Isotherms	Equations	Parameters
		qm means the adsorption amount
Langmuir	$k_L q_m C_e$	(mg/g).
	$q_e = \frac{k_L q_m C_e}{1 + k_L C_e}$	k_L (L/mg) is coefficient.
Freundlich	$q_e = k_F C_e^{1/n}$	k _F is adsorption constant (mg/g).
		$(L/mg)^{1/n}$, $1/n$ is adsorption intensity
Hill		Q _m is the maximum
	$QmCe^n$	adsorption capacity (mg/g), n and C
	$qe = \frac{QmCe^n}{C^n + Ce^n}$	are the number of ions per site
		and semi saturated concentration,
		respectively.

Table2 Adsorption isotherm models and corresponded parameters