Development of lactobionic acid conjugated-copper chelators as anticancer candidates for hepatocellular carcinoma

Xueke Zhao^{1,#}, Xiang Li^{2,3,#}, Xiaoping Huang³, Shuyu Liang², Penggen Cai¹, Yuhui Wang³, Yongming Cui¹, Wu Chen¹, Xiongwei Dong^{1*}

¹ National Local Joint Engineering Laboratory for Advanced Textile Processing and Clean Production, Wuhan Textile University, Wuhan, 430073, Hubei, P. R. China.

² College of Chemistry and Molecular Engineering, Peking University, Beijing, 100871, Beijing, P. R. China.

³ School of Chemistry, Central China Normal University, Wuhan, 430079, Hubei, P.R. China.

[#] The work has been contributed by these authors equally.

* Correspondence: xwdong@wtu.edu.cn (X.D.)

Synthesis:

Scheme S1. Synthesis of the chelators.



Y: -CH₃; -Cl; -OH; -N(CH₃)₂; -H. X: O; S.

General procedure: Synthesis of TH1, TH9 and TH10. A water solution of thiocarbohydrazide (10 mmol) was added dropwise to the solution of salicylaldehyde (for TH1, 10 mmol) in ethanol at 60-70 °C. Next, a water solution of carbohydrazide (10 mmol) was added dropwise to the solution of 4-methoxysalicylaldehyde (for TH9, 10 mmol) or 5-methsalicylaldehyde (for TH10, 10 mmol) in ethanol at 60-70 °C. Then, the mixture was refluxed for 4 h (monitored by TLC), and a large amount of precipitate was observed. The mixture was cooled to room temperature and filtrated to gain white solids. The solids were washed with methanol and water, and then dried with vacuum to get the intermediate. The crude product ware recrystallized from water/DMF or water/DMSO.

Data for TH1: yield 1.52 g (72.29%); slight yellow solid; ESI-MS: m/z 210.12 (M-H⁺), calculated: 210.26; Elem. anal. calcd for C₈H₁₀N₄OS: C 45.70, H 4.79, N 26.65, S 15.25; Found: C 45.64, H 4.68, N 26.56, S 15.18. ¹H NMR (600 MHz, DMSO-*d*₆) δ 11.36 (s, 1H), 9.86 (s, 1H), 9.74 (s, 1H), 8.32 (s, 1H), 7.97 (s, 1H), 7.20 (s, 1H), 6.86 (d, *J* = 8.1 Hz, 1H), 6.81 (s, 1H), 4.84 (s, 2H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 176.15, 156.59, 139.83, 131.73, 127.36, 120.69, 120.11, 116.85. The wave number of the Fourier Transform Infrared Spectrum (FT-IR, KBr) of compound **TH1** is 3289, 3258, 3179, 3138, 2955, 1618, 1597, 1551, 1499, 1375, 1333, 1271, 1244, 1213, 1153, 1011, 773, 635, 613, 579.

Data for **TH9**: yield 1.73 g (77.15%); white solid; ESI-MS: m/z 224.26 (M-H⁺), calculated: 224.22; Elem. anal. calcd for C₉H₁₂N₄O₃: C 48.21, H 5.39, N 24.99; Found: C 48.16, H 5.28, N 24.88. ¹H NMR (600 MHz, DMSO*d*₆) δ 10.41 (s, 1H), 8.21 (s, 1H), 7.92 (s, 1H), 7.65 (s, 1H), 7.18 (t, *J* = 7.9 Hz, 1H), 6.92-6.73 (m, 3H), 4.14 (s, 2H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 157.23, 156.45, 152.31, 130.76, 130.31, 128.14, 120.03, 116.52, 39.94. The wave number of the Fourier Transform Infrared Spectrum (FT-IR, KBr) of compound **TH9** is 3344, 3277, 3200, 3088, 2959, 1707, 1670, 1634, 1570, 1510, 1292, 1250, 1236, 1165, 1136, 1032, 806, 683, 638,573.

Data for **TH10**: yield 1.68 g (80.68%); white solid; ESI-MS: m/z 208.31 (M-H⁺), calculated: 208.22; Elem. anal. calcd for C₉H₁₂N₄O₂: C 51.92, H 5.81, N 26.91; Found: C 51.86, H 5.92, N 26.84. ¹H NMR (600 MHz, DMSO- d_6) δ 10.26 (s, 1H), 7.83 (s, 1H), 7.51 (s, 1H), 6.43 (dd, J = 12.9, 4.1 Hz, 1H), 4.12 (s, 1H), 3.73 (s, 2H). ¹³C NMR (150 MHz, DMSO- d_6) δ 145.52, 142.24, 141.35, 136.42, 113.91, 97.40, 90.43, 85.41, 23.79. The wave

number of the Fourier Transform Infrared Spectrum (FT-IR, KBr) of compound **TH10** is 3352, 3281, 3242, 3098, 2914, 1724, 1680, 1636, 1558, 1497, 1462, 1367, 1275, 1223, 1180, 957, 814, 687, 606, 573.

General procedure: Synthesis of **GT1**, **GT9** and **GT10**. Lactobionic acid (5 mmol) and NHS (N-Hydroxy succinimide, 6 mmol) were dissolved in DMF (N, N-Dimethylformamide, 30 mL), and stirred in ice water bath for 30 min. Then, EDC (1-Ethyl-3-(3-dimethylaminopropyl) carbodiimide, 6 mmol) was added, and stirred for 1 h at room temperature. Next, 5 mmol of **TH1** (1.0513 g), **TH9** (1.1205 g), **TH10** (1.0411 g) were dissolved in DMF (10 mL), and added dropwise to the solution for 72 h (monitored by TLC) at room temperature. Finally, we slowly added 3 times the volume of water to precipitate a solid, and the crude products were recrystallized from water/DMF or water/DMSO.

Data for **GT1**: yield 1.706 g (61.97%); white solid; ESI-MS: m/z 549.86 (M-H⁺), calculated: 550.54; Elem. anal. calcd for C₂₁H₃₂N₄O₁₄: C 43.63, H 5.49, N 10.18, S 5.82; Found: C 43.52, H 5.37, N 10.06, S 5.78. ¹H NMR (600 MHz, DMSO-*d*₆) δ 9.78 (d, *J* = 64.7 Hz, 2H), 9.42 (s, 1H), 8.30 (s, 1H), 7.96 (s, 1H), 7.18 (s, 2H), 6.81 (d, *J* = 27.8 Hz, 2H), 5.60-3.36 (m, 16H), 1.22 (s, 1H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 180.93, 176.39, 161.40, 144.83, 144.55, 136.56, 135.59, 125.52, 124.92, 123.86, 121.68, 110.19, 83.46, 80.69, 78.36, 76.53, 76.33, 73.24, 67.40, 65.55. The wave number of the Fourier Transform Infrared Spectrum (FT-IR, KBr) of compound **GT1** is 3169, 3140, 3001, 2982, 1618, 1562, 1535, 1485, 1452, 1396, 1362, 1319, 1285, 1200, 1178, 1034, 955, 903, 795, 746, 642, 582.

Data for **GT9**: yield 1.835g (64.98%); white solid; ESI-MS: m/z 564.86 (M-H⁺), calculated: 564.50; Elem. anal. calcd for C₂₁H₃₂N₄O₁₄: C 44.68, H 5.71, N 9.93; Found: C 44.58, H 5.67, N 9.84. ¹H NMR (600 MHz, DMSO-*d*₆) δ 10.65 (s, 2H), 8.30 (s, 2H), 7.56 (s, 1H), 6.63-6.23 (m, 4H), 5.27-4.08 (m, 9H), 3.95 (d, *J* = 8.1 Hz, 1H), 3.87-3.38 (m, 11H), 1.21 (s, 1H), 1.03 (d, *J* = 7.7 Hz, 1H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 191.00, 178.15, 176.38, 166.60, 163.45, 157.14, 117.92, 111.86, 111.26, 110.14, 106.63, 106.10, 88.96, 80.71, 78.35, 76.54, 76.33, 73.24, 67.33, 65.48, 60.33. The wave number of the Fourier Transform Infrared Spectrum (FT-IR, KBr) of compound **GT9** is 3312, 3235, 3069, 2963, 2932, 2837, 1736, 1676, 1630, 1510, 1441, 1356, 1288, 1221, 1169, 1126, 1082, 1036, 962, 835, 800, 706, 638, 615, 594, 569.

Data for **GT10**: yield 1.838 g (67.06%); white solid; ESI-MS: m/z 548.15(M-H⁺), calculated: 548.50; Elem. anal. calcd for C₂₁H₃₂N₄O₁₄: C 45.99, H 5.88, N 10.21; Found: C 45.92, H 5.78, N 10.18. ¹H NMR (600 MHz, DMSO*d*₆) δ 10.81 (s, 1H), 8.37 (s, 1H), 7.02 (d, *J* = 8.3 Hz, 1H), 6.77 (d, *J* = 8.2 Hz, 1H), 5.15 (d, *J* = 11.3 Hz, 2H), 4.94-3.39 (m, 14H), 2.23 (s, 2H), 1.22 (s, 1H), 1.04 (s, 1H). ¹³C NMR (150 MHz, DMSO-*d*₆) δ 178.15, 159.63, 157.13, 136.93, 132.89, 124.45, 121.73, 109.85, 88.91, 88.03, 80.71, 78.37, 76.70, 76.53, 76.32, 73.43, 73.24, 67.45, 65.83, 65.50, 25.23. The wave number of the Fourier Transform Infrared Spectrum (FT-IR, KBr) of compound **GT10** is 3252, 3181, 3101, 3028, 2918, 1649, 1626, 1572, 1491, 1352, 1275, 1225, 1157, 1090, 955, 826, 772, 745, 656, 567.



Figure S1.¹H NMR spectrum (600 MHz) of TH1 was recorded in DMSO-d₆.



Figure S2. ¹H NMR spectrum (600 MHz) of TH9 was recorded in DMSO-*d*₆.



Figure S3. ¹H NMR spectrum (600 MHz) of TH10 was recorded in DMSO-d₆.



Figure S4. ¹H NMR spectrum (600 MHz) of GT1 was recorded in DMSO-d₆.



Figure S5. ¹H NMR spectrum (600 MHz) of GT9 was recorded in DMSO-d₆.



Figure S6. ¹H NMR spectrum (600 MHz) of GT10 was recorded in DMSO-d₆.



Figure S7. ¹³C NMR spectrum (150 MHz) of TH1 was recorded in DMSO-d₆.



Figure S8. ¹³C NMR spectrum (150 MHz) of TH9 was recorded in DMSO-d₆.



Figure S9. ¹³C NMR spectrum (150 MHz) of TH10 was recorded in DMSO-*d*₆.



Figure S10. ¹³C NMR spectrum (150 MHz) of GT1 was recorded in DMSO-*d*₆.



Figure S11. ¹³C NMR spectrum (150 MHz) of GT9 was recorded in DMSO-*d*₆.



Figure S12. ¹³C NMR spectrum (150 MHz) of GT10 was recorded in DMSO-d₆.



Figure S13. ESI-MS spectrum of TH1.



Figure S14. ESI-MS spectrum of TH9.



Figure S15. ESI-MS spectrum of TH10.



Figure S16. ESI-MS spectrum of GT1.



Figure S17. ESI-MS spectrum of GT9.



Figure S18. ESI-MS spectrum of GT10.



Fig. S19 Fourier transform infrared spectrum comparison of the metal-coordinating moiety (TH1, 9 and 10), respective lactobionic acid conjugated-metal chelator (GT1, 9 and 10), and copper complexes. a) IR spectra comparison of TH1/GT1, TH9/GT9, and TH10/GH10. b) IR spectra comparison of GT1/GT1+Cu²⁺, GT9/GT9+Cu²⁺, and GT10/GT10+Cu²⁺. c) Key IR spectral data chelators and copper complexes in (a) and (b).



Fig. S20 a) Species distribution plots measured using potentiometric titrations for the Cu²⁺, Zn²⁺, Mn²⁺ and Fe²⁺chelator systems. The sources of Cu²⁺, Zn²⁺, Mn²⁺, and Fe²⁺ are CuCl₂, ZnCl₂, MnCl₂ and FeCl₂, respectively. Potentiometric titrations were performed for the solutions containing the chelators and equimolar amounts of metal ions at 25 °C (I = 0.1 M KCl). **b**) Spectrophotometric titrations of chelators (10 mM) with Cu²⁺ (0, 10 and 20 mM) at 25 °C in 10% fetal bovine serum (FBS). **c**) HepG2 cells without metal loading were incubated with 0, 1, 5, 10, 30 and 50 µM chelators (**GT1**, **9** and **10**) for 24 h. Then, the cell viabilities were evaluated using commercial MTS kit. For chelator treatment, the same amount of DMSO (5‰) was added as a control group, because DMSO was used as a co-solvent. All experiments were normalized by respective control data. Data are mean of triplicate samples ± SD (unpaired Student's *t* test), and all error bars are SD.



Fig. S21 a) After constructing the high-copper cell model, the HC HepG2 cells were incubated with or without 50 μ M chelators (TH1, TH9, TH10, GT1, GT9 and GT10) for another 24 h. Then, the cell viability was evaluated using commercial MTS kit. b) After constructing the high-copper cell model, the HepG2, HeLa and MCF-7 cells were incubated with or without 50 μ M GT1 for another 24 h. Then, the cell viability was evaluated using commercial MTS kit. For chelator treatment, the same amount of DMSO (5‰) was added as a control group. All experiments were normalized by respective control data. Data are mean of triplicate samples ± SD (**P* < 0.05, ***P* < 0.01, ****P* < 0.001; unpaired Student's *t* test), and all error bars are SD.



Fig. S22 HepG2 cells without metal loading (a) and MCF-10A cells (b) were incubated with 0 and 50 μ M GT1 for 24 h. For chelator treatment, the same amount of DMSO (5‰) was added as a control group, because DMSO was used as a co-solvent. Flow cytometric analysis of cell apoptosis was carried out with Annexin V and propidium iodide staining.