ethyl acetate and water. Following separation of the organic and aqueous phases, the aqueous phase was washed three times with ethyl acetate. The organic phase was evaporated and a solid was obtained (Scheme 17).

Chatzikonstantinou *et al.* (Chatzikonstantinou *et al.*, 2020) synthesised peracetylated quercetin (PQ). 200 mg of quercetin, 1.25 ml of acetic anhydride, and 15 ml of pyridine were taken and refluxed in a nitrogen atmosphere. Once the reaction was complete, it was quenched with ice water and washed with ethyl acetate and diethyl ether. Yellow solid was obtained (Scheme 18).

Mishurov *et al.* (Mishurov *et al.*, 2016) synthesised tetraglycidyl quercetin. 2 g of quercetin, 1.0 ml of epichlorohydrin, and 0.50 g of N-pentadecyl-N-benzyl-N, N-dimethyl ammonium chloride were added to DMSO and heated for 30 min at 80 °C in an argon atmosphere. A solution of aqueous NaOH was added and heated for 4 h at 80 °C. Once a concentrated



Scheme 11 Synthesis of Quercetin 3-O- β -D-glucoside and Quercetin 3-O- β -D-glucuronide.

precipitate was obtained, it was called diglycidyl quercetin since quercetin and epichlorohydrin were taken in a 1:2 ratio.

To synthesise triglycidyl quercetin, quercetin, and epichlorohydrin, they were taken in a 1:3 ratio.

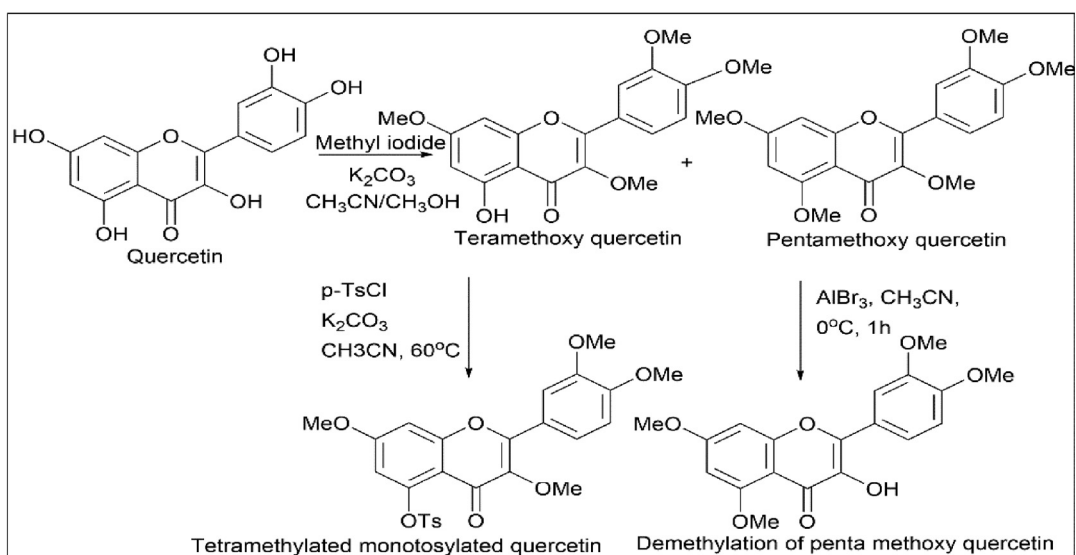
To synthesise tetraglycidyl quercetin, quercetin and epichlorohydrin were taken in a 1:4 ratio (Scheme 19).

Ayan Mukherjee *et al.* (Mukherjee *et al.*, 2019) studied the regioselectivity of quercetin. Quercetin was reacted with benzyl bromide and K_2CO_3 to give a protected form of quercetin, namely tribenzyl quercetin (60%) and tetrabenzyl quercetin (33%). Tribenzyl quercetin was reacted with methyl bromoacetate; K_2CO_3 results in a monomethyl ester substituted product at the C3' position; it was further reacted with Pd

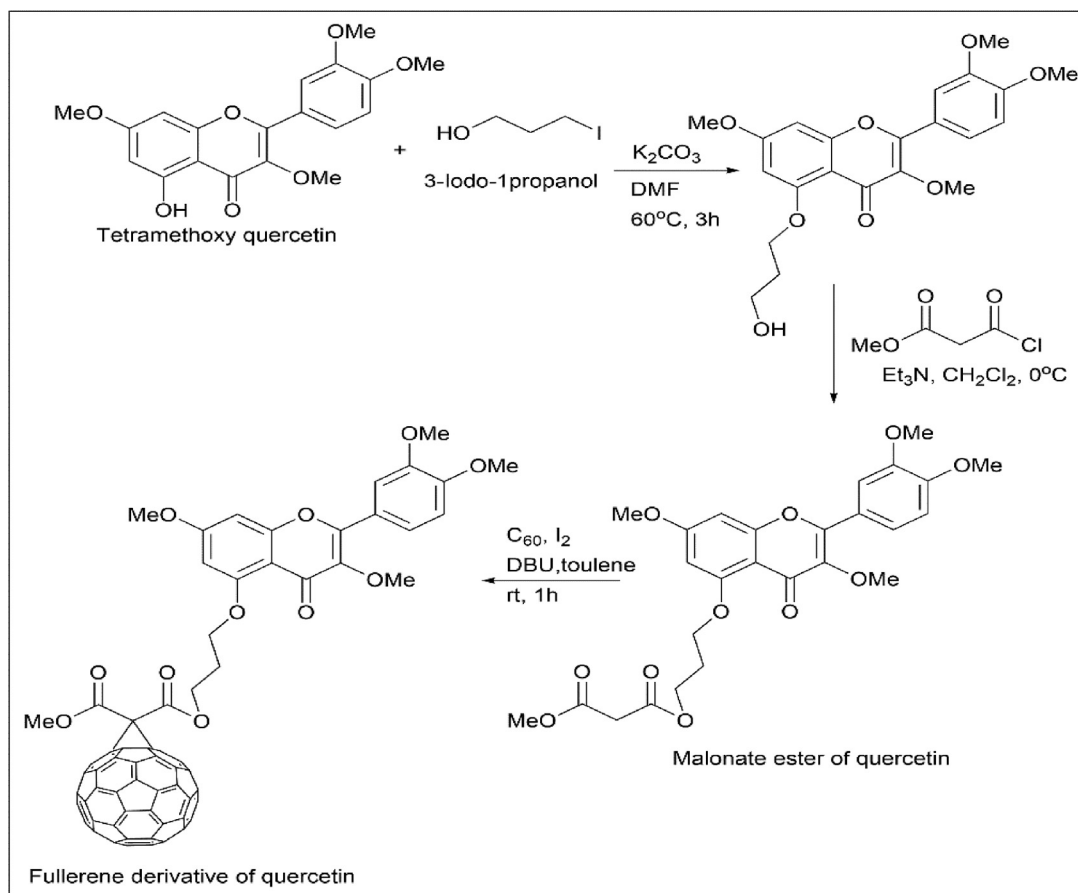
$(OH)_2/C$, THF for deprotection or debenzylation of quercetin by replacing the benzyl group with an OH group. 2 N NaOH and acetone are used to convert the methoxy group to the OH group.

Tribenzyl quercetin was treated with methyl bromo ester to give a dimethyl ester substituted product at the C3' and C5 positions, further reacting with $Pd(OH)_2/C$, THF for the deprotection or debenzylation of quercetin by replacing the benzyl group with an OH group. 2 N NaOH and acetone are used to convert the methoxy group to the OH group.

Similarly, tetrabenzyl quercetin was reacted with methyl bromoacetate, K_2CO_3 results in a monomethyl ester substituted product at the C5 position, and further reacted with



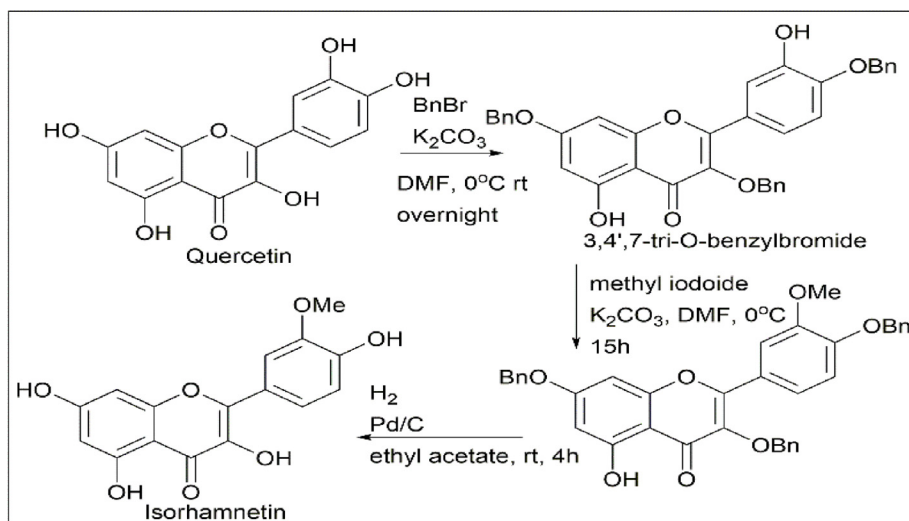
Scheme 12 Tosylation and demethylation of Quercetin.



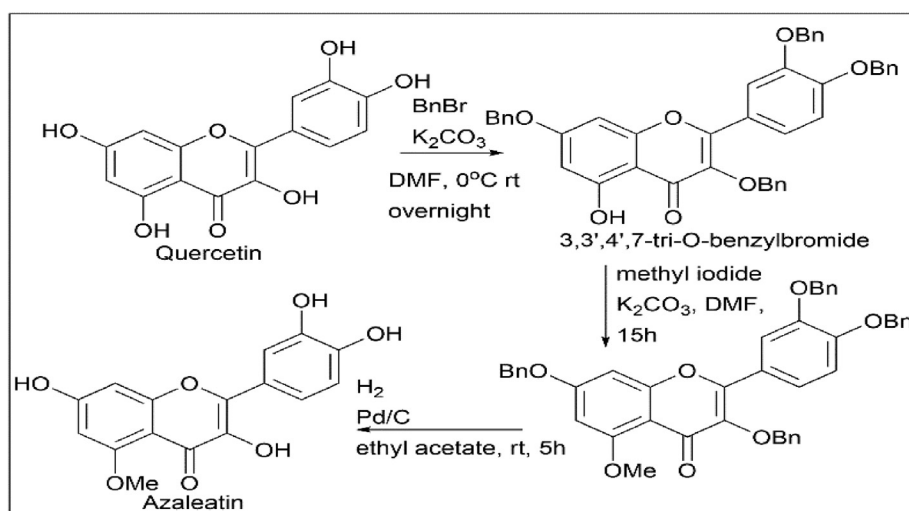
Scheme 13 Synthesis of fullerene derivative of Quercetin.

$Pd(OH)_2/C$, THF for the deprotection or debenzilation of quercetin by replacing the benzyl group with an OH group. 2 N NaOH and acetone to convert the methoxy group to the OH group (Scheme 20).

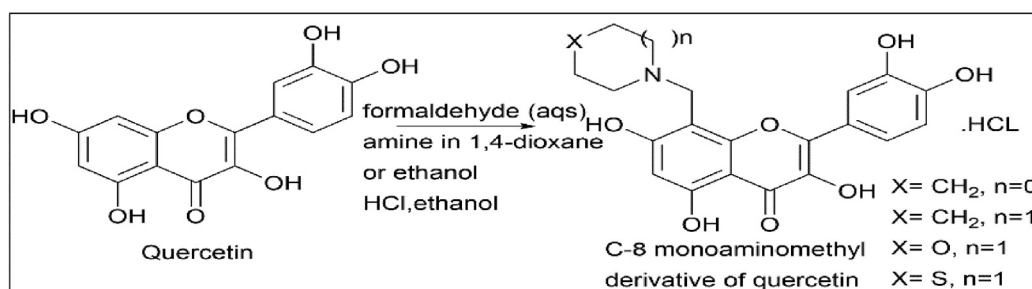
Isika *et al.* (Isika *et al.*, 2020) synthesised quercetin derivative called 2,2',2''-((2-(3,4-bis(2-amino-2-oxoethoxy)phenyl)-4-oxo-4H-chromene-3,5,7-triyl)tris(oxy))triacetamide. 2 mmol of quercetin were dissolved in 10 ml of DMF which forms a



Scheme 14 Synthesis of Isorhamnetin.



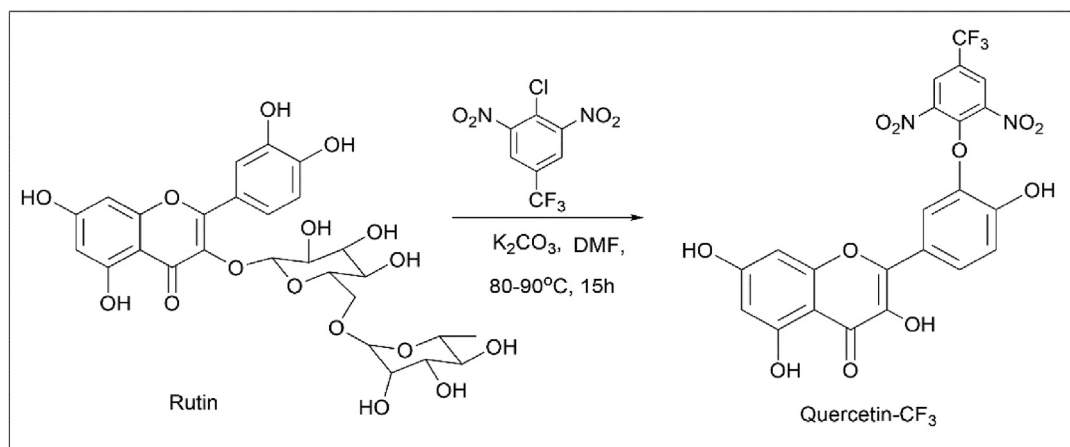
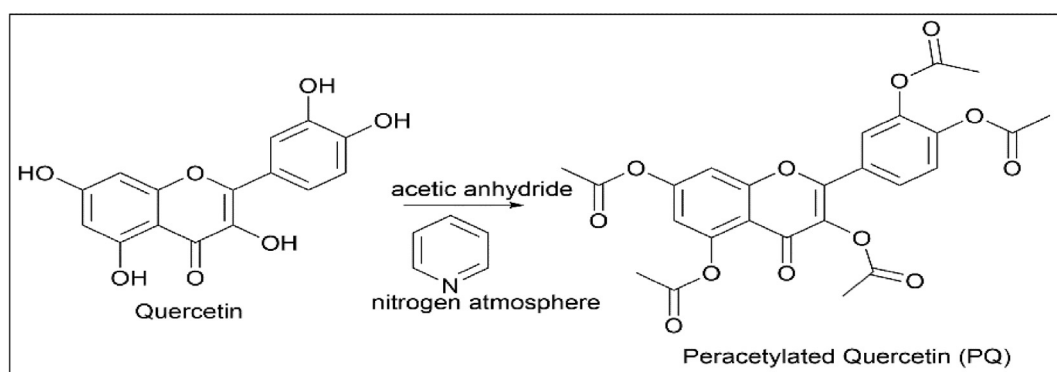
Scheme 15 Synthesis of Azaleatin.



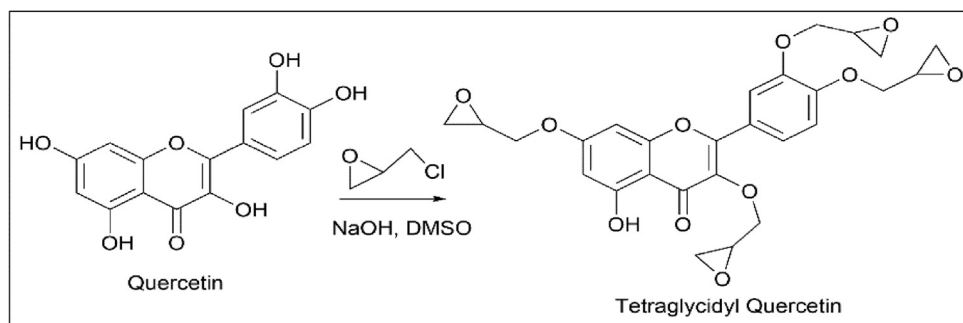
Scheme 16 Synthesis of C-8 monoaminomethyl derivative of Quercetin.

brown solution. 15 mmol of anhydrous K_2CO_3 was added and stirred at room temperature for about an hour. To this, 14 mmol of ethyl chloroacetate was added slowly and stirred at room temperature under nitrogen gas. A solid product was obtained; to this above-solid 1 mmol of THF/ H_2O (1:2) was added so that the solid will dissolve. Once dissolved,

10 mmol of $LiOH \cdot H_2O$ were added and stirred at room temperature to obtain the solid product. This above mentioned solid of about 0.7 mmol was mixed with thionyl chloride and refluxed to form an acyl chloride intermediate. This product was transferred into ice-cold ammonium hydroxide (aqueous) and stirred for two hours at cold condition and stirred at room

Scheme 17 Synthesis of Quercetin-CF₃.

Scheme 18 Synthesis of Peracetylated Quercetin.



Scheme 19 Synthesis of Tetraglycidyl Quercetin.

temperature for about 4 h to obtain 2,2',2''-((2-(3,4-bis(2-amino-2-oxoethoxy)phenyl)-4-oxo-4H-chromene-3,5,7-triyl)tris(oxy))triacetamide (Scheme 21).

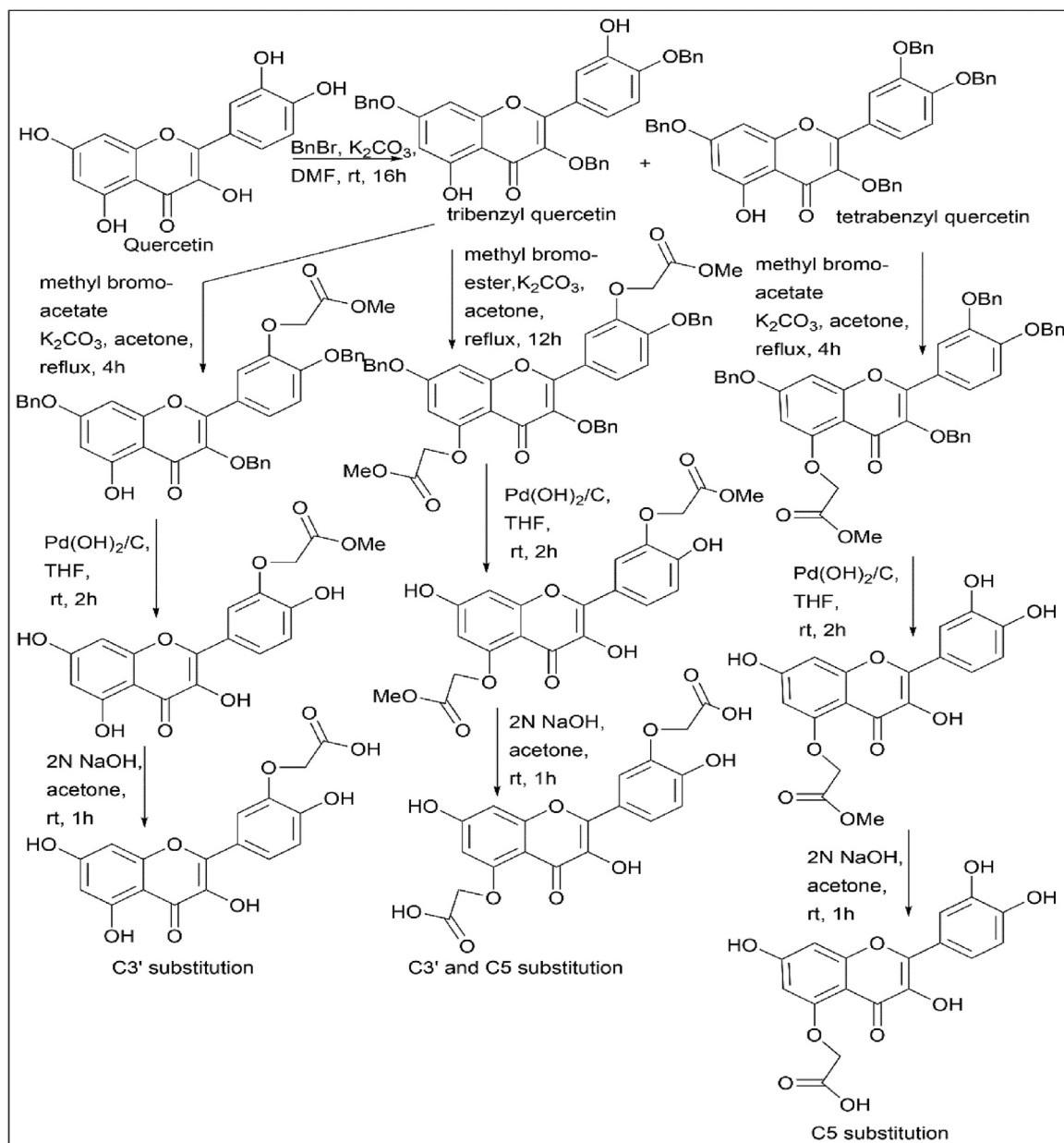
Zhong *et al.* (Zhong *et al.*, 2015) synthesised 7-O-substituted quercetin derivative. Peracetylation of quercetin using Ac₂O and pyridine further reacted with thiophenol and imidazole for regioselective deacetylation of the 7-O-Ac group at 0 °C, which gives 7-O-monodeprotected flavonol. Using substituted benzyl bromides, the 7th position was alkylated. Deprotection (deacetylation) is done using methanolic ammonia, which results in 7-O-arylmethylquercetin (Scheme 22).

E. Woznicka *et al.* (Woznicka *et al.*, 2015) synthesised quercetin-8-sulfonic acid and its sodium salt, quercetin-8-

sulfonic acid. 2 g of quercetin was taken in a round bottom flask and 8 cm³ of concentrated sulfuric acid was added. The reaction mixture was stirred at 18–20 °C for 2 h. Once the reaction is complete, ice-cold water is added to the reaction mixture, and an orange precipitate is obtained, filtered, recrystallized, and dried at room temperature.

The sodium salt of quercetin-8-sulfonic acid was obtained by neutralising the post sulfonic mixture with 20% NaOH (pH 4). The yellow precipitate was obtained, filtered, recrystallized, and dried at room temperature (Scheme 23).

Mahendra Thapa *et al.* (Thapa *et al.*, 2012) synthesised 3-aminopropyl quercetin. Quercetin was reacted with sodium carbonate and benzyl chloride in DMF to form tribenzyl quer-



Scheme 20 Regioselectivity of Quercetin derivative.

etin. Tribenzyl quercetin was treated with sodium hydride and *N*-(3-iodopropyl) phthalimide and further reacted with hydrazine, followed by hydrogen and palladium, to give 3-aminopropyl quercetin (Scheme 24).

3.3. Synthesis of quercetin metal complex

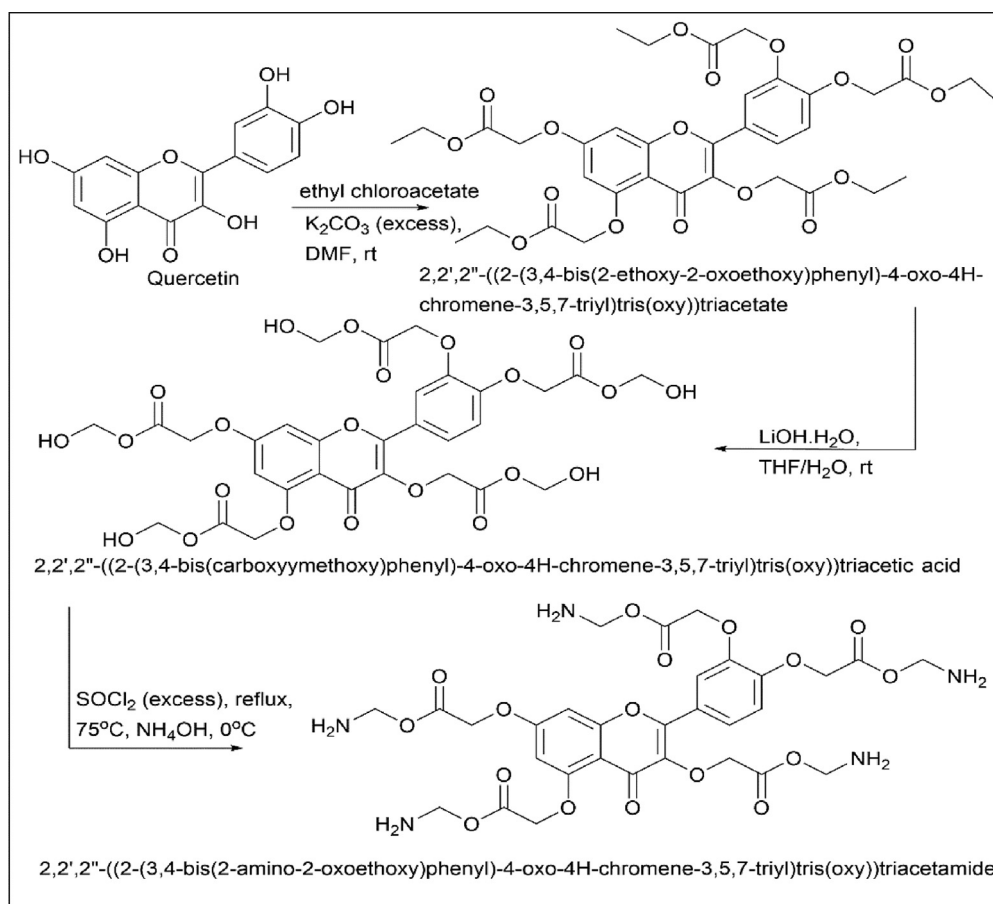
Zahra Moodi *et al.* (Moodi *et al.*, 2021) synthesised a metal complex of quercetin. Here, copper (II) has been used as a metal for complex synthesis. In this method, first a Schiff base was synthesised with quercetin and ethanolamine and the corresponding metal complex was synthesised.

The synthesis of Schiff base involves an equimolar amount of quercetin and ethanolamine. After dissolving quercetin in 7 ml of ethanol, a few drops of glacial acetic acid were added

dropwise to this ethanol amine after 30 min and refluxed at $60\text{ }^\circ\text{C}$ for 8 h with stirring. Orange crystalline precipitate was obtained (Scheme 25).

The synthesis of a metal complex with copper as the metal was achieved. 0.5 mmol of ligand was dissolved in 1 ml of 10% NaOH and in 20 ml of deionized water and stirred for 10 min at room temperature. Once an orange solution is obtained, 0.5 mmol of $\text{Cu}(\text{OAc})_2$ dissolved in 10 ml of deionized water is added dropwise under ultrasonic irradiation at room temperature and sonicated for 30 min, centrifuged and dried in vacuum at $80\text{ }^\circ\text{C}$ for 6 h (Scheme 26).

Wildson Max Barbosa da Silva *et al.* (da Silva *et al.*, 2020) worked on a metal complex of quercetin with $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$, $\text{Cu}(\text{CH}_3\text{COO})_2$, $\text{Zn}(\text{CH}_2\text{COO})_2$, NiCl_2 , and CoCl_2 with quercetin as a ligand. Quercetin was dissolved in 10 ml of methanol and



Scheme 21 Synthesis of 2,2',2''-((2-(3,4-bis(2-amino-2-oxoethoxy)phenyl)-4-oxo-4H-chromene-3,5,7-triyl)tris(oxy))triacetamide.

metal salt in 10 ml of distilled water with the stoichiometric ratio of 2:1 and stirred at room temperature for 20 min. Three drops of triethylamine were added to the reaction mixture, causing an immediate colour change, and the mixture was stirred for three hours. The reaction mixture was left at a low temperature for about two days so that precipitation was complete, filtered, and stored under vacuum in a desiccator (Scheme 27).

Alina Bravo *et al.* (Bravo and Anacona, 2001; Kalinowska *et al.*, 2021) also followed a similar method for synthesising quercetin metal complex using 2 mmol quercetin and 1 mmol metal like CoCl₂·6H₂O, MnCl₂·4H₂O, CdCl₂, or HgCl₂ in 40 ml ethanol, which was refluxed for about 10 h. It was left overnight, once a solid is formed, it was dried (Scheme 28).

Kirankumar Shastrala *et al.* (Shastrala *et al.*, 2021) synthesised in a similar manner the complexes of Zn(II), Ni(II), Co(III), Cu(II), etc prepared using respective metal chlorides. However, Mo(IV) prepared using ammonium salt, and VO(IV) was prepared using Vanadyl sulphate. Quercetin and metal were taken in a (2:1) ratio. A methanolic solution of quercetin and a methanolic or aqueous solution of metal salt were used for the reaction. The pH was adjusted to 10 by adding NaOH solution (Scheme 29).

Phakorn Papan *et al.* (Papan *et al.*, 2020) worked on the iron(III)-quercetin complex. The synthesis method involves using quercetin hydrate, of which 0.0050 mol was added to 500 ml of methanol and allowed to stir until it dissolved.

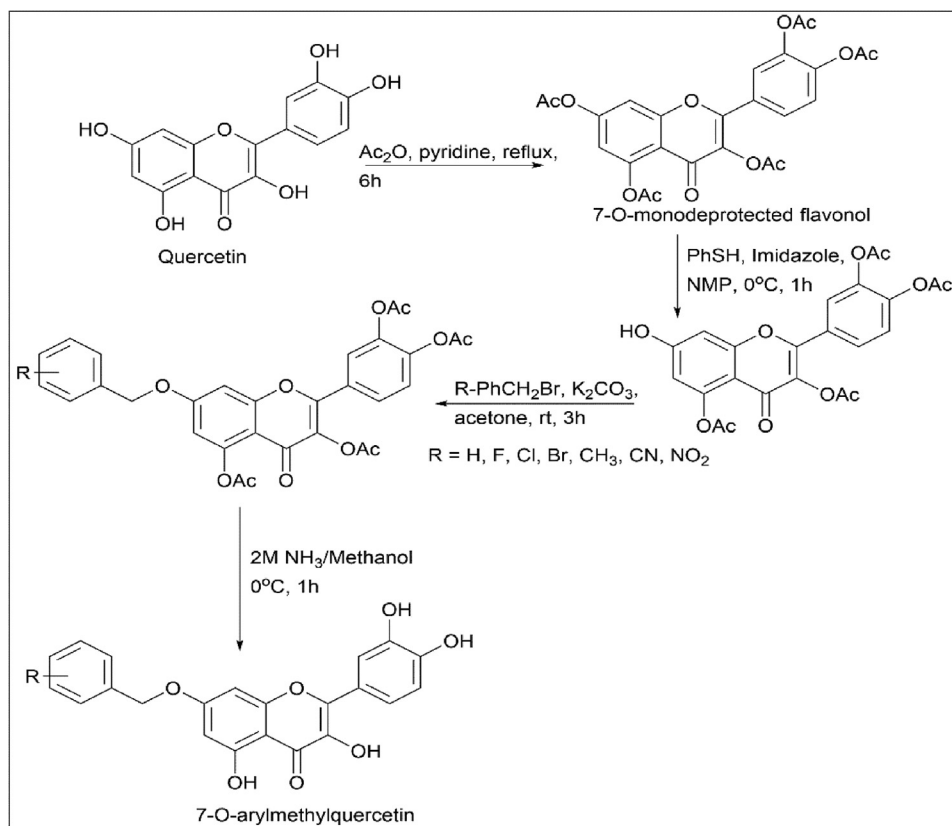
The colour of the solution is yellow. The pH of the solution was adjusted to 12 using 50% NaOH to convert it into deprotonated form (Scheme 30). Iron(III) chloride, 0.0025 mol in 500 ml ultrapure water, and added to the quercetin solution. The reaction was incubated for 2 h at 60 °C. The dark yellow product was collected and stored for future analysis (Scheme 31).

Birjess Bukhari *et al.* (Bukhari *et al.*, 2009) took a different approach by using a 1:2 ratio of quercetin and a metal like copper to form Cu-quercetin. Synthesis involves using a two-necked, round-bottom flask. 0.01 mol of quercetin was added to 20 ml of methanol, and once it had been dissolved, solid 0.02 mol of CuSO₄·5H₂O was added and stirred for 15 min at room temperature. A colour change was noticed from yellow to brownish yellow. Once the reaction is complete, the product is washed with t-butanol and dried. The yield of Cu-quercetin complex was about 77% (Scheme 32).

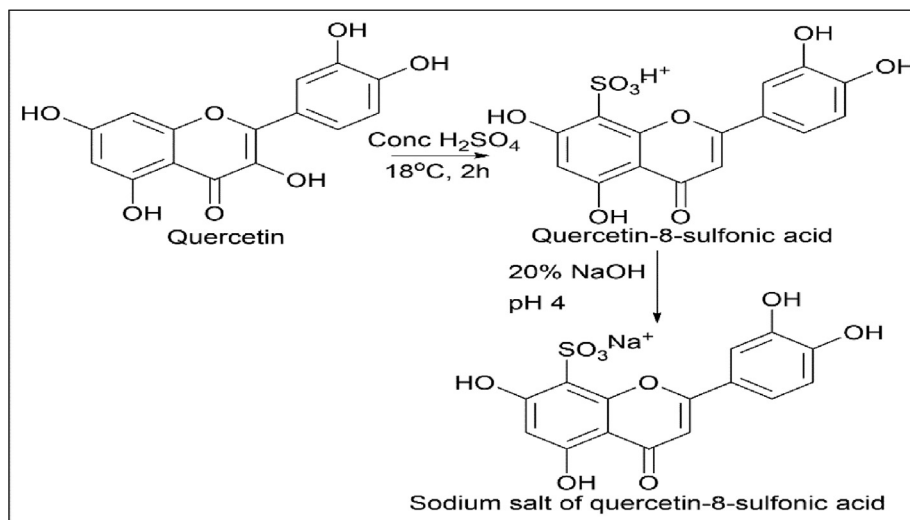
Sartaj Tabassum *et al.* (Tabassum *et al.*, 2013) worked on heterobimetallic complexes. Synthesis of monometallic complexes [Cu(Que)₂(H₂O)₂] and [Zn(Que)₂(H₂O)₂] and further heterobimetallic complexes were formed using Sn₂Cl₄.

3.3.1. Synthesis of monometallic complexes [Cu(Que)₂(H₂O)₂] and [Zn(Que)₂(H₂O)₂]

1 mmol of Cu(NO₃)₂·6H₂O / Zn(NO₃)₂ was used as a metal source, and 2 mmol quercetin was refluxed to give a monometallic complex.



Scheme 22 Synthesis of 7-O-substituted Quercetin derivative.

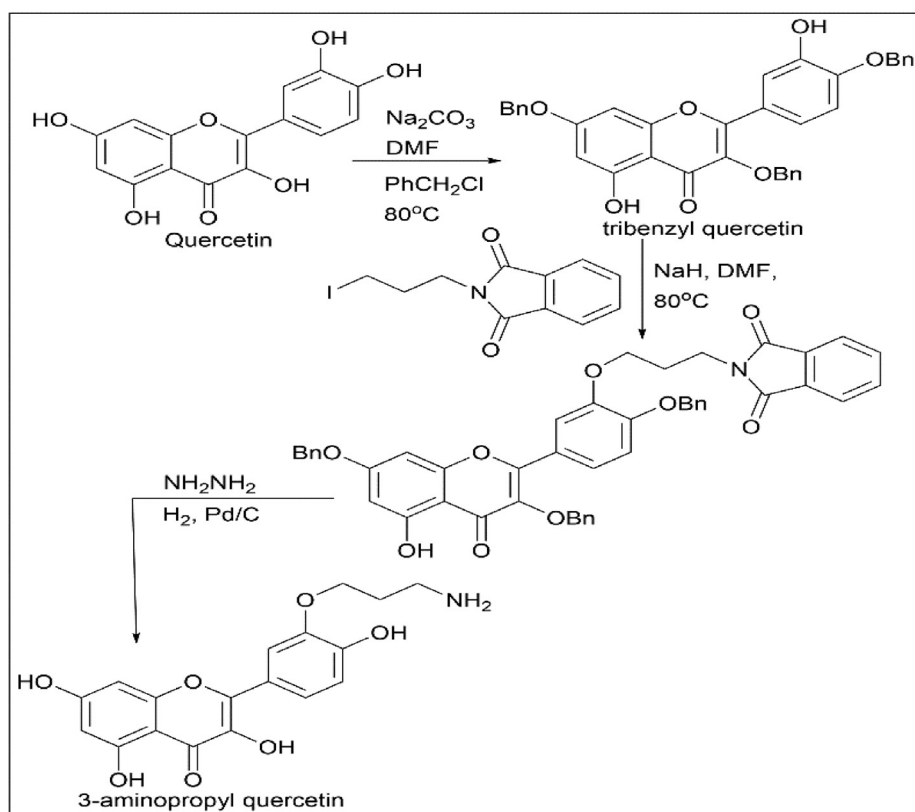


Scheme 23 Synthesis of Quercetin-8-sulfonic acid and its Sodium salt of quercetin-8-sulfonic acid.

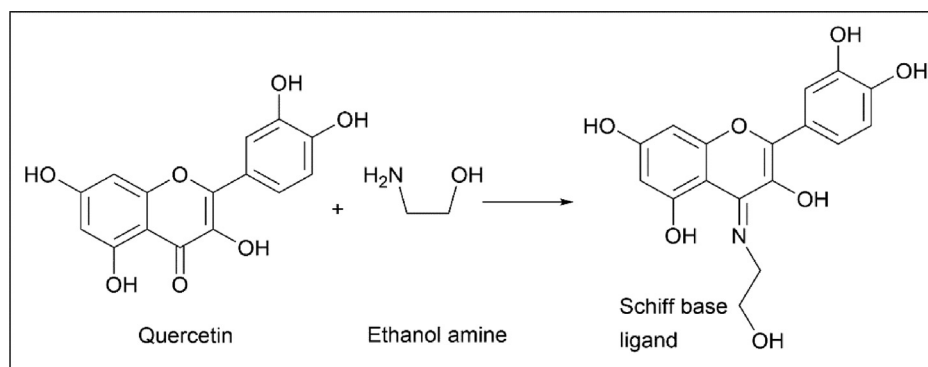
3.3.2. Synthesis of a heterobimetallic complex [$Cu(Que)_2(H_2O)_2-Sn_2Cl_4$]

1 mmol of $Cu(NO_3)_2 \cdot 6H_2O$ was dissolved in methanol, and to this 2 mmol of Sn_2Cl_4 was added dropwise and refluxed for about 6 h. The resultant brown color precipitate was obtained by adding ether to the concentrated solution. The compound was washed with ether and dried (Scheme 33). A similar procedure was carried out for $[Zn(Que)_2(H_2O)_2]$.

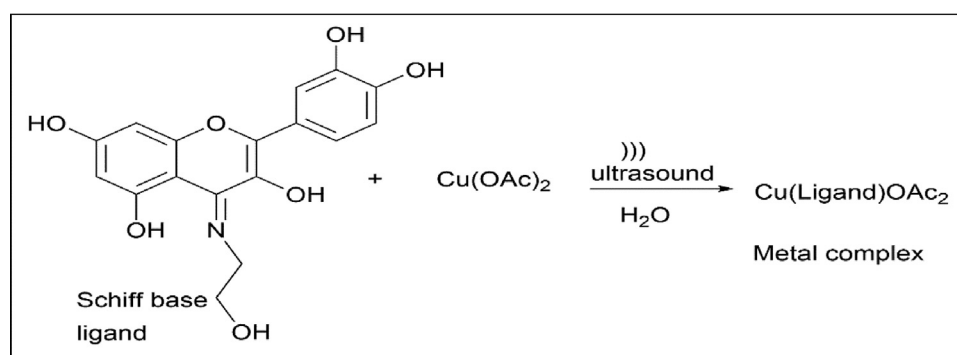
Ravichandran *et al.* (Ravichandran *et al.*, 2014) synthesized a quercetin-cadmium metal complex. 0.01 mol/L of quercetin was dissolved in 20 ml of ethanol in a 50 ml round-bottom flask. To this yellow-colored solution, 0.02 mol/L of cadmium nitrate was added quickly. After about 2 h of stirring at room temperature, the yellow colour changes to dark green, indicating the formation of a metal complex. The reaction mixture was filtered and evaporated, and a solid product was obtained. Unreacted material was washed with water (Scheme 34).



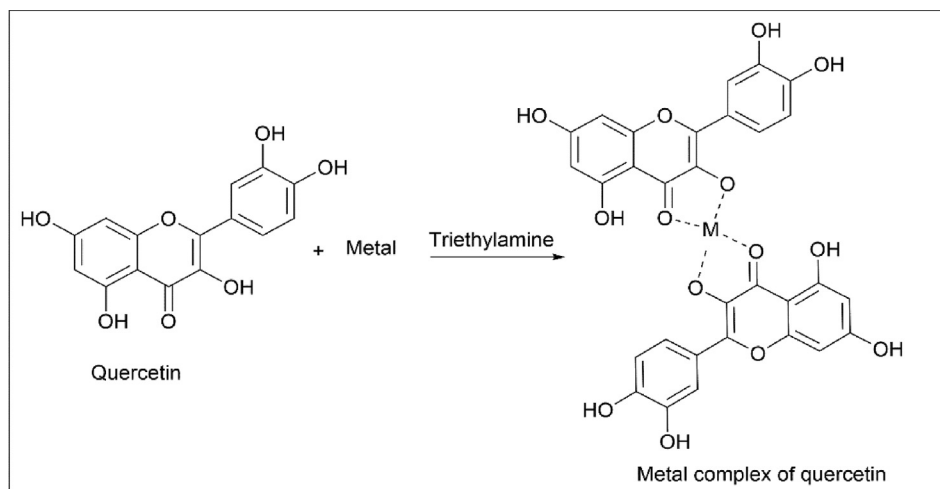
Scheme 24 Synthesis of 3-aminopropyl quercetin.



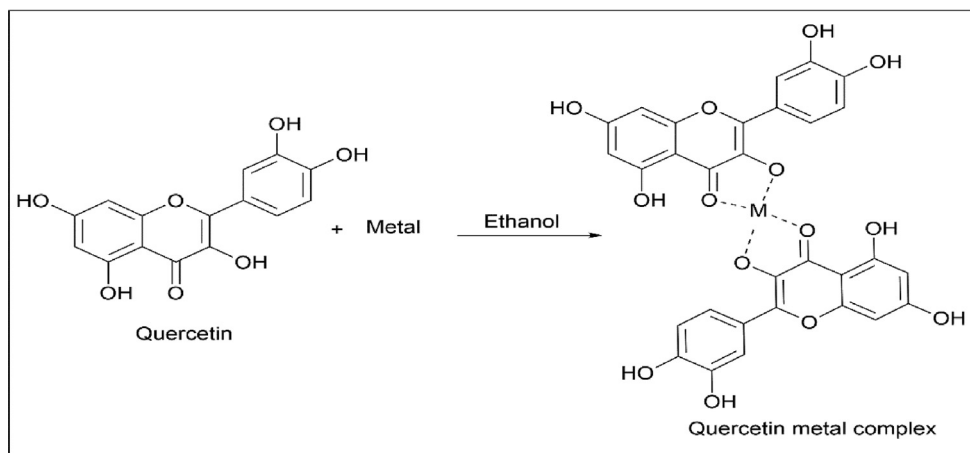
Scheme 25 Synthesis of Schiff base ligand from Quercetin.



Scheme 26 Synthesis of metal complex of Quercetin by Schiff base ligand.



Scheme 27 Synthesis of Quercetin metal complex with Triethylamine.



Scheme 28 Synthesis of Quercetin metal complex with ethanol as solvent.

4. Applications

4.1. Anti-bacterial, anti-fungal, and anti-viral

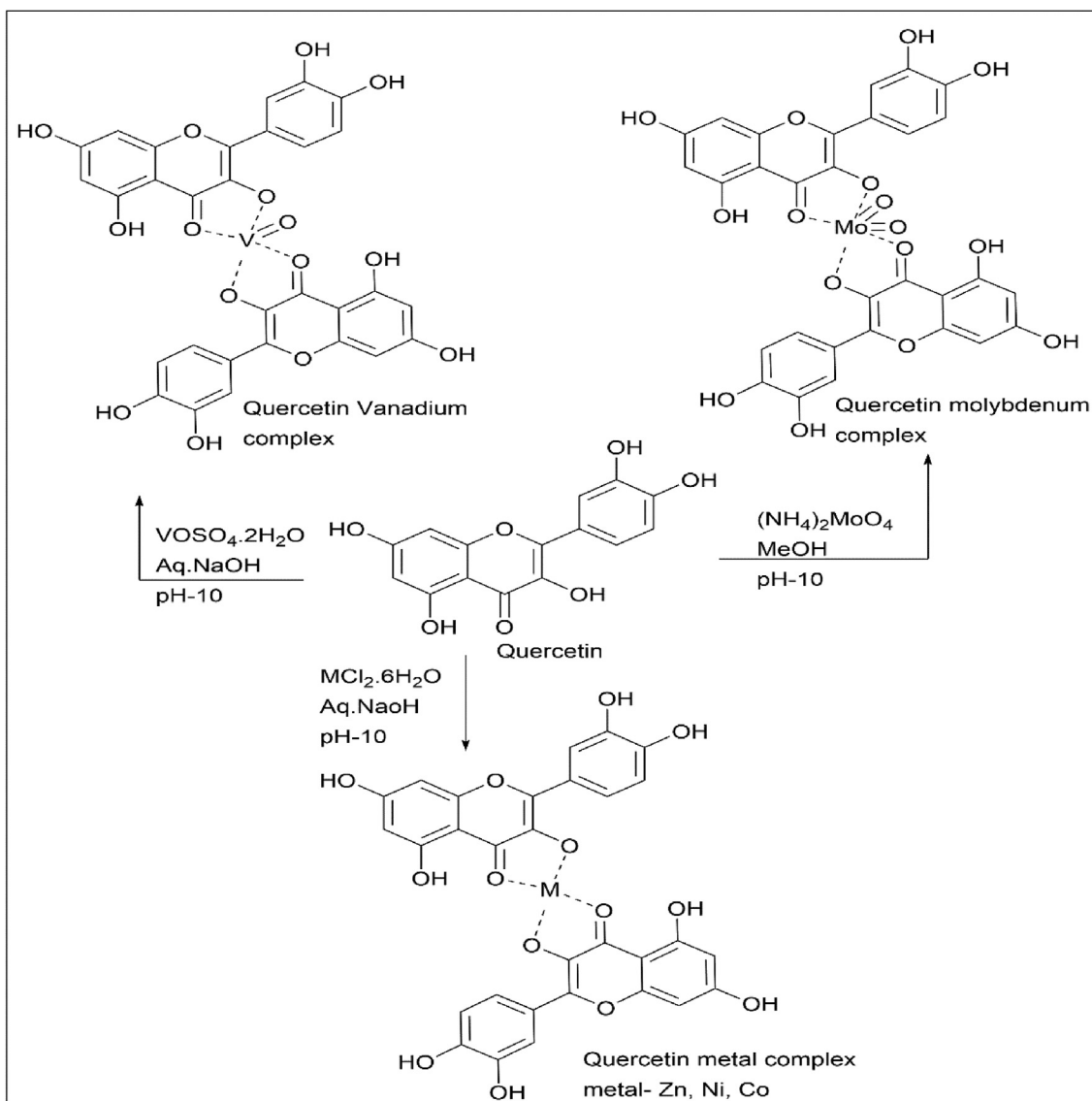
Quercetin has anti-bacterial activity against a wide range of gram-positive and gram-negative bacteria (Anand David et al., 2016a). Some gram-negative bacteria are more resistant to quercetin than gram-positive bacteria, but certain derivatives of quercetin have potent antibacterial agents against both gram-positive and negative strains of bacteria (Nguyen and Bhattacharya, 2022; S. Wang et al., 2018a). Quercetin reacts with the bacterial cell membrane, and the reaction is determined by quercetin's hydroxyl groups (Roy et al., 2022). The activity of the drug against bacteria is achieved by growing them in an agar medium with quercetin 25 $\mu\text{mol/ml}$ dissolved in DMSO and diluted to 6×10^{-3} $\mu\text{mol/ml}$ using the double dilution method. Culture media were spread by bacterial suspension (100 μl) and 100 μl of drug in drug discs. The zone of inhibition is measured after 24 h of incubation at 37 $^{\circ}\text{C}$ (Wang et al., 2018b). Minimum inhibitory concentration (MIC) will vary for different bacterial strains; for example,

Streptococcus sanguis, *Streptococcus mutans*, *Streptococcus sobrinus*, etc., have MIC values ranging from 1 to 8 mg/mL (Yi Shu, 2011) *Staphylococcus aureus* has a MIC of 20 $\mu\text{g/ml}$ (Jaisinghani, 2017). Depending on the strain of bacteria, the MIC value will change accordingly. Some bacteria are responsible for food spoiling, but when bacteria were treated with quercetin, they showed MIC, which results in killing bacteria (Montone et al., 2021).

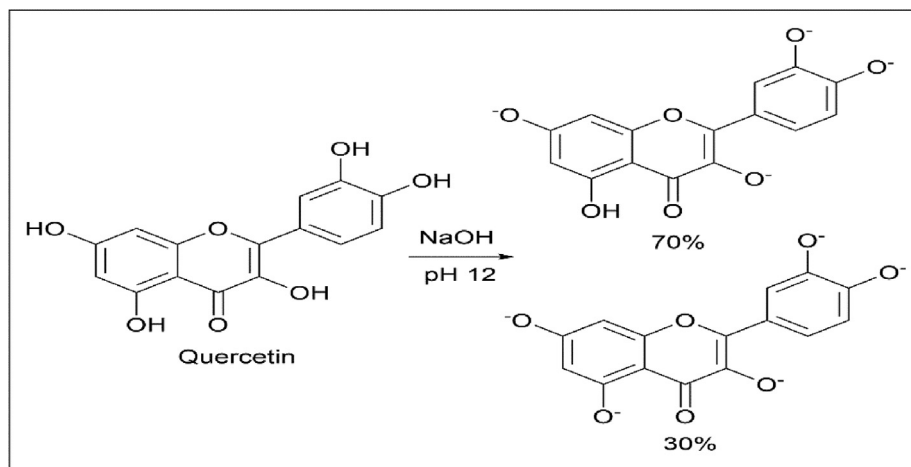
The proposed mechanism of quercetin inhibiting bacterial activities involves three main mechanisms.

- I. The perforation of quercetin damages the bacterial cytoplasmic membrane
- II. Blocking the synthesis of nucleic acids and inhibiting energy metabolism
- III. Targeting gyrase in bacteria (Ahmad et al., 2015; Osonga et al., 2019; Plaper et al., 2003).

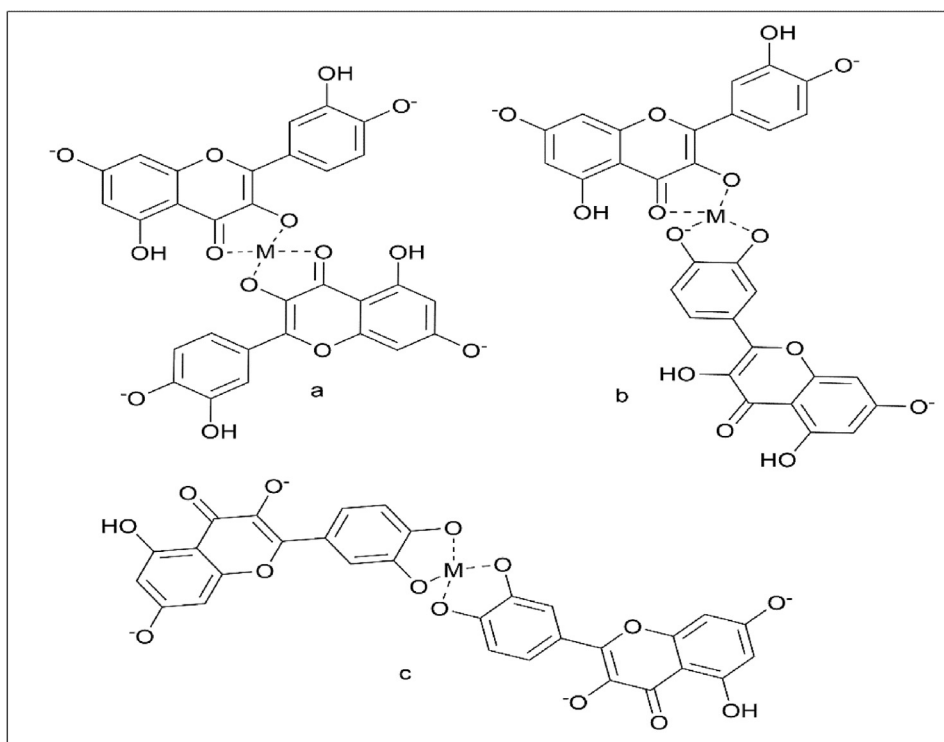
Quercetin is well known for its anti-bacterial and anti-viral activities, but its anti-fungal activities are not well documented. However, quercetin has anti-fungal activity against



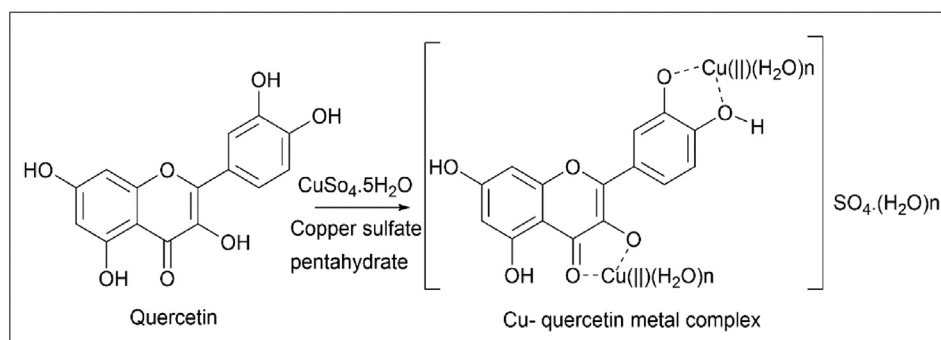
Scheme 29 Synthesis of Quercetin vanadium, molybdenum, and other metal complex.



Scheme 30 Synthesis of Deprotonated form of Quercetin.



Scheme 31 Synthesis of Iron quercetin metal complex with different attachment.



Scheme 32 Synthesis of Cu-quercetin metal complex.

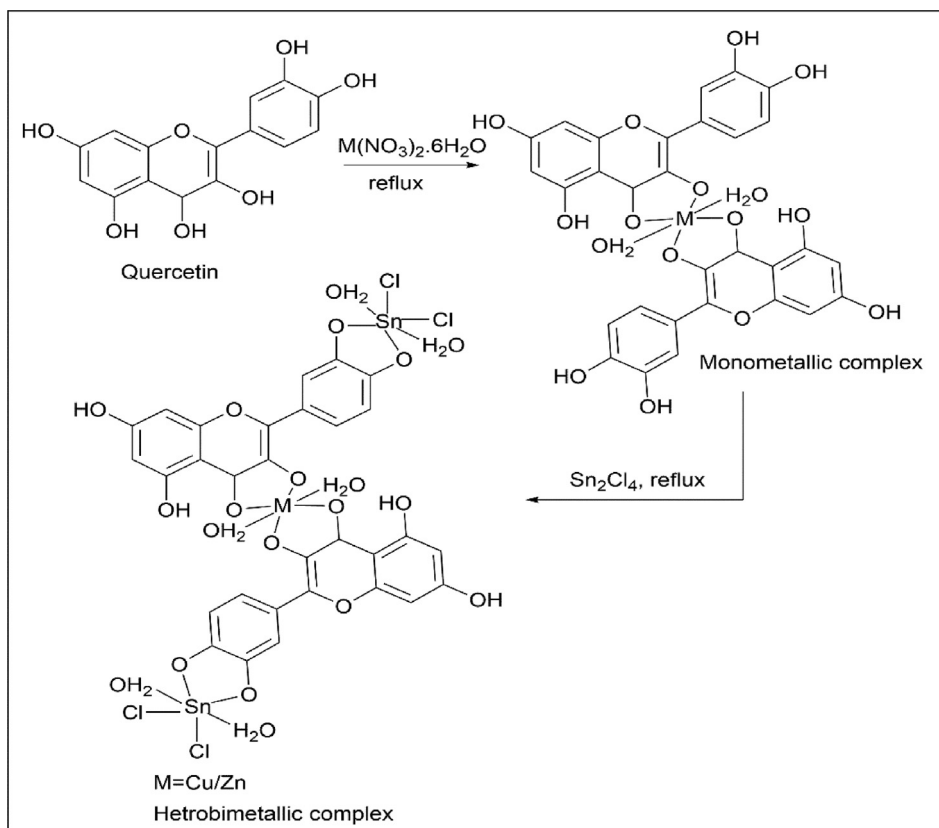
Aspergillus niger, *Aspergillus fumigatus*, etc (Batiha et al., 2020; Yin et al., 2021). Quercetin shows anti-fungal activity to manage clinical *Candida albicans* biofilms and sensitise fluconazole-resistant *Candida albicans* isolated to fluconazole (Gao et al., 2016; Shahzad et al., 2014; Singh et al., 2015).

Quercetin was studied for anti-viral activities since it has a promising effect in inhibiting protease, reverse transcriptase, and polymerase and binding to viral capsid proteins (Agrawal et al., 2020; Bachmetov et al., 2012; Shinozuka et al., 1988; Spedding et al., 1989). A certain derivative of quercetin, namely quercetin 7-rhamnoside, acts against the porcine epidemic diarrhoea virus (Choi et al., 2009). Porcine epidemic diarrhoea virus causes acute diarrhoea, dehydration, vomiting, etc (Jung et al., 2020). Quercetin inhibits the activity of the Flaviviridae family, which consists of enveloped positive-strand RNA viruses. *Pestivirus*, *Hepacivirus*, etc genera come under this category, Hepatitis C belongs to the *Hepacivirus* genus (di Petrillo et al., 2022; Roudot-Thoraval, 2021). There is

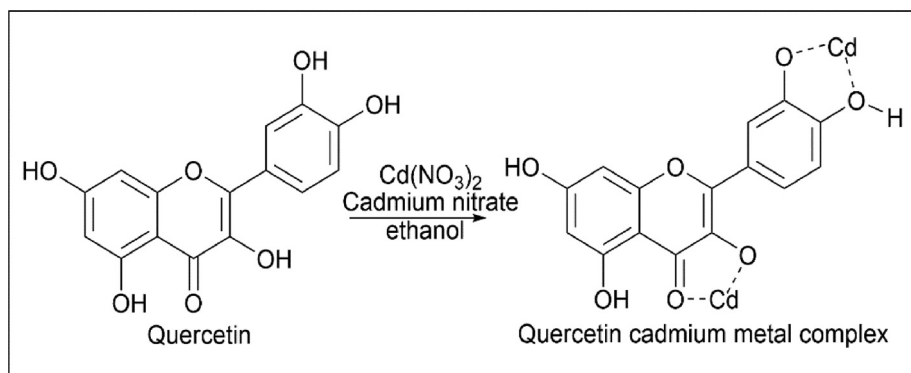
much literature supporting the antiviral properties of quercetin, both *in vitro* and *in vivo*. Quercetin has dose-dependent antiviral activity against herpes viruses (HSV-1 and HSV-2) in cell culture (Lee et al., 2017b; Lyu et al., 2005). They can inhibit respiratory viruses that are present in cultured cells, which show a protective effect on the lung (de Palma et al., 2008; Kumar et al., 2003). They also show therapeutic effects at an early stage of COVID (Kalil et al., 2021; Khan et al., 2022).

4.2. Anti-inflammatory

Quercetin was reported to have anti-inflammatory activities (Morikawa et al., 2003; Oršolić et al., 2004). Researchers have worked both *in vitro* and *in vivo* to find how efficiently quercetin works on inflammation (Kempuraj et al., 2005). Quercetin possesses both gastrointestinal cytoprotective and mast cell



Scheme 33 Synthesis of Heterobimetallic complex.



Scheme 34 Synthesis of Quercetin cadmium metal complex.

stabilising activities (B. Penissi *et al.*, 2003). It has immunosuppressive activity on dendritic cells (Huang *et al.*, 2010). They mainly inhibit AA metabolism and the production of eicosanoids (potent inflammatory mediators). Further, the effect of quercetin was examined on the COX-2 AA pathway, but no specific data were available for its potential to inhibit the COX-1 AA pathway (Lee *et al.*, 2017c; Lesjak *et al.*, 2018).

Marijia *et al.* (Azeem *et al.*, 2022) worked on derivatives of quercetin; they are quercetin-3-O-glucuronide, isorhamnetin, isorhamnetin-3-O-glucoside, tamarexitin, etc. In the inflammatory response on COX-1, results showed that anti-inflammatory activity was seen more by tamarexitin. In recent

times, COVID-19 has created a great impact in every country considering that SARS-CoV-2 is a beta coronavirus that causes severe inflammatory pneumonia. As is known, inflammation causes damage to our bodies and is even responsible for the deaths of a few people. Quercetin is a carbohydrate-free flavonoid, and this gives it a great ability to reduce inflammation (Saeedi-Boroujeni and Mahmoudian-Sani, 2021; Zhou *et al.*, 2020).

Despite the evidence that quercetin shows anti-inflammation activities, there is no proper mechanism that can explain the effect of quercetin's activity against inflammation (Chen *et al.*, 2016).

Studies have revealed that polyphenolic compounds like quercetin, curcumin, etc. can enhance the phosphorylation of AMP-activated protein kinase (AMPK) (Wang et al., 2014). Quercetin suppresses inflammation by modulating AMP-activated protein kinase (Hung et al., 2015; Qiu et al., 2018; Sato and Mukai, 2020). To study the effect of inflammation *in vivo*, studies were done in mice expressing the human CRP gene. The term effect of quercetin had a positive effect on inflammation, but in the case of *in vitro* short-term quercetin is enough to act on inflammation (Bischoff, 2008; Boots et al., 2008b; Ghosh, 1999; Kleemann et al., 2011; Rogerio et al., 2010; Zadelaar et al., 2007). Recent studies revealed that quercetin suppressed proinflammatory cytokines through attenuation of NF-kappa B and p38 expression in human mast cells (Comalada et al., 2005; Karuppagounder et al., 2016; Min et al., 2007; Nair et al., 2006). In addition to quercetin, iso-quercetin is also an effective inflammation suppressor, suggesting a potential for treating allergies (Rogerio et al., 2007). In the case of quercetin derived from plants, it can affect inflammatory mediators, and they produce secondary messengers that lead to pro-inflammatory molecule expression (Calixto et al., 2004, 2003; Cho et al., 2003; Rather and Bhagat, 2020).

4.3. Anti-diabetic

Diabetes mellitus has become a major health issue all over the world. Diabetic disease leads to an increase in blood glucose levels because the body is unable to metabolize the sugar. It is a metabolic disease characterized by chronic hyperglycemia, which results in defects in insulin secretion (Craig et al., 2009; Galtier, 2010; Kharroubi, 2015). They are caused by genetics, changes in lifestyle, etc. They are further classified as type I and type II diabetics (Canivell and Gomis, 2014; M.D., 2020; Wilkin, 2009). In type I, T cells destroy β cells of Islets of Langerhans; this leads to imbalance and disturbs equilibrium, as they are insulin-dependent (Daneman, 2006; Devendra et al., 2004; Ferrannini et al., 2010; Kilic et al., 2014). Blood glucose increases since the insulin level is low. A person suffers from polyurea, polydipsia, polyphagia, etc (Dabelea et al., 2014; DiMeglio et al., 2018).

In the case of type II they are insulin independent (Kahn, 1994); they have loss of insulin receptor function, that is, the pancreas has insulin production but becomes insensitive once it binds to the insulin receptor (Galicia-Garcia et al., 2020; Roden and Shulman, 2019; Weyer et al., 1999). As a result of the increased cell number and size, more insulin is produced. When β cell number is increasing, it is referred to as hyperplasia, and when it increases in size, it is called hypertrophy (Muir et al., 2016; Schuster, 2010). It produces the protein Amylin, which uses all the energy and nutrients. Eventually, a reverse takes place from hypertrophy to hypotrophy, that is, size decreases, β cells start dying, and insulin production decreases (Druet et al., 2006; Robertson, 1995). This can be caused by genetics as well as acquired factors.

In recent times, researchers have found that quercetin has anti-diabetic activity. When quercetin was administered over a longer duration (14–28 days and ≥ 28 days) of intervention, it had a significant effect on serum glucose levels with a shorter duration (Bule et al., 2019). When quercetin was tested on plasma glucose concentration, control rats, i.e., non-diabetic rats, had no effect, but there was a significant reduction in

the blood glucose level of diabetic rats after 8–10 days at the two doses used (Table 4) (Haddad and Eid, 2017; Vessal et al., 2003). To improve quercetin activity, scientists tried using polymer-based carriers like chitosan, liposomes, PLA, PLGA, polyphosphoesters, etc. They are non-toxic, easily synthesized, and have other advantages that help quercetin reach the target and perform the specific function (Mukhopadhyay et al., 2012; Mukhopadhyay and Prajapati, 2015; Pinto, 2010; Xu et al., 2007a; Xu et al., 2007b). Extensive preclinical and clinical studies should be carried out to make a prominent anti-diabetic drug, and it should be approved by the FDA (Yao et al., 2019; Yi et al., 2021).

4.4. Cardiovascular disease prevention

Flavonoids can deliver anticipated cardioprotective action in cardiovascular diseases and ageing (Egert et al., 2009; Peluso, 2006). Recent studies showed that quercetin plays an important role in cardiovascular protection (Deng et al., 2020). In 1993, an investigation that was performed on humans showed a positive correlation between dietary intake of flavonoids, including quercetin, and a lower risk of cardiovascular disease in both men and women (Hertog et al., 1993). Traditional Chinese medicine is widely used in Southeast Asia for cardiovascular disease. An extract from *Ginkgo biloba* leaves had the flavonoid quercetin, which showed a significant effect on cardiovascular activities. When analyzed, quercetin had pharmacokinetic action on 47 cardiovascular disease-related targets and 12 KEGG (Kyoto Encyclopedia of Genes and Genomes) signaling pathways (Wu et al., 2019b). From experiments, mice, rats, rabbits, and other animals that have the same cardiac physiology also exhibit the same functional capacity for quercetin (Chirumbolo, 2012; Patel et al., 2018; Zahedi et al., 2013). In addition to cardiovascular activity, quercetin also reduces blood pressure; certain cardio favored actions, etc (Islam et al., 2014; Lakhnpal and Rai, 2008; Larson et al., 2010; Perez-Vizcaino and Duarte, 2010).

Researchers found that quercetin has been speculated to inhibit the protein kinase involved in the Ca^{2+} sensitizing mechanism, which helps in smooth muscle contraction, which helps reduce hypertension and ease the flow of blood in blood vessels (Larson et al., 2012). They have a direct effect on vascular smooth muscle tone, where quercetin-related flavonoids exert a vasodilator effect in isolated arteries (Duarte et al., 1993b, 1993a; Fitzpatrick et al., 1993). Quercetin and its metabolites are more potent in coronary arteries when compared to conductance vessels since they are resistant (Dagher et al., 2021; Pérez-Vizcaino et al., 2002).

G1-phase of the cell cycle is blocked by quercetin, which subsequently down-regulated CDKs and cyclins and up-regulated the CDK inhibitor p21 expression in vascular smooth muscle cells (Perez-Vizcaino et al., 2006; Wang et al., 2022a). Since quercetin is abundant and researchers have worked on it *in vitro* and *in vivo*, clinical studies showed that quercetin is cardioprotective by also reducing blood pressure and having antioxidant potential, which makes way for smooth blood flow (Chen et al., 2021b; Duarte et al., 2001; Kachur et al., 2017; Manna and Jain, 2015; Mirsafaei et al., 2020). Hypertension, dyslipidemia, diabetes, and obesity are also causes of cardiovascular disease. So, quercetin has been proposed for cardiovascular disease protection since it has

Table 4 Quercetin effect on diabetics.

Flavonoid	Metabolites produced from flavonoids	Function	Mechanism of action	Model
Quercetin	(a) Quercetin-3-O-glucoside (b) Quercetin-3-O-glucoside-7-O-glucoside (c) Quercetin-3-O-galaactodise (d) Aglycone	Antihyperglycemic effect and hypolipemic effect	Inhibit insulin dependent activation of P13K, inhibit GLUT2 which reduces the absorption of glucose in small intestine, block the activity of tyrosine kinase. Improve the recovery of cell proliferation, improve glucose absorption, reduces lipid peroxidation	<i>In-vitro</i> studies involve skeletal muscle cells hepatocyte RINm5F β cells. <i>In-vivo</i> studies involves rats streptozotocin (STZ)-induced diabetic rats.

anti-thrombotic, anti-dyslipidemia, and anti-hypertension properties (Chen et al., 2021a; Feng et al., 2022; Kadoglou et al., 2021; Li et al., 2021b; Tabasco et al., 2011).

An investigation confirmed the effect of quercetin (120 mg per day) for two months on patients had a positive effect on chronic systemic inflammation who had coronary artery disease (Brunetti et al., 2014; Chekalina et al., 2018; Ishizawa et al., 2011; Knekt et al., 1996; Papakyriakopoulou et al., 2022). Also, quercetin is associated with endothelial dysfunction, which helps against cardiovascular disease by vasodilation (Bondonno et al., 2015; Costa et al., 2022; Edwards et al., 2007; Knekt et al., 2000; Kozłowska and Szostak-Węgierek, 2022). But further research must confirm the complete function and confirm the hypothesis.

4.5. Neurological disorder prevention

Quercetin has a good effect on neurodegenerative disorders, and for neurodegenerative disease, inflammation plays a key role (Alam et al., 2016; Islam et al., 2021; Sharifi-Rad et al., 2020). Early brain injury triggers neuroinflammation, which is a secondary reaction that leads to significant neuronal damage, ultimately creating different neuropathologies that act as a driving force (Glass et al., 2010; Sharifi-Rad et al., 2021; Smith et al., 2012; Tsoukalas et al., 2021). Glial cells, including microglia and astrocytes, in the central nervous system (CNS) act as an immune system that maintains the normal structure of neurons by defending against pathogens (Badshah et al., 2016; Khan et al., 2018; Williams et al., 2008). There is an issue with the potential use of quercetin, where researchers are not sure whether it passes the blood-brain barrier (BBB) in *in vivo*, but *in vitro* studies indicated that with the BBB model, quercetin enters the brain (Amanzadeh et al., 2019; Faria et al., 2010; Schaffer and Halliwell, 2012; Youdim et al., 2004). When quercetin 30 mg/kg was administered to rats, studies revealed that rats had increased memory function, also found in brain tissue, and had a positive effect on neurological studies (de Boer et al., 2005; Haleagrahara et al., 2009; Huebbe et al., 2010; PARK et al., 2018; Yang et al., 2014). Quercetin is known for its role in neurodegenerative diseases since they have the capacity to modulate microRNA (miRNA) expres-

sion (Benameur et al., 2021). As mentioned, quercetin plays a key role in Alzheimer's disease and Parkinson's disease and is a recent derivative of natural food sources such as vitamins and phytochemicals, which are safe, functional foods, etc. Since vitamin C is water soluble, a high intake of this vitamin C decreased the risk of Alzheimer's disease, and people with higher plasma concentrations of vitamin C had better memory performance (Gupta et al., 2017; Heo and Lee, 2004; Masaki et al., 2000; Perrig et al., 1997; Samini, 2019).

Neuroinflammation is also a serious issue that may be caused by some primary detrimental disorders, be induced by β -amyloid aggregates, or be due to neoplasms (hyperproliferated cells). In addition to neurons, even gliosis is involved in pathological changes related to neurodegeneration and CNS disorder (Heneka et al., 2015; Hooten et al., 2015; Lyman et al., 2014; Testa et al., 2014; Troubat et al., 2021; Wróbel-Biedrawa et al., 2022). Analysis of quercetin in the brain showed that when quercetin was administered orally in conjugated form, it was accumulated in the brain as (quercetin-3-O-glucuronide and isorhamnetin-3-O-glucuronide), but methylquercetin-3-O-glucuronide was not quantifiable (Bitan et al., 2003; Dajas et al., 2015; Ho et al., 2013; Ishisaka et al., 2011; Muñoz-Reyes et al., 2021). Autophagy is a cellular destructive mechanism involving the recycling and removal of damaged and aggregated proteins by a lysosomal degradative process. This is an integral component of the central nervous system (Siokas et al., 2021). Genes related to autophagic processes lead to neurodegeneration by mutation when quercetin is administered (Ciechanover and Kwon, 2015; Qu et al., 2014; Regitz et al., 2014).

4.6. Antioxidant

A molecule that can produce free radicals by oxidation reaction can be inhibited by antioxidants. Antioxidants can be natural or synthetic; there are many dietary antioxidants like vitamin C, vitamin E, etc (Blokhina, 2003; Geethangili and Ding, 2018). Many flavonoids have antioxidant properties; one of them is quercetin, which is known for its antioxidant properties (Flora and Pachauri, 2010; Lakhanpal and Rai, 2007). The property of quercetin to act against various diseases

is due to its antioxidant properties, because of which the study of quercetin's antioxidant property has become a research hot-spot (Boots et al., 2008a; Chen, 2006; Liao et al., 2018).

Quercetin maintains oxidative balance, exhibiting strong antioxidant activity. *In vivo* studies of quercetin showed antioxidant activity through its effect on glutathione enzymatic activity, toxicology factors, reactive oxygen species from the environment, signal transduction, etc (Ademosun et al., 2016; Jin et al., 2016; Kinaci et al., 2012; Kobori et al., 2015; Qi et al., 2022; Song et al., 2013). Quercetin's chemical structure plays an important role as an antioxidant; it has a hydroxyl group (OH) mainly in the B and C rings, which contribute more than the A ring (Zheng et al., 2017b). Studies also showed that quercetin can act as an antioxidant agent and reduce cellular levels of mixed disulfide proteins (Ashfaq et al., 2006; Ishige et al., 2001; Michala and Pritsa, 2022). SPLET (sequential proton loss electron transfer) mechanisms are important antioxidant mechanisms in which antioxidants trap free radicals. Phenolic antioxidants have confirmed the important role of SPLET in polar solvents (Leopoldini et al., 2010; Vagánek et al., 2014; Xue et al., 2014; Zheng et al., 2017a). Recently, chemical studies revealed that quercetin metal complexes have better antioxidant activity than quercetin alone (Chen et al., 2005; Xu et al., 2019). Comparing quercetin and curcumin, quercetin has a higher reduction potential and total antioxidant capacity; it reduces lipopolysaccharide-induced reactive oxygen species to a normal level and ultimately reduces lipopolysaccharide-induced NO production (Zhang et al., 2011).

In certain cases, there can be DNA damage by means of free radicals or chemicals, which cause various types of damage to our body by causing diseases such as diabetes, cancer, nerve disease, etc (Selvaraju et al., 2012; Song et al., 2020; Stone et al., 2011). So, to protect DNA from these adverse effects, natural and synthetic antioxidants are required, which can protect DNA against damage. Researchers found quercetin has antioxidant ability, which can protect DNA from these external effects, but the mechanism is still unknown (Janjua et al., 2009; Kanakis et al., 2007, 2005; Oliveira-Brett and Diculescu, 2004; Silva et al., 2008; Srivastava et al., 2016). In recent time, the effect of chalcogens on the scavenging power of quercetin was studied using functional theory; quercetin has a high peroxy radical scavenging property at pH 7–10, which is attributed to its antioxidant abilities and chelating agent (Castañeda-Arriaga et al., 2020; de Souza and De Giovanni, 2004; Galano et al., 2016; Simunkova et al., 2021; Ullah et al., 2020). Quercetin can prevent the oxidation of low-density lipoproteins by scavenging free radicals (Murota and Terao, 2003). There is further research required to enhance the bioavailability of flavonoids to exert antioxidant activity (Hollman and Katan, 1997).

4.7. Anti-cancer

Cancer is abnormal uncontrolled cell growth which spreads all over the body (metastasis) and cause damage to various organs (Chaffer and Weinberg, 2011). Cancer is caused by various reason like genetics, UV rays, CFC'S, tobacco, PVC, mutation, etc (Blackadar, 2016). Cure for cancer is chemotherapy, radiotherapy, surgery but all these methods have side effects (Baskar et al., 2012; Nurgali et al., 2018). Some studies have

indicated that phytochemicals like polyphenol, flavones, flavonoids have anti-cancer activity (Kopustinskiene et al., 2020). One such example is quercetin which has wide range of activity on lung, colon, ovarian, liver, breast, pancreatic, prostate, bladder, gastric, bone cancer, etc (Fig. 3, Table 5) (Asgharian et al., 2022b; Baruah et al., 2016; Rauf et al., 2018).

While acceptable dosages of quercetin are used to treat cancer, there are no detrimental side effects on normal cells (Tang et al., 2020). *In vitro* and *in vivo* tests of quercetin's anti-cancer efficacy revealed that it inhibits metastasis and angiogenesis and encourages apoptosis (programmed cell death) (Ingham and Schwartz, 2017).

Due to changes in lifestyle, breast cancer has recently become a more severe issue for women. The active flavonoid quercetin, which inhibits breast cancer by inhibiting signal transduction, also prevents free radicals from impairing low-density lipoproteins, so quercetin, which acts as a lipid peroxidation inhibitor, acts against cancer cells (Ahmed et al., 2021; Ezzati et al., 2020; Wang et al., 2018a). Glioblastoma multiforme (GBM) is a connective tissue malignant tumor, considered the most aggressive type of brain cancer (Afshari et al., 2019; Ostrom et al., 2015). It is important to develop potential treatments for glioblastoma. Quercetin was able to inhibit cell migration (Liu et al., 2017; Tavana et al., 2020). Collectively, gliomas are malignant brain tumors that go untreated despite the availability of multiple therapeutic options. Researchers also found that quercetin inhibits cell viability and migration of gliomas (Pan et al., 2015; Zhang et al., 2015). Lung cancer is the leading cause of death worldwide. The main reason for lung cancer is using tobacco products, smoking, air pollution, radon gas, etc (Lemjabbar-Alaoui et al., 2015). Lung cancer is divided into non-small cell lung cancer (NSCLC) and small cell lung cancer (SCLC) (de la Cruz et al., 2011). Quercetin inhibits NSCLC by quercetin-induced mitotic catastrophe, which involves perturbation of mitotic microtubules, which leads to monopolar spindle formation and failure of cytokinesis, which is caused by depletion of actin filaments by quercetin (Klimaszewska-Wiśniewska et al., 2017). Ovarian cancer is a malignant tumor that affects women's reproductive tract, and it is ranked as the seventh most prevalent female cancer across the world (Vafadar et al., 2020). Since quercetin has two benzene rings joined by a pyrone ring, it acts as two antioxidant pharmacophores and helps in removing free radicals (Dhalaria et al., 2020; Shafabakhsh and Asemi, 2019; Thomasset et al., 2007). Quercetin has a cytotoxic effect on cancer cells through several mechanisms (Vargas and Burd, 2010). Cell cycle arrest is achieved by depolymerization of cellular microtubules when quercetin directly binds to tubulin (Gupta and Panda, 2002). Just like women suffer from ovarian cancer, men also suffer from prostate cancer. Prostate cancer is a common male malignant tumor, and the incident rate of new cases is 27% (Siegel et al., 2014; YANG et al., 2015). It is the second leading malignant tumor, followed by lung cancer; it is uncontrolled growth in the prostate gland (Hussain et al., 2021). Researchers found that the flavonoid quercetin decreases the function of androgen receptors by suppressing their expression (Ghafouri-Fard et al., 2021). Morris et al. (Kim et al., 2019; MORRIS et al., 2006) studied the action of quercetin by isolating prostate cancer cells from affected tissue and treating them with quercetin. Quercetin was able to diminish the primary prostate epithelial cells by reducing DNA synthesis or micro-RNA, where cancer cells are sensitive

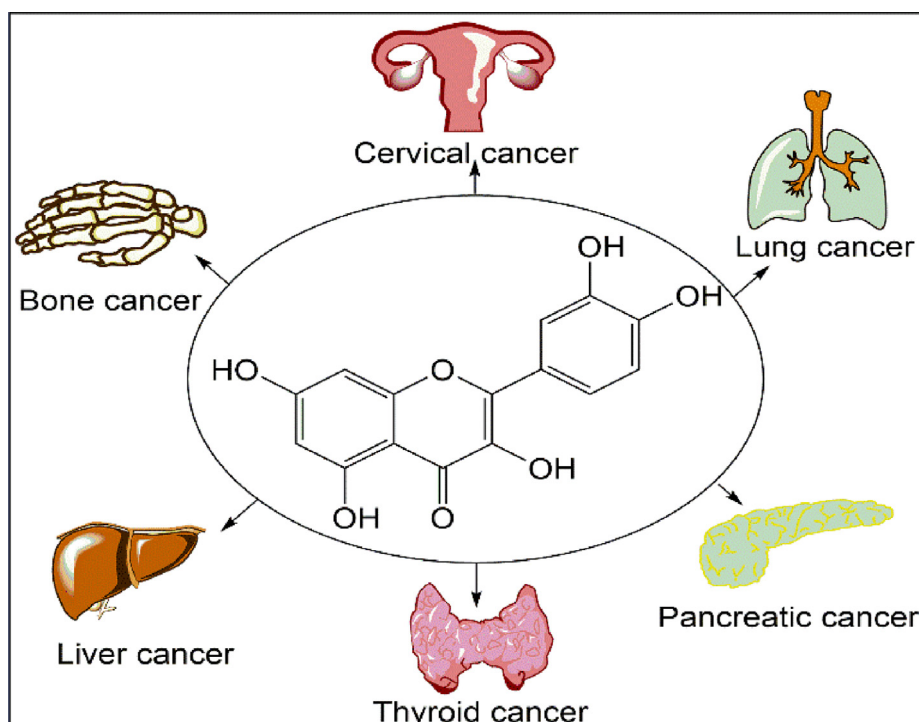


Fig. 3 Quercetin acting on various types of cancer.

Table 5 Mechanism and therapeutic action of quercetin on various cancers.

Quercetin and types	Type of cancer	Mechanism	Dose(s)	Model
Quercetin	Gastric cancer	Suppresses the growth of gastric cancer in human by provoking mitochondrial-dependent apoptosis by repression of PI3K/Akt signaling	20 μ M	<i>In vitro</i>
Quercetin targeted via phenyl boronic acid and zinc oxide nanoparticles (PBA-ZnO-Q)	Breast cancer	This can induce apoptotic cell death in breast cancer with combination of oxidative stress and mitochondrial damage	8 μ g/ml	<i>In vitro</i> and <i>in vivo</i>
Quercetin	Colorectal cancer	Promote 5-fluorouracil-induced apoptosis in MSI CRC cells via p53 regulation	1 μ M, 100 μ M	<i>In vitro</i>
Quercetin	Oral cancer	Enhanced apoptosis in human oral cancer SAS cells by mitochondria and endoplasmic reticulum-mediated signaling pathways	40 μ M	<i>In vitro</i>
Quercetin	Thyroid cancer	Decreases cell proliferation and promoted apoptosis through caspase activation and down regulation of Hsp90 expression	10 μ M	<i>In vitro</i>
Quercetin	Lung cancer	Promotes apoptosis in lung cancer via modulation of p53 posttranslational modifications and antiproliferative and antimetastatic effect on A549 non-small cell lung cancer via impact on the cytoskeleton	100 mg/kg and 74 μ M	<i>In vivo</i> and <i>in vitro</i>
Quercetin	Pancreatic cancer	Overexpression of microRNA let-7c and suppress pancreatic cancer progression	50 μ M	<i>In vivo</i> and <i>in vitro</i>
Gold-quercetin into poly (DL-lactide-co-glycolide) nanoparticles	Liver cancer	Suppress liver cancer progression by AP-2 β /hTERT, impeding caspase/cyto-cpathway, inactivating NF-kB/COX-2	30–50 μ g/ml	<i>In vivo</i> and <i>in vitro</i>
Quercetin	Prostate cancer	Converts the doxorubicin resistance of prostate cancer cells via targeting the c-met/PI3K/AKT pathway	10 μ M	<i>in vitro</i>

to chemotherapy. Pancreatic cancer is a common cancer of the gastrointestinal tract (GIT). It is characterized as exocrine pancreas and ductal adenocarcinoma tumors (Asgharian et al.,

2021; Scarpa et al., 2018). The possibility of developing pancreatic cancer is 1.5% in both genders but is generally seen in older people between the ages of 70 and 80 (Rahib et al.,

2014; Sanabria Mateos and Conlon, 2016). Quercetin inhibits cell proliferation *in vitro* and *in vivo* in the MIA PaCa-2 cell line of pancreatic cells (Angst et al., 2013). Pancreatic ductal adenocarcinoma is a highly aggressive type of tumor that has developed resistance to gemcitabine and other cytotoxic drugs, so researchers are finding new ways to develop a potent drug against pancreatic cancer, where quercetin has led them to positive results in pancreatic cancer (Fan et al., 2016; Hasegawa et al., 2014). Considering liver cancer, hepatocellular carcinoma (HCC) is the most frequent cancer (Br  chot et al., 1998). Patients also have a history of liver cirrhosis, which is caused by the Hepatitis B or C virus (Shiratori, 1995; Wu et al., 2019a). There are very few reports on how quercetin proliferates tumors, but when studied with cell lines like KIM-1, KYN-2, KYN-3, HAK-1B, HAK-2, HAK-5, HAK-6, KMCH-1, KMCH-2, they showed a synergic effect when treated with quercetin (HISAKA et al., 2020). Skin cancer is also common now due to changes in lifestyle and the use of harmful chemicals on the skin. Melanoma is a type of skin cancer that develops from neural crest-derived melanocytes (Harris et al., 2016). Melanocytes have a special feature in that they express the oxidative enzyme tyrosinase, which oxidizes quercetin into reactive adducts (Kubo et al., 2007; Metodiewa et al., 1999). Keratinocyte malignancy is generally seen over the face and arms as it is exposed to ultraviolet rays (Narayanan et al., 2010). Non-melanomatous skin cancer (NMSC) is more common than melanomas, but it is less deadly and easier to treat. Treatment with flavonoids can proliferate cancer (Pop and Diaconeasa, 2021). But researchers are working hard to find the effectiveness of quercetin on many types of cancer.

4.8. Anti-allergy

With a sudden change in environment and lifestyle, the human immune system is not able to tolerate even a simple allergen like pollen grains. Different changes in the environment will contribute to this problem, like indoor and outdoor allergens, air pollution, food, water, etc (Kawai et al., 2007; Takano and Inoue, 2017). The immune system reacts quickly when it encounters some allergens, which ultimately leads to the secretion of allergy-related mediators that generate allergic symptoms (Singh et al., 2011a). Decarboxylation of histidine produces histamine by releasing carbon dioxide; once histamine is released, it exerts its function by binding to various types of histamine receptors found in various cells throughout the body, which causes allergy (Maintz and Novak, 2007; Parsons and Ganellin, 2006; Thangam et al., 2018). Quercetin has a potential effect on allergic diseases by inhibiting histamine production. Current therapies such as β_2 -agonists and corticosteroids are limited due to their side effects, so the effect of quercetin on allergy studies such as asthma and atopic dermatitis has been extensively studied (Djukanovic, 2002; Nakamura et al., 2011; Traub and Murray, 2020; Ward, 2002).

Fewtrell and Gomperts examined the effect of quercetin on secretion from antigen sensitized mast cells and found histamine had an inhibitory effect on histamine secretion mediated by antigen, but when considering ionophores, they had very little effect (Fewtrell and Gomperts, 1977). Quercetin has an inhibitory action on mast cells because of which it causes a regression in the release of tryptase, monocyte chemoattractant protein 1 (MCP-1), and IL-6 (Mlcek et al.,

2016; Shaik et al., 2006). Allergic asthma is a chronic inflammatory lung disease. Asthma has symptoms like wheezing, airway hyperresponsiveness, and airway obstruction (He et al., 2020). When analyzed, quercetin had a positive effect when tested *in vitro* and *in vivo* by the immediate phase response and the late phase response of an allergic reaction. In the experimental studies, mice were injected intraperitoneally with 8 or 16 mg/kg/day in 200 μ l of quercetin every day, and quercetin showed a strong indication of a decrease in allergic airway inflammation and hyperresponsiveness by an anti-allergic effect in the Th1/Th2 immune response (Kumazawa et al., 2014; Lee et al., 2007; Park et al., 2009; Nutakki, 2020). In one study, quercetin was able to inhibit liver fibrosis using a murine model of experimental infection with *Schistosoma japonicum*; quercetin modulates airway remodeling in asthmatic patients and acts as a bronchodilator (Fortunato et al., 2012; Jafarinia et al., 2020; Townsend and Emala, 2013; Xu et al., 2006). Quercetin exhibits a potent effect on cellular and humoral immune functions in pre-clinical investigation with resveratrol, genistein, and epigallocatechol-3-gallate, which can modulate allergic sensitization and have a direct effect on mast cells, inhibiting mediator release (Jantan et al., 2015; Kimata et al., 2000; Kobori et al., 2016; Marzocchella et al., 2011; Yang et al., 2013). Sometimes quercetin alone may not be able to act against allergies in certain cases, but a combination of drugs can work effectively. Quercetin along with luteolin can suppress β -hexosaminidase activity as an index of anti-allergy activity (Hashimoto et al., 2006; Mizuno et al., 2017; Ueda et al., 2002; Yamashita et al., 2016). In 2017, Luo et al. (Cruz et al., 2012; Luo et al., 2018) worked on various herbs and investigated that an herb with quercetin had a special feature of bronchodilators, or relaxing the airways, in the case of asthma and certain chronic obstructive pulmonary diseases. Similar tests were mast cell-dependent; *in vivo* studies showed that mice displayed increased airway response. Patients with asthma have mast cells but not T cells or eosinophils that can localize within the bronchial smooth muscle bundles. Several mast cell mediators have unique functions in airway smooth muscle function, where mast cells involved in asthmatic airway smooth muscle involve CXCL10 and CXCR3 (Bradding, 2003; Bradding et al., 2006; Brightling et al., 2005; Christensen et al., 2006; Komolafe and Pacurari, 2022; Reuter et al., 2008; Satarkar and Patra, 2022).

4.9. Anti-depressant

Today's population is significantly impacted by depression, which is a critical global health issue. Depression is a mental disorder that relates to emotion and has neurovegetative symptoms that do not allow a person to focus on anything while damaging the quality of life (Joo, 2017; Lee et al., 2017a; Malhi and Mann, 2018). Recently, the World Health Organization (WHO) reported that worldwide, almost 280 million people will suffer from depression in 2021, and depression can lead to suicide. As every day reports, the suicide rate is increasing due to depression and is mostly seen in teenagers (Goyal et al., 2009; Grover et al., 2010; World Health Organization, 2021). Biologically, depression can be caused by nerve damage, aplastic disorder, inflammation, stress, genetics, and other psychological factors, etc (Delva and

Stanwood, 2021; Felger, 2018; Mahar et al., 2014; Ménard et al., 2016; Yohn et al., 2017).

The cure for depression is antidepressants; one such antidepressant is quercetin. The main effect of quercetin was studied in comparison with that of classic antidepressants like fluoxetine and imipramine. Since depression is associated with diabetes, studies revealed that quercetin has the potential to be employed as a therapy for depression associated with diabetes (Anjaneyulu et al., 2003; Erenmemisoglu et al., 2010; Holzmann et al., 2015; Plumb et al., 1999). A Chinese herb like Chaihu Shugansan was able to improve depression, which was much better than fluoxetine, duloxetine, etc. Quercetin with other components can improve depression synergistically, and clinical studies revealed that quercetin monotherapy is required for treating depression (Li et al., 2014; Qin et al., 2013; Wang et al., 2022b; Wang et al., 2012; Wei et al., 2021). When diabetic rats were treated with quercetin, it showed that 50 mg/kg an effective anti-depressant, but when a high dose of 100 mg/kg was used, it was ineffective. This result was obtained when a forced swim test was done on rats after administration of quercetin for 21 days and analyzed by measuring plasma adrenocorticotrophic hormone (ACTH) and corticosterone (CORT) levels (Asgharian et al., 2022a; Demir et al., 2016; Kawabata et al., 2010; Said Mohammadi et al., 2014; Vahl et al., 2005). To investigate the protective effect of quercetin on the cell viability of cultured HepG2 (Human Hepatoma) cells that were subjected to stress induced by *tert*-butyl hydroperoxide, all doses from 0.1 μ M to 10 μ M of quercetin were given to mice, and it was found that pretreatment with 10 μ M worked effectively on stress (Alía et al., 2005; Alía et al., 2006a; Alía et al., 2006b; Choi et al., 2010; Kim et al., 2013). Brain-derived neurotrophic factor (BDNF) is also associated with depression and cardiovascular disease. Estrogen receptor α (ER α) is a receptor that acts against depression and cardiovascular disease. In the tail suspension test (TST) and forced swimming test (FST), it was proven that quercetin exhibits similar behavior to ER α , which acts against depression and cardiovascular disease by reducing immobility (Chen et al., 2008; Hwang et al., 2020; Malacara et al., 2004; Wang et al., 2021; Zhu et al., 2018). Since quercetin is naturally available in many plants, researchers worked on *Coccinia indica* (Cucurbitaceae), which was already in use in Ayurveda for depression, but claims are still not validated. So, researchers worked by extracting quercetin using methanol, and the extract significantly reduced the duration of immobility at an FST dose of 400 mg/kg which did not affect locomotor activity, which confirmed its antidepressant activity (Randhawa et al., 2015). Already there are a lot of other synthetic drugs in the market for depression like clomipramine, tranylcypromine, and fluoxetine, but all these drugs have some side effects like constipation, dry mouth, restlessness, headaches, etc., because of which scientists are working from a plant source that acts as an anti-depressant and has no side effects (Dziukas and Vohra, 1991; Ghaemi et al., 2001; Majeroni and Hess, 1998; Rahmatullah et al., 2012).

4.10. Anti-arthritis and interstitial cystitis prevention

Arthritis means “disease of the joints.” It is a chronic inflammation that often co-exists with pain and structural damage (Golds et al., 1983; Senthelal et al., 2022). More than 100 dif-

ferent types of arthritis have been reported, of which osteoarthritis is the most common, followed by non-inflammatory arthritis. Sometimes our immune system acts against our body through an autoimmune response like ankylosing spondylitis, rheumatoid arthritis, psoriatic arthritis, etc (Bullock et al., 2018; Walecki, 2013; Zhu et al., 2019). Sometimes arthritis can be caused by infection, like Lyme’s arthritis (Quintero et al., 2021; Tamási and Szekanez, 2007; Wolfe and Ang, 2017). Quercetin has anti-inflammatory and antinociceptive effects. When researchers worked on women with rheumatoid arthritis, quercetin had a positive effect. In clinical studies, 50 women with rheumatoid arthritis were administered quercetin 500 mg/day for about 8 weeks, which reduced early morning stiffness, and joint count because of quercetin (Jackson et al., 2006; Javadi et al., 2017).

As discussed earlier, rheumatoid arthritis is an autoimmune disorder and a chronic inflammatory joint disease with the main pathological conditions of synovial membrane inflammation, an autoimmune response, oxidative stress, bone damage, etc (Aletaha and Smolen, 2018; Choudhary et al., 2018; Gegout et al., 1994; Quiñonez-Flores et al., 2016; Smolen et al., 2016). Studies revealed that quercetin decreased synovial inflammation by reducing joint clinical parameters and inflammatory cytokines and decreasing oxidative stress (Tang et al., 2022). In comparison with quercetin and hesperidin, quercetin had the ability to reduce edema when administered (Emim et al., 2011; Formica and Regelson, 1995; Guardia et al., 2001; Pelzer et al., 1998). Curcumin and quercetin are both antioxidant molecules with a lot of advantages, like being anti-tumor, anti-proliferative, anti-inflammatory, etc. When the inhibitory effects of both were studied on inflammation associated with arthritis using assays like neutrophil activation, synoviocyte function, and angiogenesis, both curcumin and quercetin were able to inhibit neutrophil activation, which thus confirmed that both can be suitable candidates for pre-clinical evaluation of anti-inflammatory agents (Arora et al., 2015; Banerjee et al., 2003; Bengmark, 2006; Hahm et al., 2004; Hui et al., 1997; Tudan et al., 2003).

Interstitial cystitis is a painful bladder syndrome, a chronic condition with suprapubic pain and chronic inflammation, and patients feel pain mainly in the suprapubic region with a strong sense of urgency to urinate (Gupta et al., 2015; Marcu et al., 2018; Sholan, 2020; Theoharides et al., 2008; Tyagi et al., 2018). Frequent urination may occur at night or during the day, with pain when passing urine (dysuria) and pain during sexual intercourse (dyspareunia) (Cvach and Rosamilia, 2015). Recently, it was proven that the oral bioflavonoid form of quercetin is clinically effective in men with chronic pelvic pain syndrome disorders like interstitial cystitis. Similar studies were done on interstitial cystitis in 22 patients and documented that the Cysta-Q complex (equivalent to 500 mg quercetin) for four weeks caused improvement in the patients (FORREST and SCHMIDT, 2004; Katske et al., 2001; Nickel, 2004; Winter et al., 2015). Acute cystitis is treated with antibiotic agents, but only a small amount reaches the bladder, so to reach antibiotics in the bladder, intravesical administration is required, avoiding the first-pass effect (Alidjanov et al., 2018; di Vico et al., 2020). Quercetin, when encapsulated in a micelle, which is a biodegradable nanoparticle, is a safe and efficient intravenous drug delivery system. Quercetin was able to reduce inflammatory cell infiltration of bladders with acute cystitis (Gou and Wang, 2012; Lee and Cima, 2011).

5. Drawbacks of quercetin

Despite having so many applications, mainly anti-tumor, anti-diabetic, anti-inflammatory, etc., due to the drawbacks of quercetin, the therapeutic index is limited in pharmacological applications. Since quercetin is hydrophobic in nature, it is difficult to administer it orally (Guo and Bruno, 2015; Li et al., 2009). Oral bioavailability has been illustrated to be less than 17% in animals, and in humans it is less than 2% (Gao et al., 2011; Sak, 2014). However, in natural food sources where quercetin is attached to glycoside, for example, in the case of onions, where quercetin is attached to glucose to form quercetin-3-glucoside, quercetin glucoside is more bioavailable than quercetin aglycone because of its water-soluble nature based on the octanol–water partition coefficient; however, if given to humans, a specific dosage is required (Guo et al., 2014; Lee and Mitchell, 2012; Rothwell et al., 2013; Wolfram et al., 2002). Quercetin undergoes the first-pass effect, i.e., quercetin is metabolized in the liver once it is entered into our body due to its poor absorption in the gastrointestinal tract since it is surrounded by a mucus layer with 90% water content, and it undergoes the first-pass effect (gastrointestinal instability) (Date et al., 2011; Li et al., 2021a; Macierzanka et al., 2011; NIELSEN et al., 1998). Quercetin is also associated with some drawbacks like headaches, stomachaches, and a high dosage of quercetin can lead to kidney related problems (Hu et al., 2010; Mulholland et al., 2001). Scientists are still working on its bioavailability, water solubility, first-pass effect, stability, and no side effects on humans.

5.1. How to overcome the drawbacks of quercetin

Quercetin has poor permeability and instability; numerous approaches have been undertaken that help in drug delivery (Cai et al., 2013; Ersoz et al., 2020). To ensure complete safety,

polymeric nanoparticles can act as a carrier of quercetin, and nanoparticles should have the following properties: being non-toxic, protecting the drug in the gastrointestinal tract, not altering the properties of the drug, increasing bioavailability, reaching the specific target, having a prolonged sustained release profile from the nanoparticle, controlled release, and being biodegradable (Bilia et al., 2014; Chiu et al., 2012; Khosravi Zanjani et al., 2014; Petri, 2015; Sezgin et al., 2007; Zhang et al., 2008). Fungi called *Penicillium brevicompactum* produce a secondary metabolite called mycophenolic acid, which, along with liposomes and quercetin, helps target breast cancer (Patel et al., 2020, 2016; Puel et al., 2005). To enhance its poor aqueous solubility and stability, quercetin was encapsulated in polymeric micelles. Polymeric micelles are an excellent candidate for drug delivery in cancer (Dian et al., 2014; Hamaguchi et al., 2005). In certain other cases, cancer-associated fibroblasts (CAFs) are stromal cells in the tumor environment that create a physical barrier for drugs to penetrate the tumor, making drug delivery difficult. In such cases, the liposome system of quercetin helps in the sequential delivery of drugs to the target (Duan et al., 2022; Lv et al., 2018a, 2018b; Munot et al., 2022; Tran et al., 2014; Truffi et al., 2019). Synthesizing various derivatives of quercetin, which is soluble in aqueous medium, helps in drug delivery (Kashyap et al., 2016). The metal complex of quercetin acts as a more potent anti-cancer agent than quercetin itself (Massi et al., 2017; Uivarosi and Munteanu, 2017). It was possible to increase the aqueous solubility of quercetin with rebaudioside or rubusoside treatment without altering its properties. As Ru concentration increases, the solubility of quercetin in aqueous medium also increases (Kumari et al., 2011; Nguyen et al., 2015). Nanotechnology and nanoparticle carriers have a unique property in drug delivery that helps quercetin reach the target and perform its function (Fig. 4) (Couvreur and Vauthier, 2006; Curcio et al., 2012).

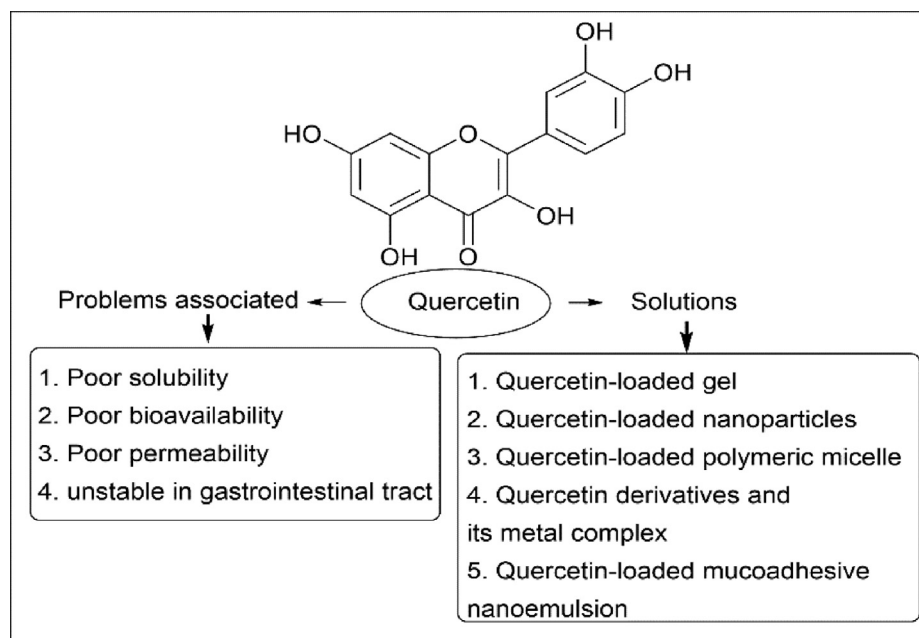


Fig. 4 Problems associated with quercetin and solutions.

6. Conclusion

Quercetin is a flavonoid with many applications, but it is still not recognized in the field of pharmacology due to the drawbacks associated with it. Researchers are working to improve its bioavailability, aqueous solubility, avoiding first pass effect, etc. The synthesis of derivatives that are soluble in water, the synthesis of metal complex and checking their therapeutic action on various diseases are being done. Researchers have found polymeric or metal nanoparticle conjugation of quercetin is best for various conditions, including cancer treatment. In the future, quercetin can be conjugated or encapsulated with some biopolymer, or the synthesis of quercetin derivatives that are hydrophilic, stable in the gastrointestinal tract, etc., can be used in the application of drug delivery that can overcome the drawbacks of quercetin and act as a potent drug.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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