

## King Saud University

## Arabian Journal of Chemistry

www.ksu.edu.sa



www.sciencedirect.com

## **ORIGINAL ARTICLE**

# Synthesis and antibacterial activity of some new 4-anilino-5-phenyl-4H-1,2,4-triazole-3-thiol derivatives



# Abd El-Badih A.G. Ghattas, Hassan M. Moustafa, Elsayed A.A. Hassanein, Bahgat R.M. Hussein \*

Chemistry Department, Faculty of Science, Sohag University, Sohag, Egypt

Received 9 January 2012; accepted 23 April 2012 Available online 30 April 2012

#### **KEYWORDS**

4-Anilino-5-phenyl-4H-1,2,4triazole-3-thiol; Mannich bases; Triazolotriazoles: Oxadiazolotriazoles; Triazinotriazoles; Antibacterial activity

Abstract 4-Anilino-5-phenyl-4H-1,2,4-triazole-3-thiol (1) reacted with formaldehyde and different amines to give Mannich bases 2a-i. Treatment of compound 1 with formaldehyde afforded the corresponding 2-hydroxymethyl derivative 3, which upon its reaction with thionyl chloride yielded the corresponding chloromethyl derivative 4. Treatment of compound 4 with some thiols gave the corresponding sulfides 5a-f. The ring closure reaction of chloromethyl derivative 4 with hydrazine hydrate, phenyl hydrazine, hydroxylamine, urea and thiourea afforded triazolo-, oxadiazolo- and triazinotriazoles 6-10, respectively.

© 2012 Production and hosting by Elsevier B.V. on behalf of King Saud University. This is an open access article under the CC BY-NC-ND license (http://creativecommons.org/licenses/by-nc-nd/3.0/).

#### 1. Introduction

Various 1,2,4-triazoles found extensive investigations due to their useful application in different areas of biological activity and as industrial intermediates such as antiasthmatic (Naito et al., 1996), antiviral (ribavirin) (De Clercq, 2004), antifungal (fluconazole) (Collin et al., 2003), antimicrobial (Kidwai et al., 2001), antibacterial (Papakonstantinou-Garoufalias

E-mail address: bahgat.ramadan@yahoo.com (B.R.M. Hussein). Peer review under responsibility of King Saud University.



et al., 2002), insecticidal (Ghorab et al., 1999), amoebicidal (Andotra and Sharma, 1988), hypnotic (Hester et al., 1971), cytotoxic (Milton, 2001) and hypotensive (Burell et al., 1994; Ghorab et al., 1996) activities. This moiety was also found in potent agonist and antagonist receptor ligands, (Wadsworth et al., 1992; Chen et al., 2001) in HIV-1 protease inhibitors (Thompson et al., 1994) and in thrombin inhibitors (Duncia et al., 1998). Along with these significant pharmaceutical uses, 1,2,4-triazole derivatives are effectively used in polymers, dyestuff, photographic chemicals and agricultural chemicals (Potts, 1961). In view of these findings and in continuation of our work in the same field (Ghattas and El-saraf, 1989; Ghattas et al. 1989, 2001, 2003; El-wassaimy et al., 1992; Abdel-Rahman et al., 1993; Moustafa, 2001), we report herein the synthesis of some new s-triazole derivatives and tested the latter for antibacterial activity.

#### http://dx.doi.org/10.1016/j.arabjc.2012.04.016

1878-5352 © 2012 Production and hosting by Elsevier B.V. on behalf of King Saud University. This is an open access article under the CC BY-NC-ND license (http://creativecommons.org/licenses/by-nc-nd/3.0/).

<sup>\*</sup> Corresponding author. Address: Postal code: 82524, Sohag, Egypt. Tel.: +20 1064382209; fax: +20 93601159.

#### 2. Experimental

Melting points are uncorrected and were determined by Kofeler melting point apparatus. IR (cm<sup>-1</sup>) spectra were recorded (KBr disk) on a Nicolet 710 FT-IR spectrophotometer. <sup>1</sup>H NMR (DMSO- $d_6$ ) spectra were recorded at 300 MHz on a Varian Gremini NMR spectrometer at the Cairo University, the chemical shift is expressed in  $\delta$  value (ppm) using TMS as an internal reference. Element analyses carried out on an elemental analyzer 240 °C. Mass spectra were performed on Micromass 7070 E spectrometer operating at 70 eV, using direct inlet.

#### 2.1. Synthesis of Mannich bases 2a-i (general procedure)

Formaldehyde (1.5 mL, 40% solution) was added to a solution of compound 1 (0.5 g, 1.8 mmol) in ethanol (15 mL) and the reaction mixture was refluxed for 1 h. The appropriate amine (0.001 mol) was added and the reaction mixture was refluxed for 4 h. After cooling, the formed precipitate was filtered and recrystallized from the appropriate solvent to give the corresponding Mannich bases 2a-i.

#### 2.1.1. 4-Anilino-5-phenyl-2-[(1,3-thiazol-2-yl-amino)methyl]-2,4-dihydro-3H-1,2,4-triazole-3-thione (2a)

White crystals from ethanol, yield 0.39 g (55%), m.p. 155 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3431 (NH), 3195 (NH), 3081 (Ar, CH), 2918 (Aliph., CH), 1592 (C=N); MS, m/z ( $I_{rel}$ ,%): 380 [M]<sup>+</sup> (10.6), 312 (21.3), 239 (27.7), 178 (51), 135 (34), 103 (100), 77 (81), 51 (72.3). Anal. for C<sub>18</sub>H<sub>16</sub>N<sub>6</sub>S<sub>2</sub> (380.48) Calcd. %: C 56.77; H 4.20; N 22.07; S 8.41. Found %: C 56.51; H 4.35; N 22.30; S 8.61.

#### 2.1.2. 4-Anilino-5-phenyl-2-[(1,3,4-thiadiazol-2-yl-

amino)methyl]-2,4-dihydro-3H-1,2,4-triazole-3-thione (**2b**) White crystals from ethanol, yield 0.7 g (98%), m.p. 113 °C; IR (KBr):  $V_{\rm max}/{\rm cm}^{-1}$ : 3415 (NH), 3203 (NH), 3072 (Ar, CH), 2940 (Aliph., CH), 1568 (C=N); MS, m/z ( $I_{rel}$ ,%): 381 [M]<sup>+</sup> (0.04), 312 (21.3), 103 (60.6), 60 (100). Anal. for C<sub>17</sub>H<sub>15</sub>N<sub>7</sub>S<sub>2</sub> (381.47) Calcd. %: C 53.47; H 3.93; N 25.69; S 8.38. Found %: C 53.33; H 3.67; N 25.53; S 8.41.

#### 2.1.3. 4-Anilino-5-phenyl-2-[(pyridin-2-yl-amino)methyl]-2,4dihydro-3H-1,2,4-triazole-3-thione (2c)

White-yellow crystals from ethanol, yield 0.65 g (94%), m.p. 175 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3428 (NH), 3240 (NH), 3015 (Ar, CH), 2954 (Aliph., CH), 1600 (C=N); MS, m/z ( $I_{rel}$ ,%): 374 [M]<sup>+</sup> (0.07), 312 (37.7), 279 (19), 181(13.2), 149 (22.6), 103 (100), 55 (96.2). Anal. for C<sub>20</sub>H<sub>18</sub>N<sub>6</sub>S (374.46) Calcd. %: C 64.06; H 4.80; N 22.43; S 8.54. Found %: C 64.15; H 4.66; N 22.52; S 8.37.

#### 2.1.4. 1-[4-(4-Anilino-5-phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione-2-yl-methylamino)phenyl]-1-ethanone (2d)

Yellow crystals from ethanol, yield 0.75 g (97%), m.p. 168 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3328 (NH), 3185 (NH), 3048 (Ar, CH), 2930 (Aliph., CH), 1657 (C=O), 1589 (C=N); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta$ , ppm: 7.95–7.58 (m, 15H, Ar-H), 6.95 (s, 2H, 2NH), 5.54 (s, 2H, N–CH<sub>2</sub>–N), 2.39 (s, 3H, CH<sub>3</sub>). Anal. for C<sub>23</sub>H<sub>21</sub>N<sub>5</sub>OS (415.51) Calcd. %: C 66.42; H 5.05; N 16.84; S 7.70. Found %: C 66.31; H 5.22; N 16.75; S 7.53.

#### 2.1.5. 4-Anilino-2-[(diethylamino)methyl]-5-phenyl-2,4dihydro-3H-1,2,4-triazole-3-thione (**2e**)

Brown crystals from ethanol, yield 0.37 g (56%), m.p. 92 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3369 (NH), 3055 (Ar, CH), 2966 (Aliph., CH), 1617 (C=N); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta$ , ppm: 7.92–7.33 (m, 11H, Ar-H + NH), 3.07–3.04 (q, 4H, CH<sub>2</sub>, J = 7.8 Hz), 2.68 (s, 2H, CH<sub>2</sub>), 1.15–1.20 (t, 6H, CH<sub>3</sub>, J = 14.7 Hz). Anal. for C<sub>19</sub> H<sub>23</sub>N<sub>5</sub>S (353.48) Calcd. %: C 64.50; H 6.50; N 19.80; S 9.00. Found %: C 64.31; H 6.72; N 19.63; S 9.12.

#### 2.1.6. 4-Anilino-2-{[ethyl(phenyl)amino]methyl}-5-phenyl-2,4dihydro-3H-1,2,4-triazole-3-thione (**2f**)

White crystals from ethanol, yield 0.68 g (90%), m.p. 107 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3424 (NH), 3048 (Ar, CH), 2964 (Aliph., CH), 1606 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.91–6.92 (m, 10H, Ar-H), 5.68 (s, NH), 5.20 (s, 2H, N– CH<sub>2</sub>–N), 3.69–3.63 (q, 2H, N–CH<sub>2</sub>, J = 18.0 Hz), 1.21–1.19 (t, 3H, CH<sub>3</sub>, J = 6 Hz). Anal. for C<sub>23</sub>H<sub>23</sub>N<sub>5</sub>S (401.52) Calcd. %: C 68.73; H 5.72; N 17.43; S 7.97. Found %: C 68.61; H 5.83; N 17.35; S 7.81.

#### 2.1.7. 4-Anilino-2-(morpholin-4-yl-methyl)-5-phenyl-2,4dihydro-3H-1,2,4-triazole-3-thione (**2**g)

White crystals from ethanol, yield 0.48 g (71%), m.p. 118 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3427.9 (NH), 3051.8 (Ar, CH), 2937, 2841 (Aliph., CH,), 1615 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.91–7.33 (m, 10H, Ar-H), 5.16 (s, NH), 5.01 (s, 2H, N–CH<sub>2</sub>–N), 3.58–3.36 (m, 4H, 2N–CH<sub>2</sub>), 2.79–2.49 (m, 4H, 2O–CH<sub>2</sub>). Anal. for C<sub>19</sub>H<sub>21</sub>N<sub>5</sub>OS (367.46) Calcd. %: C 62.04; H 5.71; N 19.05; S 8.70. Found %: C 62.23; H 5.82; N 19.31; S 8.65.

#### 2.1.8. 4-Anilino-5-phenyl-2-(piperidin-1-ylmethyl)-2,4-dihydro-3H-1,2,4-triazole-3-thione (**2h**)

White crystals from ethanol, yield 0.63 g (92%), m.p. 93 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3422 (NH), 3058 (Ar, CH), 2925, 2845 (Aliph., CH), 1608 (C=N); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta$ , ppm: 7.91–7.55 (m, 10H, Ar-H), 4.99 (s, 2H, N–CH<sub>2</sub>), 3.36 (s, 1H, NH), 2.72–2.50 (m, 4H, 2N–CH<sub>2</sub>), 1.48–1.44 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>N), 1.33–1.32 (m, 2H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>). Anal. for C<sub>20</sub>H<sub>23</sub>N<sub>5</sub>S (365.49) Calcd. %: C 65.66; H 6.29; N 19.15; S 8.75. Found %: C 65.42; H 6.33; N 19.31; S 8.61.

#### 2.1.9. 4-Anilino-1-[4-(4-anilino-3-phenyl-5-thioxo-4,5-dihydro-1H-1,2,4-triazole-1-yl-methyl) piperazino methyl-3-phenyl-4,5dihydro-1H-1,2,4-triazole-5-thione (2i)

White crystals from ethanol, yield 0.98 g (81%), m.p. 240 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3430 (NH), 3059 (Ar, CH), 2938, 2841 (Aliph., CH), 1615 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.89–7.56 (m, 20H, Ar-H), 5.0 (s, 4H, 2N–CH<sub>2</sub>–N), 3.21 (s, 2H, NH), 2.80 (m, 8H, 4N–CH<sub>2</sub>). Anal. for C<sub>34</sub>H<sub>34</sub>N<sub>10</sub>S<sub>2</sub> (646.83): Calcd. %: C 63.07; H 5.25; N 21.64; S 9.89. Found %: C 63.19; H 5.11; N 21.55; S 9.97.

#### 2.2. Synthesis of 4-anilino-2-hydroxymethyl-5-phenyl-2,4dihydro-3H-1,2,4-triazole-3-thione (3)

Formaldehyde (1.5 mL, 40% solution) was added to a solution of compound 1 (1.34 g, 5 mmol) in ethanol (15 mL) and the reaction mixture was refluxed for 1 h. The solvent was evaporated to dryness, the formed solid was collected and

recrystallized from ethanol to give compound **3**. White crystals from ethanol, yield 1.4 g (94%), m.p. 132 °C; IR (KBr):  $V_{\text{max}}$ / cm<sup>-1</sup>: 3418 (OH), 3107 (NH), 2944 (Ar, CH), 2768 (Aliph., CH), 1610 (C=N); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta$ , ppm: 7.92–7.58 (m, 10H, Ar-H), 6.06 (s, 2H, CH<sub>2</sub>), 5.4 (s, 1H, NH), 4.99 (s, 1H, OH). Anal. for C<sub>15</sub>H<sub>14</sub>N<sub>4</sub>OS (298.36) Calcd. %: C 60.32; H 4.69; N 18.76; S 10.72. Found %: C 60.41; H 4.59; N 18.65; S 10.81.

#### 2.3. Synthesis of 4-anilino-2-chloromethyl-5-phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione (4)

Thionyl chloride (10 mL) was added dropwise to compound **3** (2.98 g, 10 mol) and the reaction mixture was warmed on a water bath for 1 h. After cooling, the reaction mixture was poured into petroleum ether (100 mL, 80:100 °C). The formed precipitate was filtered and recrystallized from ethanol to give compound **4**. White crystals from ethanol, yield 2.76 g (87%), m.p. 106 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3439 (NH), 3037 (Ar, CH), 2929 (Aliph., CH), 1616 (C=N), 767 (C-Cl); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.91–7.28 (m, 10H, Ar-H), 5.41 (s, 1H, NH), 5.39 (s, 2H, CH<sub>2</sub>). Anal. for C<sub>15</sub>H<sub>13</sub>ClN<sub>4</sub>S (316.8) Calcd. %: C 56.81; H 4.10; N 17.67; S 10.10. Found %: C 56.57; H 4.13; N 17.54; S 10.21.

#### 2.4. Synthesis of compounds 5a-f (general procedure)

To a solution of compound 4 (0.5 g 1.58 mmol) in dimethylformamide (30 mL) and triethylamine (1.58 mmol), the appropriate thiol (1.58 mmol) was added. The reaction mixture was refluxed for 3 h, the solvent was evaporated under reduced pressure. The formed precipitate was collected and recrystallized from the appropriate solvent to give the corresponding sulfides 5a-f.

#### 2.4.1. 4-Anilino-5-phenyl-2-[(propylthio) methyl]-2,4-dihydro-3H-1,2,4-triazole-3-thione (**5a**)

Brown crystals from dioxane, yield 0.44 g (79%), m.p. 89 °C. IR (KBr):  $V_{max}/cm^{-1}$ : 3438 (NH), 3046 (Ar, CH), 2961, 2931 (Aliph., CH), 1615 (C=N); <sup>1</sup>H NMR (DMSO- $d_6$ ),  $\delta$ , ppm: 7.87–7.54 (m, 10H, Ar-H), 6.15 (br, 3H, CH<sub>2</sub>–N + NH), 3.206–3.157 (t, 2H, S–CH<sub>2</sub>, J = 14.7 Hz), 1.70–1.60 (m, 2H, –CH<sub>2</sub>–), 0.99–0.97 (t, 3H, CH<sub>3</sub>–, J = 8.4 Hz). Anal. for C<sub>18</sub>H<sub>20</sub>N<sub>4</sub>S<sub>2</sub> (356.5) Calcd. %: C 60.58; H 5.61; N 15.70; S 17.95. Found %: C 60.48; H 5.55; N 15.61; S 17.99.

#### 2.4.2. 4-Anilino-2-[(cyclohexylthio)methyl]-5-phenyl-2,4dihydro-3H-1,2,4-triazole-3-thione (**5b**)

White crystals from dimethylformamide, yield 0.39 g (62%), m.p. 222 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3439 (NH), 3055 (Ar, CH), 2929 (Aliph., CH), 1608 (C=N); <sup>1</sup>H NMR (DMSO $d_6$ ),  $\delta$ , ppm: 7.89–7.57 (m, 11H, Ar-H + NH), 6.36 (s, 2H, CH<sub>2</sub>), 3.59–3.27 (m, 11H, cyclohexyl). Anal. for C<sub>21</sub>H<sub>24</sub> N<sub>4</sub>S<sub>2</sub> (396.56) Calcd. %: C 63.54; H 6.05; N 14.12; S 16.13. Found %: C 63.48; H 6.22; N 14.11; S 16.25.

#### 2.4.3. 4-Anilino-5-phenyl-2-[(phenylthio)methyl]-2,4-dihydro-3H-1,2,4-triazole-3-thione (5c)

White crystals from dimethylformamide, yield 0.41 g (67%), m.p. 98 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3433 (NH), 3054 (Ar, CH), 2928 (Aliph., CH), 1581 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.82–7.31 (m, 15H, Ar-H), 5.58 (br, 3H, NH + CH<sub>2</sub>). Anal. for  $C_{21}H_{18}N_4S_2$  (390.52) Calcd. %: C 64.52; H 4.61; N 14.33; S 16.38. Found %: C 64.64; H 4.55; N 14.47; S 16.35.

# 2.4.4. 4-Anilino-2-{[(4-chlorophenyl)thio]methyl}-5-phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione (5d)

White crystals from dimethylformamide, yield 0.56 g (82.5%), m.p. 138 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3439 (NH), 3055 (Ar, CH), 2929 (Aliph., CH), 1608 (C=N); <sup>1</sup>H NMR (DMSO $d_6$ ),  $\delta$ , ppm: 7.89–7.57 (m, 11H, Ar-H + NH), 6.36 (s, 2H, CH<sub>2</sub>). Anal. for C<sub>21</sub>H<sub>17</sub>ClN<sub>4</sub>S<sub>2</sub> (424.96) Calcd. %: C 59.29; H 4.00; N 13.17; S 15.06. Found %: C 59.41; H 4.12; N 13.32; S 15.12.

# 2.4.5. 4-Anilino-2-{[(4-anilino-5-phenyl-4H-1,2,4-triazol-3-yl)thio]methyl}-5-phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione (5e)

Pale-pink crystals from dimethylformamide, yield 0.65 g (75.5%), m.p. 148 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3440 (2NH), 3028 (Ar, CH), 2932 (Aliph., CH), 1610 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.99–7.54 (m, 20H, Ar-H), 5.9 (br, 4H, N–CH<sub>2</sub> + 2NH). Anal. for C<sub>29</sub>H<sub>24</sub>N<sub>8</sub>S<sub>2</sub> (548.68) Calcd. %: C 63.42; H 4.37; N 20.41; S 11.66. Found %: C 63.35; H 4.43; N 20.37; S 11.81.

#### 2.4.6. 4-Anilino-5-phenyl-2-{[(5-phenyl-1,3,4-oxadiazol-2-

yl) thio]methyl}-2,4-dihydro-3H-1,2,4-triazole-3-thione (**5f**) Pale-white crystals from dimethylformamide, yield 0.54 g (75%), m.p. 217 °C; IR (KBr):  $V_{max}/cm^{-1}$ : 3445 (NH), 3065 (Ar, CH), 2931 (Aliph., CH), 1612 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.95–7.11 (m, 16H, Ar-H + NH), 5.43 (s, 2H, N–CH<sub>2</sub>). Anal. for C<sub>23</sub>H<sub>18</sub>N<sub>6</sub>OS<sub>2</sub> (458.55) Calcd. %: C 60.18; H 3.92; N 18.31; S 13.95. Found %: C 60.22; H 3.87; N 18.42; S 13.76.

#### 2.5. Synthesis of compounds 6–10: (general procedure)

To a solution of compound 4 (2.2 mmol) and triethylamine (2.2 mmol) in xylene (40 mL) were added hydrazine hydrate, phenyl hydrazine, hydroxylamine hydrochloride, urea and/or thiourea (2.2 mmol), respectively. The reaction mixture was refluxed for 10 h. The solvent was evaporated under reduced pressure, the formed precipitate was collected and recrystal-lized from ethanol to give the corresponding compound **6–10**.

#### 2.5.1. 7-Anilino-6-phenyl-2,7-dihydro-3H-1,2,4-triazolo[4,3-b]-1,2,4-triazole (6)

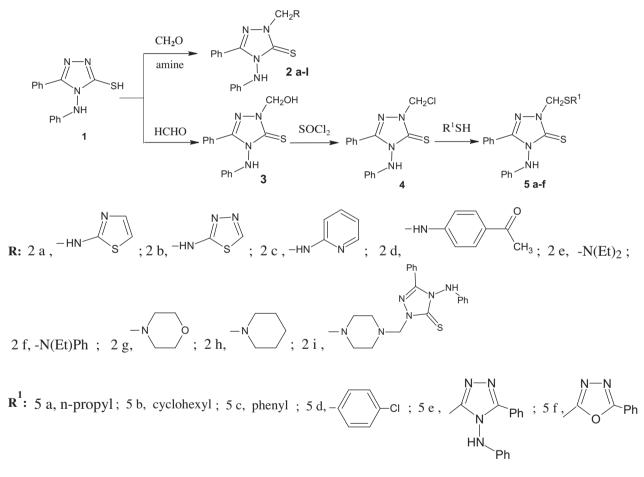
White crystals from ethanol, yield 0.68 g (97%), m.p. 200 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3106 (NH), 3015 (Ar, CH), 2932 (Aliph., CH), 1622 (C=N); MS, m/z ( $I_{rel}$ ,%): 276 [M]<sup>+</sup> (7.7), 252 (58.5), 192 (69.2), 161 (100), 133 (32.3), 104 (66.2),77 (67.7), 51 (72.3). Anal. for C<sub>15</sub> H<sub>12</sub>N<sub>6</sub> (276.3) Calcd. %: C 67.34; H 4.48; N 31.42. Found %: C 67.21; H 4.35; N 31.27.

#### 2.5.2. 7-Anilino-2,6-diphenyl-2,7-dihydro-3H-1,2,4-triazolo[4,3b]-1,2,4-triazole (7)

Dark brown crystals from ethanol, yield 0.56 g (72%), m.p. 115 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3414 (NH), 3045 (Ar, CH), 2928 (Aliph., CH), 1598 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>),  $\delta$ , ppm: 7.98–7.04 (m, 16H, Ar-H + NH), 5.54 (s, 2H, N–CH<sub>2</sub>–N). Anal. for C<sub>21</sub>H<sub>18</sub>N<sub>6</sub> (354.41) Calcd. %: C 71.10; H 5.07; N 23.7. Found %: C 71.24; H 5.11; N 23.67.

Table 1	Results of	anti-bacterial	evaluation o	f compounds	2b, 2c	e, 2e, 2	g, 2h,	2i, 4	4, 5a,	5b, 5d,	5e and
---------	------------	----------------	--------------	-------------	--------	----------	--------	-------	--------	---------	--------

Types of bacteria	Bacillus cereus			Pseudomonas aeruginosa			Escherichia coli			
	Concentrations			Concentrations			Concentrations			
Compound	10,000 ppm	30,000 ppm	50,000 ppm	10,000 ppm	30,000 ppm	50,000 ppm	10,000 ppm	30,000 ppm	50,000 ppm	
2b	0.9 cm	2.3 cm	2.6 cm	0.7 cm	1.2 cm	1.4 cm	0.4 cm	0.8 cm	1.4 cm	
2c	1.0 cm	1.7 cm	1.7 cm	-	0.5 cm	_	0.2 cm	0.4 cm	0.7 cm	
2e	0.7 cm	1.3 cm	1.8 cm	0.7 cm	1.3 cm	1.8 cm	0.5 cm	1.1 cm	1.2 cm	
2g	1.4 cm	3.1 cm	3.4 cm	0.8 cm	1.4 cm	1.9 cm	0.9 cm	1.8 cm	1.9 cm	
2h	1.7 cm	2.6 cm	3.5 cm	1.1 cm	1.4 cm	2.3 cm	1.0 cm	1.2 cm	1.7 cm	
2i	0.8 cm	0.9 cm	1.0 cm	0.8 cm	0.9 cm	1.3 cm	0.5 cm	0.8 cm	0.9 cm	
4	1.0 cm	1.1 cm	2.0 cm	0.9 cm	1.2 cm	1.5 cm	1.1 cm	1.3 cm	2.3 cm	
5a	0.5 cm	0.6 cm	0.7 cm	0.5 cm	0.7 cm	1.0 cm	0.5 cm	0.6 cm	0.8 cm	
5b	0.6 cm	0.9 cm	1.1 cm	-	-	0.5 cm	0.5 cm	0.8 cm	0.4 cm	
5d	0.7 cm	0.9 cm	0.9 cm	0.5 cm	0.7 cm	0.9 cm	0.6 cm	0.8 cm	1.0 cm	
5e	0.3 cm	0.4 cm	0.8 cm	-	-	_	0.7 cm	0.8 cm	0.6 cm	
6	0.4 cm	0.5 cm	0.7 cm	0.9 cm	1.3 cm	2.1 cm	0.5 cm	0.8 cm	0.9 cm	



Synthesis of Mannich bases and sulfides. Scheme 1

#### 2.5.3. 7-Anilino-6-phenyl-7H-1,2,4-triazolo[5,1-c]-1,2,4oxadiazole (8)

Brown crystals from ethanol, yield 0.55 g (90%), m.p. 138 °C; IR (KBr): V<sub>max</sub>/cm<sup>-1</sup>: 3422 (NH), 3022 (Ar, CH), 2932 (Aliph., CH), 1607 (C=N); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>), δ, ppm: 8.0-7.54 (m, 11H, Ar-H + NH), 5.9 (s, 2H, CH<sub>2</sub>). Anal. for C15H13N5O (279.3) Calcd. %: C 64.44; H 4.65; N 25.06. Found %: C 64.36; H 4.58; N 25.31.

#### 2.5.4. 1-Anilino-2-phenyl-1,5,6,7-tetrahydro-1,2,4-triazolo[1,5a]-1,3,5-triazin-7-one (9)

Pale-brown crystals from ethanol, yield 0.41 g (61%), m.p. 140 °C; IR (KBr): V<sub>max</sub>/cm<sup>-1</sup>: 3398 (NH), 3038 (Ar, CH), 2923 (Aliph., CH), 1720 (C=O), 1612 (C=N); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>),  $\delta$ , ppm: 8.0–7.55 (m, 12H, Ar-H + 2NH), 5.9 (s, 2H, CH<sub>2</sub>). Anal. for C<sub>16</sub>H<sub>14</sub>N<sub>6</sub>O (306.32) Calcd. %: C 62.67; H 4.57; N 27.42. Found %: C 62.51; H 4.43; N 27.56.

#### 2.5.5. 1-Anilino-2-phenyl-1,5,6,7-tetrahydro-1,2,4-triazolo[1,5-a]-1,3,5-triazine-7-thione (10)

Brown crystals from ethanol, yield 0.45 g (90%), m.p. 176 °C; IR (KBr):  $V_{\text{max}}/\text{cm}^{-1}$ : 3430 (2NH), 3091 (Ar, CH), 2924 (Aliph., CH), 1607 (C=N), 1175 (C=s); MS, m/z ( $I_{rel}$ ,%): 322 [M]<sup>+</sup> (2.77), 253 (100), 194 (10.22), 149 (10.09), 126 (41.39), 118 (19.53), 103 (12.7), 91 (18.3), 77 (26.62), 51(6.5). Anal. for C<sub>16</sub>H<sub>14</sub>N<sub>6</sub>S (322.38) Calcd. %: C 59.55; H 4.34; N 26.05. Found %: C 59.64; H 4.27; N 26.22.

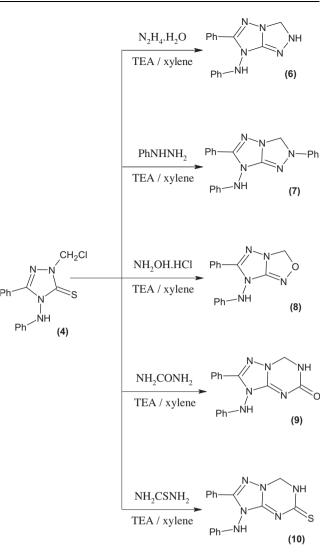
#### 2.6. Antibacterial activity

The compounds were dissolved in DMSO. In order to ensure that the solvent per se had no effect on bacterial growth or enzymatic activity, negative control tests were performed using DMSO at the same concentrations. The inhibitory effect of compounds **2b**, **2c**, **2e**, **2g**, **2h**, **2i**, **4**, **5a**, **5b**, **5d**, **5e** and **6** on the *in vitro* growth of broad spectrum of bacteria representing one gram positive bacteria, namely *Bacillus cereus* and two gram negative bacteria, namely; *Pseudomonas aeruginosa* and *Escherichia coli* was evaluated using agar diffusion method (cup and plate method) (Barry, 1976) by measuring the zone of inhibition on agar plates at three different concentrations 10,000, 30,000 and 50,000 ppm. DMSO was used as solvent control. All plates were incubated at  $37 \pm 0.5$  °C for 24 h. The zone of inhibition of compounds was measured using cm scale. The results indicated in Table 1.

#### 3. Results and discussion

#### 3.1. Chemistry

Mannich reaction on 4-anilino-5-phenyl-4H-1,2,4-triazole-3thiol (1) (Chande et al., 1993) using formaldehyde and different amines namely; 2-aminothiazole, 2-amino-1,3,4-thiadiazole, 2-aminopyridine, p-aminoacetophenone, diethylamine, N-ethylaniline, morpholine, piperidine and piperazine afforded the corresponding Mannich bases 2a-i. (cf. Scheme 1). The structure of these compounds was established on the basis of their elemental and spectral analyses. MS of compounds 2a, **2b** and **2c** showed the molecular ion peaks at  $m/z = 380 \text{ (M}^+$ , 10.6%), 381(M<sup>+</sup>, 0.04%) and 374(M<sup>+</sup>, 0.07%), respectively, (cf. experimental). 4-Anilino-2-hydroxymethyl-5-phenyl-2,4dihydro-3H-1,2,4-triazole-3-thione (3) was obtained via the reaction of compound 1 with formaldehyde, which in turn reacted with thionyl chloride to give 4-anilino-2-chloromethyl-5phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione (4). IR spectrum (KBr,  $cm^{-1}$ ) of compound **3** showed new absorption band corresponding to OH at 3418.3 cm<sup>-1</sup>. Treatment of compound 4 with the respective aliphatic, aromatic or heterocyclic thiols namely; propanethiol, cyclohexanethiol, thiophenol, 4chlorothiophenol, 4-anilino-5-phenyl-4H-1,2,4-triazole-3-thiol and 5-phenyl-1,3,4-oxadiazole-2-thiol, in boiling dimethylformamide in the presence of triethylamine, yielded the corresponding sulfides 5a-f in good yields (cf. Scheme 1). The IR spectra (KBr,  $cm^{-1}$ ) of compounds **5a-f** showed disappearance of absorption band corresponding to C-Cl at 767  $\text{cm}^{-1}$ . Condensation of chloromethyl derivative 4 with hydrazine hydrate, phenyl hydrazine, hydroxylamine, urea and thiourea in refluxing p-xylene and triethylamine gave; 7-anilino-6-phenyl-2,7-dihydro-3H-1,2,4-triazolo[4,3-b]-1,2,



Scheme 2 Synthesis of triazolo-, oxadiazolo- and triazinotriazoles.

4-triazole (6), 7-anilino-2,6-diphenyl-2,7-dihydro-3*H*-1,2, 4-triazolo[4,3-*b*]-1,2,4-triazole (7), 7-anilino-6-phenyl-7*H*-1,2,4-triazolo[5,1-*c*]-1,2,4-oxadiazole (8), 1-anilino-2-phenyl-1,5,6,7-tetrahydro-1,2,4-triazolo[1,5-*a*]-1,3,5-triazin-7-one (9) and 1-anilino-2-phenyl-1,5,6,7-tetrahydro-1,2,4-triazolo[1,5a]-1,3,5-triazine-7-thione (10) respectively (cf. Scheme 2). The structure of these compounds was proved by elemental and spectral analyses (*cf.* experimental). MS of compounds **6** and **10** showed the molecular ion peak at *m*/*z* 276 (M<sup>+</sup>, 7.7%) and 322 (M<sup>+</sup>, 2.77%), respectively.

#### 3.2. Antibacterial activity

The results of antibacterial activity are shown in Table 1. Mannich bases derivatives exhibited good activity, specially compounds 2g and 2h, showed the highest inhibitory effect against all types of bacteria excepting *E. coli* which effected highly by chloromethyl compound 4 (2.3 cm). Sulfides display good activity against Bacillus cereus and *E. coli* but possess moderate to poor activities against *P. aeruginosa*. Triazolotriazole **6** showed good activities against all types of bacteria.

The zone of inhibition of all compounds was increased by increasing the concentrations excepting compound 2c that has activity against *P. aeruginosa* at 30,000 ppm only.

#### Acknowledgments

The authors are deeply grateful to the Chemistry Department, Faculty of Science, Sohag University in Egypt for supporting and facilitating this study. Also, we extend our thanks to Dr. Rehab Mostafa Mohamed and Miss. Dalia Ahmed Abd El Raheem, Botany Department, Faculty of Science, Sohag University in Egypt for Evaluation of bacterial inhibiting effects.

#### References

- Abdel-Rahman, M.A., Ghattas, A.B.A.G., El-Saraf, G.A., Khodiary, A., 1993. Phosphorus, Sulfur and Silicon 85, 183–192.
- Andotra, C.S., Sharma, S.K., 1988. Proc. Natl. Acad. Sci. India 58A, 215.
- Barry, A.L., 1976. The Antimicrobial Susceptibility Test, Principles and Practices. Illus Lea and Febiger, Philadeliphia, PA, USA, p. 180.
- Burell, G., Evans, J.M., Hadley, M.S., Hicks, F., Stemp, G., 1994. Bioorg. Med. Chem. Lett. 4, 1285.
- Chande, M.S., Bahandari, J.D., Joshi, V.R., 1993. Indian J. Chem. 32B, 1218–1228.
- Chen, C., Dagnino, R., Huang, C.Q., McCarthy, J.R., Grigoriadis, D.E., 2001. Bioorg. Med. Chem. Lett. 11, 3165.
- Collin, X., Sauleau, A., Coulon, J., 2003. Bioorg. Med. Chem. Lett. 13, 2601.
- De Clercq, E., 2004. J. Clin. Virol. 30, 115.
- Duncia, J.V., Santella, J.B., Higley, C.A., VanAtten, M.K., Weber, P.C., Alexander, R.S., Kettner, C.A., Pruitt, J.R., Liauw, A.Y.,

Quan, Knabb, R.M., Wexler, R.R., . Bioorg. Med. Chem. Lett. 8, 775.

- El-Wassaimy, M.T., Abdel-Rahman, M., Ghattas, A.B.A.G., Abd Allah, O.A., 1992. Phosphorus, Sulfur and Silicon 70, 99.
- Ghattas, A.B.A.G., El-Saraf, G.A., 1989. Sulfur Lett. 8, 261.
- Ghattas, A.B.A.G., Abdel-Rahman, M., El-Wassaimy, M.T., El-Saraf, G.A., 1989. Rev. Roum. Chim. 34, 1987.
- Ghattas, A.B.A.G., Moustafa, H.M., Abd Allah, O.A., Amer, A.A., 2001. Synth. Commun. 31 (16), 57–66.
- Ghattas, A.B.A.G., Abd Allah, O.A., Moustafa, H.M., Amer, A.A., 2003. Eygpt. J. Chem. 46, 297–311.
- Ghorab, M.M., Abdel- Hamide, S.G., Ali, G.M., El-Sayed, H.S., Shaurub, H., 1996. Pestic. Sci. 48, 31.
- Ghorab, M.M., Abdel- Hamide, S.G., El-Gaby, M.S.A., El-Sayed, S.M., 1999. Acta Pharm. 49, 1.
- Hester, J.B., Rudzik, A.D., Kamdar, B.V., 1971. J. Med. Chem. 14, 1078.
- Kidwai, M., Sapra, P., Misra, P., Saxena, R.K., Singh, M., 2001. Bioorg. Med. Chem. 9, 217.
- Milton, N.G.N., 2001. Neurotoxicology 22, 767.
- Moustafa, H.M., 2001. Synth. Commun. 31 (1), 97.
- Naito, Y., Akahoshi, F., Takeda, S., Okada, T., Kajii, M., Nishimura, H., Sugiura, M., Fukaya, C., Kagitani, Y., 1996. J. Med. Chem. 39, 3019.
- Papakonstantinou-Garoufalias, S., Pouli, N., Marakos, P., Chytyroglou-Ladas, A., 2002. Farmaco 57, 973.
- Potts, K.T., 1961. Chem. Rev. 61, 87.
- Thompson, S.K., Eppley, A.M., Frazee, J.S., Darcy, M.G., Lum, R.T., Tomaszeck, T.A., Ivanoff, L.A., Morris, J.F., Sternberg, E.J., Lambert, D.M., Fernandez, A.V., Patteway, S.R., Meek, T.D., Metcalf, B.W., Gleason, J.G., 1994. Bioorg. Med. Chem. Lett. 4, 2441.
- Wadsworth, J.H., Jenkins, S.M., Orlek, B.S., Cassidy, F., Clark, M.S.G., Brown, F., Riley, G.J., Graves, D., Hawkins, J., Naylor, C.B., 1992. J. Med. Chem. 35, 1280.