

Supporting Information

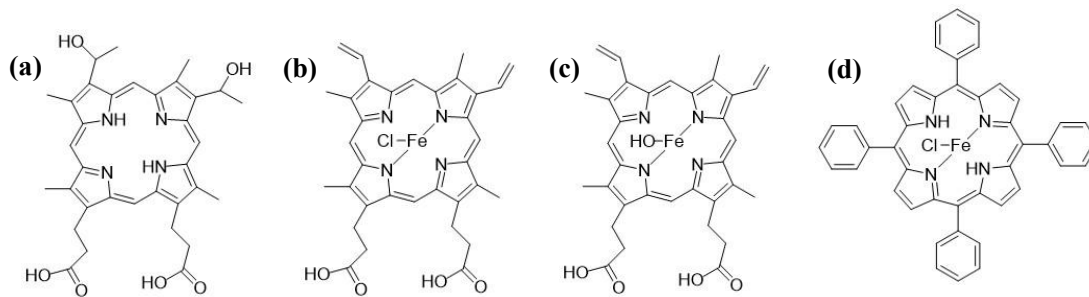
Catalytic effect of cytochrome c on H₂O₂ oxidation of a *p*-hydroxyphenyl-substituted BODIPY derivative

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2 **Figure S1.** (a) Chemical structure of Hematoporphyrin (He). (b) Chemical structure of Hemin

3 (Hc). (c) Chemical structure of Hematin porcine (Hh). (d) Chemical structure of

4 Iron(III) Meso-Tetraphenylporphine Chloride (It).

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1 *SI.1. Reagents*

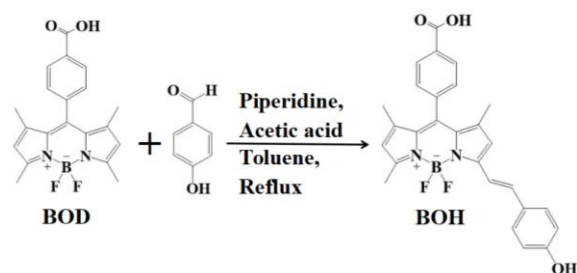
2 2,4-dimethylpyrrole, 4-carboxybenzaldehyde, Trifluoroacetic acid (TFA),
3 2,3-Dichloro-5,6-dicyano-1,4-benzoquinone (DDQ), Triethylamine, Boron trifluoride
4 diethyl etherate, *p*-hydroxybenzaldehyde, Acetic acid, Piperidine, Toluene, DMSO-d₆,
5 H₂O₂ (AR, 30 WT. % in H₂O) were purchased from Aladdin Industrial Corporation
6 (Shanghai, China).

7 *SI.2. Apparatus*

8 The ¹H NMR and ¹³C NMR spectra were examined using a Bruker 600 MHz
9 instrument, and the chemical shifts were reported on the delta scale in ppm relative to
10 DMSO-d₆ (δ = 2.50 ppm) for ¹H NMR and DMSO-d₆ (δ = 39.51 ppm) for ¹³C NMR.

11 *SI.3. Synthesis of BOH*

12 A BODIPY derivative BOD was firstly synthesized following Safacan
13 Kolemen's method [S1]. Then, BOD (2.5 mM, 0.92 g) and *p*-hydroxybenzaldehyde
14 (7.5 mM, 0.92 g) were added to a 500 mL round-bottomed flask. 200 mL of toluene,
15 0.5 mL of piperidine and 0.5 mL of acetic acid were added into the solution. The
16 mixture was heated under reflux using a Dean Stark trap. The reaction process was
17 monitored by TLC (CHCl₃: MeOH = 97:3). The starting materials were consumed,
18 the solution was cooled to room temperature and washed for three times with water.
19 The organic phase was dried with anhydrous Na₂SO₄ and evaporated by using a rotary
20 evaporator. The residue was purified through silica gel column chromatography using
21 chloroform and methanol (CHCl₃: MeOH = 97:3) as the eluant. Purple black solid
22 (BOH, 0.27 g, 23%) [S2]. BOH: ¹H NMR (600 MHz, DMSO-d₆): δ_H 8.12 (d, J = 8.1
23 Hz, 2H), 7.54 (d, J = 7.8 Hz, 2H), 7.50 (s, 1H), 7.47 (d, J = 8.5 Hz, 2H), 7.32 (d, J =
24 16.3 Hz, 1H), 6.94 (s, 1H), 6.90 (d, J = 4.3 Hz, 2H), 6.18 (s, 1H), 2.49 (s, 3H), 1.39 (s,
25 3H), 1.34 (s, 3H). ¹³C NMR (600 MHz, DMSO-d₆): δ_C 167.35, 159.67, 154.30,
26 154.03, 142.87, 141.56, 39.06, 138.98, 138.54, 138.48, 132.26, 30.75, 130.52, 129.52,
27 129.07, 127.54, 121.59, 118.75, 116.54, 115.11, 55.25, 14.72, 14.39 ppm. QTOF-MS
28 (m/z): [M+H]⁺ calc'd for C₂₇H₂₄BF₂N₂O₃: 473.1843, Found: 473.1842, Δ=0.21 ppm
29 [S3].



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2 **Figure S2.** Synthesis of the *p*-hydroxystyryl-substituted BODIPY derivative (BOH).

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4 [S1] Kolemen S., Bozdemir O., Cakmak Y.. Optimization of distyryl-Bodipy chromophores for
5 efficient panchromatic sensitization in dye sensitized solar cells. Chem. Sci. 2011;2: 949–954.

6 [S2] Lu Y., Ke H., Wang Y., Zhang Y., Li H., Huang C., Jia N.. A ratiometric
7 electrochemiluminescence resonance energy transfer platform based on novel dye BODIPY
8 derivatives for sensitive detection of lactoferrin, Biosens. Bioelectron. 2020;170:112664.

9 [S3] Xu Q., Li K., Wang P.. The reaction mechanism of *p*-hydroxystyryl-substituted BODIPY with
10 ABTS^{•+} and Fe³⁺ in solutions and in liposomes. Arab. J. Chem. 2025;18:106041.

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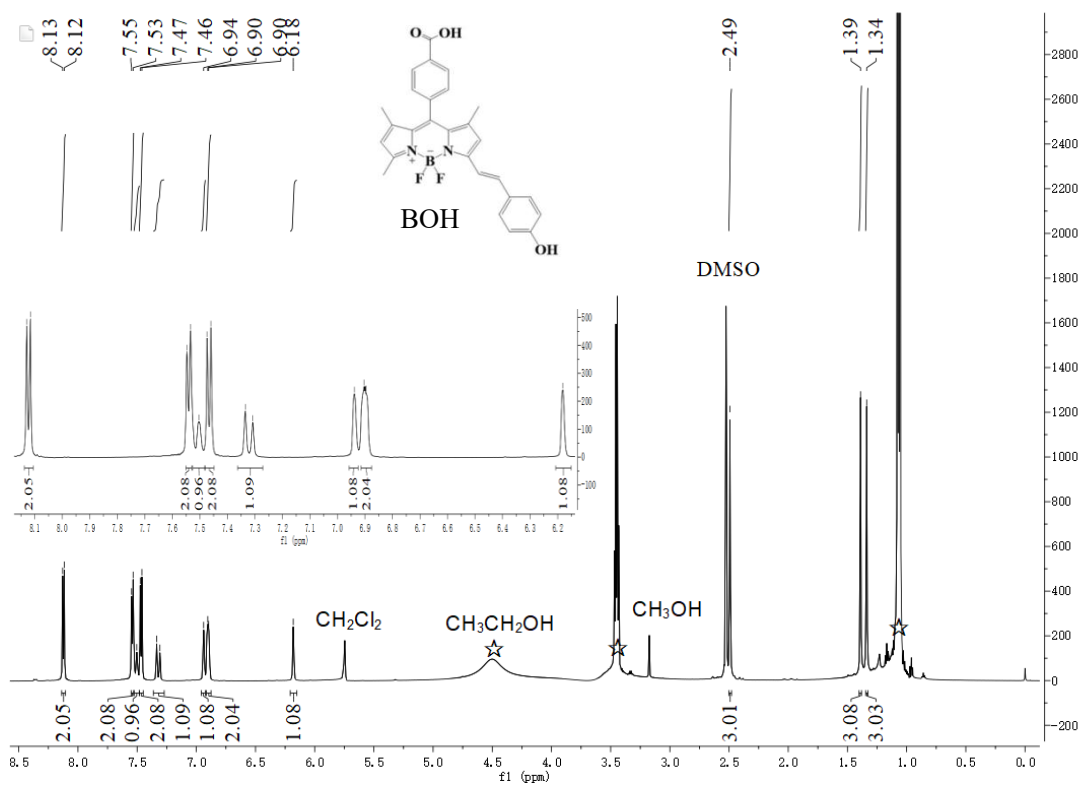
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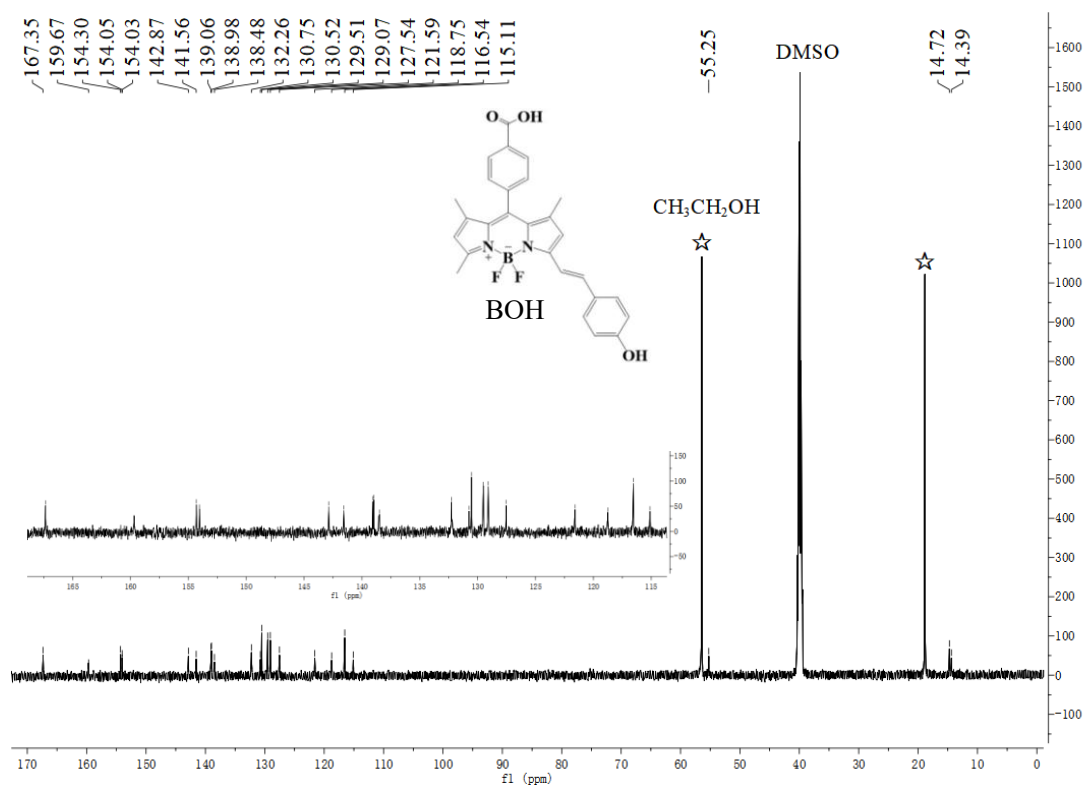
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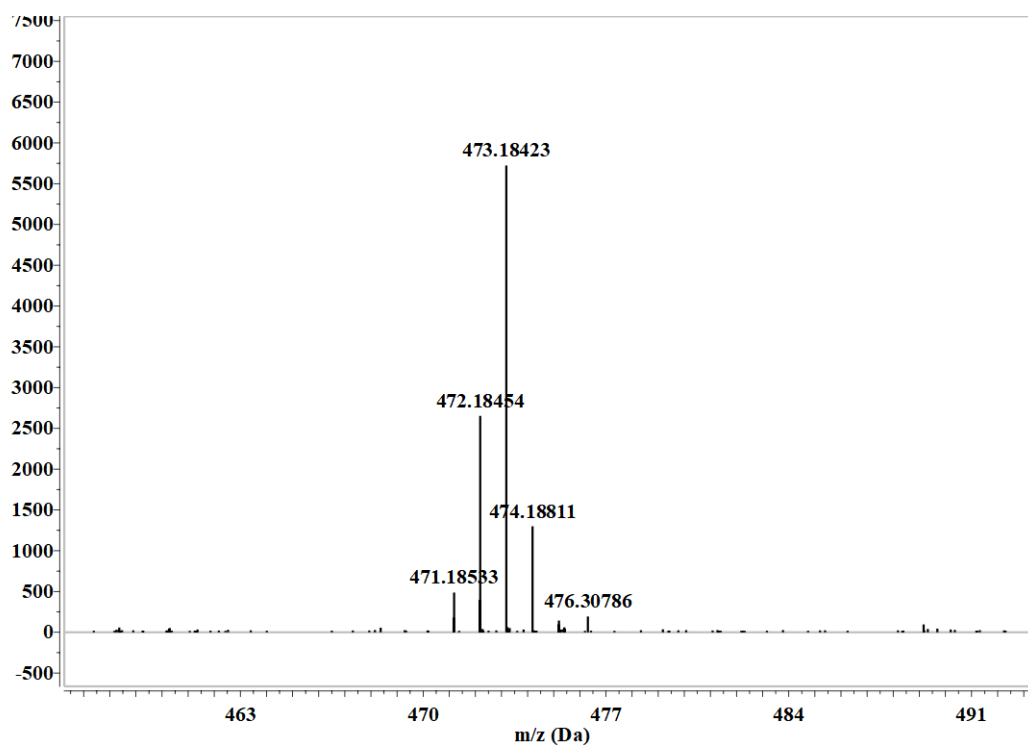
1 *S1.4. NMR spectrum and QTOF-MS spectrum of BOH*



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3 **Figure S3.** ¹H-NMR spectrum of BOH in (CD₃)₂SO.



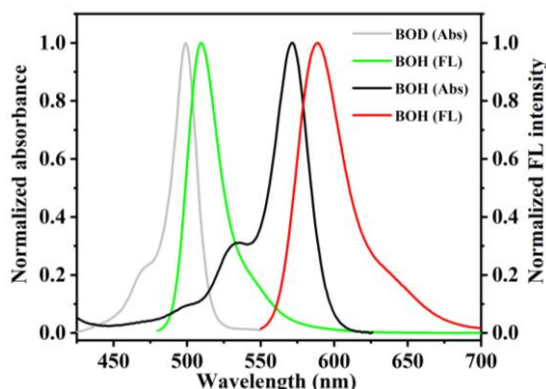
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5 **Figure S4.** ¹³C-NMR spectrum of BOH in (CD₃)₂SO.



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Figure S5. QTOF-MS spectrum of BOH.

1 *S1.5. Characteristic spectra of BOD and BOH*



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3 **Figure S6.** Normalized absorption spectra and fluorescence emission spectra ($\lambda_{\text{Ex}} = 450 \text{ nm}$) of
 4 BOD in ethanol solution and normalized absorption spectra and fluorescence emission spectra (λ_{Ex}
 5 $= 520 \text{ nm}$) of BOH in ethanol solution [S3].

6 The absorption peaks of BOD and BOH were observed at 499 nm and 572 nm in
 7 ethanol solution, respectively. The fluorescence emission peaks of BOD and BOH
 8 were observed at 509 nm and 589 nm in ethanol solution.

9 **Table S1.** Physicochemical parameters of BOH in ethanol solution [S3].

	$\varepsilon/\text{L}\cdot\text{mol}^{-1}\text{cm}^{-1}$	$\lambda_{\text{max}}^{\text{Abs}}/\text{nm}$	$\lambda_{\text{max}}^{\text{Em}}/\text{nm}$	$\Phi_{\text{FL}}/\%$	$\tau_{\text{FL}}/\text{ns}$	TEAC	Redox potential (V vs. RHE)	pKa	LogP _{Cal}
BOH	107758	572	589	79.90	4.08	1.31	0.17	4.19, 9.33	1.32

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11 [S3] Xu Q., Li K., Wang P. The reaction mechanism of *p*-hydroxystyryl-substituted BODIPY with
 12 ABTS^{•+} and Fe³⁺ in solutions and in liposomes. Arab. J. Chem. 2025;18:106041.

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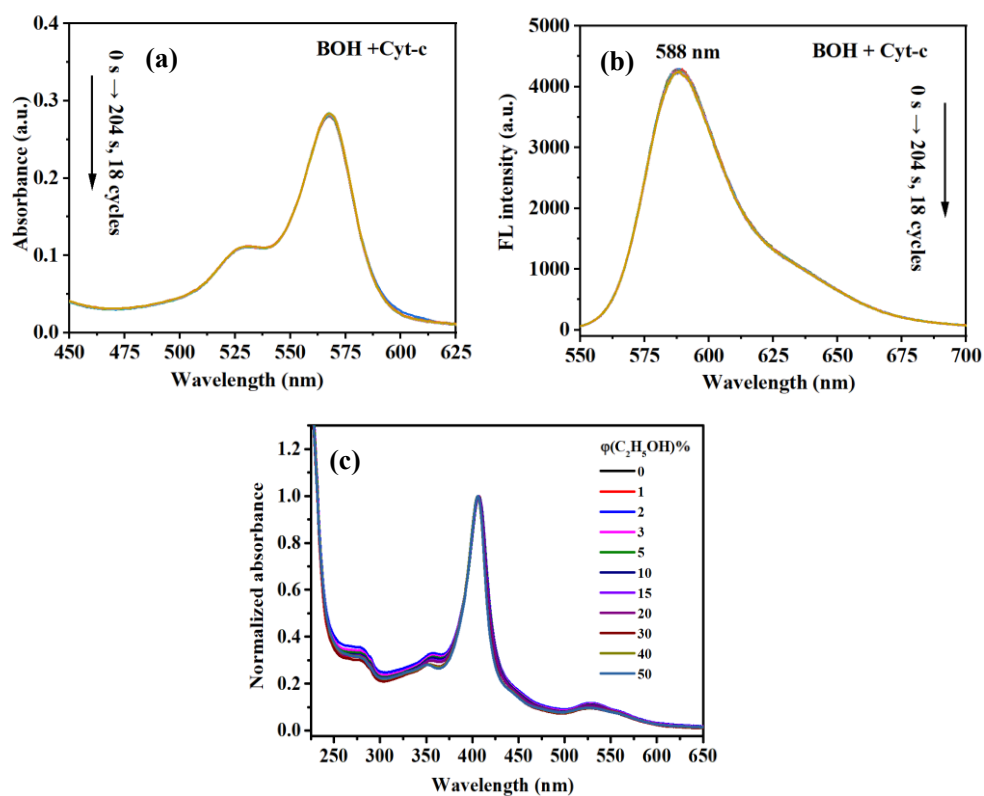
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1 *S1.6. Effect of BOH on Cyt-c*



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3 **Figure S7.** (a) The absorption spectra of BOH and Cyt-c within 18 cycles in 204s. (b)
4 Fluorescence emission spectra of BOH and Cyt-c within 18 cycles in 204s. (c) Normalized
5 absorption spectra of Cyt-c (10^{-5} M) with different volume ratio of ethanol to water.

6 The interaction between Cyt-c and BOH was investigated in this study. It
7 demonstrated that the characteristic absorption and fluorescence emission peaks of
8 BOH at 570 and 588 nm remained unchanged after the addition of Cyt-c, indicating
9 the absence of any chemical reaction between BOH and Cyt-c (Figure S7a and S7b).
10 To further understand the impact of ethanol on Cyt-c, which served as the solvent for
11 the redox reaction of $\text{BOH} + \text{Cyt-c} + \text{H}_2\text{O}_2$, experiments were conducted using a
12 mixture of ethanol and water. The absorption spectra revealed that the Soret band at
13 408 nm of Cyt-c exhibited minimal change as the percentage of ethanol increased,
14 suggesting that ethanol had a negligible effect on the α -helix secondary structure of
15 Cyt-c (Figure S7c) [S4]. The presence of a stable secondary structure of Cyt-c in the
16 ethanol solvent was conducive to the oxidation of BOH by H_2O_2 , which was catalyzed
17 by Cyt-c.

1 [S4] Gong J., Yao P., Jin L., Jiang M., Chunyu L.. Conformational transition of cytochrome c and
2 apocytochrome c induced by ethanol. Fudan Univ J Med Sci (Nat Sci) 2001;40:355–359.
3 DOI:10.3969/j.issn.0427-7104.2001.04.002.

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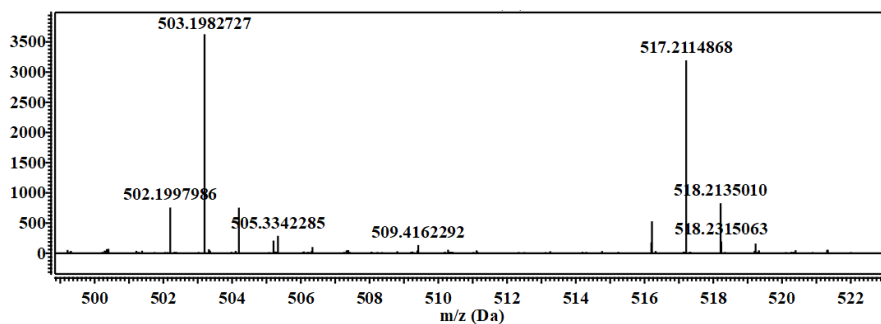
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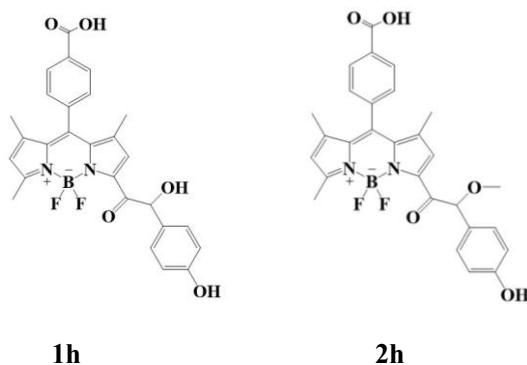
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1 *SI.7. The mass spectra of the catalysis of Cyt-c on the oxidation of BOH by H₂O₂*



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3 **Figure S8.** HPLC-QTOF-MS spectrum of BOD + Cyt-c + H₂O₂.



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6 **1h:** HPLC-QTOF-MS (m/z): [M-H]⁻ calc'd for C₂₇H₂₂BF₂N₂O₅: 503.1595, Found: 503.1983

7 **2h:** HPLC-QTOF-MS (m/z): [M-H]⁻ calc'd for C₂₈H₂₄BF₂N₂O₅: 517.1751, Found: 517.2115

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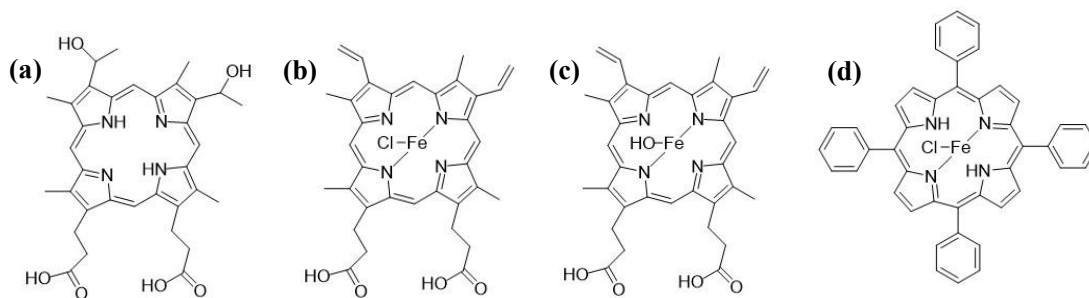
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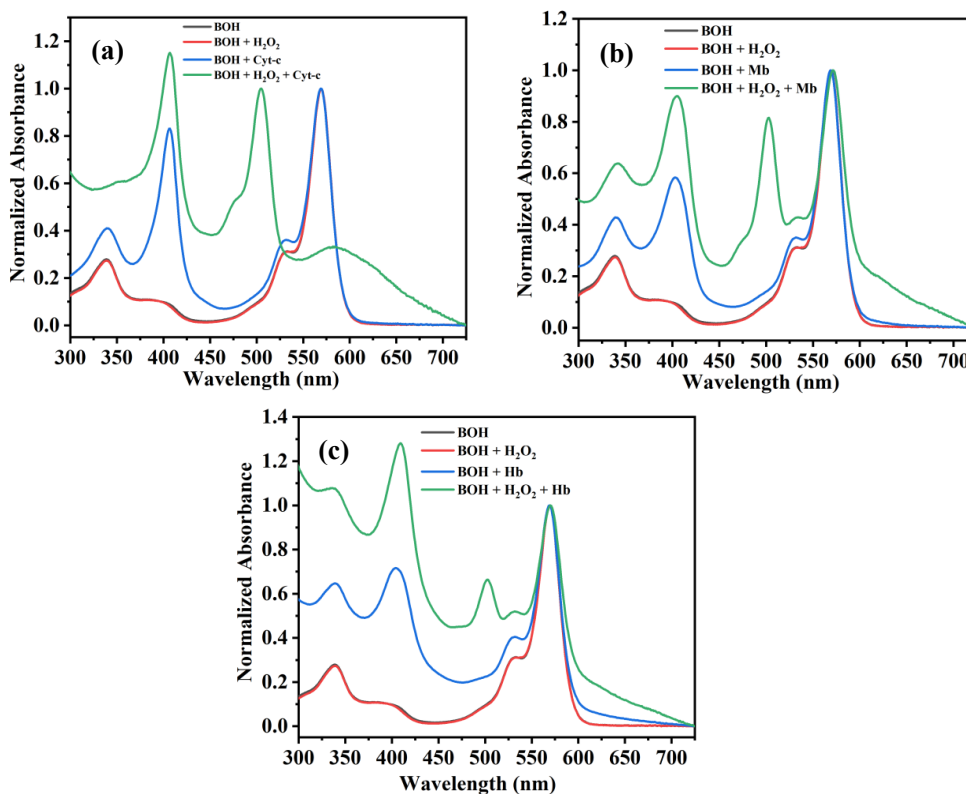
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1 *S1.8. The catalysis of hemoproteins on the oxidation of BOH by H₂O₂*



3 **Figure S1.** (a) Chemical structure of Hematoporphyrin (He). (b) Chemical structure of Hemin
4 (Hc). (c) Chemical structure of Hematin porcine (Hh). (d) Chemical structure of
5 Iron(III) Meso-Tetraphenylporphine Chloride (It).



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7 **Figure S9.** (a) The absorption spectra of BOH, BOH + H₂O₂, BOH + Cyt-c, and BOH + H₂O₂ +
8 Cyt-c . (b) The absorption spectra of BOH, BOH + H₂O₂, BOH + Mb, and BOH + H₂O₂ + Mb. (c)
9 The absorption spectra of BOH, BOH + H₂O₂, BOH + Hb, and BOH + H₂O₂ + Hb.

1 *Sl.9. Amino acid sequences of cytochrome c, myoglobin and hemoglobin*

2 *Sl.9.1. Cytochrome c (Protein Data Bank entry 1HRC)*

3 GDVEK GKKIF VQKCA QCHTV EKGKK HKTGP NLHGL FGRKT GQAPG
4 FTYTD ANKNK GITWK EETLM EYLEN PKKYI PGTKM IFAGI KKKTE REDLI
5 AYLKK ATNE HEC

6 *Sl.9.2. Myoglobin (Protein Data Bank entry 1WLA)*

7 GLSDG EWQQV LNVWG KVEAD IAGHG QEVLI RLFTG HPETL EKFDK
8 FKHLK TEAEM KASED LKKHG TVVLT ALGGI LKKKG HHEAE LKPLA
9 QSHAT KHKIP IKYLE FISDA IHHVL HSKHP GDFGA DAQGA MTKAL ELFRN
10 DIAAK YKELG FQG HEM SO4

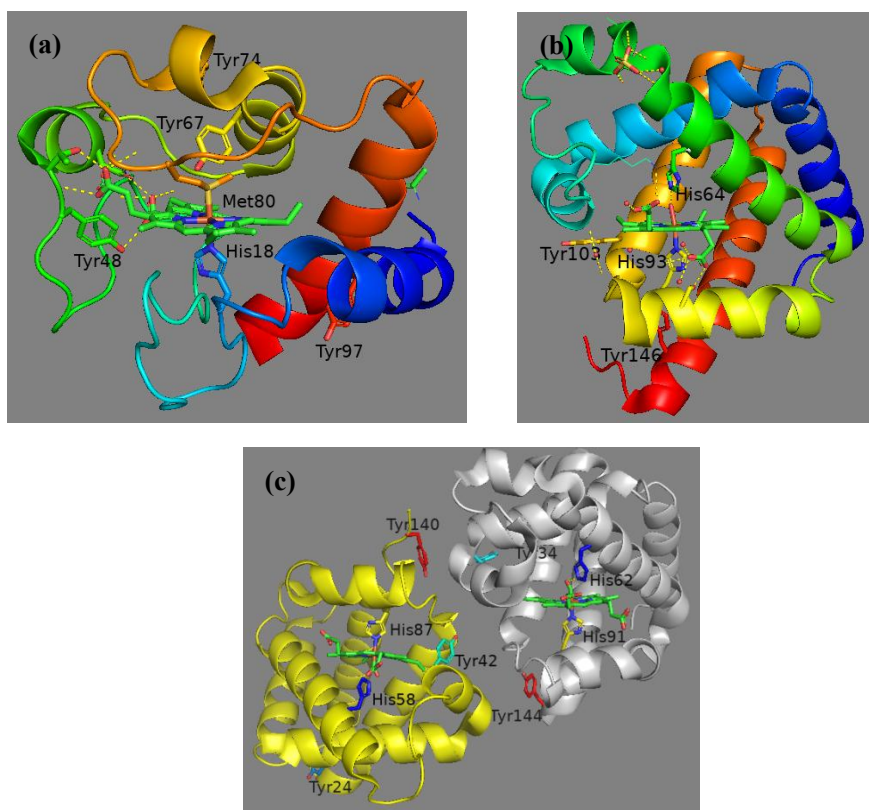
11 *Sl.9.3. Hemoglobin (Protein Data Bank entry 2QSS)*

12 VLSAA DKGNV KAAWG KVGGH AAAYG AEALE RMFLS FPTTK TYFPH
13 FDLSH GSAQV KGHGA KVAANA LTKAV EHLDD LPGA SELSD LHAHK
14 LRVDP VNFKL LSHSL LVTLA SHLPS DFTPA VHASL DKFLA NVSTV LTSKY
15 RHEM CMO

16 MLTAE EKAAY TAFWG KVKVD EVGGE ALGRL LVVYP WTQRF FESFG
17 DLSTA DAVMN NPKVK AHGKK VLDSF SNGMK HLDDL KGTFY ALSEL
18 HCDKL HVDPE NFKLL GNVLV VVLAR NFGKE FTPVL QADFQ KVVAG
19 VANAL AHRYP HEM CMO

20 VLSAA DKGNV KAAWG KVGGH AAAYG AEALE RMFLS FPTTK TYFPH
21 FDLSH GSAQV KGHGA KVAANA LTKAV EHLDD LPGAL SELSD LHAHK
22 LRVDP VNFKL LSHSL LVTLA SHLPS DFTPA VHASL DKFLA NVSTV LTSKY
23 RHEM CMO

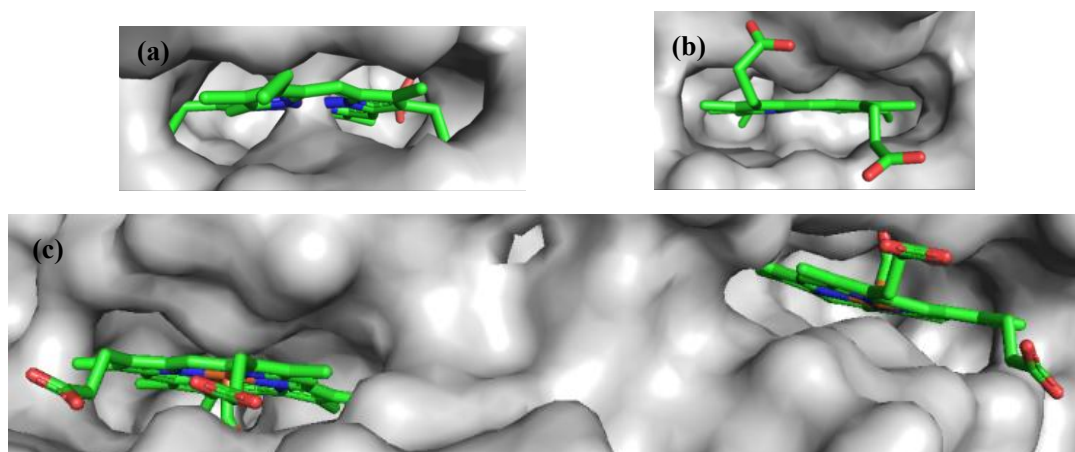
24 MLTAE EKAAY TAFWG KVKVD EVGGE ALGRL LVVYP WTQRF FESFG
25 DLSTA DAVMN NPKVK AHGKK VLDSF SNGMK HLDDL KGTFY ALSEL
26 HCDKL HVDPE NFKLL GNVLV VVLAR NFGKE FTPVL QADFQ KVVAG
27 VANAL AHRYP HEM CMO



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2 **Figure S10.** (a) Crystal structure of Cytochrome c (Protein Data Bankentry 1HRC). (b) Crystal
 3 structure of Myoglobin (Protein Data Bank entry 1WLA). (c) Crystal structure of Hemoglobin
 4 (Protein Data Bank entry 2QSS).

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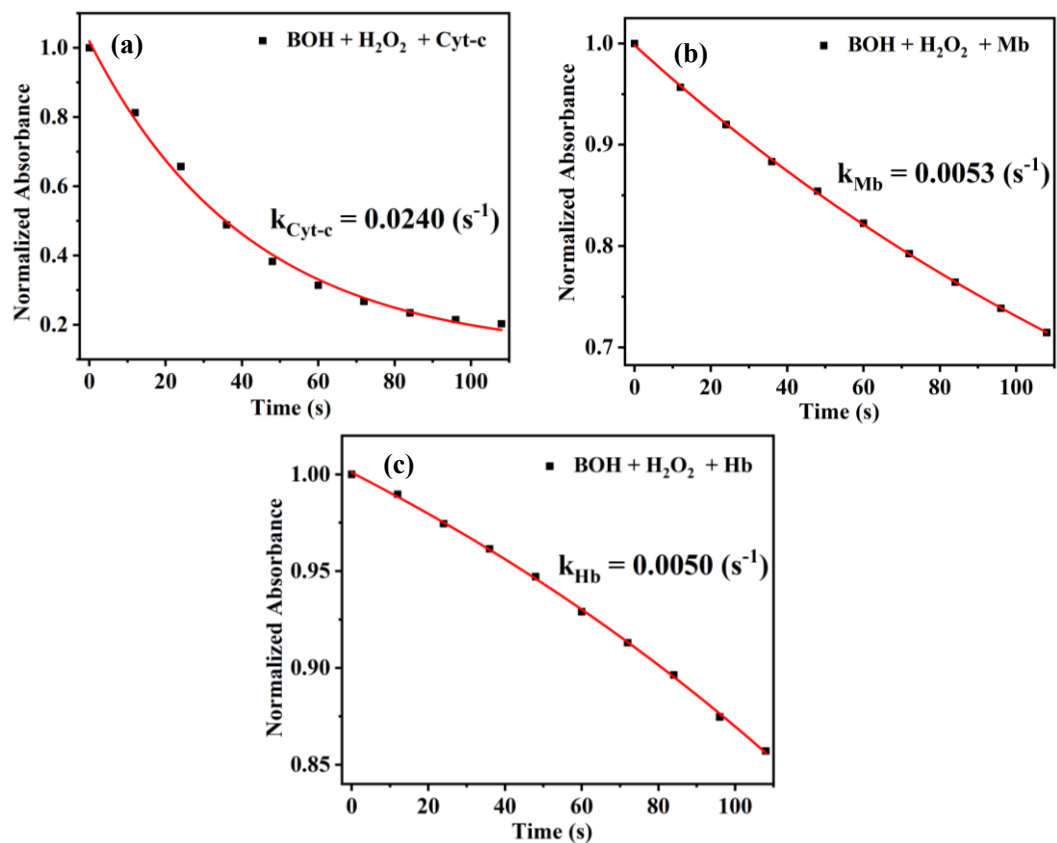


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7 **Figure S11.** (a) Cytochrome c surface (Protein Data Bank entry 1HRC). (b) Myoglobin surface
 8 (Protein Data Bank entry 1WLA). (c) Hemoglobin surface (Protein Data Bank entry 2QSS).

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2 **Figure S12. (a) Chemical kinetic of BOH + H₂O₂ + Cyt-c measured at 568 nm. (b) Chemical**
 3 **kinetic of BOH + H₂O₂ + Mb measured at 568 nm. (c) Chemical kinetic of BOH + H₂O₂ + Hb**
 4 **measured at 568 nm.**